

Synthesis and Crystal Structure of *N*-[3,5-Bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl]-*N*-ethylnitrous amide and *N*-[3,5-Bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl]-*N*-isopropyl nitrous amide

JINSEN CHEN, PEI CHEN, HUAXIN YAO and JIN ZHU*

Department of Polymer Science and Engineering, School of Chemistry and Chemical Engineering, State Key Laboratory of Coordination Chemistry, Nanjing National Laboratory of Microstructures, Nanjing University, Nanjing 210093, P.R. China

*Corresponding author: Fax: +86 25 83317761; Tel: +86 25 83686291; E-mail: jinz@nju.edu.cn, chenjinsen.mikechen@gmail.com

Received: 28 June 2014;

Accepted: 16 September 2014;

Published online: 27 April 2015;

AJC-17177

The title compound *N*-[3,5-bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl]-*N*-ethylnitrous amide (**2**, m.f. C₂₀H₃₂N₂O₅B₂) and *N*-[3,5-bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl]-*N*-isopropyl nitrous amide (**4**, m.f. C₂₁H₃₄N₂O₅B₂) have been synthesized and characterized by elemental analysis, MS, IR, ¹H, ¹³C, ¹¹B NMR spectroscopies. In addition, their crystal structures were determined by single-crystal X-ray diffraction. The crystal of **2** belongs to monoclinic, space group P2₁/n with a = 6.3220(13), b = 28.910(6), c = 12.827(3) Å, β = 102.61(3)°, Z = 4, V = 2287.8(8) Å³, D_c = 1.167 g/cm³, μ(MoKα) = 0.081 mm⁻¹, F(000) = 864, the final R = 0.0737 and wR = 0.1548 for 4209 observed reflections with I > 2σ(I), while the compound **4** crystallizes in triclinic, space group P1̄ with a = 6.8730(14), b = 10.827(2), c = 17.000(3) Å, α = 101.30(3), β = 92.07(3), γ = 92.11(3)°, Z = 2, V = 1238.5(4) Å³, D_c = 1.116 g/cm³, μ(MoKα) = 0.077 mm⁻¹, F(000) = 448, the final R = 0.0934 and wR = 0.1628 for 4546 observed reflections with I > 2σ(I).

Keywords: Synthesis, Crystal structure, Nitroso group.

INTRODUCTION

Transition metal catalyzed C-H borylation reaction has received significant attention in recent years for the synthetic utility of boronate esters and this method can avoid the need of pre-installed of halogen which make the synthetic route more straight forward¹⁻⁴. While the high selectivity of arene C-H borylation can be obtained by appropriate choice of 1,2- or 1,3-disubstituted arene substrate, a mixture of borylation products is often obtained with iridium catalyst when using mono-substituted arene⁵⁻⁷. We recently launched a program on C-H functionalization with the *N*-nitroso as the directing group^{8,9}. It is thus anticipated that the *N*-nitroso can act as a steric tool to synthesized *meta*-pinacol boronate esters through Ir-catalyzed C-H borylation. Herein we report the synthesis and crystal structure of *N*-[3,5-bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl]-*N*-ethylnitrous amide (**2**) and *N*-[3,5-bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl]-*N*-isopropyl nitrous amide (**4**) using [Ir(μ-OMe)(COD)]₂ (COD = cyclooctadiene) with 4,4'-di-*tert*-butylbipyridine (dtbpy) as the ligand.

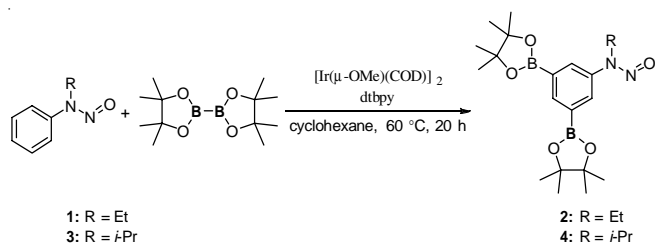
EXPERIMENTAL

[Ir(μ-OMe)(COD)]₂, bis(pinacolato) diboron, 4,4'-di-*tert*-butyl bipyridine were purchased and employed without

purification. Cyclohexane was distilled over Na/benzophenone before use. ¹H and ¹³C NMR spectra were recorded in CDCl₃ (with tetramethylsilane as an internal reference) solution on a Bruker AVANCE 400 MHz. The carbon directly attached to the boron atom was not seen by ¹³C NMR spectroscopy¹⁰. ¹¹B NMR spectra were referenced to an external BF₃·Et₂O sample in CDCl₃ (0.0 ppm). High-resolution MS (HRMS) spectra were obtained on a Thermo Fisher Scientific LTQ FT Ultra facility. Infrared (IR) spectra were recorded on a Nicolet 6700 spectrophotometer and are reported as wave number (cm⁻¹). Arene substrate **1** and **3** were prepared according to previous manuscripts^{8,9}.

Synthetic procedure: The synthesis route of title compounds **2** and **4** is outlined in **Scheme-I**. [Ir(μ-OMe)(COD)]₂ (8 mg, 0.012 mmol), dtbpy (6.4 mg, 0.024 mmol), B₂pin₂ (203 mg, 0.8 mmol)¹¹, arene substrate (0.4 mmol) and cyclohexane (2 mL) were added into a dry vial with a magnetic stir bar in a glove box. The vial was sealed and the reaction mixture was stirred at 60 °C for 20 h. Then the volatiles were removed *in vacuo* and the residue was subjected to column chromatography provided the corresponding boronate ester.

***N*-[3,5-bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl]-*N*-ethylnitrous amide (**2**):** The title compound was obtained as pale white solid in 76 % yield. R_f = 0.2

Scheme-1: Synthesis of compound **2** and **4**

(Et₂O:Hexane, 1:5). IR (neat) $\nu = 2979, 1603, 1410, 1319, 1261, 1139, 1113, 859, 707 \text{ cm}^{-1}$. ¹H NMR (400 MHz, CDCl₃) δ 8.26 (s, 1H), 8.00 (d, $J = 0.9 \text{ Hz}$, 2H), 4.11 (q, $J = 7.2 \text{ Hz}$, 2H), 1.35 (s, 24H), 1.16 (t, $J = 7.2 \text{ Hz}$, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 140.44, 140.24, 129.01, 84.16, 39.38, 24.88, 11.74. ¹¹B NMR (128 MHz, CDCl₃) δ 31.08. HRMS (DART) Calcd. for C₂₀H₃₃N₂O₅B₂: [M + H]⁺, 401.2643; found: m/z 401.2638.

N-[3,5-bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl]-N-isopropyl nitrous amide (4): The title compound was obtained as a pale white solid in 78 % yield of inseparable mixture of *syn* and *anti* isomers in 78 % yield^{8,9}. By ¹H NMR, the *syn:anti* ratio was determined to be approximately 1:0.66. R_f = 0.18 (Et₂O:Hexane, 1:5). IR (neat) $\nu = 2973, 1601, 1382, 1353, 1317, 1139, 1111, 848, 715 \text{ cm}^{-1}$. ¹H NMR (400 MHz, CDCl₃) (*syn* and *anti* isomers) δ 8.32 (d, $J = 9.1 \text{ Hz}$, 1H), 8.31 (s, 1H \times 0.66), 7.84 (d, $J = 1.0 \text{ Hz}$, 2H), 7.46 (d, $J = 1.0 \text{ Hz}$, 2H \times 0.66), 5.18 (hept, $J = 6.8 \text{ Hz}$, 1H), 4.97 (hept, $J = 6.8 \text{ Hz}$, 1H \times 0.66), 1.51 (d, $J = 6.8 \text{ Hz}$, 3H \times 0.66), 1.35 (s, 24H), 1.32 (s, 24H \times 0.66), 1.22 (d, $J = 6.9 \text{ Hz}$, 6H). ¹³C NMR (101 MHz, CDCl₃) (*syn* and *anti* isomers) δ 142.05, 141.48, 138.68, 136.28, 136.04, 134.73, 84.15, 84.08, 56.33, 46.69, 24.87, 22.23, 19.85. ¹¹B NMR (128 MHz, CDCl₃) δ 31.10. HRMS (DART) Calcd. for C₂₁H₃₅N₂O₅B₂: [M + H]⁺, 415.2799; found: m/z 415.2794.

Crystal structure determination: Crystal of the title compound **2** for X-ray structure determination was obtained by slow evaporation in ether/hexane solution at 0 °C. A pale yellow single crystal with dimensions of 0.30 mm \times 0.20 mm \times 0.10 mm was selected for X-ray diffraction analysis at 293(2) K. The data was collected on CAD4/PC with MoK α radiation ($\lambda = 0.71073 \text{ \AA}$). A total of 4602 reflections were collected in the range of $1.41 < \theta < 25.38^\circ$ at room temperature and 4209 independent reflections ($R_{\text{int}} = 0.0668$) with $I > 2\sigma(I)$ were used in the structure determination and refinements. The structure was solved by direct methods with SHELXS-97 program and refined on F² with SHELXL-97 program. The non-hydrogen atoms were refined anisotropically and the hydrogen atom were determined with theoretical calculation. A full-matrix least-squares refinement gave final $R = 0.0737$, $wR = 0.1548$ ($w = 1/[\sigma^2(F_o^2) + (0.0700P)^2 + 0.0000P]$, where $P = (F_o^2 + 2F_c^2)/3$), $R = 0.1690$ and $wR = 0.1914$ for all data. The largest parameter shift $(\Delta/\sigma)_{\text{max}}$ is 0.000. The goodness-of-fit indicator is 1.001. the maximum and minimum peaks in the final difference Fourier map are 0.353 and -0.222 e/\AA^3 , respectively. All calculation were performed on a PC with SHELXTL program¹².

Crystal of the title compound **4** for X-ray structure determination was obtained by slow evaporation in ether/hexane solution at 0 °C. A colourless single crystal with dimensions

of 0.20 mm \times 0.10 mm \times 0.10 mm was selected for X-ray diffraction analysis at 293(2) K. The data was collected on CAD4/PC with MoK α radiation ($\lambda = 0.71073 \text{ \AA}$). A total of 4956 reflections were collected in the range of $1.22 < \theta < 25.37^\circ$ at room temperature and 4546 independent reflections ($R_{\text{int}} = 0.0478$) with $I > 2\sigma(I)$ were used in the structure determination and refinements. The structure was solved by direct methods with SHELXS-97 program and refined on F² with SHELXL-97 program. The non-hydrogen atoms were refined anisotropically and the hydrogen atoms were determined with theoretical calculation. A full-matrix least-squares refinement gave final $R = 0.0934$, $wR = 0.1628$ ($w = 1/[\sigma^2(F_o^2) + (0.0400P)^2 + 0.9000P]$, where $P = (F_o^2 + 2F_c^2)/3$), $R = 0.1998$ and $wR = 0.1999$ for all data. The largest parameter shift $(\Delta/\sigma)_{\text{max}}$ is 0.000. The goodness-of-fit indicator is 1.005. The maximum and minimum peaks in the final difference Fourier map are 0.231 and -0.216 e/\AA^3 , respectively. All calculation was performed on a PC with SHELXTL program¹².

RESULTS AND DISCUSSION

The single crystal of the title compounds (**2** and **4**) were obtained by slow evaporation in ether/hexane solution. The HRMS, IR and NMR spectra are in good agreement with the formulae proposed by the X-ray crystallography.

The view of the crystal structure of the title compound **2** and **4** are shown in Figs. 1 and 2, respectively. And the packing diagrams for them were shown in Figs. 3 and 4. The selected bond lengths and bond angles are listed in Tables 1 and 2, respectively. In the structure of **2**, B(1), O(1), C(7), C(8) and O(2) atoms form a five-membered ring, as well as B(2), O(3), C(13), C(14) and O(4). The bond length of B(1)-C(6) is 1.564(5) Å, close to the typical B-C_{aryl} bond (1.556 Å)¹³. The bond length of N(2)=O(5) in **2** is 1.237(4) Å and N(2)=O(5) in **4** is 1.213(5) Å, which consistent with the bond length of double bond of N=O (1.218 Å). However, the bond length of N(1)-N(2) in **2** is 1.306(4) Å and N(1)-N(2) in **4** is 1.379(6) Å, both of which are shorter than the usual N-N bond (1.425 Å) and longer than N=N bond (1.240 Å)¹³. It is because of the partial double bond character of the bonds due to partial electron delocalization of N(2)=O(5) bond.

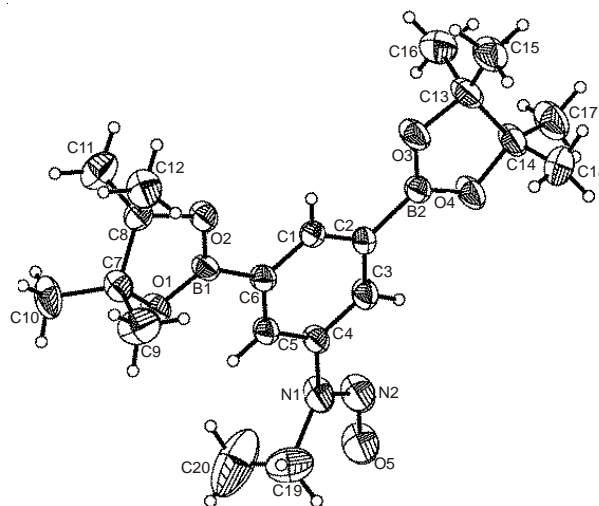
Fig. 1. Molecular structure of the title compound **2**

TABLE-1
SELECTED BOND LENGTHS (Å) AND BOND ANGLES (°) FOR COMPOUND 2

Bond	Dist.	Bond	Dist.	Bond	Dist.
O(1)-B(1)	1.350(4)	N(1)-C(4)	1.423(4)	B(2)-O(3)	1.348(4)
B(1)-C(6)	1.564(5)	N(1)-C(19)	1.688(6)	C(1)-C(2)	1.398(4)
N(1)-N(2)	1.306(4)	N(2)-O(5)	1.237(4)	-	-
Angle	(°)	Angle	(°)	Angle	(°)
N(2)-N(1)-C(4)	117.7(3)	O(2)-B(1)-C(6)	121.1(3)	C(10)-C(7)-C(8)	115.4(3)
N(2)-N(1)-C(19)	119.9(3)	B(1)-O(2)-C(8)	106.3(3)	O(2)-C(8)-C(11)	106.6(3)
C(4)-N(1)-C(19)	120.0(3)	B(1)-O(1)-C(7)	106.7(3)	O(2)-C(8)-C(12)	107.4(3)
O(5)-N(2)-N(1)	115.0(4)	O(1)-C(7)-C(10)	108.2(3)	C(11)-C(8)-C(12)	109.6(3)
C(20)-C(19)-N(1)	95.9(6)	O(1)-C(7)-C(9)	106.7(3)	O(2)-C(8)-C(7)	101.8(3)
O(1)-B(1)-O(2)	114.5(3)	O(1)-C(7)-C(8)	110.2(3)	C(11)-C(8)-C(7)	113.7(3)
O(1)-B(1)-C(6)	124.4(3)	O(1)-C(7)-C(8)	103.0(3)	C(12)-C(8)-C(7)	116.8(3)

Symmetry transformation: a: $x, y + 1, z$; b: $-x, y, -z + 1/2$; c: $-x, -y + 1, -z$

TABLE-2
SELECTED BOND LENGTHS (Å) AND BOND ANGLES (°) FOR COMPOUND 4

Bond	Dist.	Bond	Dist.	Bond	Dist.
O(1)-B(1)	1.370(5)	N(1)-C(13)	1.471(6)	B(2)-O(3)	1.357(6)
B(1)-C(16)	1.543(6)	N(1)-C(20)	1.453(5)	C(16)-C(17)	1.396(5)
N(1)-N(2)	1.379(6)	N(2)-O(5)	1.213(5)	-	-
Angle	(°)	Angle	(°)	Angle	(°)
N(2)-N(1)-C(20)	110.5(4)	O(2)-B(1)-O(1)	112.5(4)	O(1)-C(1)-C(3)	102.9(4)
N(2)-N(1)-C(13)	127.4(4)	O(2)-B(1)-C(16)	121.7(4)	C(4)-C(1)-C(3)	108.1(5)
C(20)-N(1)-C(13)	121.5(4)	O(1)-B(1)-C(16)	125.8(4)	C(2)-C(1)-C(3)	111.9(5)
O(5)-N(2)-N(1)	109.4(6)	B(1)-O(1)-C(1)	107.9(3)	O(2)-C(2)-C(5)	108.5(4)
N(1)-C(13)-C(15)	112.8(5)	B(1)-O(2)-C(2)	107.5(4)	O(2)-C(2)-C(1)	105.1(4)
N(1)-C(13)-C(14)	108.6(5)	O(1)-C(1)-C(4)	110.5(4)	C(5)-C(2)-C(1)	118.1(5)
C(15)-C(13)-C(14)	114.3(6)	O(1)-C(1)-C(2)	103.0(3)	O(2)-C(2)-C(6)	103.2(4)
O(2)-B(1)-O(1)	112.5(4)	C(4)-C(1)-C(2)	119.2(5)	C(5)-C(2)-C(6)	109.3(5)

Symmetry transformation: a: $x, y + 1, z$; b: $-x, y, -z + 1/2$; c: $-x, -y + 1, -z$

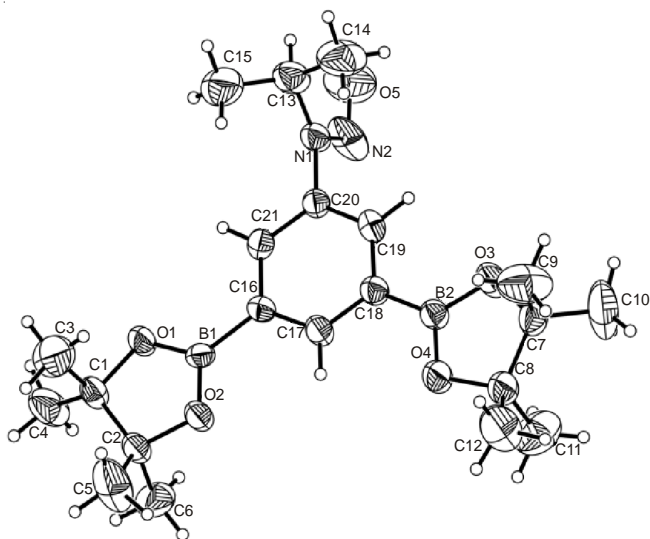


Fig. 2. Molecular structure of the title compound 4

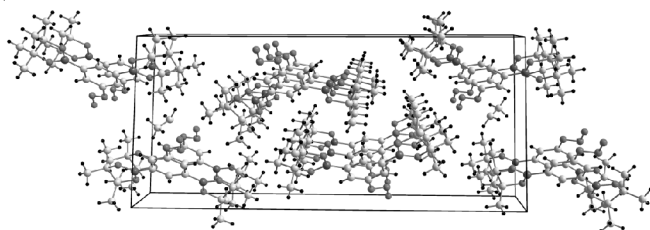


Fig. 3. Packing structure of the title compound 2

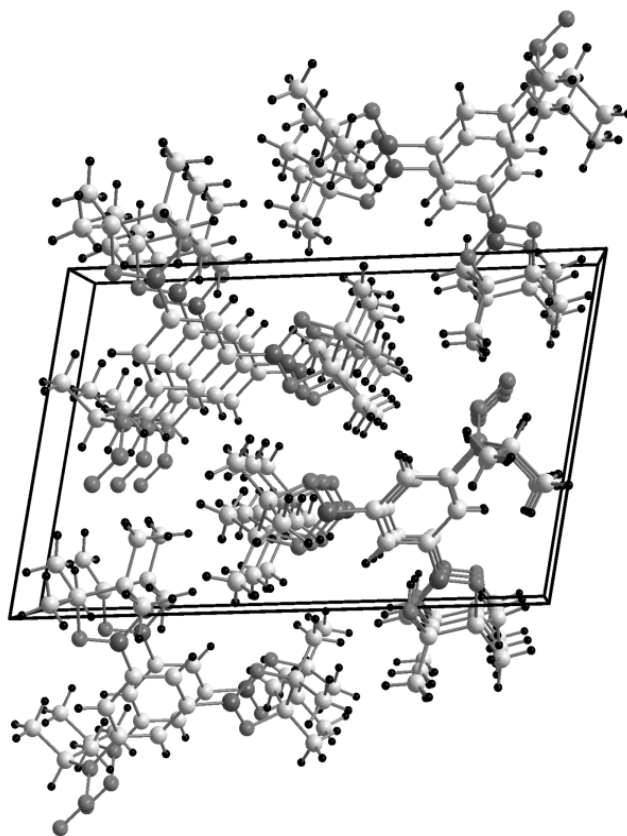


Fig. 4. Packing structure of the title compound 4

Conclusion

In summary, *N*-[3,5-*bis*(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl) phenyl]-*N*-ethylnitrous amide (**2**) and *N*-[3,5-*bis*(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-phenyl]-*N*-isopropyl nitrous amide (**4**) have been synthesized and fully characterized by elemental analysis, Mass, IR, ¹H, ¹³C, ¹¹B NMR spectroscopies. In addition, their crystal structure were determined by single-crystal X-ray diffraction. Particularly, compared to the traditional way, our synthesis route is found to be simple and efficient.

Supplementary data: CCDC-1006980, CCDC-1009207 and CCDC-1006981 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge *via* <http://www.ccdc.cam.ac.uk/conts/retrieving.html>, or from the Cambridge Crystallographic Data Centre, 12 Union Road Cambridge CB2 1EZ, UK (Fax: +44-1223-336033; or E-mail: deposit@ccdc.cam.ac.uk).

REFERENCES

1. N. Miyaura and A. Suzuki, *Chem. Rev.*, **95**, 2457 (1995).
2. I.A.I. Mkhaliid, J.H. Barnard, T.B. Marder, J.M. Murphy and J.F. Hartwig, *Chem. Rev.*, **110**, 890 (2010).
3. J.F. Hartwig, *Chem. Soc. Rev.*, **40**, 1992 (2011).
4. J.F. Hartwig, *Acc. Chem. Res.*, **45**, 864 (2012).
5. J.-Y. Cho, M.K. Tse, D. Holmes, R.E. Maleczka Jr. and M.R. Smith III, *Science*, **295**, 305 (2002).
6. G.A. Chotana, M.A. Rak and M.R. Smith III, *J. Am. Chem. Soc.*, **127**, 10539 (2005).
7. T. Ishiyama and N. Miyaura, *Pure Appl. Chem.*, **78**, 1369 (2006).
8. B. Liu, Y. Fan, Y. Gao, C. Sun, C. Xu and J. Zhu, *J. Am. Chem. Soc.*, **135**, 468 (2013).
9. B. Liu, C. Song, C. Sun, S. Zhou and J. Zhu, *J. Am. Chem. Soc.*, **135**, 16625 (2013).
10. S. Kawamorita, T. Miyazaki, H. Ohmiya, T. Iwai and M. Sawamura, *J. Am. Chem. Soc.*, **133**, 19310 (2011).
11. When using 1 equiv. of B₂pin₂, *para*-Monosubstituted Product was Obtained in 40-50 % Besides the Disubstituted Products whose Structure was Confirmed by X-ray Diffraction Analysis (CCDC 1006981 for Substrate 3 and Monosubstitute Product of **1** is oil).
12. G.H. Sheldrick, SHELXS-97 and SHELXL-97. Program for Solution of Crystal Structures. University of Gottingen, Germany (1997).
13. F.H. Allen, O. Kennard, D.G. Watson, L. Brammer, A.G. Orpen and R. Taylor, *J. Chem. Soc., Perkin Trans. II*, S1 (1987).