Asian Journal of Chemistry; Vol. 25, No. 13 (2013), 7639-7640



ASIAN JOURNAL OF CHEMISTRY

http://dx.doi.org/10.14233/ajchem.2013.14589



NOTE

Synthesis and Crystal Structure of 1,2-bis(Pyrazin-2-ylthio)ethane

M. OUYANG², X.Z. ZHAO³ and Y.M. JIANG^{1,*}

¹College of Chemistry and Chemical Engineering, Guangxi Normal University, Guilin 541004, Guangxi Province, P.R. China ²Department of Chemistry and Life Science, Hechi University, Yizhou 546300, Guangxi Province, P.R. China ³China Coal Pingshuo Coal Industry Limited Liability Company, Shuozhou 036000, Shanxi Province, P.R. China

*Corresponding author: Tel: +86 77 35822850, E-mail: ymjiang@mailbox.gxnu.edu.cn

(Received: 17 September 2012;

Accepted: 5 July 2013)

AJC-13773

A new organic compound has been synthesized and characterized by elemental analysis and X-ray diffraction. The results show that the present compound 1,2-*bis*(pyrazin-2-ylthio)ethane is a double-thioether compound. At first, we tried to obtain a Co(II) complex containing pyrazine-2-thiol ligand. Unexpectedly, two new organic ligands of 1,2-*bis*(pyrazin-2-ylthio)ethane and 1-(2-(pyrazin-2-ylthio)ethyl)-pyrazine-2(1*H*)-thione which is isomer to each other were obtained from pyrazine-2-thiol.

Key Words: Thioether, Synthesis, Structure.

For a long time, thioether-type ligands have been a focus for researchers due to their good flexibilities and versatile coordination modes. Therefore, there were a lot of reports about the crystal structure related to thioether-type compounds recently^{1,2}. As a part of our series of research about thioether-type compounds, we present herein the crystal structure of the 1,2-bis(pyrazin-2-ylthio)ethane. At first, we tried to obtain a Co(II) complex containing pyrazine-2-thiol ligand. Unexpectedly, two new organic ligands of 1,2-bis(pyrazin-2-ylthio)ethane and 1-(2-(pyrazin-2-ylthio)-ethyl)pyrazine-2(1H)-thione which is isomer to each other were synthesized from pyrazine-2-thiol. The two compounds, one of which was previously reported last year³, were separated according to their different solubility in water.

All solvents and chemicals obtained from commercial sources were of reagent grade and used without further purification. The elemental analyses for C, H, N and S were performed on a Perkin-Elmer 240 °C elemental analyzer. Crystal data were collected at 298(2) K on a Bruker SMART CCD area detector diffractometer.

Synthesis of the title compound: The syntheses process of the title compound is according to the literature³. To a solution of pyraine-2-thiol (250 mg, 2 mmol) and 1,2-dibromoethane (0.09 mL, 1 mmol) in 15 mL sodium ethoxide ethanol solution, a aqua solution containing CoCl₂·6H₂O (47.9 mg, 0.2 mmol) was added with drop. The resulting solution was stirred at 352 K for 3 h and a portion of ethanol was removed by distillation under reduced pressure, then a yellow powder

was obtained. The products were placed in distilled water and found that only a small part of the powder can dissolve in it. They were recrystallized with anhydrous ethanol to give two different colours of crystals suitable for X-ray diffraction. We found that the compound which is poorly soluble in water is the title compound. Anal. calcd. $C_{10}H_{10}N_4S_2$ for (%): C, 47.98; H, 4.02; N, 22.38; S, 25.62. Found (%): C, 48.12; H, 3.82; N, 22.56; S, 25.50.

X-ray crystallographic determination: A yellow-block single crystal of the title compound with dimensions of 0.40 mm \times 0.38 mm \times 0.19 mm was used for structure determination. Data were collected at 298(2) K on a Bruker SMART CCD area detector diffractometer equipped with a graphitemonochromatic MoK_{α} ($\lambda = 0.71073$ Å) radiation by using a ϕ -ω mode in the range of 3.73 \leq θ \leq 25.01°. The structure was solved by direct methods and refined by full-matrix leastsquares on F² using SHELXL-97 program package⁴. All nonhydrogen atoms were assigned anisotropically. The hydrogen atoms were generated theoretically and treated by a mixture of independent and constrained refinement. The final cycle of refinement converged at R = 0.0419, wR = 0.0949. Selected bond lengths and bond angles of the title compound are listed in Table-1 and the hydrogen bond lengths and bond angles of the compound in Table-2.

Crystal structure description: The title compound, $C_{10}H_{10}N_4S_2$, is isostructural with 1-[2-(pyrazin-2-ylthio)ethyl]-pyrazine-2(1*H*)-thione reported last year by our group, whose structure has been described in detail³.

7640 Ouyang et al. Asian J. Chem.

TABLE-1 SELECTED BOND LENGTHS (Å) AN D BOND ANGLES (°) OF THE TITLE COMPOUND						
Bond	Dist (Å)	Bond	Dist (Å)			
N1-C1	1.332(4)	N1-C4	1.340(4)			
S1-C1	1.758(3)	S1-C5	1.807			
Angle	(°)	Angle	(°)			
C1-N1-C4	115.6(3)	C2-N2-C3	115.5(3)			
C1-S1-C5	102.62(15)					

TABLE-2 HYDROGEN BOND LENGTHS (Å) AND BOND ANGLES (°) OF THE COMPOUND						
D–H···A	d (D-H)	d (H···A)	d (D···A)	∠ (DHA)		
C5-H5B···S1	0.97	3.09	3.785 (37)	129		
C2-H2···N1	0.93	2.68	3.482 (46)	145		
C4-H4···N2	0.93	2.83	3.603 (53)	141		
C5-H5A···N1	0.97	3.08	3.714 (13)	124		

In the compound, the plane of the two pyrazine ring from the same molecule (mean deviation from plane = 0.0 Å) are parallel to each other and the distance of the two plane is 1.477 Å (Fig. 1). The distances of C-S bonds are 1.758(3) Å and 1.807 Å, which are longer than C=S double bonds and match with the normal C-S range l. In the structure, there is short of any classical hydrogen bonds due to lack of relevant donors and acceptors for such bonds. However, there are many nonclassical hydrogen bonds such as C-H···N (C_2 -H₂···N₁, C_4 -H₄···N₂, C_5 -H₅A···N₁) and C-H···S (C_5 -H₅B···S₁), which are beneficial to the stability of the crystal structure (Fig. 2).

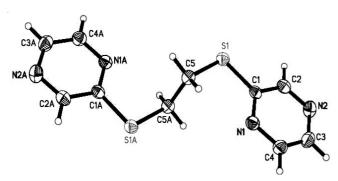


Fig. 1. Molecular view of the title compound showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 30 % probability level. H atoms are represented as small spheres of arbitrary radii

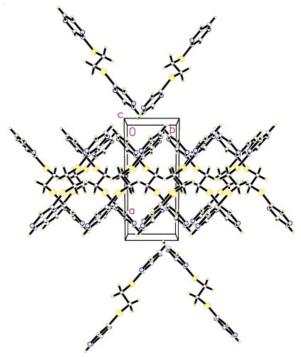


Fig. 2. Crystal packing of the title compound viewed along the c axis

ACKNOWLEDGEMENTS

This work was funded by the Guangxi Science Foundation of the Guangxi Zhuang Autonomous Region of the People's Republic of China (grant No. 2010GXNSFD 013017), the Project of Education Department of Guangxi Province (No. 201106LX593) and Youth Foundation of Hechi University (No. 2011B-N004).

REFERENCES

- H.H. Wang, C.Y. Zhang, Y. Cui and Y.B. Xie, Acta Cryst., E65, 3268 (2009).
- J.R. Black, N.R. Champness, W. Levason and G. Reid, J. Chem. Soc., Chem. Commun., 1277 (1995).
- X.Z. Zhao, J.J. Jia, M. Ou-Yang, F.P. Huang and Y.M. Jiang, *Acta Cryst.*, E67, 418 (2011).
- (a) G.M. Sheldrick, SHELXS-97. Program for X-ray Crystal Structure Solution, Göttingen University, Germany (1997); (b) G.M. Sheldrick, SHELXL-97. Program for X-ray Crystal Structure Refinement, Göttingen University, Germany (1997).
- A.B. Zolotoy, M. Botoshansky, M. Kaftory, J.R. Scheffer and J. Yang, Acta Cryst., C58, 220 (2002).