



NOTE

Synthesis and Crystal Structure of $\text{Dy}(\text{H}_2\text{O})_9(1,4\text{-dimethylbenzene})_2\text{Cl}_4\cdot\text{H}_3\text{O}$

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One-pot solvothermal reaction of $\text{DyCl}_3\cdot 6\text{H}_2\text{O}$ with 1,4-dimethylbenzene and water generates $\text{Dy}(\text{H}_2\text{O})_9(1,4\text{-dimethylbenzene})_2\text{Cl}_4\cdot\text{H}_3\text{O}$ that is characterized by single-crystal X-ray diffraction. The compound crystallizes in monoclinic, space group $P2_1/m$ with $a = 7.8378(10)$, $b = 25.103(3)$, $c = 7.9068(10)$ Å, $\beta = 119.3850(10)$, $V = 1355.5(3)$ Å³, $\text{C}_{16}\text{H}_{41}\text{Cl}_4\text{DyO}_{10}$, $M_r = 697.79$, $D_c = 1.710$ g/cm³, $\mu(\text{MoK}\alpha) = 3.194$ mm⁻¹, $F(000) = 702$, $Z = 2$, the final $R = 0.0503$ and $wR = 0.1323$ for 2157 observed reflections ($I > 2\sigma(I)$). The 2-D supramolecular layer is built up from the compound by hydrogen bonds.

Keywords: Chloride, Crystal, Dimethylbenzene, Dysprosium, Solvothermal.

1,4-Dimethylbenzene (namely, *p*-xylene) is a derivative of benzene and an important raw material for chemical industries¹⁻³. As one of the basic chemical products, the *p*-xylene has become the essential element in people's life. Hydrogen bonds have recently been applied in the crystal engineering of metal-organic coordination supramolecular compounds^{4,5}. Solvothermal reactions have been employed to prepare single crystals due to the obvious advantages^{6,7}. In this paper, we reported the preparation and crystal structure of a dimethylbenzene compound $\text{Dy}(\text{H}_2\text{O})_9(1,4\text{-dimethylbenzene})_2\text{Cl}_4\cdot\text{H}_3\text{O}$. The molecules are linked by hydrogen bonds to form 2-D supramolecular layer.

Synthesis of $\text{Dy}(\text{H}_2\text{O})_9(1,4\text{-dimethylbenzene})_2\text{Cl}_4\cdot\text{H}_3\text{O}$:

A mixture of $\text{DyCl}_3\cdot 6\text{H}_2\text{O}$ (1 mmol, 375 mg), 1,4-dimethylbenzene 3 mL and H_2O 9 mL was sealed in a 25 mL Teflon-lined stainless steel reactor and heated to 175 °C for 5 days and then cooled to room temperature spontaneously. The crystals are washed with H_2O to give the title compound in about 25 % yield (based on $\text{DyCl}_3\cdot 6\text{H}_2\text{O}$). The crystallographic data is given in Table-1.

X-ray structural determination: The X-ray diffraction data was collected at 296(2) K on a CCD area detector equipped with a graphite-monochromatized $\text{MoK}\alpha$ radiation. Data processing was conducted with a SAINT program. The

TABLE-1
CRYSTALLOGRAPHIC DATA AND STRUCTURE REFINEMENT SUMMARY

Empirical formula	$\text{C}_{16}\text{H}_{41}\text{O}_{10}\text{DyCl}_4$	Temperature (K)	296(2)
Formula weight	697.79	Radiation ($\text{MoK}\alpha$), $\lambda/\text{Å}$	0.71073
Crystal size/mm ³	$0.11 \times 0.09 \times 0.06$	μ/mm^{-1}	3.194
Crystal system	Monoclinic	$D_c/(\text{g}/\text{cm}^3)$	1.710
Space group	$P2_1/m$	$F(000)$	702
$a/\text{Å}$	7.8378(10)	Index ranges (h, k, l)	-9/8, -29/29, -9/9
$b/\text{Å}$	25.103(3)	Reflections collected/unique	7498/2234 ($R_{\text{int}} = 0.0216$)
$c/\text{Å}$	7.9068(10)	Observed reflections $I > 2\sigma(I)$	2157
$\beta/^\circ$	119.3850(10)	Data/restraints/parameters	2157/49/151
$V/\text{Å}^3$	1355.5(3)	Goodness-of-fit on F^2	1.092
Z	2	R, wR ($I > 2\sigma(I)$)	$R1 = 0.0503$, $wR2 = 0.1323$
θ range/ $^\circ$	2.98/25.02	R, wR (all data)	$R1 = 0.0517$, $wR2 = 0.1335$
$(\Delta/\sigma)_{\text{max}}$	0	Large diff. peak and hole/ $(e/\text{Å}^3)$	1.297 and -1.365

single-crystal structure was solved by direct methods with SHELXS-97 program and refined on F^2 using SHELXL-97 program by full-matrix least-squares methods with equivalent isotropic displacement parameters for all non-hydrogen atoms. The hydrogen atoms were placed according to theoretical models. CCDC: 998313.

X-ray single-crystal diffraction study reveals that the title compound crystallizes in the $P2(1)/m$ space group and is composed of $\text{Dy}(\text{H}_2\text{O})_9(1,4\text{-dimethylbenzene})_2\text{Cl}_4\cdot\text{H}_2\text{O}$ (Fig. 1). Asymmetric unit contains one crystallographically independent Dy center, one 1,4-dimethylbenzene ligand, two Cl^- ions, one H_2O and six coordinated water molecules. The Dy^{3+} center adopts nine-coordinated mode with nine O atoms from nine coordinated water molecules to construct a mono-capped square antiprism. The Dy-O bond lengths are in the normal range of 2.438(6)-2.536(7) Å (Table-2), similar to those observed in the reported Dy-compounds⁸⁻¹⁰. An interesting structure feature of the title compound is hydrogen bond interactions. As illustrated in Fig. 2, because of hydrogen bond interactions among $\text{Dy}(\text{H}_2\text{O})_9$ groups, Cl^- ions and H_2O molecules, the title compound presents a two-dimensional (2-D) supramolecular layer structure.

Bond	(Å)	Bond	(Å)
Dy(1)-OW2	2.438(6)	Dy(1)-OW4	2.468(5)
Dy(1)-OW2#1	2.438(6)	Dy(1)-OW5	2.530(7)
Dy(1)-OW3	2.462(6)	Dy(1)-OW6	2.535(7)
Dy(1)-OW3#1	2.462(5)	Dy(1)-OW7	2.536(7)
Dy(1)-OW4#1	2.468(5)		

Symmetry code: #1 $x, -y + 3/2, z$

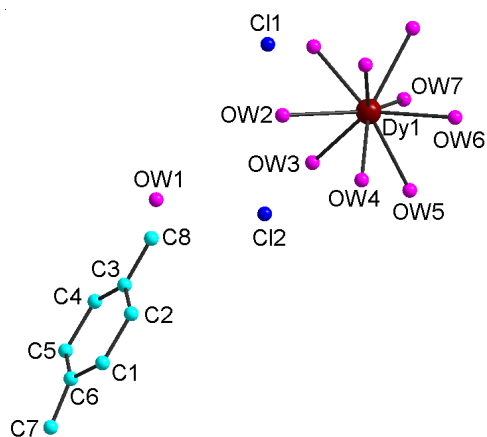


Fig. 1. Coordination environment of Dy1 and the crystal structure of the title compound

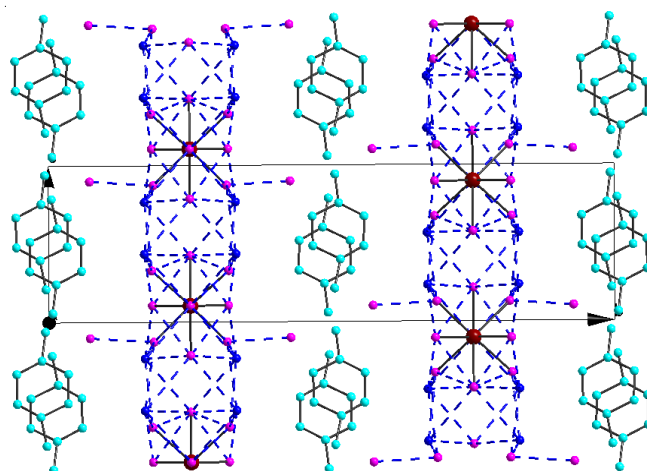


Fig. 2. A packing drawing of the title compound. The dashed lines are hydrogen bonds

Conclusion

One-pot solvothermal reaction gives $\text{Dy}(\text{H}_2\text{O})_9(1,4\text{-dimethylbenzene})_2\text{Cl}_4\cdot\text{H}_2\text{O}$ is characterized by single-crystal X-ray diffraction. The 2-D supramolecular layer is built up from the compound by hydrogen bonds.

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