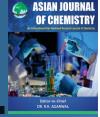


ASIAN JOURNAL OF CHEMISTRY



https://doi.org/10.14233/ajchem.2024.31196

Magnesium Powder-Catalyzed, Highly Efficient, Solvent-Free Synthesis of Amides through N-Acetylation of Amines and their Antibacterial Activity

Najma Bajracharya^{1,©}, Sunita Shrestha^{1,2,©} and Gan B. Bajracharya^{1,*,©}

¹Laboratory of Catalysis and Frontier Molecules, Faculty of Science, Nepal Academy of Science and Technology (NAST), Khumaltar, Lalitpur, Nepal

²Department of Chemistry, Tri-Chandra Multiple Campus, Tribhuvan University, Ghantaghar, Kathmandu, Nepal

*Corresponding author: E-mail: ganbajracharya@yahoo.com; gan.bajracharya@nast.gov.np

Received: 10 January 2024; Accepted: 23 February 2024; Published online: 30 March 2024; AJC-21583

Magnesium powder was found to be an efficient catalyst for the acetylation of amines (1) with Ac₂O affording corresponding amides (2) in excellent yields (> 99%). This green synthetic protocol has utilized 2.5 mol% of magnesium powder and a stoichiometric amount of Ac₂O (1.2 equiv.). Amides **2a-k** were synthesized within a short reaction time (2-3 min) at an ambient temperature under solvent free condition. The products were screened against *Escherichia coli*, *Pseudomonas aeruginosa*, *Staphylococcus aureus* and *Staphylococcus epidermidis* bacterial strains. *N*-(2-Hydroxyphenyl)acetamide (**2f**) was found most active against Gram-positive *S. aureus* and *S. epidermidis* indicating its efficacy in treating skin diseases.

Keywords: Acylation, Antibacterial assay, Green chemistry, Skin disease, Solvent-free, Organic synthesis.

INTRODUCTION

Acetamide derivatives are reported to possess several bioactivities such as antimicrobial [1-5], antitubercular [6,7], antimalarial [8], antiparasitic [9], antidiabetic [10,11], antipsychotic [12], antioxidant activities [13], *etc.* Acetamides are used also as chemotherapeutic agents for inflammation and inflammation-associated cancers [14,15]. More than 25% of the worldwide used pharmaceutical drugs possess an amidal motif [16,17]. Because of the presence of amide linkage in the chemical structures, the compounds may display specific enhanced bioactivity [18,19]. Some examples include paracetamol (antipyretic), penicillin (antibacterial), pyrazinamide (antitubercular), *etc.*

Acylation is one of the most fundamental reactions in the organic synthesis. This reaction is extensively used to protect various functional groups such as phenols, alcohols, amines, thiols and others in order to achieve synthesis of target molecules [20]. Besides using of classical acids and bases, several Lewis acid catalysts, solid acid and heteropolyacid catalysts, organocatalysts, ionic liquids and others have been used for the acylation reactions [21]. Amides are commonly produced by *N*-acylation of amines [22-24].

Recently, we reported magnesium powder as an efficient catalyst for the acetylation of phenols with acetic anhydride under neat condition and a free radical mechanism was proposed for the transformation [21]. More recently, Anbu et al. [25] also reported that the acetylation of alcohols, amines, phenols and thiols is achievable under catalyst- and solvent-free conditions involving an ionic mechanism. Significantly, amine acetylation was acheived at 60 °C in 30 min using 1.5 equiv. of acetic anhydride. Herein, the acetylation of amines (1) was accomplished at room temperature by using a stoichiometric amount of acetic anhydride, in the presence of a trace amount of magnesium powder, under air atmosphere. The reactions were completed within 2-3 min affording the corresponding amides (2) in excellent to quantitative yields just after simple work up. Moreover, obtained amides 2a-k were further evaluated for the antibacterial activity.

EXPERIMENTAL

Chemicals and reagents were purchased from different reputed commercial supplers like Fischer, Qualigens, Aldrich, Burgoyne Burbidges and Merck. Magnesium powder and media were purchased from Himedia. Thin layer chromatography

This is an open access journal, and articles are distributed under the terms of the Attribution 4.0 International (CC BY 4.0) License. This license lets others distribute, remix, tweak, and build upon your work, even commercially, as long as they credit the author for the original creation. You must give appropriate credit, provide a link to the license, and indicate if changes were made.

(TLC) was performed using pre-coated Kieselgel 60 F_{254} plates of 0.2 mm thickness (E. Merck). For product identification, GC-MS spectra were recorded using Agilent 7890A GC system coupled with an Agilent 5975 C mass selective detector and IR spectra were recorded in KBr disc using an IR Tracer-100 (Shimadzu). The measured melting points are uncorrected.

General procedure for *N*-acetylation of amines: Amine (1, 5 mmol), powder Mg (2.5 mol%, 3 mg, 0.125 mmol) and acetic anhydride (0.57 mL, 6 mmol) were charged in a reaction tube equipped with a stirring bar. The content was stirred at 25 °C and the reaction was monitored by TLC. After completion, the reaction mixture was diluted with EtOAc (40 mL), washed with brine, dried over Na₂SO₄, filtered, concentrated and finally vacuum dried. Corresponding amide 2 was obtained in chromatographically pure form. No further purification was required, except in the synthesis of compound 2i.

N-Phenylacetamide (2a): The general procedure was followed by using aniline (1a, 466 mg, 5 mmol) to obtain 2a as a colourless solid. Yield: 628 mg, 93%. m.p.: 114 °C (reported 112-114 °C) [26,27]. R_f = 0.43 (silica gel, hexane:EtOAc, 1:1). GC-MS: calcd. for C_8H_9NO (M⁺) 135.0684, found 135.1. IR (KBr, cm⁻¹): 3294 (vNH); 3194, 3138, 3064 (vCH arom.), 1666 (vCO), 1494 (δHNC), 1438 (δHCH), 1321 (δHCC+vCC) [28].

N-(4-Methoxyphenyl)acetamide (2b): The general procedure was followed by using 4-methoxyaniline (1b, 616 mg, 5 mmol) to obtain 2b as a purple solid. Yield: 719 mg, 87%. m.p.: 127-129 °C (reported 126-129 °C) [26,29]. $R_f = 0.23$ (silica gel, hexane:EtOAc, 1:1). GC-MS: calcd. for $C_9H_{11}NO_2$ (M⁺) 165.0790, found: 165.1. IR (KBr, cm⁻¹): 3244 (vNH), 3190, 3130, 3070 (vCH arom.), 1654 (vCO),1512 (δHNC), 1409 (δHCH), 1321 (δHCC + vCC).

N-(*m*-Tolyl)acetamide (2c): The general procedure was followed by using *m*-toluidine (1c, 536 mg, 5 mmol) to obtain 2c as a colourless solid. Yield: 710 mg, 95%. m.p.: 65-66 °C (reported 65–69 °C) [26,27]. $R_f = 0.43$ (silica gel, hexane: EtOAc, 1:1). GC-MS: calcd. for C₉H₁₁NO (M⁺) 149.0841, found: 149.1. IR (KBr, cm⁻¹): 3300 (vNH); 3145, 3049 (vCH arom.); 1674 (vCO); 1404 (δHCH); 1317 (δHCC + vCC).

N-(*o*-Tolyl)acetamide (2d): The general procedure was followed by using *o*-toluidine (1d, 536 mg, 5 mmol) to obtain 2d as a colourless solid. Yield: 706 mg, 95%. m.p.: 108-111 °C (reported 108-110 °C) [26,27]. R_f = 0.35 (silica gel, hexane: EtOAc, 1:1). GC-MS: calcd. for C₉H₁₁NO (M⁺) 149.0841, found: 149.1. IR (KBr, cm⁻¹): 3292 (vNH); 3194, 3034, 2980 (vCH arom.); 1651 (vCO); 1529 (δHNC); 1456 (δHCH); 1371 (δHCC + vCC).

N-(4-Hydroxyphenyl)acetamide (2e): The general procedure was followed by using 4-aminophenol (1e, 545 mg, 5 mmol) to obtain 2e as a cream white solid. Yield: 700 mg, 93%. m.p.: 170 °C (reported 168-169 °C) [26,30]. $R_f = 0.80$ (silica gel, CH₃OH). GC-MS: calcd. for $C_8H_9NO_2$ (M⁺) 151.0633, found: 151.1. IR (KBr, cm⁻¹): 3327 (vOH); 3165 (vNH); 2929, 2881 (vCH arom.); 1656 (vCO); 1510 (δHNC); 1440 (δHCH); 1325 (δHCC + vCC).

N-(2-Hydroxyphenyl)acetamide (2f): The general procedure was followed by using 2-aminophenol (1f, 545 mg, 5

mmol) to obtain **2f** as a colourless solid. Yield: 666 mg, 88%. m.p.: 206 °C (reported 207-209 °C) [26]. $R_f = 0.52$ (silica gel, hexane:EtOAc, 1:1). GC-MS: calcd. for $C_8H_9NO_2$ (M⁺) 151.0633, found: 151.1. IR (KBr, cm⁻¹): 3404 (vOH); 3080 (vNH); 2976, 2881 (vCH arom.); 1660 (vCO); 1456 (δ HNC); 1382 (δ HCH); 1330 (δ HCC + vCC).

N-(4-Bromophenyl)acetamide (2g): The general procedure was followed by using 4-bromoaniline (1g, 860 mg, 5 mmol) to obtain 2g as a colourless solid. Yield: 999 mg, 94%. m.p.: 164-166 °C (reported 166-170 °C) [26,27,31,32]. $R_f = 0.40$ (silica gel, hexane:EtOAc, 1:1). GC-MS: calcd. for C_8H_8 BrNO (M⁺) 212.9789, found: 213.0. IR (KBr, cm⁻¹): 3294 (vNH); 3186, 3115, 3053 (vCH arom.); 1672 (vCO); 1485 (δHNC); 1388 (δHCH); 1305 (δHCC + vCC).

N-(4-Nitrophenyl)acetamide (2h): The general procedure was followed by using 4-nitroaniline (1h, 690 mg, 5 mmol) to obtain 2h as a pale yellow solid. Due to poor solubility of the product, an excess of EtOAc was needed to use for its extraction during work up process. Yield: 893 mg, 99%. m.p.: 214 °C (reported 210-214 °C) [26,31]. $R_f = 0.25$ (silica gel, hexane: EtOAc, 1:1). GC-MS: calcd. for $C_8H_8N_2O_3$ (M⁺) 180.0535, found: 180.1. IR (KBr, cm⁻¹): 3275 (vNH); 3217, 3157, 3091 (vCH arom.); 1679 (vCO); 1564 (vNO); 1498 (δHNC); 1404 (δHCH); 1348 (δHCC + vCC).

N-(2-Nitrophenyl)acetamide (2i): Following the general procedure, the reactions of 2-nitroaniline (1i, 691 mg, 5 mmol) were performed at two different temperatures (25 and 100 °C) to obtain compound 2i. The product was purified by recrystallization with 75% EtOH producing yellow crystals. Yield: 827 mg, 92% (at 25 °C, reaction time 16 h). Yield: 890 mg, 99% (at 100 °C, reaction time 2 h). m.p.: 91 °C (reported 90-91 °C) [33]. R_f = 0.64 (silica gel, hexane:EtOAc, 1:1). GC-MS: calcd. for $C_8H_8N_2O_3$ (M⁺) 180.0535, found: 180.1. IR (KBr, cm⁻¹): 3373 (vNH); 3165, 3089 (vCH arom.); 1712 (vCO); 1585(vNO); 1496 (δHNC); 1431 (δHCH); 1346 (δHCC + vCC).

N-(Naphthalen-1-yl)acetamide (2j): The general procedure was followed by using 1-naphthylamine (1j, 716 mg, 5 mmol) to obtain 2j (886 mg, 96%) as a colourless solid. Yield: 886 mg, 96%. m.p.: 160 °C (reported 159-160 °C) [26]. R_f = 0.36 (silica gel, hexane:EtOAc, 1:1). GC-MS: calcd. for $C_{12}H_{11}NO$ (M⁺) 185.0841, found: 185.1. IR (KBr, cm⁻¹): 3271 (vNH); 3049 (vCH arom.); 1654 (vCO); 1500 (δHNC); 1456 (δHCH); 1342 (δHCC + vCC).

N-Benzylacetamide (2k): The general procedure was followed by using benzylamine (1k, 536 mg, 5 mmol) to obtain 2k as a colourless solid. Yield: 714 mg, 96%. m.p.: 60 °C (reported 58-60 °C) [26,34]. $R_f = 0.71$ (silica gel, CH₃OH). GC-MS: calcd. for C₉H₁₁NO (M⁺) 149.0841, found: 149.1. IR (KBr, cm⁻¹): 3292 (vNH); 3088, 3032 (vCH arom.); 1643 (vCO); 1552 (δHNC); 1446 (δHCH); 1328 (δHCC + vCC).

Antibacterial assay: The agar well diffusion method was employed to evaluate the antimicrobial activity of the synthesized amides 2a-k against four bacterial strains *viz. E. coli* (ATCC 8739), *P. aeruginosa* (ATCC 9027), *S. aureus* (ATCC 6538P) and *S. epidermidis* (ATCC 12228) [35-37]. Mueller-Hinton agar (MHA) plates were uniformly swabbed with standardized bacterial suspension (equivalent to McFarland 0.5)

prepared in Mueller-Hinton broth (MHB). Sample solutions of the amides 2a-k were prepared in DMSO with 50 mg/mL concentration each. Wells (6 mm in diameter) were bored on the media plates and then loaded $50\,\mu\text{L}$ of each sample solution in triplicates. Antibiotic gentamycin ($10\,\mu\text{g}/\text{disc}$) was used as a positive control, while DMSO was served as a negative control. After 24 h of incubation at 37 °C, the antibacterial effect was evaluated by measuring the ZOI appeared.

Next, the samples displaying antibacterial activity against the specific bacteria were further considered for the determination of minimum inhibition concentration (MIC) values by two-fold serial dilution technique, the Broth macrodilution method [38]. Briefly, the standardized bacterial suspension was diluted to 1:10 using normal saline $(1.5 \times 10^7 \text{ CFU/mL})$. A mixture of equal volumes of MHB (1 mL) and each sample solution (1 mL) was then serially diluted in another 14 sterile test tubes that contained 1 mL of MHB (total volume in each tube = 1 mL). The above bacterial suspension (50 μ L) was then inoculated (5×10^5 CFU/mL). Thereafter, the tubes were incubated at 37 °C for 24 h. The lowest concentration, at which no bacterial growth observed visually, was taken as the MIC. minimum bactericidal concentration (MBC) values were subsequently determined through sub-culture of the content from the tubes showing no bacterial growth by direct streaking

on sterile MHA plates. After incubation at 37 °C for 24 h, the plates were examined for the growth of bacteria and thus MBC values were determined.

RESULTS AND DISCUSSION

Based on the reported work [21], we developed to explore magnesium powder for the catalytic organic transformations. Now, we have investigated the N-acetylation of amines. At first, aniline (1a) was chosen as a model substrate. After a few trials, it was found that the reaction of 1a with 1.2 equiv. of Ac_2O , in the presence 2.5 mol% Mg, at 25 °C, was achieved to be completed within 2 min affording N-phenylacetamide (2a) in 93% yield (Table-1, entry 1). When the catalyst loading was reduced to 1 mol%, 2a was produced with the same yield but required a prolonged reaction time (i.e. 14 min). The acetylation of 1a can be achieved without expending catalyst producing 2a in 92% yield, after 14 min [25].

After obtaining an optimized reaction condition in hand, the scope of the reaction was studied. Amines bearing an electron donating group (OMe, Me or OH) either at *ortho-*, *meta-* or *para-*position were tolerated affording corresponding amides (**2b-f**) within 2-3 min, with excellent yields (Table-1, entries 2-6). In case of substrates **1e** and **1f** containing both the amino and hydroxyl groups, corresponding *N*-acetylation products

	Mg I	OWDER-CATALYZED N -AC $ \begin{array}{ccc} R & \longrightarrow NH_2 & + & Ac_2O \\ 1 & & (1.2 \text{ equi}) \end{array} $	2.5 mo	1% Mg 2-3 min	R—NHA 2			
Entry Substrate used		Product	Reaction	Isolated yield	Molecular ion peak	C=O	m.p. (°C)	
Linuy			time	(%)	(M ⁺)	stretching - (cm ⁻¹)	Found	Reported
1	NH ₂	NHAc	2 min	93	135.1	1666	114	112-114 [26,27]
2	MeO 1b	MeO 2b	2 min	87	165.1	1654	127-129	126-129 [26,29]
3	NH ₂	NHAc	2 min	95	149.1	1674	65-66	65-69 [26,27]
4	NH ₂	NHAc	2 min	95	149.1	1651	108-111	108-110 [26,27]
5	HO 1e	HO Ze NHAc	3 min	93	151.1	1656	170	168-169 [26,30]

862 Bajracharya et al. Asian J. Chem.

	OH NH ₂	OHNHAc						
6	ır	2f	3 min	88	151.1	1660	206	207-209 [26]
7	Br NH ₂	Br 2g	3 min	94	213.0	1672	164-166	166-170 [27,31,32]
8	$O_2N \qquad \qquad Ih$	O ₂ N NHAc	3 min	99	180.1	1679	214	210-214 [26,31]
9 10*	NO ₂ NH ₂	NO ₂ NHAc	16 h 2 h	92 99	180.1	1697	91	90-91 [33]
11	NH ₂	NHAc 2j	3 min	96	185.1	1654	160	159-160 [26]
12	NH ₂	NHAc 2k	3 min	96	149.1	1643	60	58-60 [26,34]

*Reaction temperature was 100 °C.

2e and **2f** were selectively formed over the *O*-acetylation products. *N*-Acetylation of the substrates with an electron withdrawing group (Br or NO₂) at *para*-position was also proceeded smoothly (entries 7-8). However, because of intervention of hydrogen bonding, the *N*-acetylation of 2-nitroaniline (**1i**), containing a nitro group at *ortho*-position, was found to be sluggish. The reaction required 16 and 2 h for the completion to afford *N*-(2-nitrophenyl)acetamide (**2i**) at 25 and 100 °C reaction temperature, respectively (entries 9 and 10). In optimized reaction conditions, *N*-(naphthalen-1-yl)acetamide (**2j**) was isolated in 96% yield by using 1-naphthylamine (**1j**) as substrate (entry 11). Similarly, an aliphatic amine, such as benzylamine (**1k**), was found equally facile producing **2k** in 96% isolated yield (entry 12). All the products were confirmed by recording of GC-MS spectrum, IR spectrum and melting point.

Biological activity: To determine the antibacterial activity of the synthesized amides (2a-k), wells on the bacteria inoculated MHA plates were loaded with the amide solution prepared in DMSO at 2.5 mg/well dose level. The zone of inhibition (ZOI) produced in the antibacterial assay was measured and the results are tabulated in Table-2. The growth of Gram-negative bacteria *E. coli* was inhibited by amides 2a-c, 2e, 2g and 2i

with the ZOI ranging from 10 to 16 mm (entries 1-3, 5, 7 and 9). Among the amides used, N-(m-tolyl)acetamide ($2\mathbf{c}$) has produced the highest ZOI (16.0 ± 0.58 mm) against E. coli (entry 3). N-(4-Methoxyphenyl)acetamide ($2\mathbf{b}$) has substantial antibacterial activity against all the four bacteria tested (entry 2). N-(2-Hydroxyphenyl)acetamide ($2\mathbf{f}$) was found effective against P. aeruginosa, S. aureus and S. epidermidis (entry 6). Among 11 amides tested, only compounds $2\mathbf{b}$ and $2\mathbf{f}$ were found effective against both the Gram-positive S. aureus and S. epidermidis (entries 2 and 6). A high ZOI (20.33 ± 0.33 mm) was exhibited by N-(2-nitrophenyl)acetamide ($2\mathbf{i}$) against S. aureus (entry 9). N-(o-Tolyl)acetamide ($2\mathbf{d}$) (entry 4), N-(4-nitrophenyl)acetamide ($2\mathbf{h}$) (entry 8), N-(naphthalen-1-yl)acetamide ($2\mathbf{j}$) (entry 10) and N-benzylacetamide ($2\mathbf{k}$) (entry 11) were found completely ineffective against the four bacterial strains tested.

The results of MIC and MBC values are given in Table-3. *N*-(2-Hydroxyphenyl)acetamide (**2f**) was found highly antibacterial against Gram-positive *S. aureus* and *S. epidermidis* with MIC values of 1.56 and 0.78 µg/mL, respectively (entry 6). It also displayed antibacterial effect against Gram-negative *P. aeruginosa*. Amides **2b** and **2e** have displayed bactericidal effect against *E. coli* (entries 2 and 5), while amides **2a**, **2c**, **2g**

TABLE-2 ANTIBACTERIAL ACTIVITY OF THE AMIDES (2a-k)							
	ZOI against the bacterial strains ± standard error mean (mm)						
Amides used	Gram-nega	tive bacteria	Gram-positive bacteria				
	E. coli	P. aeruginosa	S. aureus	S. epidermidis			
N-Phenylacetamide (2a)	13.33 ± 0.88	12.0 ± 1.0	-	-			
<i>N</i> -(4-Methoxyphenyl)acetamide (2b)	12.67 ± 0.88	8.67 ± 0.33	16.33 ± 1.20	8.67 ± 0.33			
N-(m-Tolyl)acetamide (2c)	16.00 ± 0.58	-	-	_			
N-(o-Tolyl)acetamide (2d)	_	-	-	_			
<i>N</i> -(4-Hydroxyphenyl)acetamide (2e)	11.67 ± 0.33	9.33 ± 0.33	-	_			
N-(2-Hydroxyphenyl)acetamide (2f)	_	12.66 ± 0.67	12.67 ± 0.33	11.00 ± 0.58			
N-(4-Bromophenyl)acetamide (2g)	10.00 ± 0	-	-	_			
N-(4-Nitrophenyl)acetamide (2h)	_	-	-	_			
N-(2-Nitrophenyl)acetamide (2i)	14.66 ± 0.33	-	20.33 ± 0.33	_			
N-(Naphthalen-1-yl)acetamide (2j)	_	_	-	_			
N-Benzylacetamide (2k)	_	_	-	_			
Gentamycin	20	24	23	26			

(-)-Sign indicates no significant ZOI was observed.

TABLE-3 MIC AND MBC VALUES OF THE AMIDES (2a-k)							
	MIC (MBC) values (μg/mL)						
Amides used	Gram-nega	tive bacteria	Gram-positive bacteria				
	E. coli	P. aeruginosa	S. aureus	S. epidermidis			
N-Phenylacetamide (2a)	6.25 (12.50)	12.50 (12.50)	-	-			
<i>N</i> -(4-Methoxyphenyl)acetamide (2b)	12.50 (12.50)	12.50 (12.50)	25.00 (25.00)	25.00 (25.00)			
N-(m-Tolyl)acetamide (2c)	3.12 (6.25)	-	-	-			
N-(o-Tolyl)acetamide (2d)	-	-	_	_			
<i>N</i> -(4-Hydroxyphenyl)acetamide (2e)	6.25 (6.25)	25.00 (25.00)	_	_			
N-(2-Hydroxyphenyl)acetamide (2f)	-	12.50 (25.00)	1.56 (1.56)	0.78 (6.25)			
N-(4-Bromophenyl)acetamide (2g)	3.12 (6.25)	-	_	_			
N-(4-Nitrophenyl)acetamide (2h)	-	_	_	_			
N-(2-Nitrophenyl)acetamide (2i)	1.56 (12.50)	_	6.25 (25.00)	_			
N-(Naphthalen-1-yl)acetamide (2j)	-	_	_	_			
<i>N</i> -Benzylacetamide (2k)	_	_	_	_			

(-)-Sign = Value not determined as no significant ZOI was observed in the Agar well diffusion assay.

and **2i** were bacteriostatic (entries 1, 3, 7 and 9). Among these amides, N-(2-nitrophenyl)acetamide (**2i**) was found more efficient exhibiting MIC value of 1.56 μ g/mL against E. coli (entry 9). Amide **2i** has also displayed effective antibacterial activity by inhibiting S. aureus with the MIC value of 6.25 μ g/mL.

Antibacterial activity of compounds 2a and 2g against E. coli and S. aureus is reported by Jagessar & Rampersaud [39]. In contrast, this work showed that the amides 2a and 2g were effective against E. coli but ineffective against S. aureus. Antibacterial activity of N-(2-hydroxyphenyl)acetamide (2f) against P. aeruginosa and S. aureus is recently reported by Hifnawy et al. [40]. In some other reports, compound 2f is reported to be ineffective against E. coli, S. aureus, methicillin-resistant S. aureus, methicillin-resistant S. epidermidis, B. subtilis, Klebsiella pneumoniae and Enterococcus faecalis [40-44]. Compound **2f** has also displayed anti-inflammatory [14,45-47] and antitumor [48] activities. It also possesses strong apoptotic activity and inhibits the growth of U87 in a dose dependent manner [49]. The antibacterial activity of N-(4-hydroxyphenyl)acetamide (2e) against E. coli has been reported by Jayadevappa et al. as well [2].

Conclusion

In conclusion, a convenient and green approach for the *N*-acetylation of amines with acetic anhydride in the presence of magnesium powder is presented. This reaction was promoted rapidly with an excellent yield at ambient temperature. Simple work up procedure could provide pure amide products and no column chromatography was needed. Nowadays, identification of new potent antibacterial agents is an immense importance task in order to combat antibacterial resistance. This study reveals that simple and readily available *N*-(2-hydroxyphenyl)acetamide (2f) could be an effective antibacterial agent which appeals for further investigations.

ACKNOWLEDGEMENTS

One of the authors, N.B. thanks Nepal Academy of Science and Technology (NAST) for providing the research assistant-ship. The authors thanks Mr. Mohan Paudel for accessing Scifinder database and Analytical Service Centre, NAST for recording of the spectra.

864 Bajracharya et al. Asian J. Chem.

CONFLICT OF INTEREST

The authors declare that there is no conflict of interests regarding the publication of this article.

REFERENCES

- T. Nomura, T. Iwaki, Y. Narukawa, K. Uotani, T. Hori and H. Miwa, Bioorg. Med. Chem., 14, 3697 (2017); https://doi.org/10.1016/j.bmc.2006.01.036
- A.S. Sheikh, H. Nadeem, M.T. Khan, A. Saeed and B. Murtaza, ACS Omega, 8, 9785 (2023); https://doi.org/10.1021/acsomega.2c05782
- M.D. Altintop, U. Abu-Mohsen, Y. Okay, R. Demirel and Z.A. Kaplancikli, *Turk. J. Pharm. Sci.*, 12, 29 (2015).
- H. Lu, X. Zhou, L. Wang and L. Jin, *Molecules*, 25, 1772 (2020); https://doi.org/10.3390/molecules25081772
- M.M. Ghorab, A.S. Alqahtani, A.M. Soliman and A.A. Askar, *Int. J. Nanomed.*, 15, 3161 (2020); https://doi.org/10.2147/IJN.S241433
- W. Ang, Y.N. Lin, T. Yang, J.Z. Yang, W.Y. Pi, Y.H. Yang, Y.F. Luo, Y. Deng and Y.Q. Wei, *Molecules*, 17, 2248 (2012); https://doi.org/10.3390/molecules17022248
- B. Tanwar, A. Kumar, P. Yogeeswari, D. Sriram and A.K. Chakraborti, *Bioorg. Med. Chem. Lett.*, 26, 5960 (2016); https://doi.org/10.1016/j.bmcl.2016.10.082
- R. Zogota, L. Kinena, C. Withers-Martinez, M.J. Blackman, R. Bobrovs, T. Pantelejevs, I. Kanepe-Lapsa, V. Ozola, K. Jaudzems, E. Suna and A. Jirgensons, Eur. J. Med. Chem., 163, 344 (2019); https://doi.org/10.1016/j.ejmech.2018.11.068
- S.K. Tipparaju, S.P. Muench, E.J. Mui, S.N. Ruzheinikov, J.Z. Lu, S.L. Hutson, M.J. Kirisits, S.T. Prigge, C.W. Roberts, F.L. Henriquez, A.P. Kozikowski, D.W. Rice and R.L. McLeod, J. Med. Chem., 53, 6287 (2010); https://doi.org/10.1021/jm9017724
- 10. A.P.G. Nikalje, S. Choudhari and H. Une, Eur. J. Exp. Biol., 2, 1302 (2012).
- Md..J Naim, Md.J. Alam, F. Nawaz, V.G.M. Naidu, S. Aaghaz, M. Sahu, N. Siddiqui and O. Alam, *Bioorg. Chem.*, 73, 24 (2017); https://doi.org/10.1016/j.bioorg.2017.05.007
- S. Kumar, A.K. Wahi and R. Singh, Trop. J. Pharm. Res., 10, 265 (2011); http://dx.doi.org/10.4314/tjpr.v10i3.6
- S. Ölgen, F. Bakar, S. Aydin, D. Nebioglu and S. Nebioglu, *J. Enzyme Inhib. Med. Chem.*, 28, 58 (2013); https://doi.org/10.3109/14756366.2011.631183
- H. Jawed, S.U.A. Shah, S. Jamall and S.U. Simjee, *Int. Immunopharmacol.*, 10, 900 (2010); https://doi.org/10.1016/j.intimp.2010.04.028
- P. Rani, D. Pal, R.R. Hegde and S.R. Hashim, *J. Chemother.*, 28, 255 (2016); https://doi.org/10.1179/1973947815Y.0000000060
- A.K. Ghose, V.N. Viswanadhan and J.J. Wendoloski, *J. Comb. Chem.*, 1, 55 (1999); https://doi.org/10.1021/cc9800071
- J.S. Carey, D. Laffan, C. Thomson and M.T. Williams, *Org. Biomol. Chem.*, 4, 2337 (2006); https://doi.org/10.1039/B602413K
- R. Wolf and S. Brenner, *Dermatology*, **189**, 1 (1994); https://doi.org/10.1159/000246749
- 19. P. Rajput and A. Sharma, J. Pharmacol. Med. Chem., 2, 22 (2018).
- A. Kokel, C. Schäfer and B. Török, Curr. Org. Synth., 16, 615 (2019); https://doi.org/10.2174/1570179416666190206141028
- G.B. Bajracharya and S.S. Shrestha, Synth. Commun., 48, 1688 (2018); https://doi.org/10.1080/00397911.2018.1459721
- D.H. Kim, J. Heterocycl. Chem., 13, 179 (1976); https://doi.org/10.1002/jhet.5570130201
- T.S. Ibrahim, I.A. Seliem, S.S. Panda, A.M.M. Al-Mahmoudy, Z.K.M. Abdel-Samii, N.A. Alhakamy, H.Z. Asfour and M. Elagawany, *Molecules*, 25, 2501 (2020); https://doi.org/10.3390/molecules25112501
- A. Ami'c and M. Molnar, Org. Prep. Proc. Int., 49, 249 (2017); https://doi.org/10.1080/00304948.2017.1320914

- N. Anbu, N. Nagarjun, M. Jacob, J.M.V.K. Kalaiarasi and A. Dhakshinamoorthy, *Chemistry*, 1, 69 (2019); https://doi.org/10.3390/chemistry1010006
- M.T. Sangole, S.M. Thorat, S.P. Deshmukh, M. Muthukrishnan, S. Shirsath and M. Mujahid, *Indian J. Chem.*, 58B, 1125 (2019).
- S.A. Trujillo, D. Peña-Solórzano, O.R. Bejarano and C. Ochoa-Puentes, RSC Adv., 10, 40552 (2020); https://doi.org/10.1039/D0RA06871C
- Y. Umar, A. Abu-Thabit, P. Jerabek and P. Ramasami, *J. Theor. Comput. Chem.*, 18, 1950009 (2019); https://doi.org/10.1142/S0219633619500093
- J. Yin, J. Zhang, C. Cai, G.J. Deng and H. Gong, Org. Lett., 21, 387 (2019); https://doi.org/10.1021/acs.orglett.8b03542
- G.R. Lloyd, D.Q.M. Craig and A. Smith, J. Pharm. Sci., 86, 991 (1997); https://doi.org/10.1021/js970137w
- J.L. Jat, P. Kumar, S. Verma, D. Chandra, V. Singh and B. Tiwari, *New J. Chem.*, 46, 14782 (2022); https://doi.org/10.1039/D2NJ02755K
- P. Lian, R. Li, X. Wan, Z. Xiang, H. Liu, Z. Cao and X. Wan, *Org. Chem. Front.*, 9, 311 (2022); https://doi.org/10.1039/D1QO01613J
- E. Kianmehr and S.B. Nasab, Eur. J. Org. Chem., 6447 (2018); https://doi.org/10.1002/ejoc.201800779
- A.K. Chakraborti and R. Gulhane, Chem. Commun., 1896 (2003); https://doi.org/10.1039/B304178F
- 35. C. Perez, M. Pauli and P. Bazerque, Acta Biol. Med. Exp., 15, 113 (1990).
- R.K. Gupta, G.M.S. Thakuri, G.B. Bajracharya and R.N. Jha, Bibechana, 18, 143 (2021); https://doi.org/10.3126/bibechana.v18i2.31234
- C.M. Nemkul, G.B. Bajracharya and I. Shrestha, J. Plant Resour., 19, 151 (2021).
- 38. Methods for Dilution Antimicrobial Susceptibility Tests for Bacteria that Grow Aerobically, Approved Standard, 9th Edn, CLSI Document M07-A9, vol. 32, No. 2, Clinical and Laboratory Standards Institute, Wayne, PA (2012).
- 39. R. Jagessar and D. Rampersaud, New York Sci. J., 1, 22 (2008).
- M.S. Hifnaway, H.M. Hassan, R. Mohammed, M.M. Fouda, A.M. Sayed, A.A. Hamed, S.F. AbouZid, M.E. Rateb, H.A. Alhadrami and U.R. Abdelmohsen, *Mar. Drugs*, 18, 243 (2020); https://doi.org/10.3390/md18050243
- Z. Shang, X. Li, L. Meng, C. Li, S. Gao, C. Huang and B. Wang, *Chin. J. Ocean Limnol.*, 30, 305 (2012); https://doi.org/10.1007/s00343-012-1075-1
- Y. Dashti, T. Grkovic, U.R. Abdelmohsen, U. Hentschel and R.J. Quinn, *Mar. Drugs*, 12, 3046 (2014); https://doi.org/10.3390/md12053046
- J. Wang, W. He, X. Qin, X. Wei, X. Tian, L. Liao, S. Liao, B. Yang, Z. Tu, B. Chen, F. Wang, X. Zhou and Y. Liu, RSC Adv., 5, 68736 (2015); https://doi.org/10.1039/C5RA10828D
- B. Peng, Q. Peng, J. She, B. Yang and X. Zhou, *Rec. Nat. Prod.*, 16, 639 (2022); http://doi.org/10.25135/rnp.312.22.01.2314
- K. Perveen, F. Hanif, H. Jawed and S.U. Simjee, *BioMed Res. Int.*, 635143 (2013); https://doi.org/10.1155/2013/635143
- 46. A. Gul, B. Kunwar, M. Mazhar, K. Perveen, S.U. Simjee, *Inflammation*, **40**, 1177 (2017);
 - https://doi.org/10.1007/s10753-017-0561-1
 A. Arif, S. Majeed, S.U. Simjee, M. Sarfaraz, S.S. Kashif, M. Lodhi, N. Khatian, A. Aziz and H.K. Majeed, *Pak-Euro J. Med. Life Sci.*, 5,
 - 309 (2022); https://doi.org/10.31580/pjmLs.v5i2.2528
- 48. A. Aziz, F. Hanif, S. Majeed, K. Iftikhar and S.U. Simjee, *Toxicol. in Vitro*, **60**, 296 (2019);
- https://doi.org/10.1016/j.tiv.2019.06.011
 F. Hanif, K. Perveen, H. Jawed, A. Ahmed, S.M. Malhi, S. Jamall and S.U. Simjee, *Cancer Cell Int.*, 14, 133 (2014);
 https://doi.org/10.1186/s12935-014-0133-5