

Synthesis and Supramolecular Structure of 6,6'-Diethoxy-2,2'-[1,1'-(hexane-1,6-diylldioxy)bis(nitrilomethylidyne)]diphenol

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The compound, 6,6'-diethoxy-2,2'-[1,1'-(hexane-1,6-diylldioxy)bis(nitrilomethylidyne)]diphenol, was synthesized by the reaction of 3-ethoxysalicylaldehyde with 1,6-bis(aminoxy)hexane in ethanol. The molecule lies across a crystallographic inversion centre (symmetry code: -x, -y, -z) and adopts an E configuration with respect to the azomethine C=N bond. Within the molecule, the two benzene rings are parallel to each other and with the distance of 3.976(4) Å. Intramolecular O-H...N hydrogen bonds are formed between the hydroxyl groups and the oxime nitrogen atoms. The distances between the hydroxyl groups and the oxime nitrogen atoms $d(O2...N1) = 2.626(3)$ Å. Intermolecular C-H...O hydrogen bonds link the title molecules into an infinite 3D supramolecular network structure.

Key Words: Bisoxime compound, Synthesis, Supramolecular structure.

INTRODUCTION

Particular attention has recently been paid to the synthesis and crystal structure of bisoxime compounds and their analogues¹⁻³. These compounds can easily form complexes with transition metal ions as versatile chelating ligands^{4,5}. Some of them or their metal complexes are used in various organic reaction processes as catalysts, models of reaction centers of metalloenzymes and nonlinear optical materials. Because the oxime-type ligands are able to resist the metathesis of C=N bonds, we can use an O-alkyloxime unit (-CH=N-O-) instead of -CH=N- group in order to improve the stability of the ligands. Meanwhile, the large electronegativity of oxygen atoms can strongly affects the electronic properties of N₂O₂ coordination sphere, which can lead to different and novel properties and structures of the resulting complexes⁶. As an extension of our previous work, we report here the synthesis and crystal structure of a new salen-type bisoxime compound 6,6'-diethoxy-2,2'-[1,1'-(hexane-1,6-diylldioxy)bis(nitrilomethylidyne)]diphenol.

EXPERIMENTAL

3-Ethoxysalicylaldehyde was purchased from Alfa Aesar and used without further purification. 1,6-Bis(aminoxy)hexane was synthesized according to an analogous method reported earlier^{7,8}. The other reagents and solvents were analytical grade reagents from Shanghai Chemical Reagent

Factory. C, H and N analyses were carried out with a GmbH VariuoEL V3.00 automatic elemental analyzer. IR spectra in the range 4000-400 cm⁻¹ were recorded on a VERTEX70 FT-IR spectrophotometer using KBr pellets. X-Ray single crystal structure was determined on a Bruker Smart 1000 CCD area detector. Melting points were measured by the use of a microscopic melting point apparatus made in Beijing Taike Instrument Limited Company and the thermometer was uncorrected.

Synthesis of 6,6'-diethoxy-2,2'-[1,1'-(hexane-1,6-diylldioxy)bis(nitrilomethylidyne)]diphenol: To a hot ethanol solution (10 mL) of 3-ethoxysalicylaldehyde (347.8 mg, 2.09 mmol) was added an ethanol solution (5 mL) of 1,6-bis(aminoxy)hexane (151.1 mg, 1.02 mmol). After the solution had been stirred at 328 K for 8 h, the reaction mixture was filtered, washed successively with ethanol and ethanol/hexane (1:4), respectively. The product was dried under vacuum to yield 259.3 mg of the title compound. Yield 57.2 %. m.p. 454-456 K. Anal. calcd. (%) for C₂₄H₃₂N₂O₆: C, 64.85; H, 7.26; N, 6.30. Found (%): C, 64.97; H, 7.05; N, 6.12. IR (KBr, ν_{max} , cm⁻¹): C=N, 1610 and Ar-O, 1255. Colourless block-like single crystals suitable for X-ray diffraction studies were obtained after several weeks by slow evaporation from a dichloromethane-methanol mixed solution of the title compound.

X-Ray structure determination: The single crystal of the title compound, with approximate dimensions of 0.38 mm × 0.15 mm × 0.11 mm was placed on a Bruker Smart 1000

diffractometer equipped with Apex CCD area detector. The diffraction data were collected using a graphite monochromated MoK α radiation ($\lambda = 0.71073 \text{ \AA}$) at 298(2) K. The structure was solved by using the program SHELXS-97 and Fourier difference techniques and refined by full-matrix least-squares method on F^2 using SHELXL-97. Details of the data collection and refinements of title compound are given in Table-1. The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were added theoretically. CCDC: 878452.

TABLE-1
CRYSTAL DATA AND STRUCTURE
REFINEMENT FOR THE TITLE COMPOUND

Empirical formula	C ₂₄ H ₃₂ N ₂ O ₆
Formula weight	444.52
Temperature (K)	298(2)
Wavelength (\AA)	0.71073
Crystal system	Monoclinic
Space group	P2 ₁ /c
Cell dimensions, (\AA , deg)	a = 4.8160(3), b = 18.6909(17), c = 13.2091(12), $\beta = 102.4990(10)$
Volume (\AA^3)	1160.84(17)
Z	2
Density (calculated) (mg/m^3)	1.272
Absorption coefficient (mm^{-1})	0.091
$F(000)$	476
Index ranges	-5 $\leq h \leq$ 5, -22 $\leq k \leq$ 22, -15 $\leq l \leq$ 8
Reflections collected	5799
Independent reflections	2046 [$R_{\text{int}} = 0.0716$]
Data/restraints/parameters	2046/0/146
Goodness of fit indicator	1.028
R [$I > 2\sigma(I)$]	$R_1 = 0.0533$, $wR_2 = 0.1158$
Largest diff. peak and hole ($e \text{ \AA}^{-3}$)	0.152 and -0.112

RESULTS AND DISCUSSION

X-Ray crystallographic analysis revealed the crystal structure of the title compound. And the structure is shown in Fig. 1.

Selected bond distances and angles are listed in Table-2. The crystal structure of the title compound is built up by only the C₂₄H₃₂N₂O₆ molecules and all bond lengths and angles are in normal ranges^{6,7}. The molecule lies across a crystallographic inversion centre (symmetry code: -x, -y, -z) and adopts an E configuration with respect to the azomethine C=N bond. Within the molecule, the two benzene rings are parallel to each other and with the distance of 3.976(4) \AA .

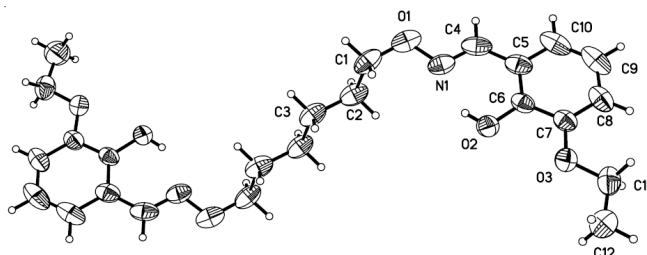


Fig. 1. Molecule structure of the title compound with atom numbering scheme. Displacement ellipsoids for non-hydrogen atoms are drawn at the 30 % probability level

In the crystal structure, intramolecular O-H \cdots N hydrogen bonds ($d(\text{O2-H2}) = 0.82 \text{ \AA}$, $d(\text{H2}\cdots\text{N1}) = 1.91 \text{ \AA}$, $d(\text{O2}\cdots\text{N1}) = 2.626(3) \text{ \AA}$, $\angle \text{O2-H2}\cdots\text{N1} = 146^\circ$) are formed between the hydroxyl groups and the oxime nitrogen atoms⁹⁻¹³. Intermolecular C-H \cdots O hydrogen bonds link the neighboring molecules into an infinite three-dimensional supramolecular network structure. The molecules of the title compound are linked by a pair of intermolecular C9-H11B \cdots O2 hydrogen bond interactions into a 1D infinite zigzag chain along the a-axis (Fig. 2). Furthermore, this linkage is further stabilized by a pair intermolecular C9-H9 \cdots O2 hydrogen bond interactions to form an infinite 2D crapy layer parallel to the bc plane (Fig. 3). To sum up, with the help of intermolecular C-H \cdots O hydrogen-bonding interactions, the crystal packing shows a self-assembling 3D supramolecular network structure¹⁴⁻¹⁶.

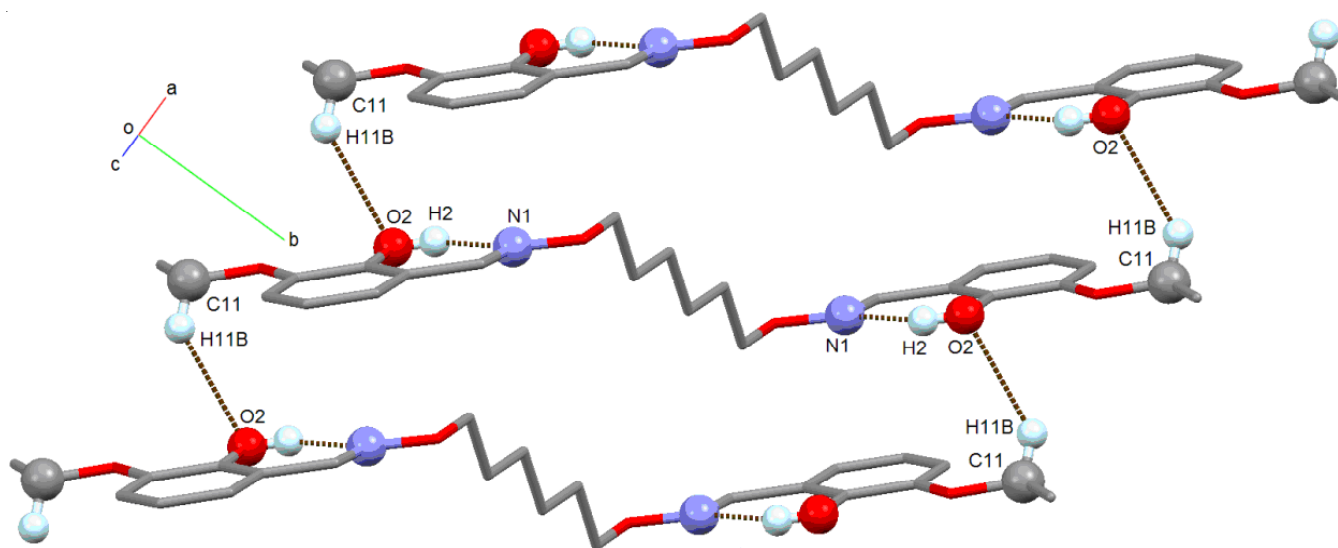


Fig. 2. Part of the supramolecular structure containing intramolecular O2-H2 \cdots N1 and intermolecular C11-H11B \cdots O2 hydrogen bonds along a axis for the title compound

TABLE-2
SELECTED BOND DISTANCES (Å) AND ANGLES (°) FOR THE TITLE COMPOUND

Bond	Lengths	Bond	Lengths	Bond	Lengths
N(1)-C(4)	1.284(4)	C(1)-C(2)	1.512(4)	C(6)-C(7)	1.405(4)
N(1)-O(1)	1.399(3)	C(2)-C(3)	1.514(4)	C(7)-C(8)	1.354(4)
O(1)-C(1)	1.428(3)	C(3)-C(3) ^{#1}	1.528(5)	C(8)-C(9)	1.373(4)
O(2)-C(6)	1.360(3)	C(4)-C(5)	1.461(4)	C(9)-C(10)	1.372(5)
O(3)-C(7)	1.356(3)	C(5)-C(6)	1.373(4)	C(11)-C(12)	1.465(4)
O(3)-C(11)	1.424(3)	C(5)-C(10)	1.398(4)	—	—
Bond	Angles	Bond	Angles	Bond	Angles
C(4)-N(1)-O(1)	111.2(3)	C(6)-C(5)-C(10)	119.7(4)	C(8)-C(7)-C(6)	119.6(3)
N(1)-O(1)-C(1)	108.8(3)	C(6)-C(5)-C(4)	121.9(3)	O(3)-C(7)-C(6)	114.8(3)
C(7)-O(3)-C(11)	117.5(3)	C(10)-C(5)-C(4)	118.4(4)	C(7)-C(8)-C(9)	120.7(4)
O(1)-C(1)-C(2)	114.0(3)	O(2)-C(6)-C(5)	123.1(3)	C(10)-C(9)-C(8)	120.9(4)
C(1)-C(2)-C(3)	113.3(3)	O(2)-C(6)-C(7)	116.9(3)	C(9)-C(10)-C(5)	119.3(4)
C(2)-C(3)-C(3) ^{#1}	113.4(3)	C(5)-C(6)-C(7)	120.0(3)	O(3)-C(11)-C(12)	109.2(3)
N(1)-C(4)-C(5)	121.3(3)	C(8)-C(7)-O(3)	125.7(4)	—	—

Symmetry transformations used to generate equivalent atoms: ^{#1}-x+2, -y+2, -z+1.

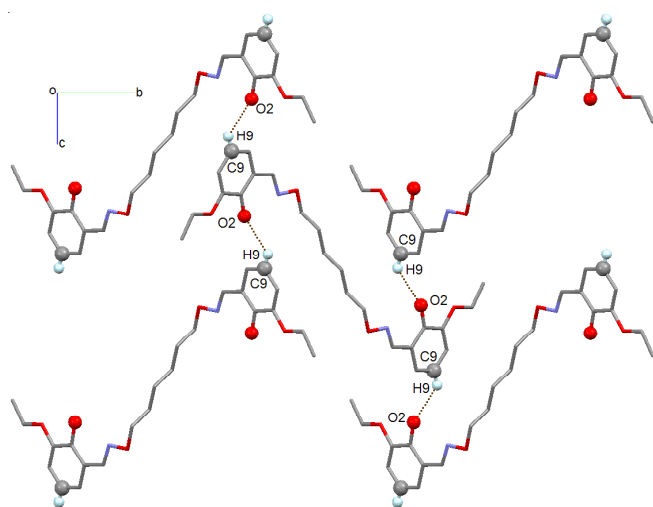


Fig. 3. Part of the supramolecular structure containing intermolecular C9-H9...O2 hydrogen bonds along bc plane for the title compound

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