



Synthesis, Growth and Characterization of Nonlinear Optical Ce-Doped L-Proline Picrate Single Crystals

T. GURUMURTHI^{1,2} and P. MURUGAKOOTHAN^{2,*}

¹Department of Physics, Easwari Engineering College, Chennai-600 089, India

²MRDL, PG and Research Department of Physics, Pachaiyappa's College, Chennai-600 030, India

*Corresponding author: E-mail: murugakoothan03@yahoo.co.in

Received: 5 September 2013;

Accepted: 27 December 2013;

Published online: 28 July 2014;

AJC-15619

A non-linear optical material L-proline picrate and cerium doped L-proline picrate were synthesized and grown as single crystals by slow evaporation method. The grown crystals were subjected to structural, elemental, thermal, optical and dielectric studies. The structural analysis reveals that pure L-proline picrate and Ce doped L-proline picrate belongs to the monoclinic crystallographic system with space group P_2_1 . Optical transparency of the grown crystals was investigated by UV-visible-NIR spectrum. The thermal analyses reveal that Ce doped L-proline picrate is thermally stable up to 162 °C. The dielectric constant and dielectric loss of the crystals were studied as a function of frequency. The non-linear optical property of the doped crystal was confirmed by the Kurtz-powder second harmonic generation test and the result is compared with pure L-proline picrate. Mechanical strength of the crystals was also carried out by Vicker's micro hardness test.

Keywords: Growth from solution, Non-linear optical crystal, X-ray diffraction, Semi organic compound, Micro hardness.

INTRODUCTION

Non-linear optical (NLO) crystals, due to their non-linear properties, have been found to have enormous applications in frequency conversion, image processing, data storage and fiber optical communication *etc.* Rare earth ions introduced as dopant during the crystal growth have important consequences on the growth kinetics, morphology and quality of crystals, improving in many cases their physical properties for specific applications. Efforts have been made to improve the NLO properties of the L-proline picrate crystal by doping with various rare earth elements. The growth, optical absorption and mechanical studies on pure L-proline picrate were studied and reported^{1,2}.

The availability of the rare earth elements is of great interest to scientists doing basic research. In the present investigation, attempts were made to study the role of Cerium as dopant on growth, physical, mechanical and optical properties, thermal, dielectric and hardness of pure L-proline picrate.

EXPERIMENTAL

Commercially available L-proline and picric acid were mixed in stoichiometric ratio and dissolved in the mixed solvent of double distilled water and acetone in 1:1 ratio to synthesize

L-proline picrate (LPP) source material. The synthesized salt was purified by repeated recrystallisation process and used for growing pure L-proline picrate. Cerium doped L-proline picrate crystal was grown from mixed solvents of water and acetone in the ratio of 1:1 using the well known solvent evaporation technique with 5 and 10 mol % of Cerium nitrate added to the L-proline picrate saturated solutions.

Optical quality crystals were collected in a period of 35 days from the supersaturated solution and are shown in Fig. 1. From the physical observations of the grown crystals the 5 mol % cerium doped L-proline picrate (Ce5 : L-proline picrate) crystals have good transparency and morphology.

Characterization: The grown single crystals of pure and Ce5 : L-proline picrate were subjected to X-ray diffraction studies using a Bruker AXS Kappa APEX II single crystal CCD diffractometer equipped with graphite monochromated $Mo(K\alpha)$ ($\lambda = 0.7107 \text{ \AA}$). EDAX studies was carried out using EDAX-AMETEK tester for the grown sample to analyze the percentage of dopants in the crystal. HIOKI 3532-50 LCR HITESTER was used for the dielectric study. The sample of size 4 mm × 4 mm × 3 mm was prepared and mounted between copper electrodes. In order to ensure good electrical contact between the crystal and the electrodes, the crystal faces were coated with silver paint. The dielectric measurements were carried out in a frequency range 100 Hz-5 MHz and temperature

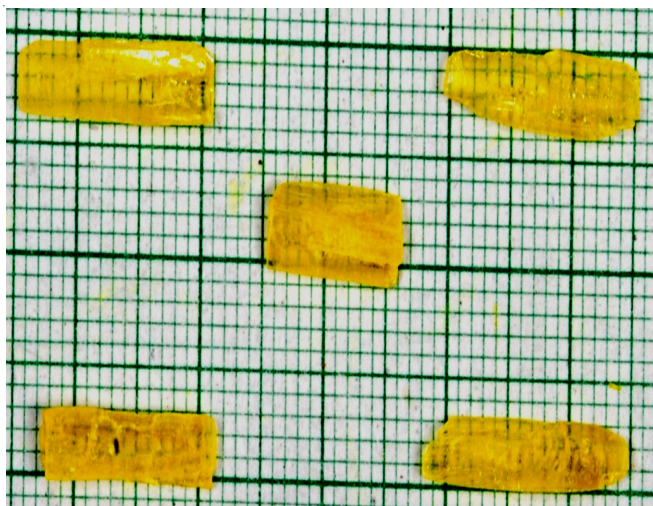


Fig. 1. As grown crystals of 5 mol % Ce doped L-prolinium picrate (Ce5 : L-prolinium picrate)

range 35-95 °C. The Thermogravimetric and differential thermal analyses (TG-DTA) response curves were drawn for pure and Ce5 : L-prolinium picrate powder sample in the temperature range from 20 to 1400 °C using the instrument NETZSCH STA 409 C at the heating rate of 10 K/min. in nitrogen atmosphere. The optical transmission spectra were recorded using Shimadzu model-1601 in the wavelength range of 300-1200 nm. The study of NLO conversion efficiency was carried out by the powder technique of Kurtz and Perry. The sample was ground into fine powder and tightly packed in a micro capillary tube. It was mounted in the path of the laser beam. A Q-switched flash lamp pumped Nd:YAG laser of power 3.2 mJ with a wavelength of 1064 nm, pulse duration of 8 ns, a repetition rate of 10 Hz and a spot size of 1 mm diameter was used for SHG test. Vickers hardness study was made on the as grown crystal using Leica-reichert polyvar2 model hardness tester fitted with a diamond indenter. Etching studies of the Ce5 : L-prolinium picrate single crystal was carried out by REICHERT POLYVAR 2-Metallurgical microscope using water-acetone as an etchant.

RESULTS AND DISCUSSION

Single crystal X-ray diffraction: Lattice parameter values of pure L-prolinium picrate (LPP) and Ce5 : L-prolinium picrate single crystals are listed in Table-1. The observed values for L-prolinium picrate are in good agreement with the reported values^{1,2}.

TABLE 1
COMPARISON OF SINGLE CRYSTAL X-RAY DATA
OF L-PROLINIUM PICRATE AND Ce5 : L-PROLINIUM
PICRATE CRYSTALS

Parameters	L-Prolinium picrate	Ce5 : L-prolinium picrate
a (Å)	10.902	10.825
b (Å)	5.352	5.432
c (Å)	12.472	12.224
V (Å ³)	687.60	682.32
System	Monoclinic	Monoclinic
β(°)	109.11	109.14
Space group	P ₂₁	P ₂₁

EDAX analysis: In order to analyze quantitatively the presence of cerium in the crystal, EDAX study was carried out for the grown sample and the percentage of dopant present in the L-prolinium picrate was confirmed and tabulated in Table-2. From EDAX analysis, it is observed that the amount of cerium atoms present inside the L-prolinium picrate lattice is very less.

TABLE 2
EDAX ANALYSIS OF CE5 : L-PROLINIUM PICRATE CRYSTAL

Element	Wt. %	At. %
Ce	3.42	0.47

Dielectric studies: The dielectric constant and dielectric loss were calculated for pure and doped crystals. Fig. 2a and 2b show the variation of dielectric constant and dielectric loss as a function of frequency. It is found that the dielectric constants of L-prolinium picrate and doped L-prolinium picrate crystals are high at lower frequencies and they decrease with increase in frequency. The trend of the dielectric constants of L-prolinium picrate and 5 mol % Ce doped L-prolinium picrate crystals are almost the same. But at fixed frequency, the

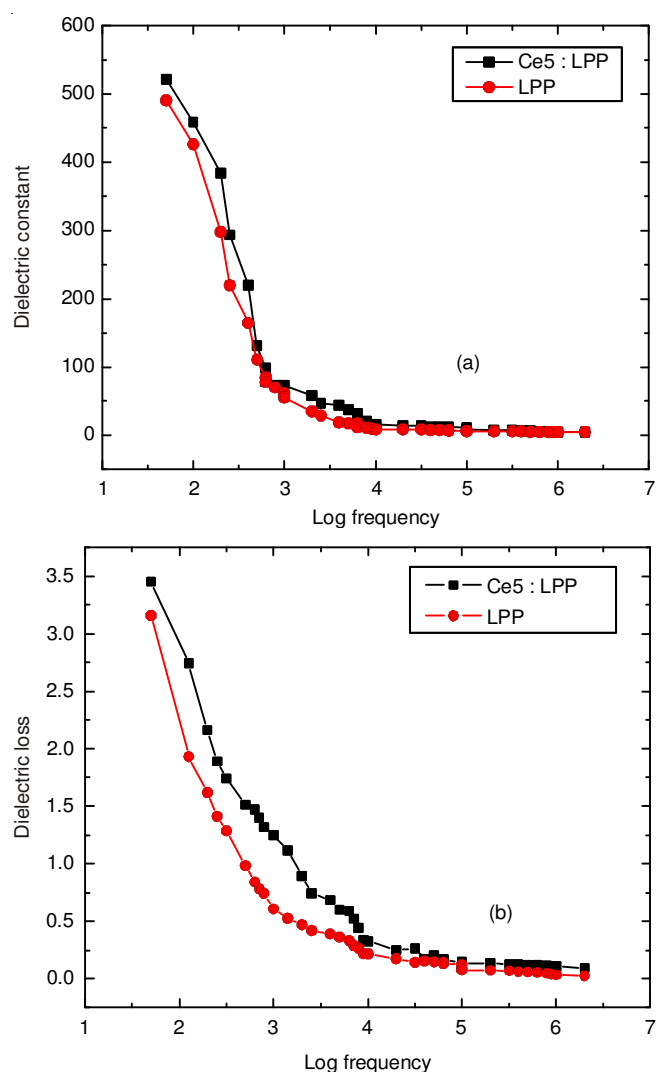


Fig. 2. (a) Dielectric constant vs. log frequency for pure and doped L-prolinium picrate crystals; (b) Dielectric loss vs. log frequency for pure and doped L-prolinium picrate crystals

dielectric constants of doped L-prolinium picrate crystals are less than that of pure one. In accordance with Miller rule, the lower value of dielectric constant is a suitable parameter for the enhancement of second harmonic generation (SHG) coefficient³. Since the dielectric constant of doped crystals is lower than that of the pure crystals they are less polarized in comparison with pure crystal. The lower polarization may be due to ineffective transportation of polarization from one molecule to its neighbor in the presence of dopants⁴.

The characteristic of low dielectric loss at high frequencies for these samples suggest that the pure and doped crystals possess enhanced quality with lesser defects⁵. For a particular frequency, the dielectric loss of doped L-prolinium picrate is slightly lesser than that of pure, which indicates that the dopant enhances the optical quality of L-prolinium picrate and reduces the defects.

Thermal analyses: The TG - DTA curves of L-prolinium picrate and Ce5 : L-prolinium picrate samples exhibits nearly similar stage of decomposition between 100 and 1200 °C as shown in Fig. 3(a) and 3(b). In order to study the influence of the dopant on the thermal stability of L-prolinium picrate, the temperature corresponding to a peak maximum of first stage of phase transition in DTA trace is taken into account for comparison. The temperature on the first state of transition for the L-prolinium picrate is found at 158 °C and for Ce5 : L-prolinium picrate it is found at 162 °C. This shows that the doped crystal possess better thermal stability compared to pure crystal.

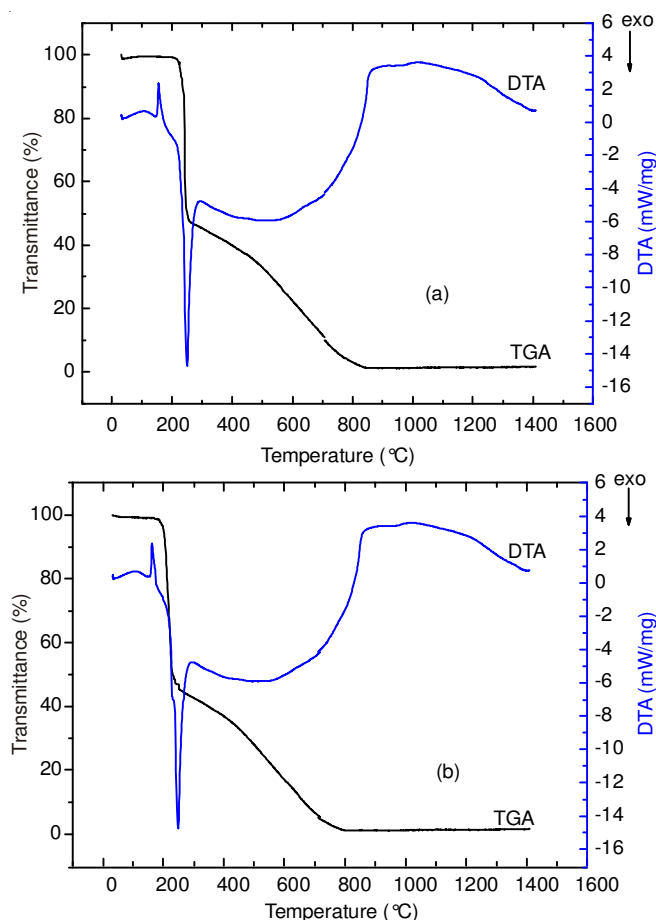


Fig. 3. (a) TG-DTA graph for L-prolinium picrate sample; (b) TG-DTA graph for Ce5 : L-prolinium picrate sample

Optical transmission study: Fig. 4 shows the optical transmission spectra of L-prolinium picrate and Ce5 : L-prolinium picrate crystals. The thickness of the sample used for this study was 1.5 mm. It is found that L-prolinium picrate crystal has transmittance of 66 % and Ce5 : L-prolinium picrate crystal has the transmittance of 63 %. The lower cut off for L-prolinium picrate and Ce5 : L-prolinium picrate crystals are 470 nm. This shows that doping the crystal with cerium did not shift the lower cut-off value, but, the percentage of transmission is decreased due to cerium addition.

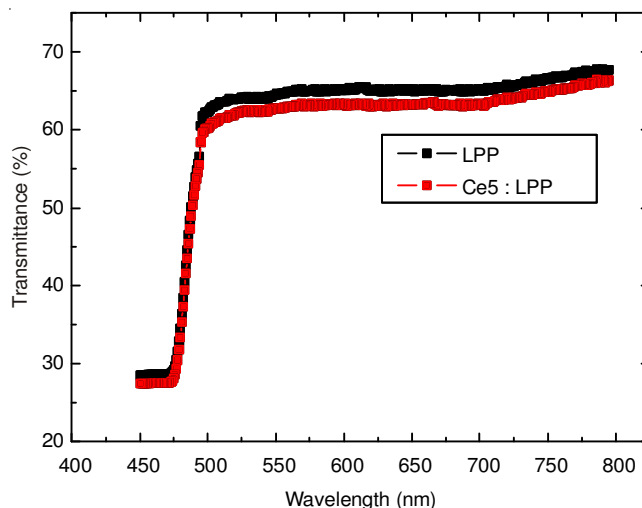


Fig. 4. Optical transmission for L-prolinium picrate and Ce5 : L-prolinium picrate crystals

Powder second harmonic generation measurement:

The study of NLO conversion efficiency was carried out by the powder technique of Kurtz and Perry⁶. Second harmonic signal of 474 mV was obtained for an input energy of 3.2 mJ/pulse for Ce5 : L-prolinium picrate. But the standard KDP sample gave a second harmonic generation signal of 15 mV/pulse for the same input energy. The results obtained by this method shows that second harmonic generation efficiency of Ce5 : L-prolinium picrate is nearly 31.6 times higher than that of KDP but pure L-prolinium picrate shows the second harmonic generation efficiency of 48 times higher than that of KDP and this is due to ineffective transportation of polarization from one molecule to its neighbor in the presence of dopants for Ce5 : L-prolinium picrate. Since the second order non linear efficiency will vary with the particle size of powder sample⁷, the care has been taken to maintain uniform particle size of source and the reference material.

Vicker's microhardness study: The good quality crystals are needed not only with good optical performance but also with good mechanical behaviour⁸ for applications. In order to study the mechanical behaviour of the grown L-prolinium picrate crystal, indentations were made on the cleaved (100) plane of pure L-prolinium picrate and Ce5 : L-prolinium picrate crystals with the applied load ranging from 5 to 100 g. The time of indentation was kept constant as 5 s for all indentations. The Vicker's hardness number was calculated using the relation⁹.

$$H_v = \frac{1.854P}{d^2} \text{ kg/mm}^2$$

where P is the applied load and d is the diagonal length. The Vicker's hardness for L-prolinium picrate and Ce5:L-prolinium picrate crystals as a function of load are shown in Fig. 5. The hardness values of L-prolinium picrate have been found to be lower than that of Ce5:L-prolinium picrate crystal. Vicker's hardness increases with increase of load till 60 g for Ce5:L-prolinium picrate crystal but upto 50 g for L-prolinium picrate crystal. The Loads above 60 g for Ce5: L-prolinium picrate and 50 g for L-prolinium picrate developed multiple cracks around the indentation mark and hardness decreases with the further increase of load. By plotting $\log P$ versus $\log d$, the value of the work hardening coefficient (n) was found to be 3.73 for L-prolinium picrate and 3.01 for Ce5 : L-prolinium picrate. According to onitsch, $1 \leq n \leq 1.6$ for hard materials and $n > 1.6$ for soft materials¹⁰. Hence it is concluded that L-prolinium picrate and Ce5:L-prolinium picrate are soft materials. In order to find the increase in strength that accompanies plastic deformation of the grown crystal, yield strength (σ_y) of the crystals was also calculated using the relation¹¹

$$\sigma_y = \frac{(H_v)}{3} 0.1^{n-2} \text{MPa}$$

where ' H_v ' is the maximum hardness and ' n ' is the work hardening coefficient. Yield strength for L-prolinium picrate and Ce5 : L-prolinium picrate crystals are found to be 0.07 MPa and 0.08 MPa, respectively.

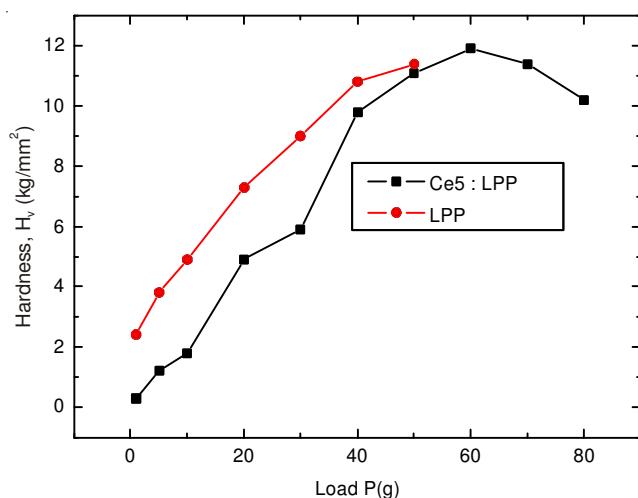


Fig. 5. Micro hardness value of L-prolinium picrate and Ce5 : L-prolinium picrate crystals

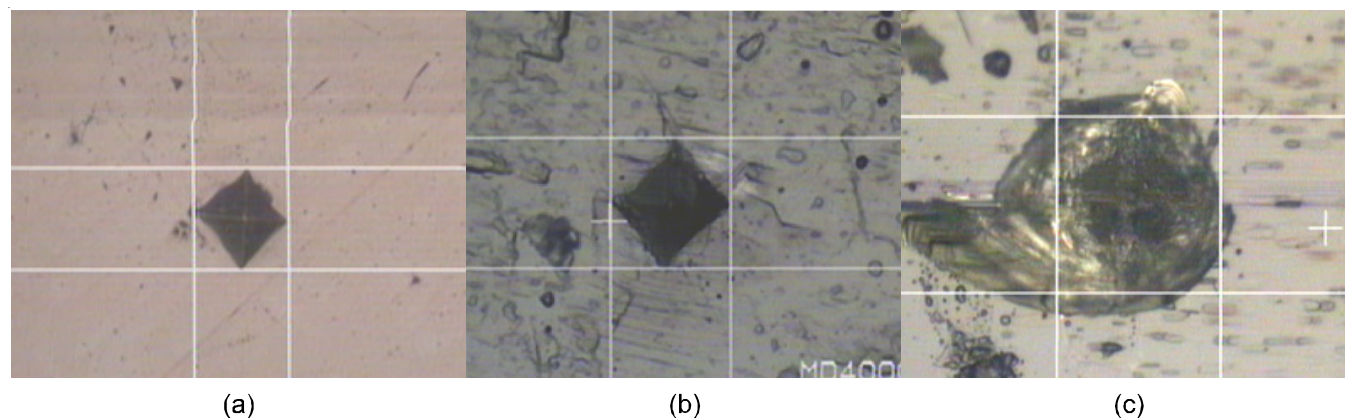


Fig. 6. Indentation image of Ce5 : L-prolinium picrate crystal for; (a) 5 g (b) 50 g and (c) 80 g

Moreover while analyzing the optical image of the indentation in the crystal for L-prolinium picrate and Ce5 : L-prolinium picrate, it is observed that mild crack started at 10 g load for both L-prolinium picrate and Ce5 : L-prolinium picrate and multicracks developed at 70 g load for Ce5 : L-prolinium picrate and 50 g load for L-prolinium picrate. Crack length was measured for various forces applied on the crystal. It was about 35 μm length from the indentation center for the load between 10 and 50 g. There were no remarkable changes found in the indentation length and crack length in this range of load. But when the applied load reached 80 g, the crystal was found to be deformed more, which can be seen in the optical image shown in the Fig. 6. Elastic stiffness constant was calculated^{4,12} from the microhardness by Wooster's empirical relation $C_{11} = H_v^{7/4}$. The maximum stiffness constants for the L-prolinium picrate and Ce5:L-prolinium picrate crystals are 67.36×10^{12} and 76.24×10^{12} Pascals, respectively. Fracture toughness (K_c) is the resistance of a material to failure from fracture starting from a preexisting crack. It was calculated using the formula $K_c = P/\beta C^{3/2}$, where C is the crack length from the center of the indentation, P is the applied load and $\beta (= 7)$ is the geometrical constant for Vicker's indenter¹³. The crack length developed with maximum hardness at 60 g applied load for Ce:L-prolinium picrate crystal was 43 μm . Hence the fracture toughness was calculated as 30,398 $\text{kg m}^{-3/2}$. Brittleness is an important property of the crystal which determines its fracture without any appreciable deformation. It is expressed in terms of brittleness index¹⁴. Brittleness index was calculated^{2,15} using the formula $B_i = H_v/K_c$ as $3.552 \times 10^{-3} \text{ m}^{-1/2}$.

Fig. 7a shows the as grown crystal surface and Fig. 7b shows the etch patterns produced for 5s. Etching produces triangular shaped etch pits. It is observed that etch pit is symmetrical in nature. The possible reason for the formations of these etch pits is due to the impurity in the crystal during growth, which introduces strain in the crystal lattice and serves as sites of dissolution¹⁶.

Conclusion

L-Prolinium picrate and cerium doped L-prolinium picrate crystals were grown from mixed solvent of water and acetone in the ratio of 1:1 by the solvent evaporation method. From the XRD analysis, it is observed that the L-prolinium picrate and Ce doped L-prolinium picrate crystals retain the monoclinic structure and the calculated lattice parameter values

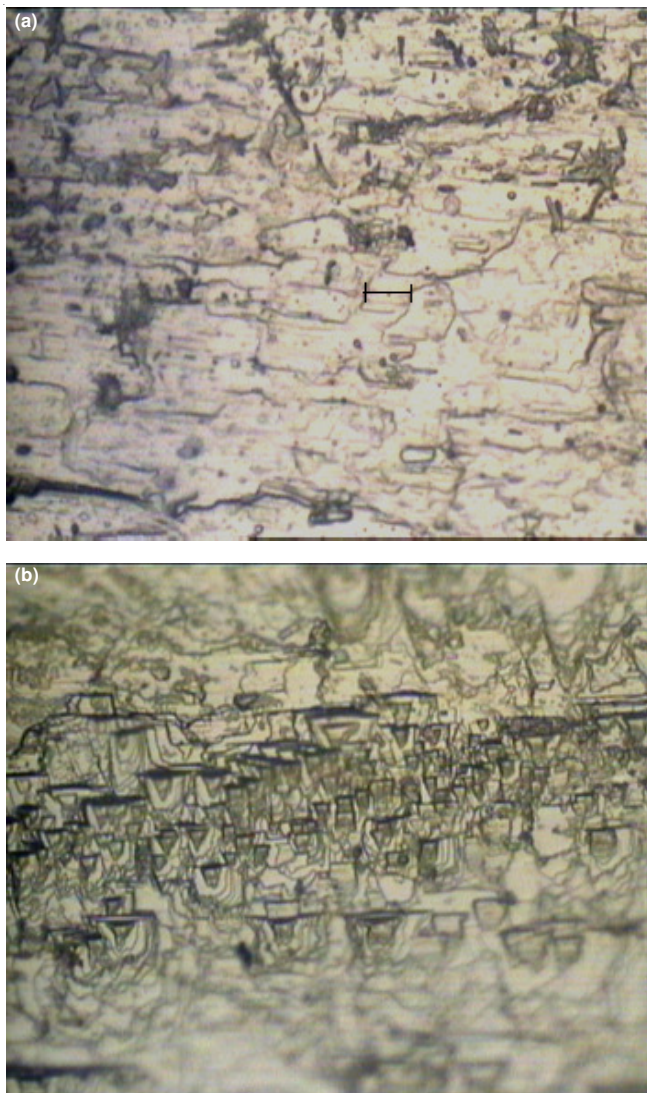


Fig. 7. (a) Surface of the as grown Ce : L-prolinium picrate (b) Etch patterns produced for 5 sec

are comparable with the reported values of L-prolinium picrate. The presence of cerium in L-prolinium picrate crystal was confirmed by EDAX analysis. Optical transmission study shows that the grown cerium doped L-prolinium picrate crystal has high transparency in the wavelength range from 470 to 800 nm. The dielectric constant and dielectric loss of Ce doped

L-prolinium picrate is found to be lesser than that of L-prolinium picrate. This shows that the doped crystal possess better optical quality with lesser defects compared to pure crystals. The thermal studies of the samples suggest that the thermal stability is better for doped crystals. Hardness study reveals that the L-prolinium picrate and Ce5 : L-prolinium picrate crystals are soft materials. Higher hardness is obtained for Ce5 : L-prolinium picrate than that of the L-prolinium picrate crystals. Yield strength, elastic stiffness, fracture toughness and brittleness index of the Ce5 : L-prolinium picrate crystals were also reported. The NLO efficiency for Ce5 : L-prolinium picrate crystal is 31.6 times higher than that of KDP crystals.

ACKNOWLEDGEMENTS

The authors acknowledged Prof. P.K. Das, Department of IPC, IISc, Bangalore, Prof. V. Ravichandran, Head of the Department, Nuclear Physics, University of Madras and IIT-SAIF, Chennai for carrying out the characterization of the sample.

REFERENCES

1. S.A. Martin Britto Dhas, G. Bhagavannarayana and S. Natarajan, *J. Cryst. Growth*, **310**, 3535 (2008).
2. T.U. Devi, N. Lawrence, R.R. Babu and K. Ramamurthi, *J. Cryst. Growth*, **310**, 116 (2008).
3. U. Von Hundelshausen, *Phys. Lett. A*, **34**, 405 (1971).
4. T.K. Kumar, S. Janarthanan, M.V.A. Raj, S. Pandi, P. Sagayaraj and D.P. Anand, *J. Phys. Chem. Solids*, **69**, 2634 (2008).
5. C. Balarew and R. Duhlev, *J. Solid State Chem.*, **55**, 1 (1984).
6. S.K. Kurtz and T.T. Perry, *J. Appl. Phys.*, **39**, 3798 (1968).
7. Y. Porter, K.M. Ok, N.S.P. Bhuvanesh and P.S. Halasyamani, *Chem. Mater.*, **13**, 1910 (2001).
8. M.K. Marchewka, S. Debrus and H. Ratajczak, *Cryst. Growth Des.*, **3**, 587 (2003).
9. A. Rahman and J. Podder, *Indian J. Phys.*, **86**, 15 (2012).
10. J. Madhavan, S. Aruna, K. Prabha, J.P. Julius, G.P. Joseph, S. Selvakumar and P. Sagayaraj, *J. Cryst. Growth*, **293**, 409 (2006).
11. M.E. Peter and P. Ramasamy, *J. Cryst. Growth*, **312**, 1952 (2010).
12. S. Suresh, A. Ramanand, P. Mani and K. Murthy, *J. Optoelectron. Biomed. Mater.*, **1**, 129 (2010).
13. S. Krishnan, C. Justin Raj, R. Robert, A. Ramanand and S. Jerome Das, *Solid-State Electron.*, **52**, 1157 (2008).
14. M. Bhat, B. Kaur, K.K. Bamzai, P.N. Kotru and B.M. Wanklyn, *J. Phys. Chem. Solids*, **65**, 1359 (2004).
15. G. Anandha Babu, G. Bhagavannarayana and P. Ramasamy, *J. Cryst. Growth*, **310**, 2820 (2008).
16. S. Mukerji and T. Kar, *J. Cryst. Growth*, **204**, 341 (1999).