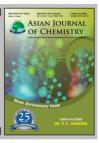




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Antioxidant Activity of Taxillus chinensis Parasitizing on Toona sinensis (A. Juss) Roem

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The present study describes the *in vitro* antioxidant activity of petroleum ether, ethyl acetate and *n*-butanol extracts of *Taxillus chinensis* living parasitically on *Toona sinensis* (A. Juss) Roem. The antioxidant activity of the extracts was assessed by using the following methods: 1,1-diphenyl-2-picrylhydrazyl and 2,2'-azo-*bis*(3-ethylbenzothiazoline-6-sulfonic acid) diamonium salt both free radical scavenging assays and reducing assays. The antioxidant activities of various extracts of *T. chinensis* from *T. sinensis* seemed to follow the same trend *i.e.*, *n*-butanol extracts > ethyl acetate extracts > petroleum ether extracts. The results showed that these extracts, especially the *n*-butanol extracts, could be considered as natural antioxidants and may be useful for curing diseases arising from oxidative deterioration.

Key Words: DPPH, ABTS, Taxillus chinensis, Host.

INTRODUCTION

Oxidative stress represents a disturbance in the equilibrium status of prooxidant/antioxidant reactions in living organisms. The excess of reactive oxygen species (ROS) can damage cellular lipids, proteins or DNA and oxidative imbalance has been implicated in a number of many diseases, including cancers, atherosclerosis and heart diseases as well as in the ageing process¹. In foods, reactive oxygen species can cause lipid peroxidation, which leads to the deterioration of the food². Although some synthetic antioxidant compounds, such as butylated hydroxytoluene (BHT) and butylated hydroxyanisole (BHA), are commonly used for increasing the shelf life of foods. These compounds have been reported to have some side effects³. Therefore, there is a good prospect in searching antioxidants from natural plants. Phenolics or polyphenols, including flavonoids have received considerable attention because of their physiological functions such as antioxidant, antimutagenic and antitumor activities⁴. Consequently, quantitative and qualitative determination of flavonoids and phenolics compounds with strong antioxidant capacity has been done for many plant extracts.

Parasites are unusal plants, more than 2500 species of higher plants are known to live parasitically on other plants. *Taxillus chinensis* belongs to the Loranthaceae family, distributed in Guangdong, Guangxi, Taiwan and Fujian province in china. Some common host trees of *T. chinensis* are mulberry, plum trees, longan trees, lychee trees and oleander trees, *etc.*

The pharmacological activities of *T. chinensis* are widely proved in antioxidant, anticancer, antibacterial and fatty acid synthase inhibitory effect⁵⁻⁸. Besides, it has been proved that the chemical constituents and biological activities of parasites plants depend to a large extent on the host plant^{9,10}. The leaves of *Toona sinensis* (A. Juss) Roem were proved to possess good antioxidant activity^{11,12}. Therefore, the *T. chinensis* living parasitically on this host trees might be a good source of powerful antioxidant. To our best of knowledge, antioxidant activity of *T. chinensis* parasitizing on *T. sinensis* has not been reported so far. In this report, the extracts of different polarity from them were prepared and studied for their antioxidant activity. 1,1-Diphenyl-2-picryl hydrazyl and 2,2'-Azo*bis*-(3-ethylbenzothiazoline-6-sulfonic acid) assay were carried out to evaluate the antioxidant activity of these extracts.

EXPERIMENTAL

Taxillus chinensis whole plants parasitizing on Toona sinensis (A. Juss) Roem was obtained from marketplace vendors from Guangxi Province, in August 2010. The voucher specimens were identified by professor Songji Wei at the Department of Zhuang Pharmacy, Guangxi traditional Chinese Medical University. 1,1-Diphenyl-2-picryl hydrazyl (DPPH) (purity 98 %) was purchased from Wako Chemicals, Japan; 2,2'-azobis-(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS) and butylated hydroxyltoluene (BHT) was purchased from Sigma Aldrich Co., St. Louis, MO, USA; Lutin standard was purchased from J & K Scientific Ltd., Beijing, China; Other

chemicals were purchased from China National Medicine Group Shanghai Corporation, Shanghai, China. All chemicals and solvents used were of analytical grade.

Preparation of extracts: The samples were initially air-dried and then reduced to small particles. The particles selected for analysis were passed through a 40-mesh screen and suspended in 95 % ethanol and the filtrate was collected for three times at every 48 h interval of a 144 h total collection period. The extracts were concentrated, suspended in deionized water and sequentially partitioned with petroleum ether, ethyl acetate and *n*-butanol to obtain three different fractions (water part was discarded). The petroleum ether extract (XPE), ethyl acetate extract (XEAE) and *n*-butanol extract (XBE) were obtained by concentrating the extract liquid under reduced pressure at 40 °C using vacuum rotary evaporator and the dry extracts were stored at -4 °C until use.

Scavenging activity on DPPH radical: The scavenging effect of different fractions on the DPPH radical was measured using a modified version of the method described by Shimada *et al.*¹³. In brief, extracts solution (100 μ L) in 95 % ethanol at different concentration (0.2-2.0 mg mL⁻¹) was added to 4 mL 0.004 % (w/v) solution of DPPH in 95 % ethanol. The reaction mixtures were incubated at 28 °C. The scavenging activities on DPPH radical were determined by measuring the absorbance at 515 nm after 10 min. The antioxidant activity was expressed as a percentage of scavenging of DPPH: SC % = [1-(absorbance of sample)/(absorbance of control)] × 100 %. The control contains all reagents except the extract. The DPPH radical scavenging activity of BHT at the same concentrate were also assayed for comparison. All tests were performed in triplicate and mean were centred.

ABTS^{+•} radical cation scavenging: The antioxidant activities of various solvent extracts in the reaction with the stable ABTS+ radical cation were determined according to the method of Re et al.¹⁴ with slight modification. The reaction between ABTS and potassium persulfate directly generates the blue/green ABTS+• chromophore, which can be reduced by an antioxidant, thereby resulting in a loss of absorbance at 734 nm. ABTS was dissolved in water to a 7 mM concentration. ABTS radical cation (ABTS^{+•}) was produced by reacting ABTS stock solution with 2.45 mM potassium persulfate (final concentration) and allowing the mixture to stand in the dark at room temperature for 12-16 h. The ABTS^{+•} solution (stable for 2 days) was diluted with phosphate buffer (2 mM, pH 7.4) to achieve an absorbance of 0.7 ± 0.05 at 734 nm. Extracts solution (20 µL) in 95 % ethanol at different concentration (0.5-2.0 mg mL⁻¹) were mixed with ABTS^{+•} solution (1.9 mL) and then absorbance was read at ambient temperature after 3 min. PBS solution was used as a control. All tests were triplicated. The radical-scavenging activity of the samples was expressed as SC % = $[(A_{control} - A_{test})/A_{control}] \times 100$ %, in which $A_{control}$ is the absorbance of the control (ABTS^{+•} solution without test sample) and A_{test} is the absorbance of the test sample (ABTS^{+•} solution plus extract).

RESULTS AND DISCUSSION

DPPH radical scavenging activity: The DPPH free radical is a stable free radical, which has been widely accepted

as a tool for estimating free radical-scavenging activities of antioxidants¹⁵. All extracts were found to be effective scavengers against DPPH radical and scavenging abilities increased steadily with concentration (Fig. 1). XBE possessed a highest scavenging capacity of 85.93 % at 2.0 mg mL⁻¹ on DPPH radical and XBE, XEAE were all superior to the positive control, BHT. By comparing the IC₅₀ (the half maximal inhibitory concentration) value of those active fractions, the free radical scavenging activities followed the order: XBE (IC₅₀: 0.31 \pm 0.016 mg mL⁻¹) > XEAE (IC₅₀: 0.58 \pm 0.022 mg mL⁻¹) > BHT (IC₅₀: 1.52 \pm 0.035 mg mL⁻¹) > XPE (IC₅₀: 20.21 \pm 0.147 mg mL⁻¹).

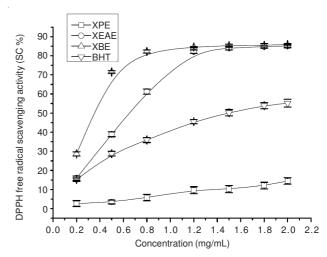


Fig. 1. DPPH radical scavenging activity of all extracts. (Results are mean \pm SD of three parallel measurements. Values are significantly different (p < 0.05), when compared to the control, BHT)

ABTS⁺ radical scavenging activity: The scavenging capacities of various extracts on ABTS⁺ radical were measured and compared (Fig. 2). And all extracts and BHT exhibited concentration-dependent ABTS+oscavenging activity. XBE aslo possessed the highest scavenging capacity on ABTS radical (97.38 %) at 2.0 mg mL⁻¹, which was a little higher than XEAE (96.42 %). Under the same experimental conditions, the SC % of XBE (0.5-2.0 mg mL⁻¹) and XEAE (0.5-2.0 mg mL⁻¹) was higher than BHT at the concentration of 0.2 mg mL⁻¹, but all extracts possessed lower capability of antioxidation than BHT (0.5-2.0 mg mL⁻¹). The ABTS radical scavenging activity of all fractions exhibited the descending order of: XBE (IC₅₀: 0.61 \pm 0.027 mg mL⁻¹) > XEAE (IC₅₀: 0.66 \pm 0.031 mg mL⁻¹) > XPE (IC₅₀: 70.55 \pm 3.48 mg mL⁻¹).

Conclusion

All the methods employed in this work demonstrated significant antioxidant properties for extracts of T. chinensis living parasitically on T. sinensis. XBE proved to be the most efficient extract. The antioxidant properties of all extracts exhibited the descending order of XBE > XEAE > XPE, indicating that there is a correspondence between them. This study is the first report on the study of antioxidant activities of T. chinensis from T. sinensis and n-butanol extract could be considered as potential natural antioxidant sources for medicinal and food applications. Since this investigation is a preliminary study, on the antioxidant mechanisms of specific flavonoids and phenolics components of T. chinensis on this

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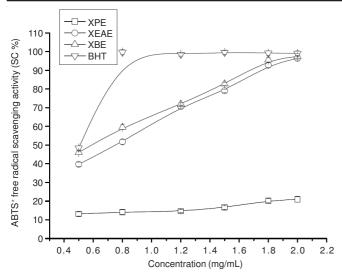


Fig. 2. ABTS⁺ radical scavenging activity of all extracts. Results are mean \pm SD of three parallel measurements. Values are significantly different (p < 0.05), when compared to the control, BHT

plant and the connection between parasites and hosts is absolutely necessary and is in progress.

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