



## Synthesis and Crystal Structure of 4,4'-Dichloro-6,6'-dibromo-2,2'-[(hexane-1,6-diylidioxy)bis(nitrilomethylidyne)]diphenol

QIONG SU<sup>1\*</sup>, PENG-FEI LIU<sup>2</sup>, YU-JIE ZHANG<sup>2</sup>, LI ZHAO<sup>2</sup>, XIU-YAN DONG<sup>2</sup> and YAN-BIN WANG<sup>1\*</sup>

<sup>1</sup>School of Chemical Engineering, Northwest University of nationalities, Lanzhou 730030, P.R. China

<sup>2</sup>School of Chemical and Biological Engineering, Lanzhou Jiaotong University, Lanzhou 730070, P.R. China

\*Corresponding author: E-mail: ybwang@126.com

(Received: 19 July 2012;

Accepted: 18 March 2013)

AJC-13130

The compound 4,4'-dichloro-6,6'-dibromo-2,2'-[(hexane-1,6-diylidioxy)bis(nitrilomethylidyne)]diphenol with the molecular formula  $C_{20}H_{20}N_2O_2Br_2Cl_2$ , is central symmetric structure and two phenyl rings in each molecule are parallel to each other. Two intramolecular O2-H2...N1 hydrogen bonds involving the hydroxyl groups and oxime N atoms generate two S(6) ring motifs in each molecule. In the crystal structure, each molecule interlinks two neighboring molecules into a 1D infinite chain supramolecule through two intermolecular C8-H8...Cl1 hydrogen bonds.

**Key Words:** Salen-type bisoxime, Synthesis, Crystal structure.

### INTRODUCTION

Salen and its analogues are most versatile chelate ligands in inorganic and organometallic chemistry. Their complexes can be used in many areas, such as medical imaging<sup>1</sup>, inorganic biochemistry<sup>2</sup>, optical materials<sup>3</sup> and catalysis<sup>4</sup>. In this paper, we report on the synthesis and crystal structure of a salen-type bisoxime compound, 4,4'-dichloro-6,6'-dibromo-2,2'-[(hexane-1,6-diylidioxy)bis(nitrilomethylidyne)]diphenol.

### EXPERIMENTAL

3-Bromo-5-chloro-2-hydroxybenzaldehyde was purchased from Alfa Aesar and used without further purification. 1,6-Bis(aminoxy)hexane was synthesized according to an analogous method reported earlier<sup>5</sup>. The other reagents and solvents were analytical grade reagents from Tianjin Chemical Reagent Factory. C, H and N analyses were carried out with a GmbH VariuoEL V3.00 automatic elemental analyzer. X-ray single crystal structure was determined on a Bruker Smart 1000 CCD area detector. Melting points were measured by the use of a microscopic melting point apparatus made in Beijing Taike Instrument Limited Company and the thermometer was uncorrected.

**General procedure:** 4,4'-Dichloro-6,6'-dibromo-2,2'-[(hexane-1,6-diylidioxy)-bis(nitrilomethylidyne)]diphenol was synthesized according to an analogous method reported earlier<sup>6</sup>. To an ethanol solution (8 mL) of 3-bromo-5-chloro-2-hydroxybenzaldehyde (158 mg, 0.67 mmol) was added an ethanol solution (3 mL) of 1,6-bis(aminoxy)hexane (49.0 mg,

0.33 mmol). The reaction mixture was stirred at 328 K for 4 h. The formed precipitate was separated by filtration and washed successively with ethanol/*n*-hexane(1:4) and *n*-hexane. The product was dried under vacuum to yield 145.1 mg of the title compound. Yield, 75.4 %. m.p. 493-494 K. Anal. calcd. (%) for  $C_{20}H_{20}N_2O_2Br_2Cl_2$ : C, 41.20; H, 3.46; N, 4.80. Found (%): C, 41.42; H, 3.54; N, 4.63.

The title compound was dissolved in chloroform/ethanol (1:1) solution and allowed to stand at room temperature for about several weeks, colourless needle-like single crystals suitable for X-ray crystallographic analysis were obtained.

**X-Ray structure determination:** The single crystal of the title compound, with approximate dimensions of 0.53 mm × 0.40 mm × 0.15 mm was placed on a Bruker Smart 1000 diffractometer equipped with Apex CCD area detector. The diffraction data were collected using a graphite monochromated  $MoK_{\alpha}$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ) at 298(2) K. The structure was solved by using the program SHELXS-97 and Fourier difference techniques and refined by full-matrix least-squares method on  $F^2$  using SHELXL-97. Details of the data collection and refinements of the title compound are given in Table-1. The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were added theoretically. CCDC:712158.

### RESULTS AND DISCUSSION

X-Ray crystallographic analysis revealed the crystal structure of the title compound. And the structure is shown in Fig. 1. Selected bond distances and angles are listed in Table-2.

TABLE-1  
CRYSTAL DATA AND STRUCTURE  
REFINEMENT FOR THE TITLE COMPOUND

Empirical formula	C <sub>20</sub> H <sub>20</sub> N <sub>2</sub> O <sub>2</sub> Br <sub>2</sub> Cl <sub>2</sub>
Formula weight	583.10
Temperature (K)	298(2)
Wavelength (Å)	0.71073
Crystal system	Triclinic
Space group	P-1
Cell dimensions, (Å, deg)	a = 4.5419(10), b = 8.0851(14), c = 15.660(2), α = 76.9600(10), β = 84.470(2), γ = 86.470(2)
Volume (Å <sup>3</sup> )	557.17(17)
Z	1
Density (calculated) (mg/m <sup>3</sup> )	1.738
Absorption coefficient (mm <sup>-1</sup> )	3.907
F <sub>(000)</sub>	290
Index ranges	-5 ≤ h ≤ 5, -9 ≤ k ≤ 5, -18 ≤ l ≤ 18
Reflections collected	2358/1946 [R <sub>(int)</sub> = 0.0477]
Independent reflections	1385
Data/restraints/parameters	1946/0/137
Goodness of fit indicator	1.063
R [I > 2σ(I)]	R <sub>1</sub> = 0.0356, wR <sub>2</sub> = 0.1026
Largest diff. peak and hole (e Å <sup>-3</sup> )	0.672 and -0.865

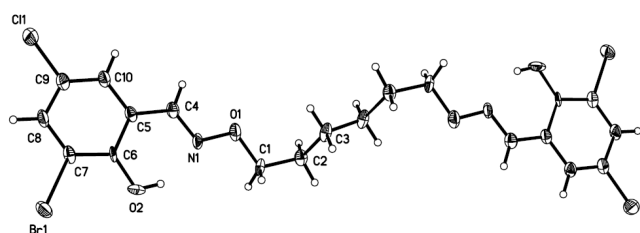


Fig. 1. Molecule structure of the title compound with atom numbering scheme. Displacement ellipsoids for non-H atoms are drawn at the 30 % probability level

The title compound crystallizes in the triclinic system, space group P-1 and the unit cell contains only one crystallographically molecule. There is a trans configuration with respect to the methylidene unit. The two phenyl rings in each molecule are parallel to each other with a perpendicular interplanar spacing of *ca.* 2.647(3) Å.

There are two intramolecular O2-H2...N1 hydrogen bonds (Table-3) involving the hydroxyl groups and oxime N atoms generate two S(6) ring motifs in each molecule<sup>7-10</sup>. In the crystal

TABLE-3  
HYDROGEN BONDS [Å, °] FOR THE TITLE COMPOUND

D-H...A	d(D-H)	d(H...A)	∠DHA	d(D...A)
O2-H2...N1	0.82	1.94	146	2.662(3)
C8-H8...Cl1	0.93	2.84	165	3.746(3)

structure, each molecule interlinks two neighboring molecules into a 1D infinite chain supramolecule through two intermolecular C8-H8...Cl1 hydrogen bonds. A view of the crystal packing for the title compound is given in Fig. 2.

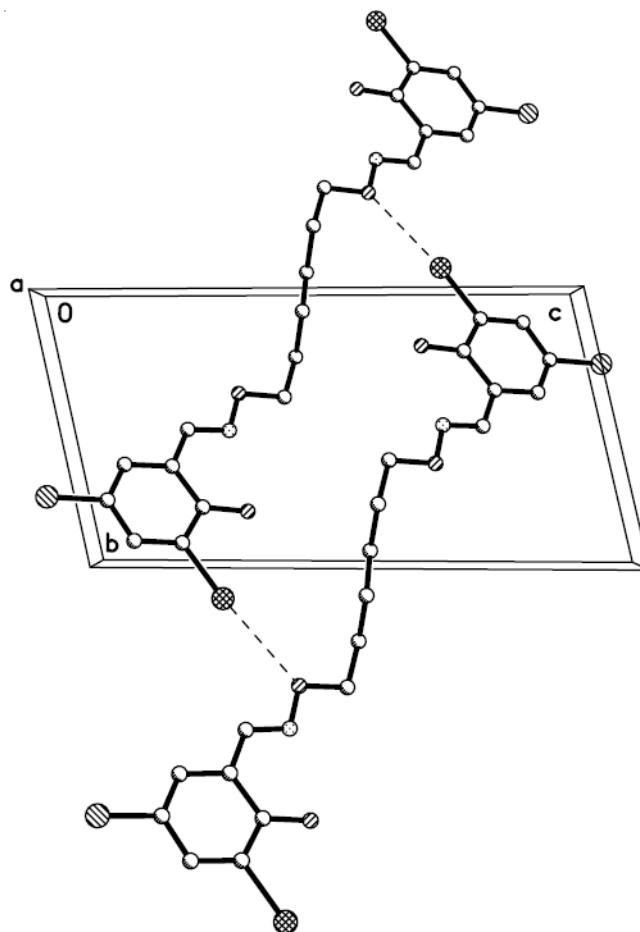


Fig. 2. A view of the crystal packing for the title compound

TABLE-2  
SELECTED BOND DISTANCES (Å) AND ANGLES (°) FOR THE TITLE COMPOUND

Bond	Lengths	Bond	Lengths	Bond	Lengths
Br(1)-C(7)	1.914(17)	C(1)-C(2)	1.49(2)	C(6)-C(7)	1.39(2)
Cl(1)-C(9)	1.74(2)	C(2)-C(3)	1.55(3)	C(7)-C(8)	1.40(3)
N(1)-C(4)	1.30(2)	C(3)-C(3)	1.49(3)	C(8)-C(9)	1.36(3)
N(1)-O(1)	1.344(19)	C(4)-C(5)	1.42(2)	C(9)-C(10)	1.34(3)
O(1)-C(1)	1.43(2)	C(5)-C(10)	1.41(3)	-	-
O(2)-C(6)	1.34(2)	C(5)-C(6)	1.45(2)	-	-
Bond	Angles	Bond	Angles	Bond	Angles
C(4)-N(1)-O(1)	115.0(15)	C(10)-C(5)-C(6)	117.7(15)	C(8)-C(7)-Br(1)	118.9(12)
N(1)-O(1)-C(1)	111.1(13)	C(4)-C(5)-C(6)	122.5(17)	C(9)-C(8)-C(7)	116.1(16)
O(1)-C(1)-C(2)	112.5(16)	O(2)-C(6)-C(7)	121.7(15)	C(10)-C(9)-C(8)	123.9(19)
C(1)-C(2)-C(3)	112.1(15)	O(2)-C(6)-C(5)	121.8(15)	C(10)-C(9)-Cl(1)	120.0(15)
C(3)-C(3)-C(2)	109.7(18)	C(7)-C(6)-C(5)	116.5(16)	C(8)-C(9)-Cl(1)	116.1(15)
N(1)-C(4)-C(5)	121.9(16)	C(6)-C(7)-C(8)	124.4(16)	C(9)-C(10)-C(5)	121.4(17)
C(10)-C(5)-C(4)	119.8(15)	C(6)-C(7)-Br(1)	116.7(14)	-	-

**REFERENCES**

1. J. Tisato, F. Refosco and F. Bandoli, *Coord. Chem. Rev.*, **135-136**, 325 (1994).
2. E.C. Niederhoffer, J.H. Timmons and A.E. Martell, *Chem. Rev.*, **84**, 137 (1984).
3. P.G. Lacroix, *Eur. J. Inorg. Chem.*, **2**, 339 (2001).
4. S. Akine, T. Taniguchi and T. Nabeshima, *Angew. Chem. Int. Ed. Engl.*, **41**, 4670 (2002).
5. W.K. Dong, G. Wang, S.S. Gong, J.F. Tong, Y.X. Sun and X.H. Gao, *Transition Met. Chem.*, **37**, 271 (2012).
6. W.K. Dong, X.N. He, H.B. Yan, Z.W. Lv, X. Chen, C.Y. Zhao and X.L. Tang, *Polyhedron*, **28**, 1419 (2009).
7. H.L. Wu, X.C. Huang, J.K. Yuan, F. Kou, F. Jia, B. Liu and Y. Bai, *Z. Naturforsch.*, **66b**, 1049 (2011).
8. W.K. Dong, Y.X. Sun, Y.P. Zhang, L. Li, X.N. He and X.L. Tang, *Inorg. Chim. Acta*, **362**, 117 (2009).
9. W.K. Dong, J.G. Duan, Y.H. Guan, J.Y. Shi and C.Y. Zhao, *Inorg. Chim. Acta*, **362**, 1129 (2009).
10. H.L. Wu, K.T. Wang, F. Kou, F. Jia, B. Liu, J.K. Yuan and Y. Bai, *J. Coord. Chem.*, **64**, 2676 (2010).