



ASIAN JOURNAL OF CHEMISTRY

http://dx.doi.org/10.14233/ajchem.2013.15021



Aliphatic Alcohol, Acid, Ester and Other Constituents from Rice Straw of Oryza sativa

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(Received: 26 December 2012;

Accepted: 18 September 2013)

AJC-14124

Eight compounds, n-octacos-9-enyl propionate (1), 1-tetratriacontanol (2), β -sitosterol (3), n-tetracontan-15 α -ol (4), tritriacontan-4,12-diene (5), n-tritetracontan-5 α -ol (6), gallic acid (7) and β -sitosterol-3-O- β -D-glucoside (8) have been isolated and identified from methanol extract of rice straw of *Oryza sativa*. The structures of these known compounds were elucidated with the help different spectroscopic techniques.

Key Words: Oryza sativa L., Gramineae, Rice straw composition.

INTRODUCTION

Rice (*Oryza sativa* L.) is the principal cereal food in Asia and the major staple of the majority of the population. It generally occurs as two types, with white and coloured hulls, although the white hulled variety is more common (85 %). The germination of rice seed is of great agricultural importance and it has long been known to be influenced by compounds present in the seed coat (hull)¹. The compounds momilactone A and B from rice hulls cause germination and growth inhibition in the roots of rice²⁻⁴. They were later found in rice leaves and straw as phytoalexins^{5,6}.

Rice straw has been applied back in larger amounts into paddy and also upland fields, especially green house croppings, as an organic material mainly for soil improvement. The degradation products of rice straw in the soils may influence the growth of crops in both nutritrional and physiological aspects. The elucidation of mechanism of humus formation from rice straw is also of importance for understanding its influence on plant growth⁷. Phenolic substances are widely distributed in various plants, including the rice plant. Some of the substances, which enter into soils from plants, cause dieback disease or other abnormal growth as inhibitors against plant growth. It was reported that p-coumaric acid and other phenolic compounds, for instance, inhibited the growth of the upland rice plant. On the other hand, Kuwatsuka and Oshima⁸ isolated and or identified p-hydroxybenzoic acid, vanillic, p-coumaric acid and ferulic acid from rice leaves. Inamatsu⁹ also found p-coumaric acid in a methanol extract of rice straw and recognized that the amount of the acid decreased during the heaping of rice straw.

The phenolic compounds were reported from rice straw on the basis of GC analysis⁷. Identification of allelopathic compounds including momilactones A and B from rice straw and their biological activity have been reported^{10,11}.

Previously reported compounds from rice straw only on the basis of HPLC or GC analysis. Because there are no reports in the literature on the isolated compounds through column chromatography of rice straw. Identification of constituents by spectroscopic analysis like IR, NMR and mass still not reported of rice straw. Identification bioactive constituents with growth or germination inhibitory properties is still required. To achieve these objectives, the aims of our research were to isolated and identified constituents from rice straw. This papers deals with the isolation and structure elucidation of known compounds from rice straw of *Oryza sativa* on the basis of IR, mass and NMR. Identification of other constituents of rice straw of *O. sativa* is in progress.

EXPERIMENTAL

All chemicals used were of analytical grade. Hexane, ethyl acetate, chloroform, methanol, ethanol, water, sulphuric acid and vanillin were purchased from Daejung Chemicals and Metals Co. Ltd, Shiheung (Gyeonggi-do) Korea. Pre-coated TLC plates (layer thickness 0.25 mm), silica gel for column chromatography (70-230 mesh ASTM) and LiChroprep RP-18 (40-63 μm) were from Merck, Darmstadt, Germany. An authentic standard of compounds were purchased from Sigma-Aldrich, St. Louis, Missourri, USA. Both 1H and ^{13}C NMR spectra were obtained on a Bruker Avance 600 high resolution spectrometer operating at 600 and 150 MHz, respectively. This

Fig. 1. Chemical structures of compounds 1-8

NMR machine was available at Seoul National University (SNU), Seoul, South Korea and all NMR spectra were recorded at SNU (Instrument, Bruker, Germany). NMR spectra were obtained in deuterated chloroform and pyridine using tetramethylsilane (TMS) as an internal standard, with chemical shifts expressed in ppm (δ) and coupling constants (J) in Hz. FAB MS data were recorded on a JMS-700 (Jeol, Japan) spectrometer instrument which was available at SNU, Seoul, South Korea. IR spectra were recorded on an Infinity

Gold FT-IR (Thermo Mattson, USA) spectrophotometer, which was available at Korea Institute of Science and Technology, Seoul, South Korea.

The straw of *O. sativa* were collected from the Konkuk University Experimental Farm, Seoul, Korea in October 2010. After harvesting, the samples was dried in the Laboratoty temperature range (25-30 °C) for 3 weeks. The voucher specimen (ILPUM variety) has been dried and deposited in the herbarium of our department.

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Preparation of extracts: Dried straws of *O. sativa* (10 kg) were immersed in methanol (MeOH) for 1 week at room temperature and concentrated under vacuum to produce an extract (78 g), which was suspended in water and extracted successively with hexane, ethyl acetate (EtOAc) and *n*-butanol (*n*-BuOH).

Isolation of compounds from hexane extract: The hexane extract was chromatographed over silica gel with hexane and ethyl acetate solvents and initially yielded three compounds *n*-octacos-9-enyl propionate (**1**, 11 mg) and 1-tetratriacontanol (**2**, 16 mg). Tritriacontan-4,12-diene (**5**, 7 mg). The separation of other fractions is in progress.

Isolation of compounds from ethyl acetate extract: The EtOAc extract (11.2 g) was subjected to normal-phase column chromatography over silica gel and yielded 40 fractions with the following eluants: fraction 1 in hexane, fractions 2-5 in hexane:EtOAc (9:1), fractions 6-11 in hexane:EtOAc (8:2), fractions 12-15 in hexane:EtOAc (7:3), fractions 16-20 in hexane:EtOAc (1:1), fractions 21-22 in EtOAc, fractions 23-28 in EtOAc:MeOH (9.5:0.5), fractions 29-32 in EtOAc:MeOH (9:1), fractions 33-34 in EtOAc:MeOH (7:3) and fractions 35-40 in MeOH. Fraction 6 was crystallized and, after purification by column chromatography, yielded β -sitosterol (3, 23) mg). This was confirmed by comparison with an authentic sample from Sigma. Fraction 11 was further purified by column chromatography over silica gel with methylene dichloride and methanol, produced two pure compounds: *n*-tetracontan-15αol (4, 6 mg), *n*-tritetracontan- 5α -ol (6, 9 mg) fraction 12, after column chromatography over silica gel and Lichroprep RP-18 (octadecyl silica [ODS]), yielded one white compound in a powder form. This was identified as gallic acid (7, 3 mg). Fraction 23, after column chromatography over silica gel with chloroform and methanol, yielded one pure compound: βsitosterol-3-O-β-D-glucoside (**8**, 12 mg).

n-Octacos-9-enyl propionate (1): Colourless solid. R_f 0.32 (CHCl₃); IR ((KBr, v_{max} , cm⁻¹): 2918, 2849, 1735, 1635, 1467, 1380, 1174, 724; ¹H NMR (CDCl₃) δ 5.34 (m, 1H, H-9), 5.32 (m, 1H, H-10), 4.06 (d, J = 6.5 Hz, 1H, H₂-1a), 4.04 (d, J = 6.5 Hz, 1H, H₂-1b), 2.29 (d, J = 7.5 Hz, 1H, H₂-2'a), 2.27 (d, J = 7.5 Hz, 1H, H_2 -2'b), 2.01 (m, 2H, CH_2 -11), 1.62(m, 2H, H₂-8), 1.58 (m, 2H, CH₂), 1.55 (m, 2H, H₂-7), 1.25 (br s, 38 H, $19 \times CH_2$), 0.89 (t, J = 6.0 Hz, 3H, Me-28), 0.86 (t, J = 7.0 Hz, 3H, Me-3'); ¹³C NMR (CDCl₃) δ : 174.23 (C-1'), 130.13 (C-9, C-10), 64.62 (C-1), 34.62 (CH₂), 32.16 (CH₂), 30.01 (CH₂), 29.93 (3×CH₂), 29.89 (2×CH₂), 29.84 (CH₂), 29.82 (CH₂), 29.76 (CH₂), 29.71 (CH₂), 29.59 (CH₂), 29.55 (CH₂), 29.51 (CH₂), 29.49 (CH₂), 29.40 (CH₂), 28.90 (CH₂), 27.44 (CH₂), 26.18 (CH₂), 25.27 (CH₂), 22.92 (CH₂), 14.34 (CH_3-28, CH_3-3') ; FAB-MS m/z (rel. int.) 464 $[M]^+(C_{31}H_{60}O_2)$ (7.4), 437 (20.2), 407 (33.9), 379 (68), 365 (8.3), 351 (9.6), 337 (9.7), 323 (9.7), 309 (9.5), 295 (11.2), 281 (10.3), 279 (15.6), 267(10.4), 253(13.2), 225(11.8), 221(9.5), 211(12.9), 197 (13.7), 183 (15.2), 169 (15.3), 155 (16.4), 149 (28.3), 141 (18.8), 127 (20.8), 113 (29.1), 97 (24.9), 85 (62.1), 71 (85), 57 (100).

1-Tetratriacontanol (2): Colouless solid, R_f 0.65 (hexane: EtOAc 8.5:1.5); m.p. 83-86 °C; $[\alpha]_D^{20}$ -0.020° (CHCl₃); IR (KBr, ν_{max} , cm⁻¹): 3450, 2917, 1465, 1215, 757; ¹H NMR (CDCl₃): 0.88 (3H, t, J = 7.0 Hz, methyl), 1.55 (2H, m, CH₂CH₂-OH),

1.25-1.33 (all CH₂ protons), 3.64 (1H, t, OH); 13 C NMR (CDCl₃): 14.33 (methyl), 63.31 (C-OH), 22.90, 24.98, 25.95, 29.30, 29.47, 29.52, 29.60, 29.83, 29.80, 29.86, 29.91, 32.15, 33.04 (all methylene carbons assignments may be interchanged); EI-MS: m/z 494 [M]⁺ (calcd. (%) for $C_{34}H_{70}O$), 476 [M-H₂O]⁺ (4.7 %), 448 (23.4), 392 (11.5), 376 (4.2), 362 (5.6), 348 (5.3), 321 (5.7), 306 (10.5), 292 (6.8), 279 (7.4), 265 (7.6), 237 (9.5), 223 (10.5), 209 (12.2), 195 (14.0), 181 (16.6), 153 (24.6), 139 (32.0), 125 (48.3), 111 (74.4), 97 (100), 71 (61.7), 57 (87.7); FAB-MS: (positive mode) m/z 495 [M+H]⁺; (negative mode) m/z 493 [M-H]⁻.

β-Sitosterol (3): R_f 0.48 (Hex:EtOAc 8:2); m.p. 139-142 °C; $[\alpha]_D^{14}$ -37° (CHCl₃); IR (KBr, v_{max} , cm⁻¹): 3429, 2936, 1644, 1462, 1376, 1057; EI-MS: m/z 414 [M]⁺ (calcd. (%) for $C_{29}H_{50}O$), 396 [M-H₂O]⁺ (70.4), 381 (37.9), 367 (10.2), 351 (8.7), 329 (38.5), 303 (31.1), 289 (11.4), 1 273 (26.2), 255 (48.5), 241 (7.6), 231 (21.7), 213 (36.2), 199 (13.7), 173 (16.1), 159 (27.6), 145 (33.0), 133 (23.2), 107 (25.2), 95 (26.7), 81 (27.0), 55 (22.9). (7.1), 213 (12.7), 201 (6.6), 200 (9.6), 189 (7.3), 157 (7.7), 133 (12.9), 105 (9.7), 91 (12.1), 81 (15.7), 55 (7.7); FAB-MS: (positive mode) m/z 315 [M + H]⁺; 1 H and 13 C NMR similar to reported literature values $^{12-14}$.

n-Tetracontan-15α-ol (4): Colourless solid; R_f 0.34 (CHCl₃:MeOH); 9.8:0.2); IR (KBr, v_{max} , cm⁻¹: 3420, 2920, 2851, 1731, 1465, 1239, 1072, 950, 830, 724; ¹H NMR (CDCl₃) δ: 3.36 (br m, W1/2 = 7.7 Hz, 1H, H-15β), 2.18 (m, 1H, CH₂), 1.89 (m, 2H, CH₂), 1.41 (m, 2H, CH₂), 1.17 (m, 2H, CH₂), 1.16 (m, 4H, 2 × CH₂), 1.13 (br s, 62 H, 31 × CH₂), 0.76 (t, *J* = 5.5 Hz, 3H, Me-1), 0.74 (t, *J* = 6.0 Hz, 3H, Me-40); ¹³C NMR (CDCl₃) δ: 77.18 (C-15), 33.26 (CH₂), 31.90 (CH₂), 29.69 (29 × CH₂), 29.64 (CH₂), 29.61 (CH₂), 29.57 (CH₂), 29.34 (CH₂), 22.90 (CH₂), 22.67 (CH₂), 16.79 (CH₃-1), 14.10 (CH₃-40); FAB MS (positive ion mode) *m/z* 579 [M+H]⁺ (C₄₀H₈₃O).

Tritriacontan-4,12-diene (5): Yellow semi-solid; IR (KBr, v_{max} , cm⁻¹): 2924, 2854, 1590, 1460, 1365, 1215, 1130, 1090, 725; ¹H NMR: (CDCl₃, 500 MHz):δ: 5.37 (1H, br s, H-4), 5.33 (2H, br s, H-5, H-12), 5.30 (1H, br s, H-13), 2.81 (1H, br m, H₂-6a), 2.76 (1H, br s, H₂-6b), 2.29 (2H, br s, H₂-3), 2.03 (2H, br s, H₂-11), 2.01 (2H, br s, H₂-14), 1.57 (2H, br m, H₂-7), 1.28 (10 H, br s, 5 × CH₂), 1.25 (26 H, br s, 13 × CH₂), 1.22 (8 H, br s, 4 × CH₂), 0.87 (6H, t, J = 6.0- Hz, Me-1, Me-33); ¹³C NMR: (CDCl₃, 500 MHz):δ: 130.44 (C-1), 130.15 (C-12), 128.23 (C-4), 128.11 (C-13), 32.02 (C-6), 30.41 (C-11), 29.92 (C-14), 29.57 (12 × CH₂), 29.21 (C-3), 28.96 (CH₂), 15.98 (Me-1), 14.32 (Me-28); FAB MS (positive mode) m/z 461 [M + H]⁺; (C₃₃H₆₄).

n-Tritetracontan-5α-ol (6): Colourless solid; R_f 0.30 (CHCl₃:MeOH; 9.8:0.2); IR (KBr, v_{max} , cm⁻¹): 3421, 2919, 2850, 1468, 1260, 1067, 980, 870, 725; ¹H NMR (CDCl₃) δ: 3.48 (brm, W_{1/2} = 6.5 Hz, 1H, H-5β), 1.58 (br s, 2H, CH₂), 1.32 (brs, 4H, 2×CH₂), 1.29 (br s, 6H, 3×CH₂), 1.25 (br s, 72 H, 36×CH₂), 0.89 (t, J = 5.5 Hz, 3H, Me-1), 0.86 (t, J = 6.0 Hz, 3H, Me-43); ¹³C NMR (CDCl₃) δ: 77.17 (C-5), 33.2 (CH₂), 31.90 (CH₂), 29.69 (33×CH₂), 29.64 (CH₂), 29.57 (CH₂), 29.35 (CH₂), 22.88 (CH₂), 22.67 (CH₂), 16.78 (CH₃-1), 14.10 (CH₃-43); FAB-MS m/z [M]⁺ (C₄₃H₈₈O) (1.5); FAB MS m/z 621 [M+H]⁺ (C₄₃H₈₉O).

Gallic acid (7): Colourlees solid; $R_f 0.45$; (CHCl₃-MeOH (9:1); m.p. 248-250 °C, yield in minor amount, M^+ 170, $C_7H_6O_5$, identical (IR, MS and NMR) with authentic sample.

β-Sitosterol-3-O-β-D-glucoside (8): R_f 0.47 (CHCl₃: MeOH 9:1); [α]_D²⁰-0.09° (pyridine); IR: (KBr, v_{max} , cm⁻¹): 3429, 2933, 1635, 1376, 1073; FAB-MS: m/z 414 [M-glucose]⁺ (calcd. (%) for C₂₉H₅₀O) (25.7 %), 396 [M-H₂O]⁺ (100), 382 (42.8), 367 (9.4), 354 (4.0), 329 (9.9), 303 (9.0), 288 (11.5), 275 (14.8), 255 (33.2), 279 (9.2), 213 (20.5), 199 (8.4), 173 (9.4), 159 (19.5), 147 (28.3), 133 (17.0), 107 (19.3), 95 (22.9), 81 (25.3), 55 (20.8); ¹H and ¹³C NMR similar to reported literature values¹⁴.

The chemical structure of all the eight compounds are given in Fig. 1.

RESULTS AND DISCUSSION

Compound 1 was obtained as a colourless compound and its molecular formula was deduced as C₃₁H₆₀O₂ from its ¹³C NMR and FAB MS m/z 464. The ¹H NMR spectrum of 1 showed two one-proton multiplets at δ 5.34 and 5.32 assigned to vinylic H-9 and H-10, respectively. Two one-proton doublets at δ 4.06 (J = 6.5 Hz) and 4.04 (J = 6.5 Hz) were attributed to oxygenated methylene H_2 -1. Two one-proton doublets at δ 2.29 (J = 7.5 Hz) and 2.27 (J = 7.5 Hz) were ascribed to C-2' methylene protons adjacent to ester groups. Two multiplets at δ 2.01 and 1.62 integrated for two protons were accounted to C-11 and C-8 methylene protons attached to the vinylic carbons. Two three-proton triplets at δ 0.89 (J = 6.0 Hz) and 0.86 (J = 7.0 Hz) were associated to C-28 and C-3' primary methyl protons, respectively. The remaining methylene protons resonated between δ 1.58-1.25. The ¹³C NMR spectrum of 1 displayed important signals for ester carbon at δ 174.23 (C-1'), oxygenated methylene carbon at δ 64.62 (C-1), methylene carbons at δ 14.34 (C-28, C-3'), vinylic carbons at δ 130.13 (C-9, C-10) and methylene carbons between δ 34.62-22.92. On the basis of this evidence the structure of 1 has been established as *n*-octacos-9-enyl propionate. The values of spectral data were compared with the help of reported values¹⁵.

Compound 2 was isolated as white granular crystals with a melting point of 83-86 °C. It exhibited a peak at m/z 476 [M-H₂O]⁺ on EI-MS, FAB-MS of compound 2 gave a positivemode $[M + H]^+$ ion peak of 495 and a negative mode $[M-H]^$ ion peak of 493. This suggested a molecular ion peak [M]⁺ of 494. The IR spectrum displayed intense absorption bands at 3450, 2917, 1465, 1215 and 757 cm⁻¹, with the band at 3450 cm⁻¹ indicating the presence of a hydroxyl group. ¹H NMR of compound 2 in CDCl₃ gave signals that appeared as one triplet at δ 0.88 (3H, t, J = 6.7 Hz) for methyl proton, a multiplet at δ 1.55 should be signal of methylene protons of CH₂CH₂OH and a triplet at δ 3.64 (CH₂OH) represented a hydroxyl proton. All remaining methylene protons appeared as a broad singlet at δ 1.25-1.33. ¹³C NMR in CDCl₃ produced carbon signals in its spectrum at δ 63.34, 33.04, 32.15, 29.92, 29.88, 29.84, 29.83, 29.66, 29.58, 29.47, 29.30, 25.96, 24.98, 22.91 and 14.33 for methylene attached to hydroxyl group, all methylenes and one methyl carbon. Analysis of compound 2 by ¹H, ¹³C and mass supported the assigned structure 2 as 1-tetratriacontanol¹⁴.

Compound 3 was isolated as colourless needle-shaped crystals that exhibited an m/z of 414 [M]⁺ (calculated for C₂₉H₅₀O), which suggests a molecular ion peak. The IR spectrum of compound 3 showed an absorption bands at 3429, 2936, 1644, 1462, 1376 and 1057 cm⁻¹, with the band at 3429 cm⁻¹ characteristic of a hydroxyl group. The ¹H NMR of compound 3 displayed a one proton broad multiplet at δ 3.51, which was assigned to a carbinol 3α-proton. Two threeproton broad signals at δ 0.63 and 0.94 were ascribed to tertiary C-18 and C-19 protons, respectively. Three doublets at δ 0.92 (J = 7.42 Hz), 0.81 (J = 7.5 Hz) and 0.91 (J = 7.15 Hz), which integrated for three protons each, were attributed to secondary C-21, C-26 and C-27 methyl functionalities, respectively. Another triplet at δ 0.88 (J = 7.5 Hz) was attributed to C-29 methyl protons and a one-proton broad multiplet at δ 5.56 for vinylic proton was attributed to C-6. The remaining methine and methylene protons resonated between δ 2.29 and 1.13. The ¹³C NMR spectrum of compound 3 showed the presence of 29 carbon signals in the molecule. The signals at δ 140.72 and 121.71 were ascribed to vinylic carbons at position C-5 and C-6. Carbon signals appeared at δ 37.31 (C-1), 31.57 (C-2), 71.80 (C-3), 42.19 (C-4), 11 31.87 (C-7, 8), 50.10 (C-9), 36.48 (C-10), 21.11(C-11), 39.81 (C-12), 42.33 (C-12 13), 56.79 (C-14), 24.32 (C-15), 28.26 (C-16), 56.11 (C-17), 11.87 (C-18),13 19.39 (C-19), 36.17 (C-20), 18.82 (C-21), 33.95 (C-22), 26.13 (C-23), 45.85 (C-24), 29.18 (C-25), 19.48 (C-26), 19.07 (C-27), 23.09 (C-28) and 12.32 (C-15 29). Based on the above results, the structure 3 was elucidated as β -sitosterol. The values of spectral data were compared with the help of reported values¹²⁻¹⁴.

Compound 4, an aliphatic alcohol was obtained as a colourless amorphous powder. The FAB-MS and ¹³C NMR data showed to corresponding to molecular formula C₄₀H₈₂O, m/z 579 [M + H]⁺. The ¹H NMR spectrum of 4 displayed a one-proton broad multiplet at δ 3.36 with half width of 7.75 Hz assigned to carbinol 15 β . Two three-proton triplets at δ 0.76 (J = 5.5 Hz) and 0.74 (J = 6.0 Hz) were attributed toterminal C-1 and C-40 primary methyl protons. The remaining methylene protons appeared between δ 2.18-1.13. The ¹³C NMR spectrum 4 showed important signals for carbinol carbon at δ 77.18 (C-15), methyl carbons at δ 16.79 (C-1) and 14.10 (C-40) and methylene carbons between δ 33.26 - 22.90. The absence of signals beyond δ 3.36 in the ¹H NMR spectrum and δ 77.18 in the ¹³C NMR spectrum indicated saturated nature of the compound devoid of any carbonyl group. On the basis of the foregoing account the structure of 4 has been elucidated as *n*-tetracontan-15 α -ol. The values of spectral data were compared with the help of reported values¹⁵.

Compound **5** was obtained from the ethyl acetate extract of rice straw. as semi-solid compound. IR spectrum showed characteristic absorption bands for unsaturation (1590 cm⁻¹) and long aliphatic chain (725 cm⁻¹). Its mass spectrum displayed a molecular ion peak at m/z 460 corresponding to molecular formula of an alkene. The ¹H NMR spectrum of **5** exhibited two one-proton broad signals at δ 5.37 and 5.30 assigned to vinylic H-4 and H-13, respectively. The ¹H NMR signals at δ 2.81 (1H br m), 2.76 (1H, br s), 2.29 (2H, br s), 2.03 (2H, br s) and 2.01 (2H, br s), were associated with the methylene protons adjacent to the vinylic carbons. Two six-proton triplets at δ

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0.87 (J=6.0 Hz) were accounted to C-1 and C-33 primary methyl protons, respectively. The remaining methylene protons resonated at δ 1.57 (2H), 1.28 (10H), 1.25 (26 H) and 1.22 (8H). The 13 C NMR spectrum displayed signals for vinylic carbons at δ 130.44 (C-5), 130.15 (C-12), 128.23 (C-4) and 128.11 (C-13), methyl carbons at δ 15.98 (C-1) and 14.32 (C-33) and the remaining carbons signals between 34.33 - 22.80. On the basis of the foregoing account the structure of **5** has been elucidated as tritriaontan-4,12-diene. The values of spectral data were compared with the help of reported values 16 .

Compound 6 was obtained as an amorphous powder. The FAB MS and ¹³C NMR data of **6** showed a peak at *m/z* 621 $[M + H]^+$ corresponding to molecular formula $C_{43}H_{88}O$. The ¹H NMR spectrum of **6** exhibited a one-proton broad multiplet at δ 3.48 with half-width of 6.5 Hz assigned to carbinol proton H-5 β . Two triplets at δ 0.89 (J = 5.5 Hz) and 0.86 (J = 6.0 Hz) integrated for three-protons each, were attri-buted to terminal C-1 and C-43 primary methyl protons. The remaining methylene protons resonated between δ 1.58-1.25. The ¹³C NMR spectrum of 6 showed important signals for carbinol carbon $(\delta 77.17)$ and methyl carbons $(\delta 16.78, C-1; 14.10 (C-43))$ and methylene carbons between δ 33.21- 22.67. The absence of any signal beyond δ 3.48 in the ¹H NMR spectrum δ 77.17 in the ¹³C NMR spectrum suggested the saturated nature of the molecule. On the basis of spectral data analysis and chemical reactions, the structure of 6 has been characterized as *n*-tritetracontan-5α-ol. The values of spectral data were compared with the help of reported values¹⁵.

Compound 7 exhibited IR absorption bands for OH (3345 cm $^{-1}$), COOH (1689 cm $^{-1}$) and double bond (1620 cm $^{-1}$) functions. An [M] $^{+}$ ion at m/z 170 in its mass spectrum suggested the molecular formula as $C_7H_6O_5$. The 1H and ^{13}C NMR data were compared with reported literature values 17 . On the basis of spectral data analysis and compared with authentic sample, the structure of 7 has been characterized as gallic acid 17 .

Compound **8** was isolated as a colourless powder that exhibited a fragmentation ion peak at an m/z of 414 [M-glucose]⁺. The IR spectrum gave an absorption bands at 3429, 2933, 1635, 1376 and 1073 cm⁻¹, with the absorption band at 3429 cm⁻¹ characteristic of a hydroxyl group. Compound **8** is a glucoside of compound **3**; other than the glucose protons and carbons, the ¹H and ¹³C NMR values were almost the same

as those of **3**. A one-proton doublet at δ 4.97 (J = 11.1 Hz) was attributed to anomeric H-1. Four one proton doublets at δ 13 4.16 (J = 9.3 Hz), 3.92 (J = 4.45 Hz), 3.85 (J = 4.85 Hz) and 3.56 (J = 4.85 Hz) and two one-protons multiplets at δ 3.83 and 3.60, were assigned to the remaining glucose protons. In ¹³C NMR, the H-1 anomeric carbon appeared at δ 103.09 and other glucose carbons appeared at δ 79.11, 78.80, 75.84, 72.22 and 65.58. Based on the above evidence, compound 8 was confirmed as β -sitosterol-3-O- β -D-glucoside. The values of spectral data were compared with the help of reported values¹⁴.

ACKNOWLEDGEMENTS

This research was supported by Basic Science Research Program through the National Research Foundation of Korea (NRF) funded by the Ministry of Education, Science and Technology (2011-0015691).

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