

Chemical Constituents from n-Butanol Extract of Rabdosia japonica var. glaucocalyx

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The *n*-butanol fraction of the dried whole plants of *Rabdosia japonica* var. *glaucocalyx* was isolated and purified by means of chromatography. Six compounds were obtained and their structures were identified by spectral analysis as glaucocalyxin G (1), arjunglucoside (2), kaempferol-3-*O*-rutinoside (3), quercetin-3-O- α -L-rhamnoside (4), rutin (5) and acacetin-7-O- β -D-glucoside (6). Compound 1 was a new *ent*-kaurane diterpenoid glycoside, compounds 2 and 3 have not been reported before from the plant source.

Keywords: Rabdosia japonica var. glaucocalyx, Chemical constituents, Glaucocalyxin G, n-Butanol fraction.

INTRODUCTION

The genus Rabdosia japonica (Burm. f.) Hara var. glaucocalyx (Maxim.) Hara is a member of the family Labiatae, subfamily Ocimoideae, tribe Plectrantheae and is mainly distributed in northeast Asia, such as China, Russian, Korea and Japan¹. It possesses some pharmacological functions, such as antitumor, antioxidation, antiinflammatory and antibacteria². Results of toxicological tests indicated that the Rabdosia *japonica* var. *glaucocalyx* is safe to human within the scope of the experimental dose³. As for the chemical constituents of the plant, diterpenoids⁴⁻⁸, flavonoids⁹⁻¹¹ and triterpenoids^{12,13} are the major constituents in the whole plant. In the course of further studies, six compounds were obtained and their structures were identified as glaucocalyxin G (1), arjunglucoside $(2)^{14}$, kaempferol-3-*O*-rutinoside $(3)^{15}$, quercetin-3-*O* - α -Lrhamnoside $(4)^{16}$, rutin $(5)^{16}$ and acacetin-7-*O*- β -D -glucoside $(\mathbf{6})^{11}$ on the basis of their NMR spectral data and by comparison of their physical properties with those reported in the literatures. Compound 1 was a new ent-kaurane diterpenoid glycoside from R. japonica var. glaucocalyx, though the diterpenoids have been regarded as the marking composition of tribe Plectrantheae, there was few report about diterpenoids glycoside of the tribe. compounds 2 and 3 have not been reported before from the plant source.

EXPERIMENTAL

ESI-MS and HRESI-MS were performed with a Mat-212 and a Micro mass Auto Spec Q-TOF spectrometers, respectively. ¹H and ¹³C NMR spectra were recorded on a Bruker DRX-500 spectrometer with tetramethylsilane as an internal standard and DMSO- d_6 as solvents. Chemical shifts were given in (ppm) values. Optical rotations was determined using a Perkin-Elmer 341 automatic polarimeter.

The plant materials were collected in June 2009 from Antu, Jilin province and identified as the dried whole plants of *Rabdosia japonica* var. *glaucocalyx* by Prof. Hanming Zhang, College of Pharmacy, Second Military Medical University. A voucher specimen (No. 20090603) has been deposited in the herbarium of College of Bio-information, Chongqing University of Posts and Telecommunications, Chongqing.

Extraction and isolation: The dried whole plants (8 kg) were chopped and extracted with 80 % EtOH three times under reflux and concentrated under vacuum to yield an EtOH extract (500 g). The extract was suspended in water and extracted successively with petroleum ether, ethyl acetate and n-butanol to obtain petroleum ether fraction (20 g), ethyl acetate fraction (220 g) and *n*-butanol fraction (60 g). The *n*-butanol fraction (55 g) was fractionated *via* silica gel column chromatography eluting with CHCl₃-MeOH-H₂O (3:1:0 to 6:4:1) to give six major fractions A_1 - A_6 . Fraction A_2 (3 g) was subjected to ODS column eluting with MeOH-H₂O (10:90 to 70:30) to give four major fractions B₁-B₄. Fraction B₂ (180 mg) was purified over silica gel and Sephadex LH-20 with MeOH repeatedly to afford compounds 1 (48 mg), 4 (30 mg) and 6 (25 mg). Fraction A₃ (1.7 g) was subjected to ODS column eluting with MeOH- $H_2O(10:90 \text{ to } 70:30)$ to give five major fractions C_1 - C_5 . Fraction C2 (122 mg) was purified over silica gel and Sephadex LH-20 with MeOH repeatedly to afford compound 2 (55 mg). Fraction C_3 (95 mg) was purified over silica gel and Sephadex LH-20 with MeOH repeatedly to afford compounds 3 (37 mg) and 5 (20 mg).

RESULTS AND DISCUSSION

Compound **1** was isolated as a grey powder with an optical rotation $[\alpha]_{20}^{D}$ -78 (c 0.1, MeOH), exhibited a quasi-molecularion peak at m/z 537 [M + Na]⁺ by ESI-MS and the molecular formula C₂₆H₄₂O₁₀ was determined by HRESI-MS (m/z 514.2777, calcd. 514.2778). And the result of its molish reaction was positive. Furthermore, six proton signals at $\delta_{\rm H}$ 3.40-4.55 suggested the existence of a sugar moiety and the ¹³C NMR and DEPT spectra displayed six aliphatic carbon signals ($\delta_{\rm C}$ 103.1(d), 74.2(d), 76.7(d), 70.0(d), 76.8(d) and 61.1(t)) due to one sugar moiety. The signal at $\delta_{\rm H}$ 4.55 (d, 1H, J = 7.5 Hz, H-1') was attributed to an anomeric proton, The coupling constant (J = 7.5 Hz) of the anomeric proton demonstrated that the glucose was in β -orientation.

The ¹³C NMR and DEPT spectra of 1 (Table-1) gave twenty six carbon signals for three methyl, six methylene, thirteen methine and four quaternary carbons, including one exocyclic carbon-carbon double bond ($\delta_{\rm C}$ 155.9 (s) and 106.3 (t)). the ¹H NMR spectrum of **1** revealed the presence of three angular methyl groups ($\delta_{\rm H}$ 0.83 (s, 3H), 0.81 (s, 3H) and 1.27 (s, 3H)). Except for the sugar moiety, there were five oxygenated methines at $\delta_{\rm H}$ 3.17 (dd, 1H, 9.5, 4.0), 3.52 (s, 1H), 3.56 (m, 1H), 4.64 (s, 1H) and 4.86 (m, 1H), respectively. Considering the structures of diterpenoids previously isolated from the plant⁴⁻⁸ and structure characteristics of the *ent*-kaurane diterpenoid, compound 1 was assigned an ent-kaurane diterpenoid glycoside and the ent-kaurane diterpenoid has five hydroxy groups and one of them with a glucose generated the glycoside. The key work was to comfirm the positions of the five hydroxy groups.

Three angular methyl groups were confirmed at C-18
$(\delta_C32.3(q)),$ C-19 $(\delta_C21.6(q))$ and C-20 $(\delta_C13.9(q)),$ respec-
tively, according to HMQC and HMBC spectra. The charac-
teristics signals of three methines (C-5, C-9 and C-13) in the
ent-kaurane diterpenoid were very clear at C-5 (δ_C 51.2(d)),
C-9 (δ_{C} 49.2(d)) and C-13 (δ_{C} 56.6(d)) and four quaternary
carbons were easily confirmed at C-4 (δ_{C} 32.4(s)), C-8 (δ_{C}
53.4(s)), C-10 (δ_C 43.0(s)) and C-16 (δ_C 155.9 (q)), respectively
in HMQC and HMBC spectra, So the positions of oxygenated
methines may be C-1, 2, 3, 6, 7, 11, 12, 14 and 15. The proton
of oxygenated methine, having HMBC correlations with C-5,
C-10 and C-20, respectively, could be confirmed as H-1.
Similarly, the correlation between H-15 (3.52 (s, 1H)) and C- $\!\!\!$
17 (δ_C 106.3(t)) in the HMBC spectrum proved that C-15 was
oxygenated methine. The correlations between H-14 ($\delta_{\rm H}4.64$
(s, 1H)) and C-15 (δ_C 72.3(t)), C-16 (δ_C 155.9(t)) in the HMBC
spectrum proved that C-14 was oxygenated methine. The
correlations between C-7 (δ_{C} 73.2(d)) and H-5 (δ_{H} 0.74 (d,
1H, 11.5)), H-9 (δ_H 1.74 (s, 1H)), respectively in the HMBC
spectrum proved that C-7 was oxygenated methine. The
correlations between H-12 ($\delta_{\rm H}4.86(m,1H))$ and C-9 ($\delta_{\rm C}49.2$
(d)), C-16 (δ_C 155.9(t)) in the HMBC spectrum proved that C-
12 was oxygenated methine. Furthermore, the correlation
between H-1' ($\delta_{\rm H}$ 4.55 (1H, d, J = 7.5 Hz)) and C-1 ($\delta_{\rm C}$ 89.8
(d)) in the HMBC spectrum, proved that C-1 was connected
with a glucose (Fig. 2). Thus, the structure of $\boldsymbol{1}$ was elucidated as
shown in Fig. 1 and named glaucocalyxin G. The stereochemistry
of 1 was deduced from its NOESY spectrum. Spatial corre-
lations were observed between H-9 and H-12, H-20 and
H-14, H-5 and H-1 as well as H-15 and H-7 (Fig. 2). Therefore,
compound 1 was denominated to $7\alpha,\!12\alpha,\!14\beta,\!15\alpha$ -tetra-
hydroxyl- <i>ent</i> -kaur-16-ene-1- <i>O</i> -β-D-glucopyranoside.

TABLE-1					
¹ H NMR AND ¹³ C NMR DATA OF COMPOUND 1					
(500 AND 125 MHz, IN DMSO-d ₆)					
Position	$\delta_{\rm H}$ (mult., J in Hz)	$\delta_{C}(mult.)$			
1	3.17 (dd, 1H, 9.5, 4.0)	89.8 (d)			
2	1.58 (m, 1H); 1.72 (m, 1H)	29.5 (t)			
3	1.16 (m, 1H); 1.32 (m, 1H)	38.8 (t)			
4		32.4 (s)			
5	0.74 (d,1 H, 11.5)	51.2 (d)			
6	1.63 (m, 1H); 1.90 (m, 1H)	26.6 (t)			
7	3.56 (m, 1H)	73.2 (d)			
8		53.4 (s)			
9	1.74 (s, 1H)	49.2 (d)			
10		43.0 (s)			
11	1.61 (m, 1H); 2.45 (d, 1H, 16.5)	26.2 (t)			
12	4.86 (m, 1H)	71.7 (d)			
13	2.39 (d, 1H, 3.5)	56.6 (d)			
14	4.64 (s, 1H)	74.2 (d)			
15	3.52 (s, 1H)	72.3 (d)			
16		155.9 (s)			
17	4.86 (s, 1H); 5.03 (s, 1H	106.3 (t)			
18	0.83 (s, 3H)	32.3 (q)			
19	0.81 (s, 3H)	21.6 (q)			
20	1.27 (s, 3H)	13.9 (q)			
1'	4.55 (d, 1H, 7.5)	103.1 (d)			
2'	3.48 (m, 1H)	74.2 (d)			
3'	3.58 (m, 1H)	76.7 (d)			
4'	3.44 (m, 1H)	70.0 (d)			
5'	3.40 (m, 1H)	76.8 (d)			
6'	2 94 (44 111 11 5 5 5), 2 70 (4 111 11 5)	61 1 (+)			



No.	R_1	R_2	R ₃	R4
3	O-Rha-(1-6)-Glc	OH	Н	OH
4	O-Rha	OH	OH	OH
5	O-Rha-(1-6)-Glc	OH	OH	OH
6	Н	O-Glc	Н	OCH ₃

Fig. 1. Structures of compounds 1-6



Fig. 2. Key HMBC and NOESY correlations of compound 1

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