

Molecular Docking Study of Novel Synthesized Pyrazole Derivatives and their Antibacterial Activity

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The current novel series of 1-[5-(furan-2-yl)-4,5-dihydro-3-(4-morpholinophenyl)pyrazol-1-yl]ethanone are synthesized from the 3-(furan-2-yl)-1-(4-morpholinophenyl)prop-2-ene-1-one by cyclization reaction. The 3-(furan-2-yl)-1-(4-morpholinophenyl)prop-2-ene-1-one chalcones are derived from the reaction of 4-morpholino acetophenone with substituted furfuraldehyde by common Claisen-Schmidt condensation reaction. The structures of the synthesized compounds skeleton are elucidated by IR, ¹H NMR and ¹³C NMR. The antimicrobial susceptibility tests of synthesized compounds are screened against *Staphylococcus aureus* and *Escherichia coli*. From the three synthesized compounds the electron-withdrawing substitution (Br) shows good effect against the Gram-positive bacteria and the electron donating substitution (CH₃) shows excellent inhibition zone against the Gram-negative bacteria at the concentration level (10 μ g/mL). The molecular docking is an effective and compounds towards 3 selected bacterial proteins (2X5O, 3TYE, 3UDI).

Keywords: Claisen-Schmidt condensation, Pyrazole, Molecular docking, Antimicrobial activities.

INTRODUCTION

Among heterocycles, pyrazole derivatives are an important class of nitrogen containing five membered heterocyclic compounds that have more attention in the field of drug discovery and agricultural research [1-3]. Pyrazole have two adjacent N-atom fused with three C-atom [4]. Pyrazoles and pyrazolones have special interest because they constitute an important class of natural and unnatural products [5]. They are rarely found in nature because due to difficulty in the formation of N-N bond by the living organisms [6]. Significant effort has been put forth in developing approaches for pyrazole synthesis [7]. The heterocyclic compounds undergo the most common synthesis methods such as cyclocondensation [7] and Michal addition [8] when react with hydrazine hydrate to form a pyrazole moiety. The pyrazole moiety is a versatile lead molecule in the pharmaceutical development because of its wide range of biological activities such as antifungal [9,10], antimicrobial [11-13], antiviral [14], anticancer [15], antitubercular [16,17], antitumor [18], antidepressant [19] and antibacterial [20].

The pyrazole moiety also used in the fluorescence, luminescence, herbicides [21], ligands [22], in the preparation of dyes [23] and as the corrosion inhibitors [8,24]. The successfully commercialized drugs contains pyrazole moiety are celexib, fipronil, sildenafil [7], pyrazofurin, lonazolac, rimonabant, ramifenazone and phenylbutazone [9]. The *bis* acetylated pyrazole [25] moiety also synthesized and it also have antimicrobial property [26].

Nowadays, infections are increased widely. The infections are caused by Gram-positive and Gram-negative pathogens [27]. For this reason, a number of drugs having antimicrobial property has to be produced. Due to this, in the present work the pyrazole derivatives are synthesized. The structure of the compounds are confirmed by IR, ¹H NMR and ¹³C NMR spectral. The synthesized compounds subjected to antimicrobial screening test and try to compile the structure activity relationship (SAR) of pyrazole derivatives with different bacterial proteins such as 3TYE, 3UDI and 2X5O.

EXPERIMENTAL

Spectral devices: The MELT-TEMP apparatus is used to find the melting point of the synthesized compounds by using open capillary method. The IR spectra characterization is carried out by using SHIMADZU FT-IR spectrometer and the KBr pellet is used as a reference. The BRUKER 400 Hz and 100 Hz instrument is used for the ¹H NMR and ¹³C NMR spectral characterization of the synthesized compounds. During NMR characterization the TMS is used as a internal reference and CDCl₃ as a solvent.

Synthesis of 3-(furan-2-yl)-1-(4-morpholinophenyl)prop-2-ene-1-ones (1a-1c): 1 mol of 4-Morpholinoacetophenone and 1 mol of substituted furfuraldehyde are weighed. The 4-morpholino-acetophenone is taken in the 250 mL beaker containing 30 mL of NaOH mixed with ethanolic solution and stirred well. After that the weighed furfuraldehyde is added to it and mixed well. The mixture is placed in the ice-cold water bath and stirred well for 30 min. The mixture is poured into a crushed ice and it is kept for a night at room temperature. Finally the solid chalcone is filtered, dried and recrystallized from ethanol. The purity of the chalcones is checked by TLC.

Synthesis of 1-[5-(furan-2-yl)-4,5-dihydro-3-(4-morpholinophenyl)pyrazol-1-yl]ethanone (4a-4c): Various substituted chalcones are taken in the 250 mL round bottom flask which contains acetic acid as a solvent. The hydrazine hydrate (1 mol) is added to it and a pinch amount of sodium acetate is also added. This mixture is refluxed for 8-10 h. The completion of the reaction is monitored by TLC. Finally, the product precipitates out as a solid, filtered and dried. The structure of the synthesized compound is confirmed by IR, ¹H NMR and ¹³C NMR spectral data. The elemental analysis is used to confirm the composition of atoms present in the synthesized compounds.

Lipinski's rule: This rule also known as Pfizer's rule of five, this is the rule to evaluate the drug likeness formulated by Christopher A. Lipinski based on oral administration of drugs [28]. This rule states that an orally active drug has no more than one violation of the following criteria, compound has (i) no more than 5 H-bond donors (the total no. of N-H and O-H bonds), (ii) No more than 10 H-bond acceptors (all N or O atoms), (iii) The molecular mass is less than 500 (Daltons) (iv) An octanol-water partition coefficient (log P) not more than 5, and (v) Not more than 120 cm³/mol of compounds molar refractivity.

Antimicrobial screening test: The inoculums and sterile swabs are constructed and sterilized. The synthesized compounds **4a-4c** are subjected to antimicrobial screening against the *Staphylococcus aureus* (Gram-positive) pathogen and *Escherichia coli* (Gram-negative) pathogen with the help of agar nutrient medium by disc-diffusion method. The maximum inhibition in minimum concentration of synthesized compounds is carried out by using micro-dilution method.

Molecular docking: Molecular docking studies have been carried out using the AutoDock Tools (ADT) version 1.5.6 and AutoDock version 4.2.5.1 docking program. The method of docking studies is followed by literature method [29].

RESULTS AND DISCUSSION

The new synthesized compound 1-[5-(furan-2-yl)-4,5-dihydro-3-(4-morpholinophenyl)pyrazol-1-yl]ethanone (**4a-4c**) are synthesized from the reaction of (*E*)-3-(furan-2-yl)-1-(4-morpholinophenyl) prop-2-en-1-one (**3a-3c**) with hydrazine hydrate in the presence of ethanol *via* Michal-addition mechanism. The compounds **3a-3c** are synthesized by Claisen-Schmidt condensation reaction between furan-2-carbaldehyde and 2-morpholino-1-phenylethanone. The pathway of the target compounds are shown in Fig. 1.

Spectral data

IR spectral values of synthesized compounds 4a-4c: The synthesized compounds 4a-4c show a characteristic band in 1645-1630 cm⁻¹ region due to the presence of amide C=O stretching frequencies. The stretching frequency around 3000 cm⁻¹ shows the peak of the aliphatic CH stretching. The aromatic CH group shows the stretching frequency from the range of 3000-2950 cm⁻¹. The absorption band of the group C-N shows band around at 1110 cm⁻¹ and the group C=N stretching presence in the range of 1450 cm⁻¹. The remaining stretching frequencies are 833.25, 821.68, 767.67, 754.17 and 744.52 cm⁻¹ due to presence of aromatic ring moiety.

¹H and ¹³C NMR spectral values of synthesized compounds 4a-4c: The ¹H NMR of the synthesized compound 4a shows the acetyl methyl protons present in the pyrazoline moiety gives the singlet signal at 2.237 ppm. The protons in the -N(CH₂)₂ group of morpholine moiety shows a doublet at 3.15 ppm and the coupling constant value, J = 4.4 Hz and the protons of



Fig. 1. Reaction scheme of the synthesized compounds 4a-4c

Compound

4a (II)

 $-O(CH_2)_2$ in the same moiety which also gives the doublet at 3.73 with the value of J = 4.4 Hz. The methylene protons (H-4a and H-4e) of the pyrazoline moiety appeared as two doublet of doublet due to multiple coupling by involving both germinal and vicinal protons. The signals of H-4a and H-4e are observed at 3.23 ppm and 3.65 ppm. The doublet of doublet of H-4a of pyrazoline moiety is assigned at 3.23 ppm with the coupling constant values $J_{4a,5a}$ = 19.6 Hz and $J_{4a,4e}$ = 4.4 Hz. In the same way, the doublet of doublet is assigned at 3.65 ppm ($J_{4e,4a}$ = 11.6 Hz and $J_{4e,5a} = 18$ Hz) for an H-4e protons. The another one doublet of doublet is observed in the range of 5.58 ppm $(J_{5a,4a} = 4 \text{ Hz and } J_{5a,4e} = 11.6 \text{ Hz})$ is due to the presence of the methine proton (H-5) of pyrazoline moiety and vicinal coupling with two magnetically non-equivalent protons of methylene group (H-4a and H-4e). Further, the H-3 proton of furan moiety gives a signal at 6.38 ppm and H-4 proton of the same ring shows doublet signal at 6.28 ppm with the J value is equal to 2.4 Hz. The H-2 and H-6 protons of phenyl ring gives the signal at 7.64 (J = 8.4 Hz) and the H-3 and H-5 of phenyl ring shows signal at 7.53 ppm.

¹³C NMR of the synthesized compound **4a**, ¹³C resonance at 47.39 ppm is assigned to the C-5 of pyrazole moiety and the signals at 38.35 ppm and 154.34 ppm are assigned for the C-4 and C-3 carbons of the pyrazole ring. The ¹³C resonance is observed at 21.64 ppm is due to the presence of acetyl methyl group which is attached in the pyrazole moiety and the signal at 167.11 ppm is assigned for acetyl C=O group is also attached in the same ring. The signal at 52.67 ppm is assigned for the N(CH₂)₂ group of morpholine moiety and in the same ring the $O(CH_2)_2$ group shows a signal at 65.87 ppm. In furan ring, the signals appeared in the positions 152.25 ppm, 106.76 ppm, 110.13 ppm and 142.13 ppm positions which is assigned to the C-2, C-3, C-4 and C-5 carbons respectively. The C-3 and C-5 position have a signal at the 114.16 ppm and the C-2 & C-6 carbons signal appeared at 127.75 ppm. The C-1 carbon of phenyl ring shows a signal at 120.96 ppm and the C-4 carbon of the same ring shows signal at the range 152.92 ppm. The ¹H NMR and ¹³C NMR results of compounds are shown in Tables 1 and 2.

Lipinski rule of compounds 4a-4c: The synthesized compounds 4a-4c obey the Lipinski rule of five. Thus, all the compounds have drug likeness property. The values are shown in Table-3.

Antibacterial activity: The synthesized compounds 4a-4c show the highest activity against Gram-positive and Gramnegative bacterial strains. According to the inhibition result, the CH₃ (e⁻ donating) substitution has wide spectrum against Gram-negative bacteria E. coli and the Br (e⁻ withdrawing) substitution has excellent inhibition against Gram-positive bacteria S. aureus. All the synthesized compounds show the best inhibition value at 10 µg/mL. The inhibition values of the synthesized compounds are given in Table-4.

 (CH_2)

		¹ H NMF	R CHEMICAL S	SHIFT VALU	TABLE-1 JES OF THE SY	YNTHESIZEI	O COMPOU	ND 4a-4c		
					Chemical Shift	ts (δ ppm)				
Compd.	Acetyl CH ₃	N(CH ₂) ₂ of morpholine moiety	O(CH ₂) ₂ of morpholine moiety	H-4a of pyrazole ring	H-4e of pyrazole ring	H-5a of pyrazole ring	H-3 of furan moiety	H-4 of furan moiety	H-2 & H- 6 proton of phenyl ring	H-3 & H- 5 proton of phenyl ring
4a (H)	2.237	3.15 (d) <i>J</i> = 4.4 Hz	3.73 (d) J = 4.4 Hz	3.23 (dd) $(J_{4a,4e} =$ 4.4 Hz & $J_{4a,5e} =$ 19.6 Hz)	3.65 (dd) ($J_{4e,4a} = 11.6$ Hz & $J_{4e,5a}$ = 18 Hz)	5.58 (dd) $(J_{5a,4a} = 4)$ Hz & $J_{5a,4e}$ = 11.6 Hz)	6.38 (d)	6.28 (d) <i>J</i> = 2.4 Hz	7.64 (d) <i>J</i> = 8.4 Hz	7.53 (d)
4b (CH ₃)	2.23 & CH ₃ in furan moiety = 2.18 Hz	3.13 (d)	3.74 (d)	3.26 (dd) $(J_{4a,4e} =$ 4.2 Hz & $J_{4a,5a} =$ 18 Hz)	3.63 (dd) ($J_{4e,4a} = 11.6$ Hz & $J_{4e,5a} =$ 17.6 Hz)	5.51 (dd) $(J_{5a,4a} =$ 4.4 Hz & $J_{5a,4e} =$ 11.6 Hz)	6.28 (d)	6.13(d) <i>J</i> = 2.4 Hz	7.64 (d) <i>J</i> = 8.6 Hz	6.99 (d) <i>J</i> = 8.8 Hz
4c (Br)	2.24	3.21 (d) J = 4 Hz	3.73 (d) J = 4.4 Hz	3.30 (dd) $(J_{4a,4e} =$ 4.8 Hz & $J_{4a,5a} = 18$ Hz)	3.68 (dd) $(J_{4e,4a} = 9.4$ Hz & $J_{4e,5a}$ = 18.4 Hz)	5.55 (dd) $(J_{5a,4a} =$ 4.4 Hz & $J_{5a,4e} =$ 11.8 Hz)	6.47 (d) J = 2.8 Hz	6.36 (d) <i>J</i> = 3.2 Hz	7.64 (d) <i>J</i> = 8.4 Hz	6.99 (d) <i>J</i> = 8.8 Hz

	¹³ C NMR CHEMICAL SHIFT VALUES OF SYNTHESIZED COMPOUND 4a-4c								
			Chemic	cal shifts (δ values i	in ppm)				
1	A aatul CH	A setul C. O	Pyrazole moiety			Morpholi	ne moiety		
	Acetyl CH ₃	Acetyl C=0	C-3	C-4	C-5	N(CH ₂) ₂	O(CH ₂)		
	21.64	167 11	154.24	20.25	47.20	52.67	65 07		

TABLE-2

Ha (11)	21.04	107.	11	154.54	50.55	T 1.57	52	.07	05.07	
4b (CH ₃)	21.69	167.	11	154.25	38.33	47.40	52	.77	65.88	
4c (Br)	21.64	167.	30	155.26	38.32	47.37	52	.70	65.87	l
			Furan moiety	,			Pheny	l ring		l
	C-2	C-3	C-4	C-5	CH ₃	C-3 & C-5	C-2 & C-6	C-1	C-4	l
4a (H)	152.25	106.76	110.13	142.13	_	114.16	127.75	120.96	152.92	l
4b (CH ₃)	151.24	106.40	110.44	150.77	66.03	114.17	127.75	121.02	152.24	
4c (Br)	152.28	109.8	112.42	114.15	-	114.15	127.82	120.79	152.28	l

	TABLE-3 LIPINSKI RULE VALES OF COMPOUNDS 4a-4c								
			Lipinski rule of five						
Compound	H-bond donors $(not > 5)$	H-bond acceptors (not > 10)	Molecular mass (not greater than 500 daltons)	log P (no greater than 5)	Molar refractivity (between 30-120) cm ³ /mol				
4 a	< 5	< 10	339.16	1.35	95.78				
4b	< 5	< 10	353.17	1.68	100.73				
4c	< 5	< 10	417.07	2.06	103.7				

Docking result: Since the molecular mechanism of the antibacterial activity of synthesized pyrazoles (4a-4c) are still unknown. The molecular docking technique is used to predict their possible targets. On the basis of literature search, three bacterial proteins are choosed. The protein enzymes used are given in the following: an enzyme MurD ligase (PDB ID: 2X5O, 3UDI) and dihydropteroate synthase enzyme (DHPS; PDB ID: 3TYE). The docking score values of the synthesized compounds are given in Table-5. The compound 4c shows an excellent binding score -7.8 and -7.5 for an enzyme MurD (PDB ID: 3UDI and 2X5O). It clearly result that the electron withdrawing substitution has excellent bacterial inhibition. The

TABLE-4 ZONE OF INHIBITION VALUES OF COMPOUNDS 4a-4c								
	Zone of inhibition (diameter)							
Compound	d Staphylococcus aureus Escherichia coli							
	10 µg/mL	5 μg/mL	2.5 μg/mL	10 µg/mL	5 µg/mL	2.5 μg/mL		
4a	19	16	12	16	14	10		
4b	16	15	8	19	12	7		
4c	22	19	13	15	13	10		









synthesized compounds **4a-4c** with MurD enzyme (PDB ID: 2X5O) have the best binding values in the range of -7.5 to -7.3. In the DHPS binding pockets the compound **4b** shows a high binding score -7.6 (the **4b** compound, the CH₂ group in the morpholine moiety form carbon-hydrogen bond with LYS A:126 and the aromatic and heteroaromatic ring form a π - σ bond which is form alkyl binding with PHE A:71). The binding scores and 2D, 3D binding structures of all the synthesized compounds among three bacterial proteins are given in Table-5.

Molecular docking studies show that the synthesized compounds could act as very good inhibitors for DHPS and MurD ligase. This study gives an idea for the designing more effective antibacterial agents in the future.

Conclusion

The target compounds are successfully synthesized *via* **Scheme-I** reaction (Michal addition). The chemical structure of the synthesized compounds are identified by IR, ¹H NMR and ¹³C NMR spectral values. The antibacterial inhibition tests are screened. From that result, the electro-negativity substituted compound like Br shows an excellent inhibition against Grampositive strain (*Staphylococcus aureus*) and the electron donating substituent like CH₃ shows the best inhibition zone against Gram-negative bacterial strain (*Escheichia coli*). Finally, the molecular docking studies help to predict the *in silico* screening of synthesized compounds with MurD enzyme and DHPS enzyme proteins. It can be concluded that the compounds **4a-4c** inhibit cell wall synthesis of bacteria there by inhibiting the growth of the bacteria.

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