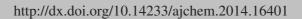




# ASIAN JOURNAL OF CHEMISTRY





## Antioxidant and Antifungal Activities of Some Agro Wastes and Their Phenolic Acid Profile

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Received: 20 September 2013;

Accepted: 30 November 2013;

Published online: 15 February 2014;

AJC-14734

In present study some selected agro wastes extracted with different solvent *i.e.* methanol, ethanol, chloroform and acetone were investigated for their antifungal and antioxidant activities and phenolic acid profile. The tested waste materials contained appreciable amount of total phenolic (0.27-198.90 GAE mg/g DW) and total flavonoids (0.02-96.20 CE mg/g DW), DPPH scavenging capacity (25.23-75.65 %), reducing power as 0.342.-1.74 (at 10 mg/mL extract concentration, absorbance at  $\lambda = 700$  nm) and percent inhibition of linoleic acid (43.23-82.34 %). Antifungal activity of selected agro wastes extracts was assessed by using disc diffusion method and MIC against *Aspergillus flavus* and *Aspergillus parasiticus* strains. Citrus leaves and pomegranate peel extract showed remarkable antifungal activity compared to other extracts with the maximum zone of inhibition 37 and 35 mm, respectively. In addition, GC/MS analysis showed the presence of seven different phenolic acids *i.e.*, gallic acid, *p*-coumaric acid, caffeic acid, ferulic acid, protocacheuic acid, cinamic acid and vanillic acid.

Keywords: Agro wastes, Aspergilllus, Antifungal activity, GC/MS analysis, Antioxidant potential.

## INTRODUCTION

Agro-industrial wastes are rich source of many important bioactive compounds, so their potential use in food industry as natural antioxidant and antimicrobial agents may be a good approach to replace synthetic preservatives. However, it may also solve the environmental problems concerning disposal of wastes and their by-products and can be a source of profit for peoples living in industrial areas<sup>1</sup>.

Recent research reports reveal that phytophenols are the most abundant secondary metabolites which are found in plants with more than 8,000 structures. They may be of simple structures like phenolic acids and complex compounds like tannins<sup>2</sup>. Wide spectrums of biological activities like antioxidant, antimicrobial, antiallergenic, anti-inflammatory, antithrombotic, anti-orthrogenic have been accredited to phenolic compounds<sup>3</sup>.

Many studies revealed the inhibitory potential of phenolics on fungal growth<sup>4,5</sup>. Kim *et al.*<sup>6</sup> demonstrated that phenolics such as vanillyl acetone, salicylic acid, vanillin, thymol and cinnamic acid may cause inhibition of *A. flavus* growth by defense system of targeting the mitochondrial oxidative stress.

At present research has focused to investigate some new antimicrobial compounds which show diverse chemical structures and better mechanisms of action<sup>7</sup>. Natural plant products may successfully replace synthetic chemicals and provide an alternative safe mode to protect food commodities

as well as other agricultural crops of nutritional significance from development and growth of fungi *i.e.*, *A. flavus* and *A. parasiticus* as well as oxidative deterioration. Many studies have confirmed the presence of bioactive compounds having good antimicrobial potential in lemon peels<sup>8</sup>, pomegranate peels<sup>9</sup> and grape seeds<sup>10</sup>.

A huge amount of wastes including mango peel, citrus peel and onion peel, garlic peel, pomegranate peel citrus and leave is produced at industrial and home level, disposal of which creates an environmental problem. It is well known that the peel and seed fractions of some vegetables and fruits have higher bioactivities components than other fractions<sup>11,12</sup>. However, a limited number of agro-wastes have been investigated for their phenolic profiles. Therefore, in the present research project characterization of active phenolic acids that can act as antifungal, antioxidant, anti-inflammatory and antibacterial agents was established by gas chromatography mass spectrometry.

The objective of present research was to study the antimicrobial and antioxidant properties of some locally producing wastes against food borne pathogens so as to reduce economic loss of cereals and food commodities by oxidation and fungal contamination.

## **EXPERIMENTAL**

Lemon peel (*Citrus limon*), pomegranate peel (*Punica grantum L.*), mango peel (*Mangiferaindica L.*), mango kernel

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(Mangiferaindica L.), lemon leaves (Citrus limon), garlic leaves (Allium sativum L.) and onion leaves (Allium cepa L.) were collected from local agricultural resources of Faisalabad, Pakistan. After washing with tap water sample were subjected to air drying.

2,2,-Diphenyl-1-picrylhydrazyl, folin-ciocalteu reagent, gallic acid, ascorbic acid, trichloro-acetic acid, aluminum chloride, sodium nitrite, ammonium thiocyanate, feric chloride, potassium fericyanate and ferous chloride were purchased from Sigma Chemical Co. (St Louis, MO, USA) and standards of phenolic acids (caffeic acid, *p*-coumaric acid, vanillic acid, ferulic acid, cinamic acid, protocacheuic and gallic acid). All other chemicals were of analytical grade *i.e.*, chloroform, ehtyalacetate, methanol, HCl, *bis*(trimethylsilyl) trifluoro-acetamide (BSTFA), tertiary butyl hydroxytoluene (TBHQ), butylatedhydroxytoluene (BHT) trimethychlorosilane (TMCS) used in this study were purchased from merck (Darmstadt, Germany). Potato dextrose agar (PDA) culture media and discs of antibiotic standard were obtained from Oxoid Ltd., (Hampshire, UK).

Extraction of antioxidant/antifungal components: Antifungal and antioxidant components from dried samples were extracted by using different solvents *i.e.* methanol, ethanol, chloroform and acetone. The extraction of powdered sample (20 g) of 80 mesh size was done at room temperature in an orbital shaker (Gallenkamp, UK) with 200 mL solvent for overnight. Supernatant and sediments were parted by filtration with Wattman no. 1 filter paper and separated residue was extracted (two time) with respective solvent and filtered then combine filtrate. Solvent was evaporated at 45 °C under reduced pressure, using a rotary evaporator (N-N Series, Eyela, Rikakikai Co. Ltd, Tokyo, Japan) and the yield of nearly dried extract was calculated. The dried extracts were stored at 4 °C, till further used.

Estimation of total phenolic contents (TPCs): The total phenolic contents of wastes extracts were estimated according the procedure of Chaovanalikit and Wrolstad , using Folin-Ciocalteu reagent. Briefly, extract (50 mg) was added in Folinciocalteu reagent (0.5 mL) and deionized water (7.5 mL). Then incubate at room temperature (10 min) and add 20 % w/v sodium carbonate solution (1.5 mL) and heat mixture for 20 min at 40 °C in water bath then cooled by icing. Absorbance was taken at 755 nm with spectrophotometer (U-2001, Hitachi Instruments Inc., Tokyo, Japan). To calculate amount of TPs calibration curve of Gallic acid at concentration range of 10-100 ppm ( $R^2 = 0.9986$ ) was used and unit to express results was Gallic acid equivalents (GAE) g/100 g of dry plant matter. Results were expressed by taking average.

Estimation of total flavonoid contents (TFCs): The amount of total flavonoid contents was assessed according to well established spectrophotometric procedure<sup>14</sup>. Briefly, 1 mL of waste extract (0.1 mg/mL) was added in a 10 mL volumetric flask containing 4 mL water. Then added 0.3 mL NaNO<sub>2</sub> (5 % solution) keep for 5 min, then 0.3 mL of 10 % solution of AlCl<sub>3</sub> was added; after next 6 min, add 2 mL of 1 M NaOH. Then add water (2.4 mL), mixed thoroughly and absorbance noted at 510 nm. Total flavonoid contents were presented as catechin equivalents (g/100 g of dry plant matter). Each sample was analyzed three times and averaged.

DPPH scavenging assay: 2,2'-Diphenyl-1-picryl-hydrazyl (DPPH) free radical scavenging activity of crude extracts was evaluated by following previous adopted procedure<sup>15</sup>. Briefly, add 1 mL of extract (25 μg/mL of dry matter in methanol) in DPPH solution that was freshly prepared at concentration 0.025 g/L and absorbance was taken at 515 nm after 0, 0.5, 1, 2, 5 and 10 min using a spectrophotometer. The un-scavenged DPPH radical (DPPH\*) was estimated from calibration curve and comparison was made between the radical scavenging activities of various extracts by using absorbance value taken at 5th min.

Antioxidant activity (linoleic acid system): Antioxidant activity of different agriculture waste extracts was also evaluated by the oxidation of linoleic acid system<sup>15</sup>. Accurately weighed 5 mg extract was mixed with linoleic acid (0.13 mL), 99.8 % ethanol (10 mL) and 0.2 M sodium phosphate buffer (10 mL) of pH 7 in a 25 mL of volumetric flask. The volume was made up with distilled water and incubated for 360 h at 40 °C. From peroxide value extent of oxidation of linoleic acid was estimated using earlier reported thiocyanate protocol<sup>16</sup>. According to this 10 mL of ethanol (75 % v/v), 0.2 mL of ammonium thiocyanate solution (30 % w/v in water), 0.2 mL of sample solution and 0.2 mL of ferrous chloride (FeCl<sub>2</sub>) solution (20 mM in 3.5 % HCl; v/v) was added sequentially and mixed for 3 min. Absorption was noted at 500 nm by spectrophotometer (U-2001, Hitachi Instruments Inc., Tokyo, Japan). Control contained all reagents except sample and butylatedhydroxytoluene (BHT) synthetic antioxidant was taken as positive control. Percent inhibition of linoleic acid oxidation was estimated by using the formula given below:

100-[(Abs. increase of sample at 360 h/Abs. increase of control at 360 h)  $\times$  100]

**Determination of reducing power:** The antioxidant potential of different waste materials was also investigated by measuring their reducing power following a reported of Yen *et al.*<sup>16</sup>, with some amendments. Concentrated extract (2.5-10 mg) was added ina 5 mL sodium phosphate buffer of pH 6.6 (0.2 M) and 5.0 mL of potassium ferricyanide (1 %); then heating of mixture was done at 50 °C for 20 min. Then add 5 mL of trichloroacetic acid (10 %) and then centrifuged at 980 g at 5 °C in a refrigerated centrifugefor 10 min (CHM-17; Kokusan Denki, Tokyo, Japan). From upper layer 5 mL of mixture was taken and added 5 mL of water following 1 mL of ferric chloride (0.1 %) solution and absorbance was noted at 700 nm by spectrophotometer (U-2001, Hitachi Instruments Inc., Tokyo, Japan). Three analysis were performed at each sample and results were averaged.

**Fungal strains:** The pure strains of *A. flavus* and *A. parasitic* were obtained from National Institute for Biotechnology and Genetic Engineering (NIBGI), Faisalabad, Pakistan. Strains were cultured at potato dextrose agar (PDA) (Oxoid, UK) at 28 °C for overnight and maintained at 4 °C.

**Disc diffusion method:** Antifungal activity of waste extracts was tested by disc diffusion method <sup>17</sup> with slight modifications. The autoclaved PDA medium was transferred in sterilized petri dishes and was inoculated with particular test fungi. The media was allowed to solidify then paper discs of 6 mm in diameter containing 50  $\mu$ L ofwaste extract were placed at PDA surface. As positive reference flumequine was used whereas for

negative control disc without sample extract was used. The plates were placed in incubator (28 °C) for 48 h and inhibitory zones (mm) were measured by zone reader.

Minimum inhibitory concentration (MIC): Antifungal activity of waste extracts was also assessed by measuring minimum inhibitory concentration (MIC) following the method of Misra and Dixit<sup>18</sup>, with slight modification. A serial dilution of extract (10-100 µg/mL) was made in microplate with one growth control. SDB (160 uL) was added on to the microplate with 20 µL of tested solution. Then 20 uL culture of  $5 \times 10^5$  CFU/mL of the *A. flavus* and *A. parasiticus* was poured on to the separate micro plate. The plates were kept in incubator for 24 h at 28 °C and then to stop the growth in controls plates were transferred at 22 °C. By micro titer plate reader optical density was measured at 620 nm and MICs were calculated.

**Extraction and derivatizationof phenolic acids for GC/MS analysis:** Phenolic acids were extracted/hydrolyzed and derivatized according to the adopted procedure of Tokusoglu *et al.*<sup>19</sup>. 25 mL acidified methanol containing HCl (1 % v/v) and TBHQ (0.5 mg/mL) was added in 5 g plant material along with 5 mL of HCl (1.2 M,) and refluxed for 2 h with carefully stirring. Then cooled mixture to room temperature and centrifuged at 5000 rpm for 10 min. Separate upper layer and to remove air bubbles mixture was sonicated for 5 min.

Phenolic acids were further extracted with ethyl acetate (1:1, v/v) and fraction obtained from extraction was treated with NaHCO<sub>3</sub> (0.5 M) (1:1, v/v) for three times and evaporation was done under nitrogen flow. The dried extract was dissolved in ethyl acetate and solution was again dried by Na<sub>2</sub>SO<sub>4</sub> for 5 min. For silation dried solution (400  $\mu$ L) was transferred to a vial and added 100  $\mu$ L bis(trimethylsilyl) trifluoroacetamide (BSTFA) +1 % trimethychlorosilane (TMCS). The vial was heated in water bath at 70 °C for 15 min. From the silylated mixture 1  $\mu$ L was analyzed by GC-MS.

Gas chromatography-mass spectrometry (GC-MS): The GC/MS used to analyze silylated samples was Shimadzu gas chromatograph (GC/MS (6890N) Agilent-Technologies, California, USA) coupled to the Shimadzu mass spectrometer (QP 2010 PLus). Separation occurred in HP-5MS, fused silica capillary column (30 m × 0.25 mm, 0.25 µm) with the injection volume of 1 µL in a "split" mode (split ratio, 1: 100) and Helium (p = 100 kPa) was used as carrier gas. The temperature of injector (auto injector) was set 250 °C, oven was 280 °C, interface was 280 °C and detector (scanning mode) was operated at 50-550 m/z @1.5 scan/sec. The integration of GC-MS system was carried out using lab solutions-GC-MS software. Phenolic acids were identified by comparing their retention time with

those of authentic compound that were eluted under same conditions and with spectral data obtained from Wiley 8 library.

**Statistical analysis:** The three different samples of each agro waste were assayed. All the measurements were made in triplicate and the results were reported as mean  $(n = 3 \times 3) \pm SD$   $(n = 3 \times 3)$ . Two way analysis of variance ANOVA was performed to analyze the data using Minitab 2000 versions 13,2, Statistical software (Minitab Inc. Pennysalvania, USA) at 5% significance level.

#### RESULTS AND DISCUSSION

The yields of antioxidant and antifungal componentsextracted from various waste materials using different solvents are presented in Table-1. Among different extraction solvents used in present study methanol showed highest yield ranged from 12.30 to 57.71 % followed by ethanol (11.69 -52.69 %), acetone (10.21-25.35 %) and chloroform (7.11-11.39 %). The highest extract yield (57.71 %) was achieved with absolute methanol from pomegranate peels while lowest was obtained (7.11 %) by chloroform from citrus peel. Results showed that there was a significant difference (P < 0.05) among the yields obtained from different wastes with the different solvent. The highest extract yield (57.71 %) from pomegranate peels and mango peel 32.22 % as found in our present study were found somewhat higher than those investigated by Sultana et al.<sup>20</sup>. They reported 29.9 % yield for pomegranate peel, 24.6 % for banana peel, 21.5 % for citrus peel and 16.4 % for apple peel with 80 % methanol. Differences in yields obtained from different agro wastes with various solvents depend on availability of different extractable component, nature of soil and agroclimatic conditions<sup>21</sup>. According to our results methanol and ethanol exhibited highest yield compared to acetone and chloroform. Methanol is usually recommended to be a better solvent for the extraction of antioxidant compounds because of polarity and availability. The effectiveness and ability of methanol may increase using water as a co solvent especially when extraction of antioxidants of various kinds is required<sup>15</sup>.

Total phenolic and total flavonoid contents: The total phenolic and total flavonoid contents investigated for different agro waste using different solvents are shown in Table-2. The interest in the phenolics has increased outstandingly due to their prominent free radical scavenging activity<sup>22</sup> therefore mostly total phenolic assay is used to assess the antioxidant potential of plant extracts. Many researchers demonstrated a well established correlation between antioxidant activity and phenolic profile<sup>23</sup>.

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	EXTRACT YIELD	(g/100 g DW) OF VARIOU	S AGRO WASTES	
Plant used	Methanol	Ethanol	Chloroform	Acetone
Mango peel	32.22±0.67 <sub>d</sub> <sup>c</sup>	25.51±0.51 <sub>c</sub> <sup>b</sup>	11.10±0.23 <sub>a</sub> <sup>c</sup>	21.93±0.45 <sub>c</sub> <sup>d</sup>
Pomegranate peel	50.71±1.10 <sub>d</sub> <sup>d</sup>	$57.69 \pm 1.32_{d}^{d}$	11.39±0.22 <sub>a</sub> <sup>c</sup>	25.35±0.61 <sub>b</sub> <sup>d</sup>
Citrus peel	14.92±0.29° a	19.40±0.43 <sub>d</sub> <sup>a</sup>	$7.11\pm0.15_{a}^{b}$	11.50±0.23 <sub>b</sub> <sup>b</sup>
Citrus leaves	23.16±0.46 <sub>d</sub> <sup>b</sup>	20.81±0.42 <sub>d</sub> <sup>a</sup>	11.60±0.24 <sub>a</sub> <sup>d</sup>	19.21±0.41°
Onion leaves	15.20±0.32 <sub>d</sub> <sup>a</sup>	16.20±0.33 <sub>d</sub> <sup>a</sup>	$8.21\pm0.14_{a}^{b}$	12.91±0.27 <sub>c</sub> <sup>b</sup>
Garlic leaves	12.30±0.26 <sub>d</sub> <sup>a</sup>	11.69±0.25 <sub>c</sub> <sup>a</sup>	$8.42\pm0.17_{a}^{b}$	10.21±0.21 <sub>b</sub> <sup>a</sup>
Mango kernel	27.65±0.51 <sub>d</sub> <sup>b</sup>	25.35±0.53 <sub>d</sub> <sup>b</sup>	10.61±0.23 <sub>a</sub> <sup>c</sup>	13.65±0.27 <sub>a</sub> <sup>b</sup>

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TABLE-2 TOTAL PHENOLICS (GAE mg/g DW) AND FLAVONOID (CE mg/g DW) CONTENTS OF VARIOUS AGRO WASTES									
Plant used		Total pheno	l contents		Total flavonoid contents				
Plant used	Methanol	Ethanol	Chloroform	Acetone	Methanol	Ethanol	Chloroform	Acetone	
Mango peel	116.80±2.10 <sub>d</sub> <sup>c</sup>	122.60±2.59 <sub>d</sub> <sup>c</sup>	14.34±0.29 <sub>a</sub> <sup>c</sup>	31.41±0.74 <sub>a</sub> <sup>b</sup>	90.89±1.74 <sub>d</sub> <sup>d</sup>	93.12±2.00 <sub>d</sub> <sup>d</sup>	6.64±0.14 <sub>a</sub> <sup>b</sup>	11.76±0.20 <sub>a</sub> <sup>b</sup>	
Pomegranate peel	198.90±3.92 <sub>d</sub> <sup>d</sup>	193.20±3.15 <sub>d</sub> <sup>d</sup>	24.43±0.49 <sub>a</sub> <sup>d</sup>	60.91±1.31 <sub>b</sub> <sup>c</sup>	96.20±1.19 <sub>d</sub> <sup>d</sup>	83.91±0.17 <sub>d</sub> <sup>d</sup>	14.69±0.29 <sub>a</sub> <sup>d</sup>	39.61±0.78 <sub>b</sub> <sup>d</sup>	
Citrus peel	158.79±1.61 <sub>d</sub> <sup>d</sup>	163.26±3.00 <sub>d</sub> <sup>d</sup>	18.61±0.39 <sub>a</sub> <sup>c</sup>	43.71±0.94 <sub>b</sub> <sup>b</sup>	25.33±0.48 <sub>d</sub> <sup>b</sup>	28.67±0.69 <sub>d</sub> <sup>b</sup>	9.32±0.19 °	12.49±0.21 <sub>a</sub> <sup>b</sup>	
Citrus leaves	192.41±2.99 <sub>d</sub> d	169.16±2.92 <sub>d</sub> <sup>d</sup>	21.62±0.41 <sub>a</sub> <sup>d</sup>	59.23±1.21 <sub>a</sub> <sup>c</sup>	31.34±0.65 <sub>d</sub> <sup>b</sup>	21.02±0.51 <sub>b</sub> <sup>a</sup>	10.27±0.22 <sub>a</sub> <sup>c</sup>	14.36±0.29 <sub>a</sub> <sup>b</sup>	
Onion leaves	12.85±0.29c <sup>a</sup>	16.12±0.32 <sub>d</sub> <sup>a</sup>	$7.26\pm0.17_{a}^{b}$	12.56±0.29 <sub>c</sub> <sup>a</sup>	5.61±0.11 <sub>c</sub> <sup>a</sup>	8.12±0.17 <sub>d</sub> <sup>a</sup>	2.96±0.04 <sub>a</sub> <sup>a</sup>	4.67 ±0.08 <sub>b</sub> <sup>a</sup>	
Garlic leaves	6.97±0.13 <sub>d</sub> <sup>a</sup>	4.81±0.08 <sub>b</sub> <sup>a</sup>	3.52±0.06 <sub>a</sub> <sup>a</sup>	6.10±0.14 <sub>d</sub> <sup>a</sup>	4.10±0.08 d	3.85±0.06 <sub>d</sub> <sup>a</sup>	2.13±0.03 <sub>a</sub> <sup>a</sup>	3.16±0.05 <sub>b</sub> <sup>a</sup>	
Mango kernel	63.89±1.32 <sub>d</sub> <sup>b</sup>	69.24±1.24 <sub>d</sub> <sup>b</sup>	9.61±0.17a <sup>b</sup>	19.52±0.39 <sub>a</sub> <sup>a</sup>	45.56±0.90 <sub>d</sub> <sup>b</sup>	32.62±0.72 <sub>c</sub> <sup>b</sup>	3.96±0.07 <sub>a</sub> <sup>b</sup>	6.76±0.13 <sub>a</sub> <sup>a</sup>	

Values are mean  $\pm$  SD of all samples analyzed individually in triplicate (P < 0.05).

Results showed that methanolic and ethanolic extract showed good level of TPCs and TFCs followed by acetone however, chloroform extract showed least amount. This is in agreement with the findings of Sidhuraju and Becker<sup>24</sup> who reported that methanol and ethanol are efficient solvent to extract antioxidant compounds. According to data TPCs of different agro waste extracts varied significantly (P < 0.05) (6.97-198.90 GAE mg/g DW) in methanol, (4.81-193.20 GAE mg/g DW) ethanol, (6.10 to 60.91 GAE mg/g DW) acetone and in chloroform (3.52- 24.43 GAE mg/g DW). However amounts of TPCs assessed for methanolic and ethanolic extracts were found to be significantly higher (P < 0.05) than for acetone and chloroform extracts.

Among the various investigated agro wastes, extracts of fruits peels were found to contain relatively higher contents of TP and TF. Our findings correlate with the report of Li *et al.*<sup>25</sup> that a higher amount of antioxidants like phenolics, anthocynins and flavonol present in peels. In present study a remarkable high amount of TPs for methanolic pomegranate peel extract (198.90 GAE mg/g DW) was found to be higher than previously reported for aqueous pomegranate peel extract (161.25 GAE mg/g)<sup>26</sup> while citrus peel (163.26 GAE mg/g DW) and mango peels (116.80 GAE mg/g DW) were found comparable to those reported earlier for citrus sinesus var. Navel peel (160.3 GAE mg/g)<sup>27</sup> and for mango peels (54.67-109.70 GAE mg/g)<sup>3</sup>. Where as among the leaves highest level of TPC was observed in citrus leaves (192.41 GAE mg/g DW) while lowest in garlic leaves (6.97 GAE mg/g).

The amount of total flavonoids among different waste extracts varied significantly (P < 0.05) ranging 4.10-96.20 CE mg/g DW in methanol, 3.85-83.91 CE mg/g DW in ethanol, 3.16 to 39.61 CE mg/g DW in acetone and 2.13-14.69 CE mg/g DW in chloroform (Table-2). The remarkable high amounts of total flavonoids in pomegranate peel (96.20 mg/g CE) and

mango peel extracts (93.12 mg/g CE) were found to be significantly higher (P < 0.05) than those reported earlier by Kanatt *et al.*<sup>26</sup> for pomegranate peel extract (7.57 mg/g CE) and Dabrosca *et al.*<sup>28</sup> for limonella apple peel (47.80 mg/100 g). The type of plant part, nature of soil, growing stages, maturity at the stage of harvest and application of different treatments at post-harvest level may affect the presence and amount of phenolics and flavonoids in plant. Moreover, the difference in gene expression in different parts of a plant would cause to affect the process of synthesis and accumulation of different components at different ratio or different ratio of same component which actually reflects antioxidant potential and other biological activities of plant<sup>29</sup>.

**DPPH radical scavenging assay:** The DPPH radical has been extensively applied method to assess the antioxidant potential of food items, such as vegetables, olive oils, fruits, juices and wines etc. The antioxidant efficacy of a plant material can be accessed from its potential to donate hydrogen to DPPH free radical which in turn changes to stable diamagnetic molecule<sup>30</sup>. Its high sensitivity for active ingredients enable it to accommodate a large number of samples in a very short time even at low concentration, this has increased its use for measuring radical scavenging activity of different plant extracts<sup>31</sup>.

The DPPH radical scavenging activities of various investigated agro waste extracts in different solvents are shown in Table-3. The potential of different wastes extracts to scavenge free radical varied 31.92-75.65 % in methanol, 34.45-71.24 % in ethanol, 29.76-67.31 % in acetone and 25.23-50.51 % in chloroform. All the investigated plants showed appreciable range to scavenge free radical however, highest efficacy was exhibited by pomegranate peel (50.51-75.65 %) followed by citrus leaves (45.11-64.38 %), citrus peel (44.72-57.23 %), onion leaves (42.64-57.37 %) garlic leaves (39.22-54.83 %) mango peel (41.42-46.56 %) and mango kernel (25.23-31.92 %).

TABLE-3
DPPH RADICAL SCAVENGING ACTIVITY (%) AND INHIBITION OF
LINOLEIC ACID OXIDATION (%) OF VARIOUS AGRO WASTES

Plant used	DP	PH radical scave	enging activity	(%)	Inhibition of linoleic acid oxidation (%)			1 (%)
Piant used	Methanol	Ethanol	Chloroform	Acetone	Methanol	Ethanol	Chloroform	Acetone
Mango peel	46.56±1.12 <sub>c</sub> <sup>b</sup>	52.52±1.02 <sub>d</sub> <sup>c</sup>	41.42±0.93 <sub>a</sub> <sup>c</sup>	43.21±0.93 <sub>a</sub> <sup>b</sup>	62.23±1.22 <sub>c</sub> <sup>b</sup>	66.66±1.38 <sub>d</sub> <sup>c</sup>	58.92±1.12 <sub>a</sub> <sup>c</sup>	61.76±1.28 <sub>b</sub> <sup>c</sup>
Pomegranate peel	$75.65\pm1.52_{d}^{d}$	$71.24\pm1.43_{d}^{d}$	50.51±0.12 <sub>a</sub> <sup>d</sup>	67.31±1.21 <sub>c</sub> <sup>d</sup>	82.34±1.73 <sub>d</sub> <sup>d</sup>	$79.87\pm1.42_{d}^{d}$	66.21±1.15 <sub>a</sub> <sup>d</sup>	75.98±1.39 <sub>c</sub> <sup>d</sup>
Citrus peel	57.23±1.25 <sub>d</sub> <sup>c</sup>	60.61±1.29 <sub>d</sub> <sup>c</sup>	44.72±0.92 <sub>a</sub> <sup>d</sup>	54.46±1.03 <sub>c</sub> <sup>c</sup>	73.81±1.37 <sub>d</sub> <sup>d</sup>	70.71±1.38 <sub>c</sub> <sup>d</sup>	61.42±1.20 <sub>a</sub> <sup>d</sup>	66.54±1.30 <sub>b</sub> <sup>c</sup>
Citrus leaves	64.38±1.48 <sub>d</sub> <sup>c</sup>	62.62±1.21 <sub>d</sub> <sup>d</sup>	45.11±0.93 <sub>a</sub> <sup>d</sup>	59.97±1.14 <sub>d</sub> <sup>d</sup>	76.21±1.47 <sub>d</sub> <sup>d</sup>	73.16±1.51 <sub>c</sub> <sup>d</sup>	$64.13\pm0.71_{a}^{d}$	69.41±1.25 <sub>b</sub> <sup>d</sup>
Onion leaves	57.37±1.23 <sub>d</sub> <sup>c</sup>	61.13±2.10 <sub>d</sub> c	42.64±0.88 a	53.34±1.12 <sub>c</sub> <sup>c</sup>	55.13±1.09 <sub>d</sub> <sup>b</sup>	50.54±1.06 <sub>c</sub> <sup>a</sup>	43.23±1.05 <sub>a</sub> <sup>a</sup>	47.45±0.94 <sub>b</sub> <sup>a</sup>
Garlic leaves	54.83±1.54 <sub>d</sub> <sup>c</sup>	50.54±1.90°	39.22±0.83 <sub>a</sub> <sup>c</sup>	44.47±0.94 <sub>b</sub> <sup>b</sup>	67.04±1.32 <sub>d</sub> <sup>c</sup>	65.78±1.31 <sub>d</sub> <sup>c</sup>	52.01±1.21 <sub>a</sub> <sup>b</sup>	55.32±1.03 <sub>b</sub> <sup>b</sup>
Mango kernel	31.92±0.16 <sub>c</sub> <sup>a</sup>	34.45±0.89 <sub>d</sub> <sup>a</sup>	25.23±1.13 <sub>a</sub> <sup>a</sup>	29.76±0.71 <sub>b</sub> <sup>a</sup>	47.56±0.80 <sub>b</sub> <sup>a</sup>	55.43±1.39 <sub>d</sub> <sup>b</sup>	43.49±0.91 <sub>a</sub> <sup>a</sup>	51.79±1.09 <sub>c</sub> <sup>b</sup>

Values are mean  $\pm$  SD of all samples analyzed individually in triplicate (P < 0.05).

Highest antioxidant potential of pomegranate peels in present study (75.65 %) was found in agreement with apple peel (78 %) $^{28}$  but was lower than that reported for pomegranate peel $^{20}$ .

Inhibition (%) of linoleic acid peroxidation of different agro wastes: Selected agro wastes extracts were also assessed for their antioxidant potential by their potential to inhibit peroxidation in linoleic acid system using thyocyanate method<sup>16</sup>. Table-3 revealed appreciable range of inhibition of peroxidation from 43.23 to 82.34 %. In general methanolic extracts exhibited a higher level of inhibition of peroxidation than the other solvent extracts. A significant difference was also observed among degree of % inhibition presented by different agro wastes. Methanolic extracts of pomegranate peels (82.34 %), citrus leaves (76.21 %) and citrus peels (73.81 %), were found to be more effective to inhibit peroxidation than the garlic leaves (67.04 %), mango peel (62.23 %), onion leaves (55.13 %) and mango kernel extracts (47.56 %). This variation in degree of percent inhibition among the different waste extracts can be attributed to the variation in the amount of total phenolics. According to many reports a good correlation was established between total phenols, total flavonoid contents and percent inhibition of peroxidation which represents antioxidant activity<sup>16,20</sup>.

High degree of percent inhibition of linoleic acid in present study was exhibited by pomegranate peels (82.34%) and citrus peels (73.81%) found to be slightly lower than earlier reported for pomegranate peels and citrus peels (96.4 and 86.9%)<sup>20</sup>. Variation in degree of percent inhibition may be due to the differences in varieties, solvent used and the applied extraction method<sup>20</sup>. According to Robbins<sup>22</sup>, the inhibition of peroxidation of linoleic acid would be due presence of antioxidant compounds such as xanthones, flavans, flavonols, diantrquinones and some other phenolic compounds in extracts.

Reducing power of agro waste extracts: Table-4 shows the trends of reducing potential of different agro wastes extracts. The reducing power (absorbance at  $\lambda = 700$  nm) of tested agro waste extracts varied from 0.342-1.74 among different solvents at concentration of 10 mg/mL and followed the order of effectiveness as: pomegranate peels (1.74) (methanolic extract) > citrus leaves (1.20) (methanolic extract) > mango peel (0.994) (ethanolic extract) > citrus peel (0.994) (methanolic extract) > mango kernal (0.897) (ethanolic extract) > garlic leaves (0.701) (methanolic extract) > onion leaves (0.623) (methanolic extract). In our present study, the organic solvent extracts of waste materials with higher level of TPC exhibited good reducing potential. Among the different antioxidant components of a plant reducing potential is mostly associated

with the concentration of total phenolics. Many reports reveal that the extracts with greater amount of total phenolicmay exhibit greater reducing potential<sup>20,24</sup>.

In our present work, reducing power of agro wastes were found in close agreement with those noted for four verities of mango  $(0.42-1.27)^{32}$  and residues of six fruits  $(0.31-1.54)^{33}$  but were found lower than the leaves of *Moringa oleifera* (2.45) and *Aloe barbadensis* leaves  $(2.01)^{34}$ .

Antifungal activity: As methanolic and ethanolic extracts of agro waste showed good antioxidant activity so they were evaluated for antifungal potential against two strains Aspergillus *i.e.*, Aspergillus flavus and Aspergillus parasiticus (Figs. 1 and 2). In general inhibitory zones of ethanolic extracts against the A. flavus (14-37 mm) and A. parasiticus (14-38 mm) were found to be comparable to those measured by methanolic extracts (12-32 mm and 11-35 mm), respectively. Statistical analysis showed that there was no significant difference between antifungal potential exhibited by ethanolic and methanolic extracts against the tested fungi A. parasiticus and A. flavus. However, a significant (P < 0.05) variation among antifungal activities of different extracts was detected.

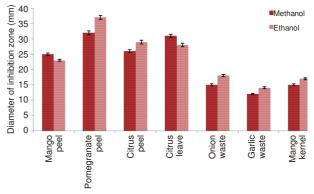


Fig. 1. Antifungal activity of some agro waste against Aspergillus flavus

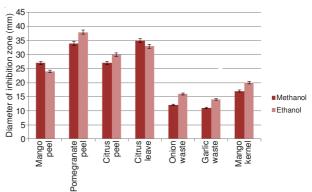


Fig. 2. Antifungal activity of some agro waste against Aspergillus parasiticus

REDU	CING POWER (ABSORBANG	TABLE-4 CE AT $\lambda = 700 \text{ nm}$ ) OF VAR	IOUS AGRO WASTES EXT	TRACTS			
Plant used	Plant used Methanol Ethanol Chloroform Acetone						
Mango peel	0.852±0.02 <sub>c</sub> <sup>b</sup>	0.994±0.03 <sub>d</sub> <sup>b</sup>	0.599±0.01 <sub>a</sub> <sup>b</sup>	0.732±0.02 <sub>b</sub> <sup>c</sup>			
Pomegranate peel	$1.740\pm0.04_{\rm d}^{\rm d}$	$1.590\pm0.03_{c}^{d}$	$1.040\pm0.02_{\rm a}^{\rm c}$	$1.290\pm0.03_{b}^{d}$			
Citrus peel	$0.993 \pm 0.03_{d}^{b}$	0.754±0.02 <sub>c</sub> <sup>a</sup>	0.521±0.01 <sub>a</sub> <sup>b</sup>	$0.621\pm0.02_{b}^{b}$			
Citrus leaves	$1.200\pm0.02_{\rm d}^{\rm c}$	1.080±0.01 <sub>c</sub> <sup>b</sup>	0.723±0.02 <sub>a</sub> <sup>b</sup>	$0.943\pm0.03_{b}^{c}$			
Onion leaves	$0.623\pm0.01_{\rm d}^{\rm a}$	0.521±0.02 <sub>c</sub> <sup>a</sup>	0.342±0.01 <sub>a</sub>	0.421±0.01 <sub>b</sub> <sup>a</sup>			
Garlic leaves	$0.701\pm0.02_{\rm d}^{\ a}$	$0.687 \pm 0.01_{\rm d}^{\rm a}$	0.343±0.01 <sub>a</sub>	0.519±0.01 <sub>c</sub> <sup>a</sup>			
Mango kernel	0.699±0.02 <sub>b</sub> <sup>a</sup>	0.897±0.05 <sub>d</sub> <sup>b</sup>	0.543±0.01 <sub>a</sub> <sup>b</sup>	0.601±0.03 <sub>b</sub> <sup>b</sup>			
Values are mean ± SD of a	all samples analyzed individua	Illy in triplicate $(P < 0.05)$ .					

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The data showed that highest inhibitory effect was exhibited by ethanolic extracts of pomegranate peels with maximum DIZ value of 37 mm against *A. flavus* while least was observed by garlic leaves (14 mm). Where as, citrus peels, citrus leaves and mango peels also showed appreciable effects with DIZ values of 29, 28 and 23 mm, respectively, while onion leaves, mango kernel and garlic leaves extracts exhibited a moderate effect with DIZs values of 18, 17 and 143 mm, respectively. On the other hand for *A. parasiticus* pomegranate peels, citrus leaves and citrus peels showed excellent inhibitory effect with DIZs values 38, 33 and 30 mm, respectively, mango peel and mango kernel exhibited moderate inhibitory effect of 24 and 20 mm and least inhibitory effect was shown by onion and garlic leaves.

In case of methanolic extracts the highest inhibitory effect was presented by pomegranate peels (32 mm) while garlic leaves showed lowest (12 mm) inhibitory potential against *A. flavus*. Citrus leaves, citrus peels and mango peels also presented appreciable inhibitory potential with DIZs values 31, 26 and 25 mm. Where as, for *A. parasiticus* highest inhibitory zone was measured by citrus leaves (35 mm) followed by pomegranate peels (34 mm), citrus peel (27 mm), mango peel (27 mm), mango kernel (17 mm), onion leaves (12 mm) and garlic leaves (11 mm).

Antifungal potential of agro wastes extracts was also assessed by determing their MICs. Various agro waste extracts showed a range of MICs values from 135 to 1340 mg/mL against *A. flavus* and 99 to 1407 mg/mL against *A. parasiticus* (Table-5). It was found that citrus leaves (methanolic extract) were most effective with lowest MIC value of 99 mg/mL against *A. parasiticus* and garlic leaves extract were least effective with highest MIC value of 1407 mg/mL. On the other hand pomegranate peels (ethanolic extract) were found most effective for *A. flavus* than other extracts with lowest MIC value of 135 mg/mL.

The remarkable inhibitory effect of citrus leaves and pomegranate peels (MIC value 99 and 107 mg/mL) as observed

against *A. parasiticus* was found to be noticeably lower than that noticed for gold mohar leaves and flowers (45 and 65 mg/mL) against *A. nigar*<sup>29</sup>. Although the antifungal potential of citrus leaves, mango peel and citrus peels observed in present analysis for *A. flavus* and *A. parasiticus* was found to be relatively greater than those measured for onion bulb (17.3 mm), scallion (12.6 mm) against *A. niger* and singara rind (15 mm) against *A. flavus* but were comparable to pomegranate rind (23 mm) and bakeri garlic (22 mm)<sup>35</sup>.

According to our findings highest inhibitory potential of pomegranate peels extract against *A. parasiticus* and *A. flavus* was found to be comparable with the earlier report of Orak *et al.*<sup>36</sup> who also noticed extraordinary inhibitory effect for different genotypes of pomegranate peels against *A. parasiticus*. According to Reddy *et al.*<sup>37</sup> report antifungal activity of pomegranate may be attributed to the presence of ellagic acid and punicalagins in pomegranate juice byproducts. However, a comparison among efficiencies exhibited by different waste extracts against two fungal species showed that *A. parasiticus* was resistant greater than *A. flavus*. This might be due to the effects that antifungal compounds or antifungal modes are not identical for different *Aspergillus* species.

**Phenolic acids:** In present analysis remarkable high antioxidant and antifungal activity of investigated leaves and peels against the tested fungi might be due to the existence of flavonoids and phenolic acids like gallic acid, cholorogenic acid, caffeic acid, 3-hydroxybenzoic acid and protocatechuic acid<sup>38</sup>. So the extracts which exhibited remarkable antifungal and antioxidant activities were further evaluated for their phenolic acid profile by GC/MS technique.

The analysis of agro waste revealed that total of 7 phenolic acids like gallic acid, vanillic acid, cinamic acid, protocacheuic acid, caffeic acid, ferulic acid and p-coumaric acid were identified on the basis of availability of standard chemicals and spectral data (Table-6). A significant variation (P < 0.05) was detected in concentration of phenolic acids among different wastes. Gallic acid was detected in almost all waste

TABLE-5
MINIMUM INHIBITORY CONCENTRATION OF METHANOL AND ETHANOL
EXTRACTS AGAINST Aspergillus flavus AND Aspergillus parasiticus

Plant used	Aspergill	us flavus	Aspergillus parasiticus		
riant useu	Methanol	Ethanol	Methanol	Ethanol	
Mango peel	415±9.98° a	480±8.11 <sub>d</sub> <sup>b</sup>	325±7.23 <sub>a</sub> <sup>a</sup>	452±9.01 <sub>c</sub> <sup>b</sup>	
Pomegranate peel	185±3.74 <sub>d</sub> a	135±3.75 <sub>b</sub> <sup>a</sup>	149±4.78 <sub>c</sub> <sup>a</sup>	107±2.76 <sub>a</sub> <sup>a</sup>	
Citrus peel	387±7.32 <sub>d</sub> <sup>a</sup>	342±7.21 <sub>c</sub> <sup>a</sup>	310±7.56° a	216±5.43 <sub>a</sub> <sup>a</sup>	
Citrus leaves	290±5.91 <sub>b</sub> <sup>a</sup>	$721\pm14.70_{d}^{b}$	99±19.80 <sub>a</sub> <sup>a</sup>	160±4.78 <sub>a</sub> <sup>a</sup>	
Onion leaves	953±19.70 <sub>b</sub> °	759±16.60 <sub>a</sub> <sup>c</sup>	1295±25.4 <sub>d</sub> <sup>d</sup>	981±19.80 <sub>b</sub> <sup>d</sup>	
Garlic leaves	1340±27.50 <sub>c</sub> <sup>d</sup>	$1190\pm24.40_{a}^{d}$	1407±29.70 <sub>d</sub> <sup>d</sup>	1235±25.60 <sub>a</sub> <sup>d</sup>	
Mango kernel	953±18.90 <sub>d</sub> <sup>c</sup>	868±17.60 <sub>d</sub> °	835±17.4 <sub>d</sub> <sup>c</sup>	465±9.67 <sub>a</sub> <sup>b</sup>	

Values are mean  $\pm$  SD of all samples analyzed individually in triplicate (P < 0.05); Minimum inhibitory concentration MIC (mg/mL).

TABLE-6
CONTENTS OF PHENOLIC ACIDS (mg/g OF DRY MATTER) OF SOME AGRO WASTES QUANTIFIED BY GC-MS

	Gallic acid	Cinamic acid	Vanalic acid	Protocacheuic acid	Ferulic acid	Caffeic acid	P-Coumaric acid
Mango peel	1.89±0.40	ND	ND	0.26±0.04	ND	0.17±0.02	0.04±0.01
Citrus peel	2.25±0.75	$0.11 \pm 0.01$	ND	$0.14\pm0.03$	ND	1.17±0.20	0.01±0.00
Pomegranate peel	6.11±0.92	0.21±0.05	$0.26\pm0.04$	ND	0.12±0.02	ND	$0.43\pm0.07$
Citrus leaves	0.96±0.03	ND	0.81±0.02	0.03±0.007	ND	0.17±0.02	$0.08\pm0.00$

Values are mean  $\pm$  SD of three separate experiments

extracts in significant (P < 0.05) higher amount ranging from 0.96 to 6.11 mg/g of dry matter followed by p-coumaric acid (0.01 to 0.43 mg/g dry matter) and caffeic acid (0.07 to 1.17 mg/g of dry matter) whereas cinamic, protocatechuic and vanillic acids were found in few of extracts in small amount ranging from 0.11-0.21, 0.03-0.26 and 0.26-0.81 mg/g of dry matter, respectively and ferulic acid was detected at a small level 0.012 mg/g of dry matter only in pomegranate peels.

Among the peels of different fruits as investigated in present analysis, pomegranate peels were found to be richest in gallic acid and p-coumaric acid at concentration of 6.11 and 0.43 mg/g following vanillic acid (0.26 mg/g), cinamic acid (0.21 mg/g) and ferulic acids (0.12 mg/g). Elango et al.<sup>39</sup> reported the presence of gallic and ellagic acid in pomegranate peel determined by GC-MS but they did not report the presence of other acids in their study that were detected in present work i.e., p-coumaric acids, vanillic, cinamic and ferulic acids. Citrus peels were also found to contain highest amount of gallic acid (2.25 mg/g) and caffeic acid (1.17 mg/g), followed by protocatechuic acid (0.14 mg/g) and small amount of p-coumaric acids (0.01 mg/g). Citrus leaves also contained all those acids that were detected in peels except cinamic acid but their concentration in leaves was found quite lower than peels. Mango peels were also found to contain appreciable amounts of gallic acid, protocacheuic acid, caffeic acid, p-coumaric and at levels of 1.89, 0.26 0.17 and 0.04 mg/g, respectively.

However, there are numerous differences among the phenolic acids profile found in the present work and those reported earlier. These differences may be accredited to the differences in plant habitat, methodology of sample preparation and analysis. So it is necessary to use various analytical techniques to determine plant metabolites<sup>40</sup>. It was found that wastes showed high contents of phenolic acids also displayed potent antioxidant as well as antifungal activities. A good correlation exists between phenolic compounds and antifungal activity which is in agreement with the findings of Martin *et al.*<sup>1</sup>.

#### Conclusion

In conclusion, the ethanolic and methanolic extracts of different agricultural wastes have demonstrated a broad spectrum of antioxidant and antifungal activities against tested fungi. Gas chromatography mass spectrometry analysis showed the presence of various phenolic acids with variable concentration. On the basis of the results, it would be possible to use waste extracts to formulate new products as food preservatives to be used in food industry replacing synthetic antioxidants and antifungal agents.

## **ACKNOWLEDGEMENTS**

The authors are grateful to the Higher Education Commission for providing financial support under the project "Protection of cereal grains against aflatoxin contamination using plant materials".

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