

Design and Synthesis of New Steroid-Oxazin Derivative Using Some Strategies

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In this study, a steroid-oxazin derivative was synthesized using several strategies; the first stage involve the reaction of β -naphthol with ethylenediamine in presence of formaldehyde to form the compound **3** (3-(1*H*-naphtho[1,2-*e*][1,3]oxazin-2(3*H*)-yl)propan-1-amine). The second stage was achieved by reaction of **3** with dehydroisoandrosterone 3-sulfate to form the compound **5** (10,13-dimethyl-17-[2-(1*H*-naphtho[1,2-*e*][1,3]oxazin-2-yl)-ethylimino]-2,3,4,7,8,9,10,11,12,13,14,15,16,17-tetradecahydro-1*H*-cyclopenta[a]phenanthrene-3-sulfonic acid as catalyst. Finally, the third stage involve the synthesis of 17-{(3-chloro-2-oxo-cyclobutyl)-[2-(1*H*-naphtho[1,2-*e*][1,3]oxazin-2-yl)-ethyl]amino}-10,13-dimethyl-2,3,4,7,8,9,10,11,12,13,14,15,16,17-tetradecahydro-1*H*-cyclopenta[a]phenanthrene-3-sulfonic acid (**7**) by reaction of **5** with chloroacetyl chloride in presence of triethylamine. The structure of compounds obtained was confirmed by elemental analysis, spectroscopy and spectrometry data. In conclusion, this method offers some advantages such as good yields, simple procedure, low cost and ease of workup.

Keywords: Steroid, Oxazin, β-naphthol, Ethylenediamine.

INTRODUCTION

Oxazine derivatives are very important heterocyclic compounds with several biological activities^{1,2}. Therefore, diverse oxazine derivatives have been developed; e.g., the synthesis of 1,2-dihydro-1-arylnaphtho[1,2-e][1,3]oxazine-3-one derivatives³ using the three-component system (β naphthol, benzaldehyde and urea) in presence of HClO4-SiO₂. Other studies showed the condensation of 2-naphthol with heteroarylaldehydes or substituted benzaldehydes in the presence of ammonia⁴. Also, severalsteroid-oxazine derivatives have been developed, such as the compound 4-aza-4-ethyl- 17β -hydroxy-2-oxa-5a-estrane which was prepared by the reaction of 4-aza-l, 17β -dihydroxy-l,2-seco-5 α -estranewith paraformaldehyde in benzene to reflux⁵. Other reports showed the synthesis of a steroid-oxazin derivative (5'R-3\beta-acetoxy- 17β -[2-phenyl-4,5-dihydro-oxazol-5-yl]androst-5-ene) by the reaction of (20 R)-3\beta-acetoxy-21-azidopregn-5-en-20-ol with benzaldehyde using $BF_3 \cdot OEt_2$ as catalyst⁶. In addition, other experimental data showed the synthesis of N-methyl-3β,5epoxyimino-5 β -cholest-1-ene by the reaction of (Z)-3 β - acetoxy-5,10-secocholest-1(10)-en-5-one with N-methylhydroxyl-aminehydro-chloride⁷. Other studies showed the synthesis of 2'-phenyl)-androstano-[16,17-e]-3',6'-dihydrol',2'-oxazineby the reaction of 10,13-dimethyl-17-vynyl-2,34,5,6,7,8,9,10,11,12,13,14,15-tetra-decahydro-1*H*cyclopenta[a]phenanthrene with nitrosobenzene⁸. All these experimental results show several procedures which are available for synthesis of several oxazin-derivatives; nevertheless, expensive reagents and special conditions are required. Therefore, in this study a new steroid-oxazin derivative was synthetized using some strategies.

EXPERIMENTAL

The compounds evaluated in this study were purchased from Sigma-Aldrich Co. Ltd. The melting points for the different compounds were determined on an Electrothermal (900 model). Infrared spectra (IR) were recorded using KBr pellets on a Perkin Elmer Lambda 40 spectrometer. ¹H and ¹³C NMR spectra were recorded on a Varian VXR-300/5 FT NMR spectrometer at 300 and 75.4 MHz in CDCl₃ using TMS as

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internal standard. EIMS spectra were obtained with a Finnigan Trace GCPolaris Q. spectrometer. Elementary analysis data were acquired from a Perkin Elmer Ser. II CHNS/0 2400 elemental analyzer.

Synthesis of 3-(1H-naphtho[1,2-e][1,3]oxazin-2(3H)yl)propan-1-amine (3) (Fig. 1): A solution of β -naphthol (200 mg, 1.39 mmol), ethylenediamine (172 mg, 2.78 mmol) and 1 mL of formaldehyde in 10 mL of methanol was stirring to reflux for 6 h. The reaction mixture was evaporated to dryness under reduced pressure. The obtained solid was washed with water, yielding 70 % of product, m.p. 160-162 °C; IR (KBr, v_{max} , cm⁻¹) = 3380, 1170, 1158; ¹H NMR (300 MHz, CDCl₃) $\delta_{\rm H}$:1.68 (t, 2H, J = 6.70), 1.70 (broad, 2H), 252 (m, 2H), 2.56 (t, 2H, 6.70), 4.10-4.20 (m, 2H), 4.80-4.82 (m, 2H), 7.08-7.78 (m, 6H) ppm.¹³C NMR (75.4 Hz, CDCl₃) δ: 30.68 (C-16), 39.30 (C-17), 50.78 (C-4), 53.66 (C-15), 82 (C-2), 111.20 (C-5), 117.80 (C-7), 120.34 (C-11), 123.30 (C-13), 126.24 (C-12), 126.60 (C-10), 127.88 (C-14), 128.34 (C-8), 133.10 (C-9), 152.34 (C-6) ppm. EI-MS *m/z*: 242.10 (M + 10). Anal. Calcd. for C₁₅H₁₈N₂O: C, 74.35; H, 7.49; N, 11.56; O, 6.60. Found: C, 74.32; H, 7.48.



Fig. 1. Synthesis of 3-(1*H*-naphtho[1,2-*e*][1,3]oxazin-2(3*H*)-yl)propan-1-amine (3). Reaction of β-naphtol (1) with ethylenediamine (2) to form the compound 3. i = formaldehyde/methanol

Synthesis of 10,13-dimethyl-17-[2-(1H-naphtho[1,2e][1,3]oxazin-2-yl)-ethylimino]-2,3,4,7,8,9,10,11,12,13, 14,15,16,17-tetradecahydro-1H-cyclopenta[a]phenanthrene-3-sulfonic acid (5) (Fig. 2): A solution of compound 3 (100 mg, 0.41 mmol), dehydroisoandrosterone 3-sulfate (146 mg, 0.41 mmol) and boric acid (50 mg, 0.82 mmol) in 10 mL of CHCl₃:CH₃OH (3:1) was stirring to room temperature for 48 h. The reaction mixture was evaporated to dryness under reduced pressure. The obtained solid was washed with water, yielding 66 % of product, m.p. 180 °C; IR (KBr, v_{max} , cm⁻¹) = 3320, 1172, 1160; ¹H NMR (300 MHz, CDCl₃) $\delta_{\rm H}$:0.98 (s, 3H), 1.01 (s, 3H), 1.05-1.66 (m, 5H), 1.74-1.90 (m, 4H), 2.04-2.50 (m, 9H), 2.66 (t, 2H, J = 7.0), 2.70 (m, 1H), 3.52 (t, J = 7.0), 2.70 (m, 1H), 3.52 (t, J = 7.0), 3.52 (t, J = 7.02H, J = 7.0), 3.74 (m, 1H), 4.30-4.40 (m, 2H), 5.00-5.10 (m, 2H), 5.54 (d, 1H, J = 5.35), 7.02-7.70 (m, 6H), 8.22 (broad, 1H) ppm. 13 C NMR (75.4 Hz, CDCl₃) δ : 15.92 (C-35), 19.20 (C-36), 20.50 (C-23), 21.80 (C-26), 25.74 (C-32), 27.50 (C-25), 29.64 (C-21), 31.70 (C-27), 32.20 (C-24), 32.91 (C-34), 34.71 (C-31), 39.44 (C-30), 41.21 (C-19), 47.22 (C-22), 50.40 (C-4), 50.80 (C-16), 53.66 (C-20), 53.80 (C-15), 55.12 (C-33), 82.33 (C-2), 113.58 (C-5), 118.40 (C-7), 120.82(C-1), 122.30 (C-28), 123.32 (C-13), 126.30 (C-12), 127.90 (C-14), 128.40 (C-8), 128.78 (C-10), 131.66 (C-9), 137.30 (C-29), 151.70 (C-6), 176.80 (C-18) ppm. EI-MS m/z: 562.26 (M + 10). Anal. Calcd. for $C_{33}H_{42}N_2O_4S$: C, 70.43; H, 7.52; N, 4.98; O, 11.37; S, 5.70. Found: C, 70.50; H, 7.50.



Fig. 2. Synthesis of 10,13-dimethyl-17-[2-(1*H*-naphto[1,2-*e*][1,3]oxazin-2-yl)-ethylimino]-2,3,4,7,8,9,10,11,12,13,14,15,16,17-tetradeca-hydro-1*H*-cyclopenta[a]phenanthrene-3-sulfonic acid (5). Reaction of 3-(1*H*-naphtho[1,2-*e*][1,3]oxazin-2(3*H*)-yl)propan-1-amine (3) with dehydroisoandrosterone 3-sulfate (4) to form the compound 5. ii = boric acid/methanol

Synthesis of 17-{(3-chloro-2-oxo-cyclobutyl)-[2-(1*H*-naphtho[1,2-e][1,3]oxazin-2-yl)-ethyl]amino}-10,13dimethyl-2,3,4,7,8,9,10,11,12,13,14,15,16,17-tetradecahydro-1*H*-cyclopenta[a]phenanthrene-3-sulfonic acid (7) (Fig. 3): A solution of compound 5 (100 mg, 0.18 mmol), chloroacetyl chloride (29 μ L, 0.36 mmol) and triethylamine (50 μ L, 0.36 mmol) in 10 mL of methanol was stirring to room temperature for 48 h. The reaction mixture was evaporated to dryness under reduced pressure. The obtained solid was washed with water, yielding 70 % m.p. 220-222 °C; IR (KBr, v_{max}, cm⁻¹) = 1720, 1345, 1164; ¹H NMR (300 MHz, CDCl₃) δ_{H} :



Fig. 3. Synthesis of 17-{(3-chloro-2-oxo-cyclobutyl)-[2-(1*H*-naphto[1,2-e][1,3]oxazin-2-yl)-ethyl]amino}-10,13-dimethyl-2,3,4,7,8,9,10, 11,12,13,14,15,16,17-tetradecahydro-1*H*-cyclopenta[a]phenan-threne-3-sulfonic acid (7). Reaction of 10,13-dimethyl-17-[2-(1*H*-naphto[1,2-e][1,3]oxazin-2-yl)-ethylimino] 2,3,4,7,8,9,10,11,12,13, 14,15,16,17-tetradecahydro-1*H*-cyclopenta-[a]phenanthrene-3-sulfonic acid (5) with chloroacetyl chloride (6) to form the compound 7. iii = triethylamine/methanol

0.72 (s, 3H), 0.86 (m, 1H), 1.00 (s, 3H), 1.10-1.60 (m, 9H), 1.66-1.70 (m, 2H), 1.68 (m, 1H), 1.82 (m, 1H), 1.96 (m, 1H), 2.00-2.42 (m, 6H), 2.50 (t, 1H, J = 6.51), 2.52 (m, 1H), 2.56 (t, 1H, J = 6.51), 3.10 (t, 2H, J = 6.51), 3.62 (m, 1H), 3.70 (m, 1H),1H), 4.30-4.41 (m, 2H), 4.58 (m, 1H), 5.00-5.10 (m, 2H), 5.56 (m, 1H), 7.00-7.78 (m, 6H), 8.16 (broad, 1H) ppm.¹³C NMR (75.4 Hz, CDCl₃) δ: 16.58 (C-39), 19.22 (C-40), 21.02 (C-23), 25.18 (C-26), 25.70 (C-32), 27.32 (C-25), 31.48 (C-21), 31.58 (C-27), 32.70 (C-36), 32.90 (C-34), 34.17 (C-31), 36.40 (C-24), 39.40 (C-30), 46.48 (C-19), 49.90 (C-15), 50.33 (C-4), 50.46 (C-16), 52.34 (C-22), 53.36 (C-20), 56.12 (C-33), 62.86 (C-37), 67.70 (C-18), 72.28 (C-35), 82.40 (C-2), 111.28 (C-5), 118.40 (C-7), 120.78 (C-11), 122.30 (C-28), 123.36 (C-13), 126.30 (C-12), 127.90 (C-14), 128.40 (C-8), 128.80 (C-10), 131.66 (C-9), 137.28 (C-29), 151.70 (C-6), 202.10 (C-38) ppm.EI-MS *m/z*: 666.28 (M + 10). Anal. Calcd. for C₃₇H₄₇ClN₂O₅S: C, 66.60; H, 7.10; Cl, 5.31; N, 4.20; O, 11.99; S, 4.81. Found: C, 66.56; H, 7.08.

RESULTS AND DISCUSSION

In this study we report a straight forward route for synthesis of steroid-oxazin derivative (7) using several strategies. It is noteworthy that there are studies which indicate a convenient route for the preparation of oxazin-derivatives by the reaction of substituted phenols with formaldehyde and primary aliphatic amines^{9,10}. Therefore, in first stage was synthetized the compound 3-(1H-naphtho[1,2-e][1,3]oxazin-2(3H)-yl)propan-1amine (3) by the reaction of β -naphthol and ethylenediamine in presence of formaldehyde. The ¹H NMR spectrum of 3 shows signals at 1.68, 2.52-2.56 ppm for methylene groups involved in the arm bound to both amino groups; at 1.70 ppm for amino group; at 4.10-4.82 ppm for oxazine ring; at 7.08-7.78 ppm for phenyl groups. Other data indicate that ¹³C NMR spectrum of 3 contains peaks at 30.68-39.30 and 53.66 ppm for methylene groups involved in the arm bound to both amino groups; at 53.66-11.20 ppm for oxazine ring; at 117.80-152.34 ppm for phenyl groups. Finally, the presence of compound **3** was further confirmed from mass spectrum which showed a molecular ion at m/z 242¹⁰.

On the other hand, in the second stage was achieved by reaction of the compound 3 with dehydro iso and roster one 3-sulfate resulting animino bond formation involved in the compound 5 (10,13-dimethyl-17-[2-(1H-naphtho[1,2e][1,3]oxazin-2-yl)-ethylimino]-2,3,4,7,8,9,10,11,12,13,14, 15,16,17-tetradeca-hydro-1H-cyclopenta[a]phenanthrene-3sulfonic acid). It is important to mention that many procedures for the synthesis of imino groups are described in the literature¹¹⁻¹³; nevertheless, in this study boric acid was used as a catalyst, because it is not an expensive reagent and no special conditions for its use are required¹⁴. The ¹H NMR spectrums of **5** shows signals at 0.98 and 1 for methyl groups; 1.05-2.50, 2.70, 3.74 and 5.54 ppm for protons involved steroid nucleus; at 2.66 and 3.52 ppm for methylene groups involved in the arm bound to both amino groups; at 4.30-5.09 ppm for oxazine ring; at 7.02-7.70 ppm for phenyl groups; at 8.22 for hydroxyl group. Other results indicate that¹³C NMR spectrum of 5 contains peaks at 15.92 and 19.20 ppm for methyl groups; at 20.50-47.22, 53.66, 55.12, 122.30 and 137.30 ppm for protons involved in steroid nucleus; at 50.40, 82.30-113.58 ppm for methylene groups of oxazine ring; at 118.40-120.82, 123.32-131.66 and 151.70 ppm for phenyl groups; at 176.68 ppm for imino group. Finally, the presence of compound 5 was further confirmed from mass spectrum which showed a molecular ion at m/z 562.26.

The third stage step was achieved by the reaction of 5 with chloroacetyl chloride to form a cyclobutanonegroup involved in compound 7. It is important to mention that many procedures for the formation of cyclobutanone derivatives are known in the literature¹⁵⁻¹⁸, nevertheless, expensive reagents and special conditions are required. Therefore, in this study chloroacetyl chloride was used to form a cyclobutanonegroup involved in the compound 7. The ¹H NMR spectrum of compound 7 shows signals at 0.72 and 1 ppm for methyl groups; at 0.86, 1.10-1.70, 1.82, 2.00-2.42, 2.52, 3.70 and 5.56 ppm; at 1.68, 1.96, 362 and 4.58 ppm for cyclobutanone group; at 2.50, 2.56 and 3.10 ppm for arm bound to both amino groups; at 4.30-4.41 and 5.00-5.10 ppm for oxazine ring; at 7.02-7.78 ppm for phenyl groups; at 8.16 ppm for hydroxyl group.Other data indicate that ¹³C NMR spectrum of 7 contains peaks at 16.58 and 19.22 ppm for methyl groups; at 21.02-31.58, 32.90-46.48, 52.34-56.12, 67.70, 122.38 and 137.22 ppm for steroid nucleus; at 32.70, 62.86 and 72.28 ppm for cyclobutanone ring; at 49.90 and 50.46 ppm for arm bound to both amino groups; at 50.38 and 82.40-11.28 ppm for oxazine group; at 118.40-120.78, 123.36-131.66 and 151.70 ppm for phenyl groups; at 202.10 for ketone group. Finally, the presence of compound 7 was further confirmed from mass spectrum which showed a molecular ion at m/z 666.20.

REFERENCES

- 1. M. Kuehne, E.A. Konopka and B.F. Lambert, *J. Med. Chem.*, **5**, 281 (1962).
- 2. A. Chaskar, V. Vyavhare, V. Padalkar, K. Phatangare and H. Deokar, *J. Serb. Chem. Soc.*, **76**, 21 (2011).
- 3. H. Abbastabar Ahangar, G.H. Mahdavinia, K. Marjani and A. Hafezian, *J. Iran. Chem. Soc.*, **7**, 770 (2010).
- 4. Z. Turgut, E. Pelit and A. Köycü, *Molecules*, 12, 345 (2007).
- 5. D.M. Piatak and E. Caspi, J. Org. Chem., **31**, 3935 (1966).
- D. Ondré, J. Wölfling, I. Tóth, M. Szécsi, J. Julesz and G. Schneider, Steroids, 74, 1025 (2009).
- M. Rajkovic, L. Lorenc, I. Juranic, Z. Vitnik and M. Mihailovic, *Tetrahedron*, 55, 6681 (1999).
- R. Skoda-Földes, K. Vándor, L. Kollár, J. Horváth and Z. Tuba, *J. Org. Chem.*, 64, 5921 (1999).
- 9. W.J. Burke, R.P. Smith and C. Weatherbee, J. Am. Chem. Soc., 74, 602 (1952).
- W.J. Burke, M.J. Kolbezen and C.W. Stephens, J. Am. Chem. Soc., 74, 3601 (1952).
- 11. A. Shirayev, I. Moiseev and S. Karpeev, Arkivoc, 199 (2004).
- 12. D. Uppiah, M. Bhowon and S. Jhaumeer, E-J. Chem, 6, 195 (2009).
- 13. M. Hania, E- J. Chem., 6, 629 (2009).
- L. Figueroa-Valverde, F. Díaz-Cedillo, E. García-Cervera, E. Pool-Gómez and M. López-Ramos, *Bulgarian Chem. Comm*, 45, 71 (2013).
- 15. G.K. Kole, G.K. Tan and J.J. Vittal, Org. Lett., 12, 128 (2010).
- 16. J. Panda and S. Ghosh, Tetrahedron Lett., 40, 6693 (1999).
- 17. G.K. Kole, G.K. Tan and J.J. Vittal, J. Org. Chem., 76, 7860 (2011).
- Y. Okada, T. Minami, S. Yahiro and K. Akinaga, J. Org. Chem., 54, 974 (1989).