

NOTE

Synthesis and Characterization of 3-(Pyridin-2-ylmethyl)pentane-2,4-dione

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A new compound, 3-(pyridin-2-ylmethyl)-pentane-2,4-dione was synthesized successfully and characterized by ¹H NMR, IR, MS and elemental analysis.

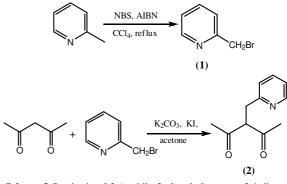
Key Words: 3-(Pyridin-2-ylmethyl)pentane-2,4-dione, 2,4-pentanedione, N-bromosuccinimide.

2,4-Pentanedione, also named acetylacetone, is an important chemical intermediate and used widely for the synthesis of pharmaceuticals, feed additives and catalysts *etc.*¹. Further more, 2,4-pentanedione and its derivatives can coordinate with many metal ions to obtain metal chelates, which can catalyze a number of reactions^{2,3}. For example, 2,4-pentanedione chelated nickel complex can catalyze the oligomerization or polymerization of ethylene, propylene, styrene, diene, cycloolefins, MMA *etc.*⁴⁻⁶.

Recently, we synthesized a new compound 3-(pyridin-2ylmethyl)pentane-2,4-dione, which is derived from 2,4pentanedione by introducing pyridinylmethyl group into 3position carbon. This tridentate ligand can also coordinate with metal compounds and the catalytic performance of the obtained metal complexes may be improved due to the introduced pyridinyl group. The synthetic route of 3-(pyridin-2-ylmethyl)pentane-2,4-dione was shown in **Scheme-I.**

The title compound was characterized by ¹H NMR, IR, MS and element analysis and the structure was confirmed. From ¹H NMR, it was found that this compound may own two isomers and the molar ratio of two isomers is 2:1 by calculating the integral areas, which is shown in Fig-1.

Preparation of 2-(bromomethyl)pyridine (1)⁷⁻⁹**:** *N*bromosuccinimide (8.9 g, 0.05 mol), 2-methylpyridine (4.7 g, 0.05 mol) and AIBN (0.3 g) were dissolved in CCl₄ (80 mL) and refluxed for 8 h. After filtration of the suspension, the filtrate was neutralized by Na₂CO₃ solution and washed twice with 100 mL H₂O, then dried by MgSO₄. The solution was concentrated under reduced pressure and the residue was purified by column chromatography on silica gel (eluent: petroleum ether : ethyl acetate = 2: 1) to give 2.67 g (31 %) of 1; ¹H NMR (400 MHz, CDCl₃): δ 4.53 (s, 2H,-CH₂), 7.18 (m, 1H, Py-H), 7.41 (m, 1H, Py-H), 7.66 (m, 1H, Py-H), 8.55 (m, 1H, Py-H); m.p. 117 °C. Anal. calcd. for C₆H₆NBr: C, 41.88; H, 3.52; N, 8.13. Found: C, 41.91; H, 3.55; N, 8.11.



Scheme-I Synthesis of 3-(pyridin-2-ylmethyl)pentane-2,4-dione

Preparation of 3-(pyridin-2-ylmethyl)pentane-2,4dione (2): A mixture of 2,4-pentanedione (2 g, 0.02 mol), sodium carbonate (4.15 g, 0.03 mol) and acetone (60 mL) were refluxed for 5 h. Then 2-(bromomethyl)pyridine (3.5 g, 0.02 mol) and KI (1.5 g) were added and refluxed again for 8 h. The mixture was filtered and the solvent was removed *in vacuo*. After adding water, the mixture was extracted three times with Et₂O and filtrated after being dried by MgSO₄. Then the solvent was removed to leave light red liquid 3.25 g (85 %). ¹H NMR (400 MHz, CDCl₃): δ 8.59 (m, 1H, Py-H), 7.24-7.32 (m, 3H, Py-H), 3.94 (s, 1H, -CH), 3.47 (s, 2H, -CH₂), 2.37 (m, 6H, -CH₃); IR (cm⁻¹): 3055, 2986, 2925, 1727, 1589, 1473, 1430, 1359, 1241, 1151, 1097, 1044, 996, 943, 847, 755. MS(EI)(m/z): 191 (M⁺). Anal. calcd. for C₁₁H₁₃NO₂: C, 69.08; H, 6.85; N, 7.33. Found: C, 69.05; H, 6.89; N, 7.31.

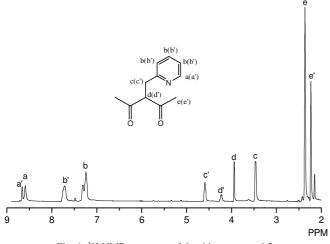


Fig. 1. ¹H NMR spectrum of the title compound 2

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REFERENCES

S. Miquel, J. Chem. Educ., 66, 779 (1989).

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- 2. G. Xie and C. Qian, J. Polym. Sci. Part A: Polym. Chem., 46, 211 (2008).
- 3. G. Xie, Y. Li, J. Sun and C. Qian, *Inorg. Chem. Commun.*, **12**, 796 (2009).
- 4. F.J. Muzzio and D.G. Loffler, Acta Chim. Scanda., 124, 403 (1987).
- 5. M. Puckert and W. Keim, Organometallics, 2, 594 (1983).
- 6. V. Muller, W. Keim and C. Kruger, *Angew. Chem. Int. Ed.*, **28**, 1011 (1989).
- 7. S.H. Louis, J.S. Michael and K.D. Peter, Inorg. Chem., 42, 4346 (2003).
- 8. H. Marcel and S.S. Ulrich, J. Org. Chem., 67, 8269 (2002).
- 9. I. Ryosuke, U. Naoki and H. Tsunehiko, *Bioorg. Med. Chem.*, **14**, 3563 (2006).