



## Synthesis and Evaluation of Some Novel Benzimidazole Derivatives Bearing Thiazolidinone Moiety as Potential Antimicrobial Activity

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A new series 3-(3-hydroxy-4-(1*H*-benzo[d]imidazol-2-yl)phenyl)-2-substituted phenylthiazolidin-4-one (**4a-h**) were synthesized by the cycloaddition of 2-(benzo[d]imidazol-2-yl)-5-(4-substituted benzylideneamino)phenol (**3a-h**) with thioglycolic acid in the presence of anhydrous zinc chloride in 1,4 dioxane and the compounds (**4a-h**) further coupled with diazonium salt of sulphonamide to give diazo substituted of thiazolidinone. The structures of these newly synthesized were characterized by elementary analysis, IR, <sup>1</sup>H NMR and the synthesized compounds were evaluated *in vitro* for their preliminary antibacterial activity against various bacterial viz: *E. coli*, *S. aureus*, *S. typhi* and *Pseudomonas species* and of their antifungal activities versus *Aspergillus niger* and *Aspergillus oryzae*. Antibacterial and antifungal activities of each compound were compared with standards amoxicillin and fluconazole, respectively. The results revealed that some compounds were exhibited moderate to good antibacterial activity against both Gram (+) and Gram (-) bacteria whereas same compounds were exhibited moderate to weak antifungal action against both fungal strains.

**Key Words:** 4-Aminosalicylic acid, Thiazolidinone, Benzene-1,2-diamine, Diazotization, Antibacterial activity.

### INTRODUCTION

In the past few decades, benzimidazole and its derivative have received much attention due to their chemotherapeutic values. The benzimidazole ring is an important pharmacophore in modern drug discovery. Benz[b]imidazole chemically defined as benzene fused with imidazole heterocyclic ring. Regarding structural activity relationship of benzimidazole, many literature reports have revealed that the substitution at 1, 2, and 5 position of benzimidazole is very important for enhancing antimicrobial activities. 2-(Aryl substituted) benzimidazole were exhibited wide range of pharmacological activity including antibacterial<sup>1</sup>, antiviral<sup>2</sup>, antitumor<sup>3</sup> and antiinflammatory<sup>4</sup>. 4-Thiazolidinone is derivatives of saturated ring of thiazole with carbonyl group at the fourth position. The chemistry and pharmacological action of 4-thiazolidinone have great interest to medicinal chemistry because of its derivative possess various biological activities such as antibacterial<sup>5</sup>, antifungal<sup>6</sup>, anticonvulsant<sup>7</sup>, antitubercular<sup>8</sup>, anticancer<sup>9</sup>, anti-nflamatory<sup>10</sup>, antihistamine<sup>11</sup>, antioxidant<sup>12</sup>, antihypertension<sup>13</sup>, analgesic<sup>14</sup> and hypoglycemic activity<sup>15</sup>. As per literature survey the azo dyes have been widely used in dyeing textile fibers, biomedical studies, advanced applications in organic synthesis and shows variety of biological activities including antibacterial<sup>16,17</sup> and pesticide activities<sup>18</sup>. Synthesized azo dyes prepared by diazotization

of a primary aromatic amine, followed by coupling with one or more nucleophiles. Amino, hydroxyl and active methylene groups are commonly used coupling components<sup>19</sup>. The azo dye sulphonamide of antibacterial pro-drug such as protonosil was the first effective chemotherapeutic agents that could be used systemically for the cure of bacterial infection in humans.

In the new drug design, the development of hybrid molecules through the combination of different pharmacophores in one frame may lead to compounds with interesting pharmacological properties. The co-administration of the chemical entities with insertion of sulphonamide group, acting by different mechanisms may have a synergistic effect and resulting in a higher activity than each of the components. The above mentioned both heterocyclic rings individually have been vast biological properties, which on prompted the synthesis of some privileged hybrid molecules by connecting the main structural unit of the benz[b]imidazole ring system with the 4-thiazolidinone bearing sulphonamide group and screened their antimicrobial activity.

### EXPERIMENTAL

The chemicals used in the present studies were of synthetic grade, Merck company Ltd. The products were characterized by IR, <sup>1</sup>H NMR & mass spectroscopy. The melting points were

determined by open capillary method and uncorrected. The  $^1\text{H}$  NMR spectra were measured in  $\text{CDCl}_3$  solutions on a Bruker 400 MHz spectrometer. The Mass spectra were recorded on Shimadzu LS-MS 2010 spectrometer. The purity of prepared compounds was checked by TLC using silica gel. Benzimidazole and thiazolidinone prepared by known literature method<sup>20</sup>.

**General synthesis of 5-amino-2-(benzo[d]imidazol-2-yl)phenol (2):** In round bottom flask, a mixture of 4-amino-2-hydroxybenzoic acid (0.01 mol), benzene-1,2-diamine (0.01 mol) and polyphosphoric acid (PPA) (15 mL) were heated in an oil-bath at 200 °C for 3 h. The reaction mixture was cooled and poured on to the 100 mL crushed ice water, neutralize with aqueous ammonia. The separated solid was filtered, washed with water and dried to obtain desired product. The crude white precipitate was purified by dissolving 10 % HCl and reprecipitated by addition of aqueous ammonia. The resulting solid was filtered and recrystallized from ethyl acetate. IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3448 (O-H), 1530 (C=C), 1621 (C=N), 1278 (C-N), 894 (Ar-H).

**Synthesis of 2-(benzo[d]imidazol-2-yl)-5-(4-substituted benzylideneamino)phenol (3a-h):** The corresponding aromatic aldehyde (0.01 mol) was added to stirred solution of 5-amino-2-(1H-benzo[d]imidazol-2-yl)phenol (**1**) (0.01 mol) of absolute ethanol and the mixture was refluxed for 2 h. After cooling, the reaction mixture was filtered and kept for 24 h and finally the crystal of desired Schiff base product of 2-(1-methyl-benzo[d]imidazol-2-yl)-5-(4-substituted benzylidene-amino)phenol (**2a-h**) recrystallized from ethanol.

**2-(1H-Benzo[d]imidazol-2-yl)-5-(benzylideneamino)-phenol (3a):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3030, 765 (Ar-H), (3350 (N-H), 3450 (Ar-OH), 1200 (C-O), 1565 (N=CH), 1360 (C-N);  $^1\text{H}$  NMR ( $\delta$ ) 8.56 (s, 1H, N=CH), 9.61 (s, 1H, Ar-OH), 6.89-7.86 (m, 12H, Ar-H).

**2-(1H-Benzo[d]imidazol-2-yl)-5-(4-chlorobenzylidene-amino)phenol (3b):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3035, 810 (Ar-H), 3350 (N-H), 3450 (Ar-OH), 1200 (C-O), 1565 (N=CH), 1403 (C-N), 783 (Ar-Cl);  $^1\text{H}$  NMR ( $\delta$ ) 8.56 (s, 1H, N=CH), 9.61 (s, 1H, Ar-OH), 6.89-7.77 (m, 11H, Ar-H).

**2-(1H-Benzo[d]imidazol-2-yl)-5-(4-methoxybenzylideneamino)phenol (3c):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3050, 815 (Ar-H), 3350 (N-H), 3455 (Ar-OH), 1200 (C-O), 1560 (N=CH), 1375 (C-N), 1250 (Ar-OCH<sub>3</sub>);  $^1\text{H}$  NMR ( $\delta$ ) 8.56 (s, 1H, N=CH), 9.61 (s, 1H, Ar-OH), 6.89-7.86 (m, 11H, Ar-H) 3.83 (s, 3H Ar-OCH<sub>3</sub>).

**2-(1H-Benzo[d]imidazol-2-yl)-5-(4-nitrobenzylidene-amino)phenol (3d):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3050, 815 (Ar-H), 3350 (N-H), 3350 (Ar-OH), 1200 (C-O), 1560 (N=CH), 1380 (C-N), 1520, 1345 (Ar-NO<sub>2</sub>);  $^1\text{H}$  NMR ( $\delta$ ) 8.56 (s, 1H, N=CH), 9.61 (s, 1H, Ar-OH), 6.89-7.86 (m, 11H, Ar-H).

**2-(1H-Benzo[d]imidazol-2-yl)-5-(2-methylbenzylidene-amino)phenol (3e):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 2985 (Ar-CH<sub>3</sub>), 3050, 735 (Ar-H), 3450 (N-H), 3350 (Ar-OH), 1200 (C-O), 1560 (N=CH);  $^1\text{H}$  NMR ( $\delta$ ) 8.56 (s, 1H, N=CH), 9.61 (s, 1H, Ar-OH), 6.89-7.86 (m, 11H, Ar-H) 2.83 (s, 3H, CH<sub>3</sub>).

**5-[(4-(1H-Benzo[d]imidazol-2-yl)-3-hydroxyphenyl-imino)methyl]-2-methoxyphenol (3f):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3050, 855 (Ar-H), 3350 (N-H), 3460 (Ar-OH), 1200 (C-O),

1235 (Ar-OCH<sub>3</sub>) 1560 (N=CH);  $^1\text{H}$  NMR ( $\delta$ ) 8.56 (s, 1H, N=CH), 9.61 (s, 1H, Ar-OH), 6.89-7.86 (m, 11H, Ar-H) 3.83 (s, 3H, OCH<sub>3</sub>).

**Synthesis of 2-aryl substituted thiazolidin-4-ones (4a-h):** A mixture of Schiff base (0.001 mol) and thioglycolic acid (0.001 mol) dissolved in 1, 4 dioxane (20 mL), a pinch of anhydrous zinc chloride (0.5 mg) was added and refluxed for 8 h. The reaction was then cooled and the resulting solid was washed with sodium bicarbonate solution and final compound 3-[4-(1H-benzo[d]imidazol-2-yl)-3-hydroxyphenyl]-2-aryl substituted thiazolidin-4-one (**4a-f**) recrystallized from absolute ethanol.

**3-[4-(1H-Benzo[d]imidazol-2-yl)-3-hydroxyphenyl]-2-phenylthiazolidin-4-one (4a):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3030 (Ar-H), 3350 (N-H), 3460 (Ar-OH), 1700 (C=O), 1405 (C-N), 1610 (C=N), 1120 (C-S);  $^1\text{H}$  NMR ( $\delta$ ) 6.86-7.58 (m, 12H, Ar-H), 9.61 (s, 1H, Ar-OH), 12.56 (s, 1H, -NH-benzimidazole), 6.44 (s, 1H, CH-thiazolidinone), 3.9 (s, 2H, CH<sub>2</sub>-thiazolidinone);  $m/z$ : 387.10 (100.0 %), 388.11 (24.1 %).

**3-[4-(1H-Benzo[d]imidazol-2-yl)-3-hydroxyphenyl]-2-(4-chlorophenyl)thiazolidin-4-one (4b):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3021 (Ar-H), 3350 (N-H), 3450 (Ar-OH), 1403 (C-N), 1610 (C=N), 783 (Ar-Cl), 1695 (C=O), 1118 (C-S);  $^1\text{H}$  NMR ( $\delta$ ): 7.16-7.67 (m, 11H, Ar-H), 9.61 (s, 1H, Ar-OH), 12.56 (s, 1H, -NH-benzimidazole), 6.44 (s, 1H, CH-thiazolidinone), 4.0 (s, 2H, CH<sub>2</sub>-thiazolidinone);  $m/z$ : 421.07 (100.0 %), 423.06 (36.5 %).

**3-[4-(1H-Benzo[d]imidazol-2-yl)-3-hydroxyphenyl]-2-(4-methoxyphenyl)thiazolidin-4-one (4c):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3011 (Ar-H), 3390 (N-H), 1420 (C-N), 1602 (C=N), 1214 (Ar-OCH<sub>3</sub>), 1684 (C=O), 1109 (C-S);  $^1\text{H}$  NMR ( $\delta$ ): 6.86-7.84 (m, 11H, Ar-H), 9.61 (s, 1H, Ar-OH), 3.83 (s, 3H, Ar-OCH<sub>3</sub>), 12.56 (s, 1H, -NH-benzimidazole), 6.44 (s, 1H, CH-thiazolidinone), 3.9 (s, 2H, CH<sub>2</sub>-thiazolidinone);  $m/z$ : 417.11 (100 %), 418.12 (25.2 %).

**3-[4-(1H-Benzo[d]imidazol-2-yl)-3-hydroxyphenyl]-2-(4-nitrophenyl)thiazolidin-4-one (4d):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3077 (Ar-H), 3340 (N-H), 3450 (Ar-OH), 1417 (C-N), 1598 (C=N), 1515, 1343 (Ar-NO<sub>2</sub>), 1701 (C=O), 1105 (C-S);  $^1\text{H}$  NMR ( $\delta$ ): 6.86-7.58 (m, 11H, Ar-H), 9.61 (s, 1H, Ar-OH), 12.56 (s, 1H, -NH-benzimidazole), 6.44 (s, 1H, CH-thiazolidinone), 4.0 (s, 2H, CH<sub>2</sub>-thiazolidinone);  $m/z$ : 432.09 (100.0 %), 433.09 (26.2 %).

**3-[4-(1H-Benzo[d]imidazol-2-yl)-3-hydroxyphenyl]-2-o-tolylthiazolidin-4-one (4e):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3023 (Ar-H), 3359 (N-H), 3460 (Ar-OH), 1393 (C-N), 1604 (C=N), 2919, 2857 (Ar-CH<sub>3</sub>), 1700 (C=O), 1117 (C-S);  $^1\text{H}$  NMR ( $\delta$ ): 6.86-7.58 (m, 11H, Ar-H), 9.61 (s, 1H, Ar-OH), 12.56 (s, 1H, -NH-benzimidazole), 2.83 (s, 3H, Ar-CH<sub>3</sub>), 6.44 (s, 1H, CH-thiazolidinone), 4.0 (s, 2H, CH<sub>2</sub>-thiazolidinone).

**3-[4-(1H-Benzo[d]imidazol-2-yl)-3-hydroxyphenyl]-2-(3-hydroxy-4-methoxyphenyl)thiazolidin-4-one (4f):** IR (KBr,  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ ): 3033 (Ar-H), 3324 (N-H), 3455 (Ar-OH), 1394 (C-N), 1604 (C=N), 1208 (Ar-OCH<sub>3</sub>), 3356 (Ar-OH), 1719 (C=O), 1108 (C-S);  $^1\text{H}$  NMR ( $\delta$ ): 6.86-7.58 (m, 10H, Ar-H), 9.61 (s, 1H, Ar-OH), 3.83 (s, 3H, Ar-OCH<sub>3</sub>), 12.56 (s, 1H, -NH-benzimidazole), 6.44 (s, 1H, CH-thiazolidinone), 4.0 (s, 2H, CH<sub>2</sub>-thiazolidinone).

**3-[4-(1H-Benzo[d]imidazol-2-yl)-3-hydroxyphenyl]-2-[4-(dimethylamino)phenyl]thiazolidin-4-one (4g):** IR (KBr,  $\nu_{\max}$ ,  $\text{cm}^{-1}$ ); 3033 (Ar-H), 3324 (NH), 3460 (Ar-OH), 1719 (C=O), 1108 (C-S), 2790, 2730 (Ar-N ( $\text{CH}_3$ )<sub>2</sub>);  $^1\text{H}$  NMR ( $\delta$ ); 6.86-7.84 (m, 11H, Ar-H), 9.61 (s, 1H, Ar-OH), 3.03 (s, 6H, Ar-N ( $\text{CH}_3$ )<sub>2</sub>), 12.56 (s, 1H, -NH-benzimidazole), 6.44 (s, 1H, CH-thiazolidinone), 4.0 (s, 2H,  $\text{CH}_2$ -thiazolidinone).

**3-[4-(1H-Benzo[d]imidazol-2-yl)-3-hydroxyphenyl]-2-(furan-2-yl)thiazolidin-4-one (4h):** IR (KBr,  $\nu_{\max}$ ,  $\text{cm}^{-1}$ ); 3150 (furan-yl-H), 3324 (N-H), 3460 (Ar-OH), 1715 (C=O), 1110 (C-S);  $^1\text{H}$  NMR ( $\delta$ ); 6.89-7.67 (m, 7H, Ar-H), 9.61 (s, 1H, Ar-OH), 12.56 (s, 1H, -NH-benzimidazole), 6.26-7.67 (m, 3H, furan-yl-H), 6.5 (s, 1H, CH-thiazolidinone), 3.9 (s, 2H,  $\text{CH}_2$ -thiazolidinone).

**Synthesis of 4-[(3-(1H-benzo[d]imidazol-2-yl)-2-hydroxy-6-(4-oxo-2-substitutedphenylthiazolidin-3-yl)phenyl) diazenyl]benzene sulfonamide (5a-h):** A cold solution of 2.5 mL of sodium nitrite was added drop wise to ice cold solution of sulphanilamide in conc. HCl and water. The temperature of the reaction was maintained up to 0-5 °C during addition. When addition was completed, the solution was kept for 5 min. with occasional stirring to complete the diazotization. Then, it was poured into an ice cold solution of 2-aryl substituted thiazolidin-4-one in 10 % (20 mL) sodium hydroxide solution. The reaction mixture was allowed to stand in an ice bath for 10-15 min. The coloured products obtained were filtered, washed with water and finally dried. The entire product individually was recrystallized from 50 % ethanol.

**4-[(3-(1H-Benzo[d]imidazol-2-yl)-2-hydroxy-6-(4-oxo-2-phenylthiazolidin-3-yl)phenyl) diazenyl]benzene sulfonamide (5a):** IR (KBr,  $\nu_{\max}$ ,  $\text{cm}^{-1}$ ); 3030 (Ar-H), 3350 (N-H), 3460 (Ar-OH), 1700 (C=O), 1405 (C-N), 1610 (C=N), 1120 (C-S), 1556 (N=N), 1317, 1147 ( $\text{SO}_2$ );  $^1\text{H}$  NMR ( $\delta$ ) 6.86-7.58 (m, 11H, Ar-H), 9.82 (s, 1H, Ar-OH), 12.56 (s, 1H, -NH-benzimidazole), 6.44 (s, 1H, CH-thiazolidinone), 4.0 (s, 2H,  $\text{CH}_2$ -thiazolidinone), 7.25 (s, 2H,  $\text{NH}_2$  of sulphonamide), 8.1-8.25 (m, 4H, Aryl azo sulphonamide).

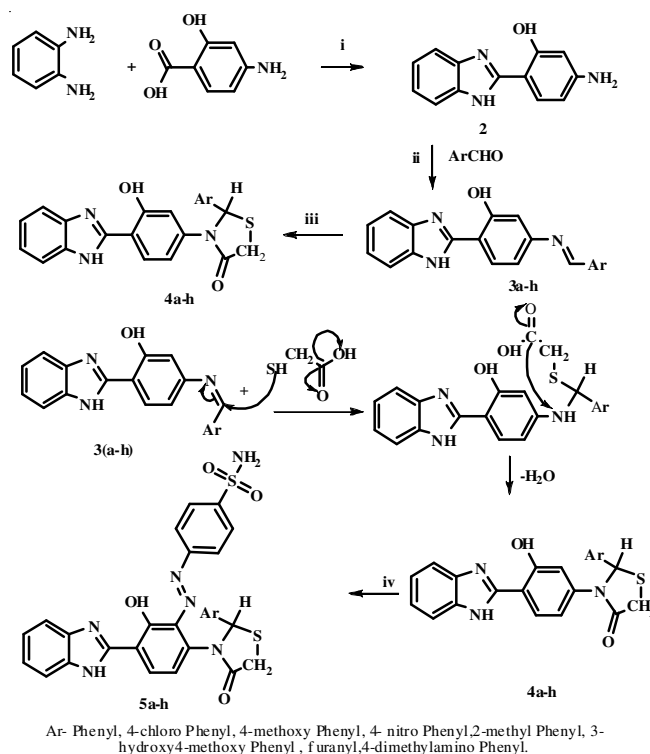
**4-[(3-(1H-Benzo[d]imidazol-2-yl)-2-hydroxy-6-(4-oxo-2-chlorophenylthiazolidin-3-yl) phenyl) diazenyl]benzene sulfonamide (5b):** IR (KBr,  $\nu_{\max}$ ,  $\text{cm}^{-1}$ ); 3030 (Ar-H), 3350 (N-H), 3460 (Ar-OH), 1700 (C=O), 1405 (C-N), 1610 (C=N), 1120 (C-S), 1556 (N=N), 1317, 1147 ( $\text{SO}_2$ );  $^1\text{H}$  NMR ( $\delta$ ) 6.86-7.58 (m, 10H, Ar-H), 9.82 (s, 1H, Ar-OH), 12.56 (s, 1H, -NH-benzimidazole), 6.44 (s, 1H, CH-thiazolidinone), 4.0 (s, 2H,  $\text{CH}_2$ -thiazolidinone), 7.25 (s, 2H,  $\text{NH}_2$  of sulphonamide), 8.1-8.25 (m, 4H, arylazosulphonamide),  $m/z$ : 604.08 (100.0 %).

**Antimicrobial activities:** All the synthesized compounds (3-5) were evaluated for their antibacterial activity against *Escherichia coli* (MTCC-723), *Staphylococcus aureus* (ATCC-29513), *Pseudomonas aeruginosa* (MTCC-1688) *Salmonella typhi* (recultred) and of their antifungal activity against *Aspergillus niger* (MTCC-281), *Aspergillus oryzae* (recultred) by the cup-plate method. Holes of 6 mm diameter were punched carefully using a sterile cork borer and these were filled with test solution different concentrations 1000, 750 and 500  $\mu\text{g/mL}$ . The plates were incubated at 37 °C for 24 h and 72 h in case of antibacterial and antifungal activity, respectively. The diameter of the zone of inhibition (mm) was measured for all the test compounds and results were compared with the

standard drugs amoxicillin and fluconazole for antibacterial and antifungal activity, respectively.

## RESULTS AND DISCUSSION

The starting material 2-substituted benzimidazole (2) was prepared according to a reported procedure through the reaction of benzene-1,2-diamine with appropriate *p*-amino salicylic acid<sup>22</sup>. The compound 2 further on condensation with various selected aromatic aldehydes furnished Schiff bases, 2'-(benzo[d]imidazol-2-yl)-5'-(4-substituted benzylideneamino) phenol (3a-h). The 5-membered heterocyclic ring such as thiazolidinone in compound 4 was incorporated by the cycloaddition of compounds (3a-h) with thioglycolic acid in the presence of anhydrous aluminum chloride as a catalyst to give 3-(3-hydroxy-4-(1H-benzo[d]imidazol-2-yl)phenyl)-2-substituted phenylthiazolidin-4-one (4a-h). The cycloaddition mechanism of reaction was mentioned in Scheme-I. The azo sulphonamide group in compound 5 was introduced by coupling with diazonium salt of sulphanilamide furnished products 4-((3-(1H-benzo[d]imidazol-2-yl)-2-hydroxy-6-(4-oxo-2-substituted phenylthiazolidin-3-yl)phenyl) diazenyl)benzene sulfonamide (5a-h). Finally the products were recrystallized with glacial acetic acid and water in (1:3). The purity was checked by the preparative TLC using silica Gel and a structure of the individual products were characterized by C, H, N analysis, IR and  $^1\text{H}$  NMR spectral data. The physical data and the percentage yield of the synthesized compounds are given in Table-1.



**Scheme-I:** Reagents: (i) polyphosphoric acid, 3h, 86 %; (ii) ethanol, 2h, 75 %; (iii) thioglycolic acid in anhydrous zinc chloride (iv) sodium nitrite, HCl/0-5 °C

The following peaks confirmed the formation of thiazolidinone. The peaks at 1720-1715, 1410-1398 and 1120-1110 in FTIR ( $\text{cm}^{-1}$ ) have shown the groups of C=O, C-N, C-S of thiazolidinone respectively. The  $^1\text{H}$  proton of imidazole is

TABLE-1  
 PHYSICAL DATA OF SYNTHESIZED COMPOUNDS **3(a-h)** AND **4(a-h)**

Compounds	Aryl	m.f. (m.w.)	Elemental analysis (%): Found (calculated)			Yield (%)
			C	H	N	
<b>3a</b>	Phenyl-	C <sub>20</sub> H <sub>15</sub> N <sub>3</sub> O (311)	76.66 (76.62)	4.82 (4.75)	13.52 (13.50)	75
<b>3b</b>	4-Chloro phenyl	C <sub>20</sub> H <sub>14</sub> N <sub>3</sub> OCl (347)	64.86 (64.85)	4.06 (4.00)	12.08 (12.00)	65
<b>3c</b>	4-Methoxy phenyl	C <sub>21</sub> H <sub>17</sub> N <sub>3</sub> O <sub>2</sub> (343)	73.45 (73.35)	4.99 (4.88)	12.24 (12.20)	80
<b>3d</b>	4-Nitro phenyl	C <sub>20</sub> H <sub>14</sub> N <sub>4</sub> O <sub>3</sub> (358)	67.03 (66.08)	3.94 (3.90)	15.63 (15.50)	75
<b>3e</b>	4-Methyl phenyl	C <sub>21</sub> H <sub>17</sub> N <sub>3</sub> O (327)	77.04 (76.82)	5.84 (5.55)	12.83 (9.90)	85
<b>3f</b>	3-Hydroxy 4-methoxy phenyl	C <sub>21</sub> H <sub>17</sub> N <sub>3</sub> O <sub>3</sub> (359)	70.18 (70.15)	4.77 (4.72)	11.69 (11.60)	45
<b>3g</b>	Furanyl-	C <sub>18</sub> H <sub>13</sub> N <sub>3</sub> O <sub>2</sub> (303)	71.28 (71.15)	2.91 (2.90)	13.50 (13.48)	85
<b>3h</b>	4-Dimethyl amino	C <sub>22</sub> H <sub>20</sub> N <sub>4</sub> O (356)	74.14 (74.10)	5.66 (5.60)	15.17 (15.15)	80
<b>4a</b>	Phenyl	C <sub>22</sub> H <sub>17</sub> N <sub>3</sub> O <sub>2</sub> S (387)	68.20 (68.15)	4.42 (4.38)	10.85 (10.75)	60
<b>4b</b>	4-Chloro phenyl	C <sub>22</sub> H <sub>16</sub> N <sub>3</sub> O <sub>2</sub> SCl (421)	62.63 (62.53)	3.86 (3.80)	9.96 (9.83)	85
<b>4c</b>	4-Methoxy phenyl	C <sub>23</sub> H <sub>19</sub> N <sub>3</sub> O <sub>3</sub> S (417)	66.17 (66.10)	4.59 (4.55)	10.08 (10.00)	75
<b>4d</b>	4-Nitro phenyl	C <sub>22</sub> H <sub>16</sub> N <sub>4</sub> O <sub>4</sub> S (432)	61.10 (61.05)	3.73 (3.63)	14.80 (14.75)	60
<b>4e</b>	4-Methyl phenyl	C <sub>23</sub> H <sub>19</sub> N <sub>3</sub> O <sub>2</sub> S (401)	68.81 (68.75)	4.77 (4.67)	10.47 (10.40)	55
<b>4f</b>	3-Hydroxy 4-methoxy phenyl	C <sub>23</sub> H <sub>19</sub> N <sub>3</sub> O <sub>4</sub> S (433)	63.73 (63.6)	4.42 (4.35)	9.69 (9.65)	60
<b>4g</b>	Furanyl-	C <sub>20</sub> H <sub>15</sub> N <sub>3</sub> O <sub>3</sub> S (373)	63.65 (63.55)	4.01 (3.90)	11.13 (11.05)	45
<b>4h</b>	4-Dimethyl amino phenyl	C <sub>24</sub> H <sub>22</sub> N <sub>4</sub> O <sub>2</sub> S (430)	66.96 (66.80)	5.15 (5.10)	13.01 (12.90)	50
<b>5a</b>	Phenyl	C <sub>28</sub> H <sub>22</sub> N <sub>6</sub> O <sub>4</sub> S <sub>2</sub> (570)	58.93 (58.85)	3.86 (3.75)	14.63 (14.55)	75
<b>5b</b>	4-Chloro phenyl	C <sub>28</sub> H <sub>21</sub> N <sub>6</sub> O <sub>4</sub> S <sub>2</sub> Cl (605)	55.58 (55.50)	3.50 (3.45)	13.89 (13.84)	-
<b>5c</b>	4-Methoxy phenyl	C <sub>29</sub> H <sub>24</sub> N <sub>6</sub> O <sub>5</sub> S <sub>2</sub> (600)	57.99 (57.85)	4.03 (3.85)	13.99 (13.85)	-
<b>5d</b>	4-Nitro phenyl	C <sub>28</sub> H <sub>21</sub> N <sub>7</sub> O <sub>6</sub> S <sub>2</sub> (615)	54.63 (54.55)	3.44 (3.33)	15.93 (15.86)	-

observed as a singlet at 12.56 ppm apparently due to deshielding caused by the benz[b]imidazole ring. The infrared spectrum of compounds **3a-h** indicated the absence of the NH<sub>2</sub> absorption band and contains characteristic absorption band at 1685-1630 cm<sup>-1</sup>, a typical of the stretching vibration of the carbon-nitrogen double bond and evidently the formation of Schiff

base, which further confirmed the presence of  $\delta$  8.56 ppm with respect to -N=CH, interpreted by NMR. The <sup>1</sup>H NMR spectrum of compounds **4a-h** revealed the presence of a singlet at  $\delta$  4.0 ppm with respect to CH<sub>2</sub> moiety of thiazolidinone. IR spectrum of compounds **5a-f** revealed the presence of characteristic absorption bands at 1317, 1149 cm<sup>-1</sup> and 3200, 3260

 TABLE-2  
 ANTIBACTERIAL PROPERTIES OF THE SYNTHESIZED COMPOUNDS

Compound	Concentration ( $\mu$ g/mL) (zone of inhibition in mm)											
	<i>Escherichia coli</i>			<i>Staphylococcus aureus</i>			<i>Pseudomonas aeruginosa</i>			<i>Salmonella typhi</i>		
	1000	750	500	1000	750	500	1000	750	500	1000	750	500
<b>3a</b>	9	8	7	9	8	6	8	8	7	5	3	2
<b>3b</b>	12	11	10	12	11	10	6	5	5	5	2	2
<b>3c</b>	10	10	9	11	11	10	9	9	8	NA	NA	NA
<b>3d</b>	12	12	11	14	13	11	10	9	8	4	3	2
<b>3e</b>	10	9	8	9	8	7	7	6	5	NA	NA	NA
<b>3f</b>	5	5	4	7	6	6	7	7	5	NA	NA	NA
<b>3g</b>	8	7	6	8	7	6	7	6	5	NA	NA	NA
<b>3h</b>	7	6	5	8	7	6	7	6	5	3	2	2
<b>4a</b>	10	9	8	10	9	7	7	7	6	4	3	2
<b>4b</b>	12	11	11	14	13	12	12	11	10	6	5	5
<b>4c</b>	9	8	7	10	8	7	7	6	5	5	3	2
<b>4d</b>	13	12	11	15	14	12	9	8	7	3	2	2
<b>4e</b>	7	6	5	6	6	5	7	6	5	2	2	2
<b>4f</b>	9	8	8	10	9	8	7	6	5	4	4	3
<b>4g</b>	5	5	4	6	5	4	7	6	5	5	4	4
<b>4h</b>	6	6	5	6	5	4	7	6	5	3	2	1
<b>5a</b>	9	8	6	9	8	7	7	6	5	2	2	1
<b>5b</b>	14	13	12	14	13	12	13	12	11	5	4	3
<b>5c</b>	10	10	9	10	9	8	8	7	6	3	3	2
<b>5d</b>	13	12	11	14	13	12	8	7	7	5	4	2
<b>5e</b>	6	5	5	7	6	5	7	6	5	3	2	1
<b>5f</b>	6	5	4	7	6	5	6	5	4	3	2	1
<b>5g</b>	5	4	3	6	5	4	7	5	4	NA	NA	NA
<b>5h</b>	6	5	4	7	6	5	6	5	4	NA	NA	NA
Std.	17	16	15	18	17	16	15	14	13	15	15	14

Std. Amoxicillin; NA; no activity.



cm<sup>-1</sup> for SO<sub>2</sub> and N-H *str.* of sulphonamide, respectively. Compounds **5a-h** showed N=N additional IR peak at 1560-1550 cm<sup>-1</sup>, which confirmed the formation of azosulphonamide. The phenolic -OH group of entire compounds were chemically detected by the treatment with FeCl<sub>3</sub> solution, which gives characteristic colour.

**Antimicrobial activities:** The synthesized derivatives were screened for their *in vitro* antibacterial and antifungal activities against *P. aeruginosa*, *S. aureus*, *S. typhi*, *E. coli*, *Aspergillus niger* and *Aspergillus oryzae* using ampicillin and fluconazole as a reference standard drugs. The results of both antibacterial and antifungal activity screening of the tested compounds were summarized in Tables 2 and 3.

TABLE-3  
ANTIFUNGAL PROPERTIES OF THE  
SYNTHESIZED COMPOUNDS

Compound	Conc. (µg/mL) (zone of inhibition in mm)					
	<i>Aspergillus niger</i>			<i>Aspergillus oryzae</i>		
	1000	750	500	1000	750	500
<b>4a</b>	7	6	5	8	7	6
<b>4b</b>	10	10	9	11	10	10
<b>4c</b>	5	4	3	8	7	5
<b>4d</b>	6	6	5	5	5	4
<b>4e</b>	6	6	5	4	3	1
<b>4f</b>	7	6	5	4	3	3
<b>4g</b>	5	4	3	4	3	2
<b>4h</b>	4	3	2	5	4	4
<b>5a</b>	5	4	3	5	4	3
<b>5b</b>	7	6	4	7	6	5
<b>5c</b>	10	9	8	10	9	8
<b>5d</b>	8	7	5	5	4	3
<b>5e</b>	8	7	6	7	6	5
<b>5f</b>	7	6	5	7	6	5
<b>5g</b>	NA	NA	NA	NA	NA	NA
<b>5h</b>	6	5	4	6	5	4
Fluconazole	16	16	15	18	17	15

The results of antimicrobial screening revealed that some of the tested compounds showed moderate to good bacterial inhibition whereas all the compounds were exhibited moderate to weak fungal inhibition. As per structural discussion of these products studies and explained that the different substituent on the aromatic ring present in the synthesized compounds exhibited a significant influence on the biological activity. The presence of electron-withdrawing group on the aromatic ring increases the antibacterial activity of tested compounds compared to compounds having electron donating groups. Compounds **3b**, **3d**, **4b**, **4d**, **5b**, **5d** and **5c** have shown good zone of inhibition against *S. aureus* and *E. coli* with three different concentrations. The compounds such as 2-(4-chloro-phenyl)-3-(3-hydroxy-4-(1*H*-benzo[d]imidazol-2-yl)phenyl)thiazolidin-4-one, 3-(3-hydroxy-4-(1*H*-benzo[d]imidazol-2-yl)phenyl)-2-(4-nitro-phenyl)thiazolidin-4-one and 4-((3-(1*H*-benzo[d]imidazol-2-yl)-6-(2-(4-chlorophenyl)-4-oxothiazolidin-3-yl)-2-hydroxy-phenyl)diazanyl)benzene sulfonamide exhibited more potent antibacterial activity against four bacterial strains and the compound 3-(4-(1*H*-benzo[d]imidazol-2-yl)-3-hydroxy-phenyl)-2-*o*-tolylthiazolidin-4-one was no activities against all

bacterial and fungal strains. The tested compounds such as **4b** and **5b** were excellent inhibition activity against *S. aureus*, *E. coli* and *P. aeruginosa*. The compounds **3a**, **3c**, **3f**, **4a**, **4c** and **5a** were found to moderate to weak activities against all bacterial stains but all the synthesized Schiff's base and thiazolidinone derivatives products were ineffective against *S. typhi*. These results revealed the importance of benzimidazolyl and sulphomoyl substituted 4-thiazolidinone ring systems as basic structural constituents in the synthesized compounds. The substitution of chloro, nitro and sulphonamide group C-4 position of the phenyl ring and azosulphonamide bearing 4-thiazolidinone ring presence to be very important for antibacterial effect. Among all the antifungal tested synthesized compounds **4b** and **5c** were found to be moderate activity against fungal strains.

## Conclusion

Present research work involves synthesis of benzimidazole bearing 4-thiazolidinone derivative to explore their antimicrobial activity. All the compounds were recrystallized, checked the purity by TLC and functionally interpreted by IR and <sup>1</sup>H NMR. In general from the above results, it was concluded the compounds having chlorine, nitro and sulphonamide at C-4 position of the substituted phenyl which enhanced the activity due to presence electron withdrawing functional groups.

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