## NOTE

# Synthesis and Crystal Structure of a Novel Square-Hole-Shaped Molecule Mercury(II): $\mathrm{C}_{50} \mathrm{H}_{48} \mathrm{Hg}_{2} \mathrm{~N}_{2} \mathrm{O}_{10} \mathrm{~S}_{4}$ 

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#### Abstract

A novel complex $\left(\mathrm{C}_{50} \mathrm{H}_{48} \mathrm{~N}_{2} \mathrm{O}_{10} \mathrm{~S}_{4} \mathrm{Hg}_{2}\right)$ was synthesized by pyridyl-pyrimidine dithioethers and $\mathrm{Hg}(\mathrm{OAc})_{2}$ in $\mathrm{CH}_{3} \mathrm{OH}$. As the square-holeshaped molecule, the title compound shows five-coordinate core $\left(\mathrm{HgO}_{3} \mathrm{~N}_{1} \mathrm{~S}_{1}\right)$ which is distorted dipyramid geometry. The crystal is trioclinic, space group P-1 with unit cell parameters: $\mathrm{a}=10.10(6) \AA, \mathrm{b}=11.14(5) \AA, \mathrm{c}=13.41(6) \AA, \mathrm{a}=106.3(4)^{\circ}, \beta=103.3(4)^{\circ}, \gamma=99.4(4)^{\circ}, \mathrm{V}=$ $1367(13) \AA^{3}, \mathrm{Z}=1, \mathrm{Mr}=1506.46, \mathrm{Dc}=1.830 \mathrm{~g} / \mathrm{cm}^{3}, \mu=5.830 \mathrm{~mm}^{-1}, \mathrm{~F}(000)=736, \mathrm{R}=0.0592$, $\mathrm{wR}=0.1403$ for 6870 reflections with I > $2 \sigma(\mathrm{I})$.


Key Words: Pyridyl-pyrimidine dithioether, van der Waals forces.

Two kinds of compounds were prepared in our earlier report $^{1}$, in which, as ligand, pyridyl-pyrimidine dithioether displayed different frameworks with different mercuric(II) salt for anions. With in-depth study, a novel square-hole-shaped macromolecular mercury(II) complex has been synthesized, whose crystal structure has emerged through elemental analysis and X-ray diffraction analysis.

All reagents were of analytical grade and were used without further purification. Analyses for carbon, hydrogen and nitrogen were performed on a Perkin-Elmer $140^{\circ} \mathrm{C}$ analyzer. Infrared spectra ( $4000-400 \mathrm{~cm}^{-1}$ ) were recorded with a Bruker Vector 22 FT-IR spectrophotometer on KBr disks.

Synthesis: The ligand (L) was synthesized with 1,3dibromopropane and 4-(pyridin-4-yl)pyrimidine-2-thiol according to literatures ${ }^{2-5}$, which then reacted with $\mathrm{Hg}(\mathrm{OAc})_{2}$ in $\mathrm{CH}_{3} \mathrm{OH}$. The mixture was refluxed for 0.5 h and then cooled to room temperature. The solution was filtered, then title complex was obtained by slow evaporation of the filtrate at room temperature in $70 \%$ yield, whose structure has been determined by X-raycrystallography. The formula of crystals is $\mathrm{C}_{50} \mathrm{H}_{48} \mathrm{~N}_{2} \mathrm{O}_{10} \mathrm{~S}_{4} \mathrm{Hg}_{2}(\mathbf{1})$ of which structure is showed in Fig. 1.

Anal. calcd. for $\mathrm{C}_{50} \mathrm{H}_{48} \mathrm{~N}_{2} \mathrm{O}_{10} \mathrm{~S}_{4} \mathrm{Hg}_{2}$ : C: 43.95; H: 3.54; N : $2.05 \%$. Found: C: 43.92; H: 3.54; N: $2.06 \%$. IR (KBr, $\nu_{\text {max }}$, $\mathrm{cm}^{-1}$ ) : 3432 (m), 1593 (w), 1562 (s), 1541 (m), 1480 (w), 1401 (m), 1382 ( s ), 1351 (m), 1322 (m), 1203 (m), 1026 (w), 831 (w).


Fig. 1. Molecular structure of the mercury complex

Crystal structure determination: A single crystal of compound with dimensions of $0.08 \mathrm{~mm} \times 0.20 \mathrm{~mm} \times 0.40$ mm was selected for the crystallographic data collection at 291(2)K and structure determination on a Bruker SMART CCD-4 K diffractometer employing graphite-monochromated $\mathrm{MoK}_{\alpha}$ radiation $(\lambda=0.71073 \AA)$. A total of 6870 reflections were collected in the range of $1.65^{\circ} \leq \theta \leq 25.0^{\circ}$, of which 4738 reflections were unique with $\mathrm{R}_{\mathrm{int}}=0.093$. The data were collected using SMART and reduced by the program SAINT. All the structures were solved by direct methods and refined by full-matrix least squares method on $\mathrm{F}^{2}$ obs by using SHELXTL-PC software package. Non-hydrogen atoms were placed in geometrically calculated positions. Hydrogen atoms were added according to theoretical model. The final full-

| TABLE-2 |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | INTRAMOLECULAR HYDROGEN BOND DISTANCES ( $(\AA)$ AND ANGLES $\left({ }^{\circ}\right)$ |  |  |  |  |  |  |
| Type $(\mathrm{D}-\mathrm{H} \cdots \mathrm{A})$ | $\mathrm{d}(\mathrm{D}-\mathrm{H})$ | $\mathrm{d}(\mathrm{H} \cdots \mathrm{A})$ | $\angle(\mathrm{DHA})$ | $\mathrm{d}(\mathrm{D} \cdots \mathrm{A})$ | A |  |  |
| $\mathrm{C} 9-\mathrm{H} \cdots \cdots \mathrm{N} 6$ | 0.9300 | 2.5800 | 134.00 | $3.29(3)$ | $\mathrm{x},-1+\mathrm{y},-1+\mathrm{z}$ |  |  |
| $\mathrm{C} 10-\mathrm{H} 10 \mathrm{~A} \cdots \mathrm{O} 2$ | 0.9700 | 2.5100 | 148.00 | $3.37(3)$ | $1-\mathrm{x},-\mathrm{y}, 1-\mathrm{z}$ |  |  |
| $\mathrm{C} 16-\mathrm{H} 16 \cdots \mathrm{O} 5$ | 0.9300 | 2.5300 | 122.00 | $3.12(2)$ | $1-\mathrm{x},-\mathrm{y}, 2-\mathrm{z}$ |  |  |

matrix least-squares refinement including 354 variable parameters for 3395 reflections with $\mathrm{I}>2 \sigma(\mathrm{I})$ and converged with unweighted and weighted agreement factors of:

$$
\begin{equation*}
\mathrm{R}_{1}=\Sigma\left(\left\|\mathrm{F}_{0}|-| \mathrm{F}_{\mathrm{c}}\right\|\right) / \Sigma\left|\mathrm{F}_{0}\right|=0.0592 \tag{1}
\end{equation*}
$$

and

$$
\begin{equation*}
\mathrm{wR} \mathrm{R}_{2}=\left\{\Sigma\left[\mathrm{w}\left(\mathrm{~F}_{0}^{2}-\mathrm{F}_{\mathrm{C}}^{2}\right)^{2}\right] / \Sigma \mathrm{w}\left(\mathrm{~F}_{0}^{2}\right)^{2}\right\}^{1 / 2}=0.1403 \tag{2}
\end{equation*}
$$

where $\mathrm{w}=1 /\left[\sigma^{2}\left(\mathrm{~F}_{0}{ }^{2}\right)+(0.0584 \mathrm{P})^{2}\right]$ and $\mathrm{P}=\left(\mathrm{F}_{0}{ }^{2}+2 \mathrm{~F}_{\mathrm{C}}{ }^{2}\right) / 3$. The maximum and minimum peaks on the final difference Fourier map are corresponding to 1.63 and $-1.17 \mathrm{e} / \AA^{3}$, respectively.

The parameters of selected bonds and hydrogen bonds are shown in Tables 1 and 2 correspondingly. The molecular structure and packing diagram of the compound are shown in Figs. 1 and 2, respectively. The title compound crystallizes in the triclinic system of $\mathrm{P}-1$ space group. The shape of complex is square-hole shaped. In $\mathrm{HgO}_{3} \mathrm{~N}_{1} \mathrm{~S}_{1}$ core, Hg atom is coordinated by three O atoms and one N atom as well as one S atom in a distorted triangular dipyramid geometry, in which the two O atoms ( $\mathrm{O} 1, \mathrm{O} 3$ ) are from different acetate anion and one O atom is from coordinated water molecule, the N atom and S atom are from distinct ligands. The axial bond angle $(\mathrm{O} 5-\mathrm{Hg} 1-$ S 1 a ) is $165.8(4)^{\circ}$. The bond length (Hg1-S1_a) is 3.068(19) $\AA$ that is similarly longer with named classical coordination bond. In addition, the two $\mathrm{Hg}-\mathrm{O}$ bond lengths ( $\mathrm{Hg}-\mathrm{O} 1$ and $\mathrm{Hg}-\mathrm{O} 3$ ) are not same. The pyrimidyl and pyridyl unit in the middle of chelate ring of the complex are planar for intermolecular hydrogen bonds respectively. The chelate ring shows a square-hole because non-axial bond angles around Hg (II) are about $90^{\circ}$.

## TABLE-1

SELECTED BOND DISTANCES ( $\AA$ ) AND ANGLES $\left({ }^{\circ}\right)$

| Hg1-O1 |  |  |  |
| :---: | :---: | :---: | :---: |
| Hg1-O3 | $2.253(17)$ | S1_a-Hg1-O1 | $87.5(5)$ |
| Hg1-O5 | $2.529(17)$ | S1_a-Hg1-O3 | $91.4(5)$ |
| Hg1-N3 | $2.356(16)$ | S1_a-Hg1-N3 | $92.4(5)$ |
| Hg1-S1_a | $3.068(19)$ | Hg1_a-S1-C1 | $99.4(6)$ |



Fig. 2. Packing diagram of the title compound via van der Waals forces

## Conclusion

With transforming the distance of two sulfur atom and the substitutional position of nitrogen in ligand, which is pyridyl-pyrimidin dithioether, the new mercuric(II) complex is still macrocycle, but the cycle is made of two central ions and two part of ligand-moleculars.

Supplementary material: Crystallographic data for the structure reported in this communication have been deposited with the Cambridge crystallographic data center as supplementary publication No. CCDC 760047.

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