Design, Synthesis, *in silico* Analysis and Pharmacological Evaluation of Coumarin Derived Mercaptotriazole Derivatives as MCL-1 Inhibitors

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Design and synthesis of new series of 4-amino-5-mercapto-4*H*-1,2,4-triazoles connected to coumarin ring and assessed for cytotoxity and the study of docking studies was done against MDA-MB-231 cell lines. Using substituted diols, a series of new coumarin linked 4-amino-5-mercapto-4*H*-1,2,4-triazoles derivatives were synthesized from substituted diols. Moreover, the elucidation of the newly synthesized mercaptotriazole derivatives was achieved by spectral data. An MTT assay was employed to evaluate the antineoplastic activity against triple-negative breast cancer (TNBC) and the synthesized compounds demonstrated promising, potent, and innovative inhibitory effects, highlighting their potential for the development of new anticancer therapies. Compounds **8h**, **8d**, **8e**, **8b**, **8i**, **8j** and **8a** exhibit high binding energies with target receptor MCL-1 enzyme of the MDA-MB231 cells, where remaining compounds exhibit good binding interactions with target protein. *In silico* Docking studies verified the uncompetitive inhibition of MCL-1 enzyme.

Keywords: Coumarins, Mercaptotriazole, Chromen-2-one derivatives, MCL-1 enzyme, Antineoplastic activity, In silico studies.

INTRODUCTION

Breast cancer is one of the leading causes of cancerrelated morbidity and mortality among women worldwide. Neoplasia, characterized by the uncontrolled proliferation of cells, underlies the development of breast cancer and arises primarily from genetic mutations in normal breast epithelial cells [1]. If cancerous cells are undetected early, they may spread to other organs and invade the nearby tissues, increasing the risk of mortality and complicating the treatment procedure. By raising the chance of a successful therapy course and breast-conserving surgery, early identification of the breast cancer significantly improves the long-term outcomes [2]. Despite their widespread usage, traditional screening methods and the clinical breast examinations have their limitations and the decreased sensitivity. The risk of breast cancer was identified by risk assessment techniques and patients who are at high risk might benefit from taking the drugs that lowers risk [3]. Menopausal state influences the choice for medication. Genetic and the inherited predisposition were among the risk factors related to breast cancer. Locoregional therapy is the cornerstone of breast cancer treatment and is considered curative when the disease is diagnosed at an early stage. If necessary, systemic therapy might be given either prior the surgery [4].

Mercaptotriazole derivatives are well-known chemical scaffolds exhibiting diverse biological activities, including antimicrobial, antibacterial and antifungal effects [5-8]. Beyond these properties, they serve as versatile intermediates in the synthesis of various heterocyclic drugs with significant physiological activity [9,10]. Leveraging these features, novel compounds are designed by integrating triazole and coumarin pharmacophores into a single molecular framework. Such hybridization is anticipated to enhance pharmacological efficacy, potentially offering new avenues for anticancer drug development. In this study, we explored the binding potential of the designed

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triazole-coumarin hybrids against MCL-1 using molecular docking with the crystal structure (PDB ID: 3WIX) [11], aiming to identify promising candidates for triple-negative breast cancer (TNBC) therapy.

EXPERIMENTAL

Analytical grade of 98% pure chemicals and solvents were brought from Sigma-Aldrich and used without further purification. Thin-layer chromatography (TLC) plates were prepared using 0.5 mm thick aluminum sheets (Merck) precoated with silica gel 60 F₂₅₄ sheets. To record an infrared (IR) spectra, KBr pellet method was used using an FTIR spectrometer (model: Thermo-Scientific Nicolet iS5/iS10). ¹H NMR spectra was obtained using a VARIAN NMR spectrometer operating at 500 MHz. The melting points of the synthesized compounds were determined using a Remi electronic melting point apparatus and are uncorrected.

General procedure

Synthesis of 4-amino-5-mercapto-4*H*-1,2,4-triazole-2*H*-chromen-2-one derivatives (8a-j): The titled mercapto-triazole derivatives were synthesized *via* a modified literature procedure [12]. Substituted diols were reacted with ethyl aceto-acetate at 70-80 °C for 30 min using Eaton's reagent as

catalyst to form coumarin derivatives. These intermediates were treated with ethyl bromoacetate in DMF with K₂CO₃ at 25 °C for 8 h, followed by sequential reactions with hydrazine hydrate, CS₂ in the presence of KOH, and a final hydrazine hydrate reflux to yield the target compounds (**Scheme-I**). The products were purified by recrystallization from alcohol.

5-((4-Amino-5-mercapto-4*H***-1,2,4-triazole-3-yl)methoxy-8-ethyl-4-methyl-2***H***-chromen-2-one (8a**): White solid, yield: 88%, m.p.: 256-268 °C; ¹H NMR (500 MHz, CHCl₃- d_6 , δ ppm): 1.08 (t, 3H), 1.36 (s, 2H), 1.58 (2H), 2.45-2.55 (s, 3H), 5.33 (s, 2H), 6.22 (s, 1H), 7.10 (s, 1H), 7.38 (s, 1H), 13.73 (s, H); 13 C NMR (125 MHz, CHCl₃- d_6 , δ ppm): 14.1, 22.3, 23.8, 31.0, 66.9, 112.3, 113.1, 124.8, 128.7, 150.5, 151.8, 155.9, 160.8, 166.9, 188.75; HRMS: m/z for C₁₇H₂₀N₄O₃S ([M+H]⁺): 360.13; found 360.43.

5-((4-Amino-5-mercapto-4*H***-1,2,4-triazole-3-yl)methoxy-8-hexyl-4-methyl-2***H***-chromen-2-one (8b):** White solid, yield: 73%, m.p.: 280-295 °C; ¹H NMR (500 MHz, CHCl₃- d_6 , δ ppm): 1.08 (t, 3H), 1.36 (s, 2H), 1.58 (2H), 2.45-2.55 (s, 3H), 5.33 (s, 2H), 6.22 (s, 1H), 7.10 (s, 1H), 7.38 (s, 1H), 13.74 (s, H); ¹³C NMR (125 MHz, CHCl₃- d_6 , δ ppm): 14.1, 22.7, 23.8, 31.3, 31.3, 31.9,66.9, 112.3, 112.7, 113.5, 124.8, 128.9, 150.8, 151.9, 152.3, 160.9, 166.7; HRMS: m/z for C₁₉H₂₄N₄O₃S ([M+H]⁺): 388.16, found 388.26.

Scheme-I: Synthesis of 5-((4-amino-5-mercapto-4H-1,2,4-triazole-3-yl)methoxy-8-ethyl-4-methyl-2H-chromen-2-one derivatives

Synthesis of 8-amino-5-((4-amino-5-mercapto-4*H*-1,2,4-triazole-3-yl)methoxy)-4-methyl-2*H*-chromen-2-one (8c): White solid, yield: 75%, m.p.: 268-288 °C, 1 H NMR (500 MHz, CHCl₃- d_6 , δ ppm): 2.45-2.55 (t, 3H), 5.23 (s, 2H), 5.32 (s, 2H), 5.670 (s, 2H), 6.28 (s, 1H), 6.85 (s, 1H), 13.93 (s, 1H). 13 C NMR (125 MHz, CHCl₃- d_6 , δ ppm): 23.3, 66.9, 112.3, 113.1, 114. 8, 116.7, 135.5, 138.8, 138.9, 148.8, 151.9, 155.5, 160.7, 166.7; HRMS: m/z for $C_{13}H_{13}N_5O_3S$ ([M+H] $^+$): 319.07, found 319.34.

7-Amino-5-((4-amino-5-mercapto-4*H***-1,2,4-triazole-3-yl)methoxy)-4-methyl-2***H***-chromen-2-one (8d): White solid, yield: 62%, m.p.: 220-240 °C; ¹H NMR (500 MHz, CHCl₃-d_6, δ ppm): 2.45 (t, 3 H), 5.23 (s, 2H), 5.32 (s, 2H), 5.60 (s, 2H), 6.18 (s, 1H), 6.20 (1 H), 6.28 (1H), 13.79 (s, 1H); ¹³C NMR (125 MHz, CHCl₃-d_6, δ ppm): 23.8, 66.9, 92.8, 96.1, 109.8, 112.7, 150.1, 151.8, 155.5, 158.8, 160.8, 166.9; HRMS: m/z for C₁₃H₁₃N₅O₃S ([M+H]⁺): 319.07, found 319.07.**

5-((4-Amino-5-mercapto-4*H***-1,2,4-triazole-3-yl)methoxy)-4,7-dimethyl-2***H***-chromen-2-one (3e):** White solid, yield: 65%, m.p.: 260-270 °C; 1 H NMR (500 MHz, CHCl₃- d_6 , δ ppm): 1 H NMR (500 MHz, CHCl₃- d_6 , δ ppm): 2.45 (t, 3H), 2.45 (s, 1H) 5.33 (s, 2H), 5.66 (2H), 6.22 (s, 1 H), 6.76 (s, 1H), 13.79 (s, 1H); 13 C NMR (125 MHz, CHCl₃- d_6 , δ ppm): 21.3, 23.8, 31.0, 66.9, 109.8, 112.5, 113.8, 116.7, 140.5, 151.2, 151.7, 155.9, 157.8, 160.8, 166.9; HRMS: m/z for C₁₄H₁₄N₄O₃S ([M+H] $^+$): 318.08, found 318.08.

5-((4-Amino-5-mercapto-4*H***-1,2,4-triazole-3-yl)methoxy)-7-hydroxy-4-methyl-2***H***-chromen-2-one (8f**): White solid, yield: 76%, m.p.: 242-255 °C; ¹H NMR (500 MHz, CHCl₃- d_6 , δ ppm): 2.46 (t, 3H), 5.33 (s, 2H), 5.62 (s, 2H), 6.207 (s, 2H), 6.35 (s, 1H), 6.47 (1 H),10.29 (1H), 13.23 (s, 1H); ¹³C NMR (125 MHz, CHCl₃- d_6 , δ ppm): 23.8, 66.9, 92.3, 98.1, 112.8, 151.5, 154.8, 155.7,155.9, 159.4, 160.8, 166.9; HRMS: m/z for C₁₃H₁₂N₄O₄S ([M+H]⁺): 320.06, found 320.16.

5-((4-Amino-5-mercapto-4*H***-1,2,4-triazole-3-yl)methoxy)-7-bromo-4-methyl-2***H***-chromen-2-one (8g**): White solid, yield: 85%, m.p.: 236-245 °C; ¹H NMR (500 MHz, CHCl₃- d_6 , δ ppm): 2.45 (t, 3H), 5.33 (s, 2H), 5.65 (s, 2H), 7.30 (s, 1H), 7.43 (s, 1H), 13.73 (s, 1H); 13 C NMR (125 MHz, CHCl₃- d_6 , δ ppm): 23.8, 66.9, 112.3, 112.5, 114.3, 118.1, 151.5, 153.8, 155.9, 160.2, 166.6,166.8; HRMS: m/z for C₁₃H₁₁BrN₄O₃S ([M+H]⁺): 381.97, found 383.22.

5-((4-Amino-5-mercapto-4*H***-1,2,4-triazole-3-yl)methoxy)-8-bromo-4-methyl-2***H***-chromen-2-one (8h): White solid, yield: 75%, m.p.: 254-265 °C; ¹H NMR (500 MHz, CHCl₃-d_6, δ ppm): 2.45 (t, 3H), 5.33 (s, 2H), 5.67 (s, 2H), 6.20 (s, 1H), 6.89 (s, 1H), 7.62 (1H), 13.73 (s, 1H); ^{13}C NMR (125 MHz, CHCl₃-d_6, δ ppm): 23.8,109.9, 111.3, 112.5, 114.8, 130.6, 147.5, 151.8, 155.9, 157.0, 160.8, 166.9; HRMS: m/z for C₁₃H₁₁BrN₄O₃S ([M+H]⁺): 383.97, found 383.22.**

5-((4-Amino-5-mercapto-4*H***-1,2,4-triazole-3-yl)methoxy)-8-chloro-4-methyl-2***H***-chromen-2-one (8i): White solid, yield: 72%, m.p.: 291-294 °C; ¹H NMR (500 MHz, CHCl₃-d_6, δ ppm): 2.46 (t, 3H), 5.33 (s, 2H), 5.67 (s, 2H), 6.27 (s, 1H), 6.98 (s, 1H), 7.48(1H), 13.73 (s, 1H); ^{13}C NMR (125 MHz, CHCl₃-d_6, δ ppm): 23.8, 66.9, 112.3, 112.5, 113.8, 122.2, 129.7, 146.5, 151.8, 155.9,156.2, 160.8, 166.9; HRMS: m/z for C₁₃H₁₁ClN₄O₃S ([M+H]⁺): 338.02, found 338.77.**

5-((4-Amino-5-mercapto-4*H***-1,2,4-triazole-3-yl)methoxy)-7-chloro-4-methyl-2***H***-chromen-2-one (8j):** White solid, yield: 76%, m.p.: 291-294 °C; ¹H NMR (500 MHz, CHCl₃- d_6 , δ ppm): 2.43 (t, 3H), 5.33 (s, 2H), 5.67 (s, 2H), 6.20 (s, 1H), 6.85 (s, 1H), 6.89 (1H), 13.73 (s, 1H). ¹³C NMR (125 MHz, CHCl₃- d_6 , δ ppm): 23.8, 66.9, 111.9, 112.1, 113.8, 116.6, 117.7, 134.5, 151.8, 152.9, 155.8, 159.4, 160.9; HRMS: m/z for C₁₃H₁₁Cl₂N₄O₃S ([M+H]⁺): 338.02, found 338.77.

Molecular docking: Docking studies were performed to identify potent breast cancer drugs by observing the molecular binding affinities of designed structures with MCL-1 receptor protein. The selection of the target receptor for docking is based on several factors such as resolution below 2.0 Å, the structure should be elucidated by X-ray diffraction techniques, consists of co-crystallized ligand and it does not have any breaks in the particular 3D structure of the protein.

MCL-1 target enzyme is the recently exploring target for identification of breast cancer agents comprising a bischalcone nucleus. The crystalline structure of the MCL-1 target receptor protein was obtained from protein databank which is having the resolution of 1.9 Å with PDB ID: 3WIX [11]. AutoDock 4.2.6 software was used to know the interactions of the designed 3D-structured mercaptotriazoles with the active site of 3WIX [13]. ChemSketch, version 2022.1.2, ACD/ Labs software used to design the structures and then converted into respective 3D design. By applying molecular mechanics they underwent energy minimizations required for the molecular docking and for preparing of corresponding pdb files. Using default parameters Grid-based docking studies was carried by and docking studies were performed by considering 7-(4-carboxyphenyl)-3-[3-(naphthalen-1-yloxy)propyl]pyrazolo[1,5-a]pyridine-2-carboxylic acid as reference ligand at MCL-1 active site. Discovery Studio visualizers v21.1 is used to check the 2D and 3D binding interactions of ligands with receptor protein [14].

Anticancer activity: MDA-MB-231 human breast cancer epithelial cells were cultured in Dulbecco's Modified Eagle Medium (DMEM) or RPMI-1640, supplemented with 10% FBS and 1% penicillin–streptomycin and maintained at 37 °C in a humidified atmosphere containing 5% CO₂. Cells were seeded into 96-well plates at a density of 5,000-10,000 cells per well in 100 μ L of complete medium and allowed to adhere for 24 h. The cells were then treated with varying concentrations of the test compounds and incubated for 24, 48 and 72 h. Etoposide was used as a positive control, while the vehicle served as a negative control.

RESULTS AND DISCUSSION

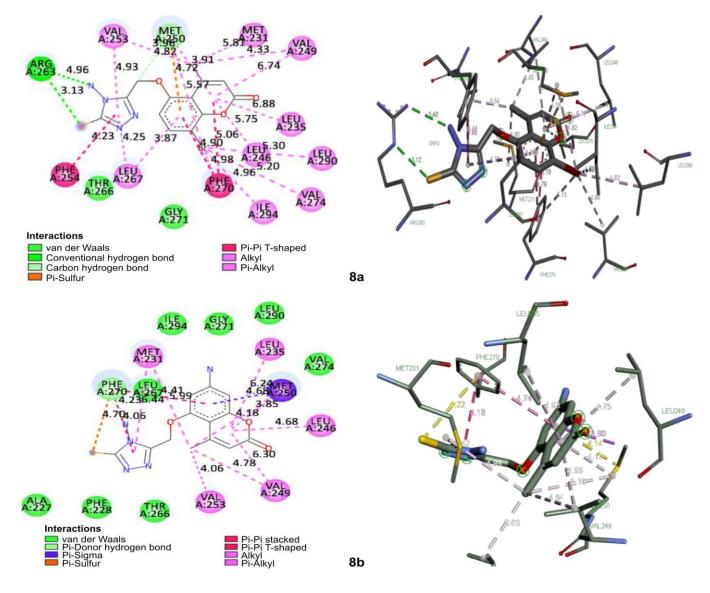
Novel coumarin based mercaptotriazole derivatives (**8a-j**) were synthesized using various aromatic aldehydes following the outlined procedures and then characterized by the spectroscopic methods were evaluated for the anticancer activity against the cell lines MDA-MB-231. Molecular docking studies at the MCL1 target protein active site region gives data of the binding energies (kcal/moL), hydrogen bond length, number of the hydrogen bonds and the interacted amino acid residues, shown in Table-1.

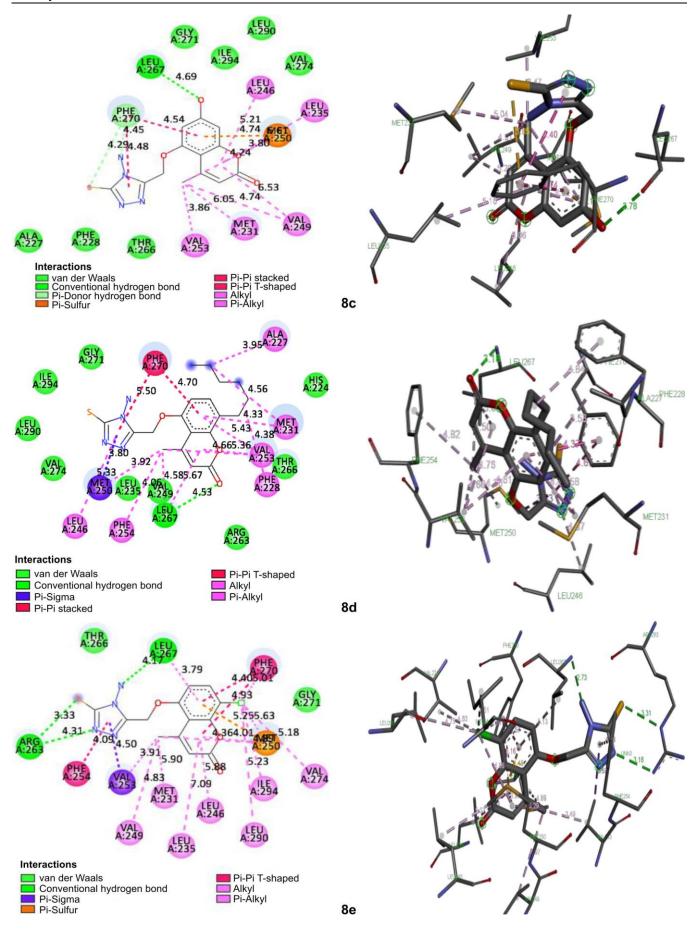
TABLE-1 BINDING ENERGIES, H-BOND INTERACTIONS, AMINOACID RESIDUES, HYDROGEN BOND LENGTHS PROFILES OF SYNTHESIZED COMPOUNDS 8a-j					
Compound	Binding energies (kcal/mol)	H-bond interactions (Å)	Amino acid residues	Hydrogen bond length (Å)	
8a	-9.82	1	LEU267	4.32	
8b	-9.88	1	LEU267	4.53	
8c	-9.62	2	ARG263, LEU267	2.94, 4.90	
8d	-9.92	0	No	_	
8e	-9.92	2	ARG263, LEU267	4.84, 3.88	
8f	-9.77	1	LEU267	4.69	
8g	-9.60	1	LEU267	4.05	
8h	-10.21	2	ARG263, LEU267	3.13, 4.96	
8i	-9.88	3	ARG263, LEU267	3.33, 4.31, 4.17	
8j	-9.85	1	ARG263	4.07	
Reference	-12.25	1	ARG263	2.70	

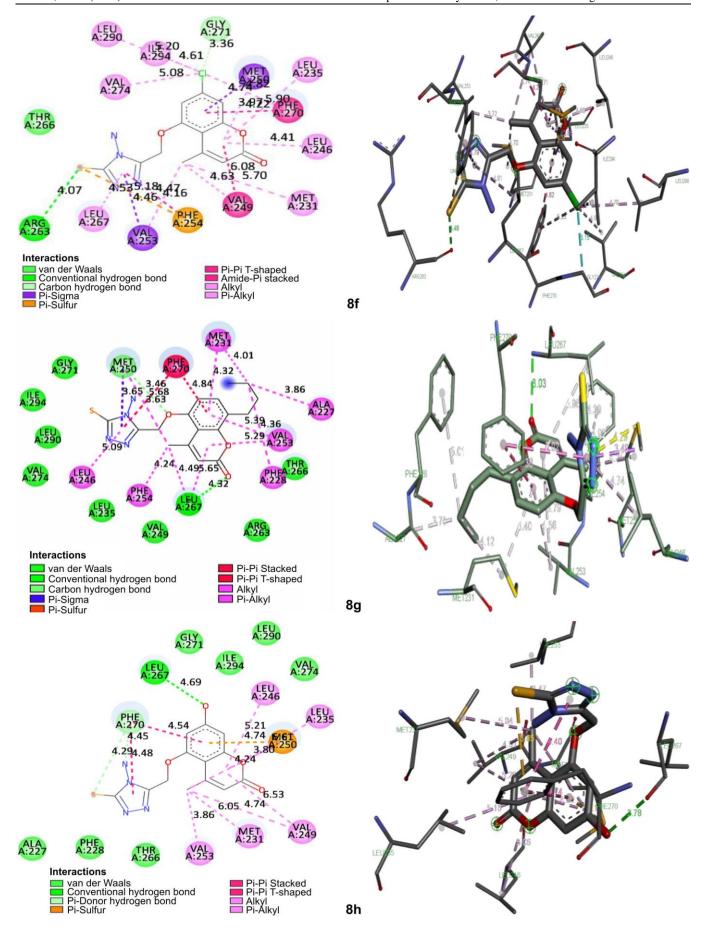
In comparison to the reference ligand, 7-(4-carboxyphenyl)-3-[3-(naphthalen-1-yloxy)propyl]pyrazolo[1,5-a]pyridine-2-carboxylic acid, which exhibited a binding energy of -12.27 kcal/mol, the synthesized compounds demonstrated strong and comparable binding affinities toward the MCL-1 receptor protein (PDB ID: 3WIX). Among the series, compound 8h

showed the highest binding affinity with a docking score of -10.21 kcal/mol, indicating the most favourable interaction with the active site of MCL-1.

Fig. 1 illustrates the 2D and 3D structures of the MCL-1 receptor and the binding modes of the reference ligand and compounds **8a-j** within the active site. Compound **8h** emer-







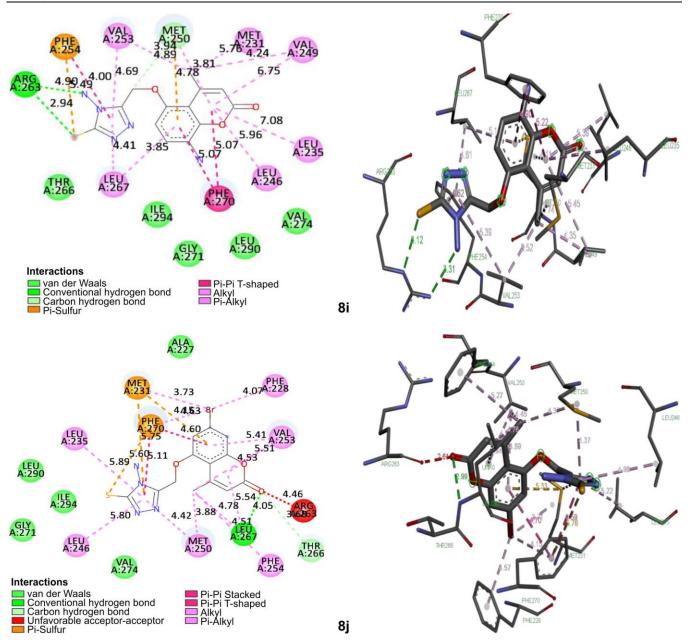


Fig. 1. 2D and 3D interactions of compounds 8a-j at target protein 3WIX

ged as the best-performing derivative, which can be attributed to the presence of a *para*-substituted halogen group, enhancing hydrophobic and electrostatic interactions with key active site residues. This was followed by compounds **8d** and **8e**, both exhibiting docking scores of -9.92 kcal/mol, suggesting that both electron-withdrawing (-Cl) and electron-donating (-NH₂) substituents significantly contribute to improve the binding affinity.

Compounds **8b** and **8i** also showed favourable docking scores of -9.88 kcal/mol, indicating that *para*-substitution on the coumarin phenyl ring, particularly with electron-donating or moderately lipophilic groups, promotes efficient binding. Similarly, compounds **8j** and **8a** displayed good binding affinities with scores of -19.85 and -9.82 kcal/mol, respectively, suggesting that bulky alkyl and halogen substituents at the *para* position are well tolerated within the MCL-1 binding pocket.

Compounds **8f** and **8c** exhibited moderate binding affinities (-9.77 and -9.62 kcal/mol, respectively), indicating that *meta*-hydroxyl or *para*-amino substitutions provide reasonable stabilization through hydrogen bonding interactions. In contrast, compound **8g** showed the lowest docking score (-9.60 kcal/mol), likely due to steric hindrance associated with *meta*-bromine substitution, which may limit optimal interaction with the active-site residues.

Overall, the docking analysis clearly demonstrates that *para*-substituted coumarin derivatives exhibit superior binding affinity toward the MCL-1 receptor, highlighting the critical role of substitution pattern and electronic effects in modulating protein–ligand interactions. These findings support the potential of the synthesized compounds as promising MCL-1 inhibitors and provide valuable insights for further structural optimization.

Anticancer activity: The anticancer effectiveness of hybrids 8a-j on MDA-MB-231 (breast cancer) cell lines was assessed using an MTT assay. The MDA-MB-231 cell line, a well-established triple-negative breast cancer (TNBC) model, was employed to evaluate the antiproliferative potential of the synthesized compounds using the MTT assay. This colorimetric assay measures cellular metabolic activity based on the reduction of the yellow tetrazolium salt 3-(4,5-dimethyl-thiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) to insoluble purple formazan crystals by viable cells.

Following treatment, MTT solution (5 mg/mL in PBS) was added to each well and incubated to allow metabolically active cells to reduce the dye to formazan. The culture medium was subsequently removed and the formed crystals were dissolved using DMSO solvent. Absorbance was measured using a microplate reader at 570 nm, with a reference wavelength of 630 nm. Cell viability was directly proportional to the measured absorbance. All experiments were performed in triplicate (n = 3) under sterile conditions to ensure reproducibility and accuracy.

Dose-response curves were generated by plotting percentage cell viability against compound concentration, enabling the determination of IC₅₀ values, defined as the concentration required to inhibit 50% of cell growth. Among the evaluated derivatives, halogen-substituted compounds exhibited enhanced antiproliferative activity. Notably, the *para*-bromo coumarin derivative **8h** showed the highest potency with an IC₅₀ value of 3.30 μ M, whereas the *meta*-bromo analogue **8g** displayed significantly lower activity (IC₅₀ = 15.15 μ M), indicating the critical influence of substitution position.

Compounds **8b**, **8i** and **8j** demonstrated strong anticancer activity with IC₅₀ values of 4.36, 4.76 and 4.97 μ M, respectively, while compound **8a** exhibited moderate potency (IC₅₀ = 5.35 μ M). Moreover, compounds **8d** and **8e** showed notable antitumor activity with IC₅₀ values of 6.33 and 6.39 μ M, respectively (Table-2). The enhanced activity of these derivatives may be attributed to their ability to form favourable hydrogenbonding interactions with intracellular targets, thereby improving binding affinity and cytotoxic efficacy.

TABLE-2
INHIBITORY CONCENTRATIONS (IC50) DATA OF
THE SYNTHESIZED COMPOUNDS 8a-j

Compound	IC ₅₀ (μM)	Compound	IC ₅₀ (μM)
8a	5.35	8g	15.15
8b	4.36	8h	3.30
8c	14.83	8i	4.76
8d	6.33	8j	4.97
8e	6.39	Reference	1.01
8f	12.89		

Structure–activity relationship (SAR) analysis revealed that *para*-substitution on the coumarin moiety, particularly with bromo (-Br), amino/alkyl and chloro or hydroxyl containing groups, consistently resulted in improved anticancer activity. In contrast, compounds **8f** (IC₅₀ = 12.89 μ M) and **8c** (IC₅₀ = 14.83 μ M) showed comparatively lower cytotoxicity, although they retained measurable antiproliferative effects and demonstrated binding affinity toward target proteins. Overall, the

results highlight the importance of substituent type and position in modulating anticancer potency against TNBC cells and support the potential of these derivatives as promising anticancer leads.

Conclusion

A series of coumarin based mercaptotriazoles (8a-j) were synthesized by reacting different diol derivatives with ethylacetoacetate in the presence of Eatons reagent as solvent to form coumarin intermediates followed by series of reactions on cyclization yield final compounds. The resulting products were recrystallized with ethanol. The synthesized compounds were thoroughly characterized using ¹H NMR, ¹³C NMR, FTIR and mass spectrometry, confirming their proposed molecular structures. All evaluated derivatives exhibited notable anticancer activity against the tested cell lines. Structureactivity relationship (SAR) analysis revealed that the position of substituents and their electron-donating or electron-withdrawing nature play a critical role in modulating anticancer potency. Among the series, compound 8h demonstrated the most pronounced antineoplastic activity against the MDA-MB-231 breast cancer cell line. In addition, molecular docking studies were carried out against the MCL-1 receptor protein, providing insights into the binding interactions and supporting the experimental anticancer findings. The combined boilogical and computational results suggest that these compounds possess promising therapeutic potential. However, further investigations into their molecular mechanisms, structure optimization and in vivo evaluation are necessary to fully elucidate their mode of action and to advance these molecules as potent lead candidates for the treatment of breast cancer.

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CONFLICT OF INTEREST

The authors declare that there is no conflict of interests regarding the publication of this article.

DECLARATION OF AI-ASSISTED TECHNOLOGIES

During the preparation of this manuscript, the authors used an AI-assisted tool(s) to improve the language. The authors reviewed and edited the content and take full responsibility for the published work.

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