

Study on the Structure-Activity Relations of Brominated Hydroxy Diphenyl Ethers Derivatives with Anilines

BIN LUO¹, HUA-HONG ZHOU², KANG-KANG YU¹, KAI-QUN WU^{1,*}, TIAN CHEN^{2,**} and YU-LIANG WANG¹

¹Faculty of Chemistry, Sichuan University, Chengdu 610065, P.R. China ²Department of Pathogenic Biology, School of Biomedical Science, Chengdu Medical College, Chengdu 610083, P.R. China

^{*}Corresponding author: Tel: +86 13708058490, E-mail: wukaiqun@tom.com; luobinhxscu@126.com ^{**}Corresponding author (For biological activities assay): Tel: +86 13540047681, E-mail: tianchen66@gmail.com

(Received: 3 April 2012;

Accepted: 14 January 2013)

AJC-12712

In order to study the influence of anilines on antibacterial activity, eight novel brominated hydroxy diphenyl ethers derivatives were designed and synthesized. The antibacterial activities of the new compounds were tested *via* agar-well diffusion method *in vitro* under different concentrations. The results showed the derivatives had antibacterial activities at the concentration 50 µg/mL against *Staphylococcus aureus* SC and *Staphylococcus aureus* ATCC26112.

Key Words: Brominated hydroxy diphenyl ethers, Structure-activity, Anilines, Antibacterial activity.

INTRODUCTION

Brominated 2-hydroxy diphenyl ethers was first isolated from the marine sponge Dysidea herbacea by Sharma et al.¹ (Fig. 1, 1) and it was reported that brominated 2-hydroxy diphenyl ethers could inhibit gram positive bacteria and gram negative bacteria^{1,2}. Afterwards, many literatures have focued on the isolation and synthesise new brominated hydroxy diphenyl ethers and their antibacterial activities³⁻⁵. These studies showed that the hydroxy was necessary for keeping antibacterial activity^{6,7} and fluorine as well as nitro substituent were favourable for enhancing antibacterial activity^{8,9}. They also emplyed that some synthetic derivatives had a special antibacterial activities such as a series of brominated dihydroxy diphenyl ethers derivatives have been designed and synthesized with good antibacterial activities^{10,11} (Fig. 1, 2). However, dioxins were found during the separation of natural brominated 2hydroxy diphenyl ethers^{12,13} (Fig. 1, **3**). In order to avoid the generation of dioxins a series of brominated 3-hydroxy (Fig. 1, 4) and 4-hydroxy (Fig. 1, 5) diphenyl ethers derivatives were designed and synthesized. The results showed that brominated 3-hydroxy and 4-hydroxy diphenyl ethers derivatives had also a good antibacterial activities and will never form dioxins14,15.

In order to avoid the generation of dioxins and to study the influence of anilines on antibacterial activity systematically, a series of novel brominated 4-hydroxy diphenyl ethers derivatives containing aniline have been designed and synthesized. The antibacterial activities of the target compounds have been tested *via* the agar-well diffusion method *in vitro*. The results showed that target compounds had some antibacterial activities against *Staphylococcus aureus* SC, *Staphylococcus aureus* ATCC26112. The synthetic route is shown in **Scheme-I**.



Fig. 1. Brominated hydroxy diphenyl ethers and related derivatives



EXPERIMENTAL

Melting points were recorded on an XRC-1 melting point apparatus (Sichuan University Instrument Inc., Chengdu, China) without being corrected. ¹H NMR spectra were run on a Varian INOVA-400 spectrometer (Varian Inc., Palo Alto, CA, USA) with CDCl₃ as the solvent and TMS as the internal standard. Mass spectra were recorded with an Agilent 6210 (DOF-MAS) spectrometer (Agilent Inc., Santa Clara, CA, USA) using the electrospray ionization (ESI) method. IR spectra were recorded with a Perkin-Elmer 16PC-FT instrument (Perkin-Elmer Inc., Norwalk Conn, CA, USA). Compounds **1** and **2** were commercially available, compounds **3** and **4** were synthesized according to the literature¹⁶.

General preparation procedure of 5a-b: A mixture of 4 (1 mmol) and NaOH (5 mmol) in solvent (20 mL, dry ethanol:water = 3:1) was stirred at 0-5 °C, then dimethyl sulfate (6 mmol) was added dropwise. Reaction solution was stirred at 0-15 °C for 24 h. Then the pH was adjusted to 7 with diluted hydrochloric acid, 20 mL water was added and the mixture was extracted with ethyl acetate. The organic layer was filtered, dried and evaporated *in vacuo* to give the crude product. The crude product was chromatographed on silica gel using petroleum ether/ethyl acetate (16:1) as eluent to afford a pure product (yield 50-55 %).

General preparation procedure of 5c-g: A mixture of 4 (1 mmol), dry Na₂CO₃ (3 mmol) and RBr (3-6 mmol) in dry DMF (20 mL) was stirred at 60~110 °C for 3~24 h, cooled. 20 mL water was added and the mixture was extracted with ethyl acetate. The organic layer was filtered, dried and evaporated *in vacuo* to give the crude product. The crude product was chromatographed on silica gel using petroleum ether/ethyl acetate (16:1) as eluent to afford a pure product (yield 60-92 %).

General preparation procedure of 6a-g: A mixture of **5a-g** (1 mmol) and 40 % HBr (6 mmol) in acetic acid (20 mL) was refluxed at 125 °C for 9~11 h, cooled, the pH was adjusted to 7 with saturated NaHCO₃ solution and the mixture was extracted with ethyl acetate. The organic layer was filtered, dried and evaporated *in vacuo* to give the crude product. The crude product was chromatographed on silica gel using petroleum ether/ethyl acetate (4:1) as eluent to afford a pure product (yield 45-85 %).

General preparation procedure of 7a-h: 6a-g (1 mmol) was dissolved in ethyl acetate/dichloromethane (1:1, 20 mL) and the mixture was heated to reflux. Liquid bromide (6 mmol) was dissolved in ethyl acetate/dichloromethane (1:1, 4 mL) and the solution was added dropwise. The reaction mixture was refluxed for 7 h, cooled, the reaction mixture was washed by 5 % Na₂S₂O₃, saturated NaHCO₃ solution, water. The organic layer was filtered, dried and evaporated *in vacuo* to give the crude product. The crude product was chromatographed on silica gel using petroleum ether/ethyl acetate (20:1) as eluent to afford a pure product (yield 45-55 %).

5a, 5c are known compounds reported in literature¹⁷ and were identified by melting point test, ¹H NMR and IR, all data was the same with the literature.

3-Chloro-4-(4-methoxyphenoxy)-*N*,*N***-dimethylbenzen-amine (5b):** Yellow solid; yield:52.3 %, m.p.: 52-54 °C; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 2.93 (6H, s), 3.77 (3H, s), 6.58 (1H, dd, *J*₁= 8.8 Hz, *J*₂= 2.8 Hz), 6.77 (1H, d, *J*= 2.8 Hz), 6.81-7.03 (5H, m); IR (KBr, v_{max}, cm⁻¹) 3040, 2936, 2834, 1607, 1562, 1499, 1443, 1234, 1202, 1039, 961, 879, 828, 772.

3-Chloro-4-(4-methoxyphenoxy)-*N*,*N***-diethylbenzenamine (5d):** Yellow oilyliquid; yield: 80.2 %; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 0.93 (6H, t, J = 7.6 Hz), 3.19 (4H, q, J = 7.6 Hz), 3.77 (3H, s), 6.53 (1H, dd, $J_1 = 8.8$ Hz, $J_2 = 2.8$ Hz), 6.66 (1H, d, J = 2.8 Hz), 6.81-7.00 (5H, m); IR (KBr, v_{max} , cm⁻¹) 3046, 2970, 2835, 1607, 1556, 1498, 1376, 1356, 1232, 1200, 1039, 874, 826, 794, 756.

4-(4-Methoxyphenoxy)-*N*,*N*-**dipropylbenzenamine** (**5e**): Yellow oilyliquid; yield: 65.3 %; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 0.92 (6H, t, *J* = 7.6 Hz), 1.59 (4H, m), 3.19 (4H, t, *J* = 7.6 Hz), 3.83 (3H, s), 6.60 (2H, d, *J* = 8.8 Hz); 6.81-6.92 (6H, m); IR (KBr, v_{max}, cm⁻¹): 3045, 2960, 2873, 2835, 1612, 1500, 1462, 1370, 1225, 1100, 1037, 873, 824,746.

4-(4-Methoxyphenoxy)-*N*,*N*-dibutyl-benzenamine (5f): Yellow oily liquid; yield: 60.3 %; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 0.95 (6H, t, *J* = 7.2 Hz), 1.34 (4H, m), 1.50 (4H, m), 3.22 (4H, t, *J* = 7.6 Hz), 3.80 (3H, s), 6.60 (2H, d, *J* = 8.8 Hz), 6.81-6.90 (6H, m); IR (KBr, v_{max}, cm⁻¹): 3044, 2957, 2868, 1611, 1500, 1462, 1369, 1277, 1227, 1105, 1037, 824.

3-Chloro-4-(4-methoxyphenoxy)-*N*,*N***-dibenzylbenzenamine (5g):** Colourless oilyliquid; yield: 91.3 %; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 3.77 (3H, s), 4.67 (4H, s), 6.56 (1H, dd, J_1 = 8.8 Hz, J_2 = 3.2 Hz), 6.79-6.91 (6H, m), 7.07-7.39 (10H, m); IR (KBr, v_{max} , cm⁻¹): 3029, 2946, 2833, 1604, 1497, 1452, 1393, 1361, 1230, 1200, 1030, 957, 883, 821, 733, 696. **4-[4-(Dimethylamino)phenoxy]phenol (6a):** White solid; yield: 84.7 %, m.p.: 133-135 °C; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 2.93 (6H, s), 4.61 (1H, s), 6.76 (2H, d, J = 8.4 Hz), 6.85 (4H, d, J = 8.8 Hz), 6.92 (2H, d, J = 8.8 Hz); IR (KBr, v_{max}, cm⁻¹): 3430, 3043, 2895, 2813, 1603, 1502, 1451, 1366, 1303, 1227, 1145, 1096, 1051, 932, 871, 826.

4-[2-Chloro-4-(dimethylamino)phenoxy]phenol (6b): White solid; yield: 83.6 %, m.p.: 127-129 °C; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 2.94 (6H, s), 4.53 (1H, s), 6.61-6.83 (5H, m), 6.91 (2H, d, *J* = 8.8 Hz); IR (KBr, v_{max}, cm⁻¹) 3097, 2921, 2821, 1600, 1495, 1443, 1355, 1237, 1142, 1052, 948, 879, 822, 782, 509.

4-[4-(Diethylamino)phenoxy]phenol (6c): White solid; yield: 82.0 %, m.p.: 113-115 °C; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 1.16 (6H, t, *J* = 3.6 Hz), 3.31 (4H, q, *J* = 6.8Hz), 4.87 (1H, s), 6.76 (2H, d, *J* = 8.8 Hz), 6.86 (4H, d, *J* = 8.8 Hz), 6.89 (2H, d, *J* = 9.2 Hz); IR (KBr, v_{max}, cm⁻¹): 3385, 3041, 2972, 2931, 1607, 1500, 1448, 1354, 1224, 1094, 1010, 872, 820.

4-[2-Chloro-4-(diethylamino)phenoxy]phenol (6d): Yellow oilyliquid; yield: 79.5 %; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 1.16 (6H, t, *J* = 7.2 Hz), 3.32 (4H, q, *J* = 7.2Hz), 4.58 (1H, s), 6.53 (1H, dd, *J*₁ = 8.8 Hz, *J*₂ = 2.8 Hz), 6.70 (1H, d, *J* = 2.8 Hz), 6.75 (1H, d, *J* = 9.2 Hz), 6.80 (2H, d, *J* = 8.8 Hz), 6.89 (2H, d, *J* = 8.8 Hz); IR (KBr, v_{max}, cm⁻¹): 3390, 3033, 2972, 2931, 1605, 1557, 1505, 1356, 1233, 1096, 1045, 1013, 924, 874, 828, 768, 693, 607, 512.

4-[4-(Dipropylamino)phenoxy]phenol (6e): Yellow oilyliquid; yield: 73.8 %; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 0.92 (6H, t, *J* = 7.6 Hz), 1.59 (4H, m), 3.19 (4H, t, *J* = 7.6 Hz), 4.60 (1H, s), 6.60 (2H, d, *J* = 8.8 Hz); 6.75 (2H, d, *J* = 8.8 Hz), 6.84-6.87 (4H, m); IR (KBr, v_{max}, cm⁻¹): 3404, 3046, 2962, 2931, 2873, 1608, 1500, 1460, 1364, 1222, 1098, 1046, 873, 818,757.

4-[4-(Dibutylamino)phenoxy]phenol (6f): Yellow oilyliquid; yield: 70.1 %; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 0.95 (6H, t, *J* = 7.2 Hz), 1.34 (4H, m), 1.50 (4H, m), 3.22 (4H, t, *J* = 7.6 Hz), 6.61 (2H, d, *J* = 8.4 Hz), 6.75 (2H, d, *J* = 8.8 Hz), 6.85-6.89 (4H, m); IR (KBr, v_{max}, cm⁻¹): 3044, 2957, 2868, 1611, 1500, 1462, 1369, 1277, 1227, 1105, 1037, 824.

4-[2-Chloro-4-(dibenzylamino)phenoxy]phenol (6g): Colourless oilyliquid; yield: 45.3 %; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 4.30 (4H, s), 4.60 (1H, s), 6.63 (1H, d, *J* = 8.8 Hz), 6.76 (1H, d, *J* = 3.2 Hz), 6.88-7.01 (5H, m), 7.28-7.39 (10H, m); IR (KBr, v_{max}, cm⁻¹): 3335, 3030, 2874, 1604, 1504, 1449, 1357, 1250, 1095, 1053, 1006, 959, 884, 836, 801, 737, 698, 405.

2,6-Dibromo-4-[4-(dimethylamino)phenoxy]phenol (**7a):** Yellow solid; yield:15.3%; m.p.: 126-128 °C; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 2.78 (6H, s), 5.36 (1H, s), 6.88 (2H, d, *J* = 2.8 Hz), 7.01 (2H, d, *J* = 5.6 Hz), 7.16 (2H, s); IR (KBr, v_{max}, cm⁻¹) 3503, 3069, 2929, 2866, 2784, 1592, 1498, 1415, 1324, 1267, 1212, 1185, 1034, 916, 883, 817, 781, 569; HR-MS (ESI): Calcd. for C₁₄H₁₄NO₂Br₂ [M+H]⁺: 385.9392, 387.9372, 389.9351; Found: 385.9381, 387.9363, 389.9345. **2,6-Dibromo-4-[3-bromo-5-chloro-4-(dimethylamino)phenoxy]phenol (7b):** White solid; yield: 47.7 %, m.p.: 109-111 °C; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 2.80 (6H, s), 5.68 (1H, s), 7.08 (1H, s), 7.15 (1H, s), 7.20 (1H, s), 7.30 (1H, s); IR (KBr, v_{max}, cm⁻¹): 3284, 3081, 2960, 2873, 2841, 2797, 1593, 1564, 1464, 1400, 1355, 1317, 1216, 1184, 1134, 1072, 1045, 970, 932, 887, 852, 801, 768, 733, 681, 580, 517, 466, 422; HR-MS (ESI): Calcd. for C₁₄H₁₂NO₂Br₃Cl [M+H]⁺: 497.8107, 499.8087, 501.8067; Found: 497.8091,499.8069, 501.8048.

2,6-Dibromo-4-[4-(diethylamino)phenoxy]phenol (7c): White solid; yield: 48.9 %, m.p.: 103-105 °C; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 1.17 (6H, t, *J* = 6.8 Hz), 3.34 (4H, q, *J* = 6.8 Hz), 5.58 (1H, s), 6.66 (2H, d, *J* = 8.8 Hz), 6.88 (2H, d, *J* = 9.2 Hz), 7.08 (2H, s); IR (KBr, v_{max}, cm⁻¹): 3501, 3076, 2970, 2928, 1606, 1510, 1465, 1402, 1320, 1263, 1218, 1150, 1011, 934, 820, 792, 742, 576; HR-MS (ESI): Calcd. for C₁₆H₁₈NO₂Br₂ [M+H]⁺: 413.9705, 415.9685, 417.9664; Found: 413.9698, 415.9706, 417.9661.

2,6-Dibromo-4-[2-chloro-4-(diethylamino)phenoxy]phenol (7d): Yellow solid; yield: 54.3 %; m.p.: 106-108 °C; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 1.18 (6H, t, *J* = 7.2 Hz), 3.33 (4H, q, *J* = 7.2 Hz), 5.57 (1H, s), 6.53 (1H, d, *J* = 8.8 Hz), 6.68 (1H, d, *J* = 2.4 Hz), 6.91 (1H, d, *J* = 8.8 Hz), 7.02 (2H, s); IR (KBr, v_{max}, cm⁻¹): 3517, 3073, 2969, 2927, 1607, 1562, 1503, 1466, 1403, 1356, 1318, 1276, 1233, 1196, 1150, 1071, 1042, 1016, 930, 800, 752, 695, 580, 436; HR-MS (ESI): Calcd. for C₁₆H₁₇NO₂Br₂Cl [M+H]⁺: 447.9315, 449.9295, 451.9274; Found: 447.9305, 449.9284, 451.9267.

2,6-Dibromo-4-[4-(dipropylamino)phenoxy]phenol (**7e**): Yellow oilyliquid; yield: 48.7 %; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 0.93 (6H, t, *J* = 7.2 Hz), 1.63 (4H, m, *J* = 7.6 Hz), 3.22 (4H, t, *J* = 7.6 Hz), 5.55 (1H, s), 6.60 (2H, d, *J* = 9.2 Hz), 6.86 (2H, d, *J* = 9.2 Hz), 7.07 (2H, s); IR (KBr, v_{max}, cm⁻¹): 3505, 3077, 2960, 2872, 1607, 1565, 1510, 1465, 1402, 1321, 1218, 1150, 1101, 1047, 932, 819, 744, 549; HR-MS (ESI): Calcd. for C₁₈H₂₂NO₂Br₂ [M+H]⁺: 442.0018, 443.9998, 445.9977; Found: 442.0013, 443.9999, 445.9976.

2,6-Dibromo-4-[4-(dibutylamino)phenoxy]phenol (7f): Yellow oilyliquid; yield: 40.2 %; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 0.96 (6H, t, *J* = 7.6 Hz), 1.36 (4H, m), 1.57 (4H, m), 3.24 (4H, t, *J* = 7.6 Hz), 5.56 (1H, s), 6.60 (2H, d, *J* = 8.8 Hz), 6.86 (2H, d, *J* = 8.8 Hz), 7.08 (2H, s); IR (KBr, v_{max}, cm⁻¹): 3511, 3078, 3044, 2956, 2866, 1608, 1565, 1510, 1465, 1402, 1369, 1321, 1215, 1149, 1109, 1051, 1008, 931, 818, 782, 744, 697, 578; HR-MS (ESI): Calcd. for C₂₀H₂₆NO₂Br₂ [M+H]⁺: 470.0331, 472.0311, 474.0290; Found: 470.0334, 472.0306, 474.0297.

2,6-Dibromo-4-[5-bromo-2-chloro-4-(dibenzylamino)phenoxy]phenol (7g): White solid; yield: 49.9 %, m.p.: 180-182 °C; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 4.67 (4H, s), 5.64 (1H, s), 7.02 (1H, s), 7.08 (3H, d, *J* = 6.8 Hz), 7.15-7.29 (10H, m); IR (KBr, v_{max}, cm⁻¹): 3476, 3381, 3074, 2923, 2853, 1605, 1568, 1451, 1409, 1383, 1309, 1278, 1218, 1189, 1155, 1067, 973, 922, 848, 779, 737, 713, 582, 532, 493, 454; HR-MS (ESI): Calcd. for C₂₆H₂₀NO₂Br₂Cl [M+H]⁺: 649.8733, 651.8713, 653.8693; Found: 649.8718, 651.8699, 653.8683.

2,6-Dibromo-4-[2-bromo-4-(dimethylamino)phenoxy]phenol (7h): White solid; yield: 45.4 %, m.p.: 120-122 °C; ¹H NMR spectrum (400 MHz; d₁-CDCl₃; TMS): δ (ppm) = 2.78 (6H, s), 5.70 (1H, s), 6.90 (1H, dd, J_1 = 8.8 Hz, J_2 = 2.8 Hz), 7.08 (1H, d, J = 8.8 Hz), 7.14 (1H, s), 7.2 (2H, d, J = 2.8 Hz); IR (KBr, v_{max} , cm⁻¹): 3418, 3078, 2954, 1586, 1559, 1464, 1403, 1338, 1263, 1213, 1153, 1034, 937, 864, 789, 752, 669,584, 544; HR-MS (ESI): Calcd. for C₁₄H₁₃NO₂Br₃ [M+H]⁺: 463.8497, 465.8477, 466.8456; Found: 463.8488, 465.8466, 466.8443.

Biological assay: Wild *Staphylcoccus aures* SC and standard *Staphylcoccus aures* ATCC26112 were used in the studies. The antibacterial activities of the target compounds *in vitro* were tested *via* agar-well diffusion method. Every sample (1000 μ g) was dissolved with ethanol (1 mL) and diluted to 50 μ g/mL with ethanol. The resulting solution was added to the flat plate of an Oxford cup which was covered with culture and kept at 37 °C for 24 h. The results of average diameters of the bacteriostatic circle are listed in Table-1.

RESULTS AND DISCUSSION

The bioactivities of compounds **7a-h** against *S. aureus* ATCC26112 and *S. aureus* SC are shown in Table-1. The results showed that all target compounds exhibited antibacterial activity at the concentration of 50.0 μ g/mL.

TABLE-1		
ANTIBACTERIAL ACTIVITY OF THE TARGET COMPOUNDS		
(7a-h) AT THE CONCENTRATION OF 50.0 µg/mL TOWARD		
S. aureus ATCC26112 and S. aureus SC		
Compound	Diameter of inhibition zone (mm)	
	S. aureus ATCC26112	S. aureus SC
7a	11	13
7b	9	10
7c	8	8
7d	11.5	11
7e	13	12
7f	7	9
7g	7	9
7h	8.5	10
Triclosan	22	20
Ethanol ^b	-	_

a: Concentration of triclosan: 20.0 µg/mL; as the positive control;

b: negative control: ethanol; diameter of the well in each plate: 6 mm

During the preparation of **5a-g**, by-products increased as the reduction of activities of RBr. So it's necessary to increase reaction temperature and time. Otherwise, RBr were vapourized along with long-playing heating, so more RBr needed during reaction process.

As shown in **Scheme-I**, the intermediates **6a-g** were synthesized by **5a-g** with HBr in HAc. On this condition the benzyl located in the nitrogen would fall off partially, so the yield of the intermediate **6g** was very low.

Conclusion

In conclusion, eight novel brominated hydroxy diphenyl ethers derivatives were designed and synthesized. Compared to 4-hydroxy (Fig. 1, **5**) diphenyl ethers derivatives¹⁵ with nitro group which exhibited good antibacterial activity at the concentration of $10 \mu g/mL$, all target compounds have similar structures with them, but their antibacterial activities are much worse. This made a further proof that nitro group is important for enhancing antibacterial activity. We will continue to study structure-activity relations of brominated hydroxy diphenyl ethers derivatives.

ACKNOWLEDGEMENTS

The authors appreciated the financial support from the National Science Foundation of China (No. 21072135), the antibacterial activity test of Chengdu Medical College (cx20100037) and the ¹H NMR analysis by Sichuan University Analytical & Testing Center.

REFERENCES

- G.M. Sharma, B. Vig and P.R. Burkholder, Marine Technology Society, Washington D.C. (1969).
- 2. G.M. Sharma and B. Vig, Tetrahedron Lett., 13, 1715 (1972).
- 3. B. Carté and D.J. Faulkner, *Tetrahedron*, **37**, 2335 (1981).
- 4. R.S. Norton, K.D. Croft and R.J. Wells, Tetrahedron, 37, 2341 (1981).
- E.N. Segraves, R.R. Shah, N.L. Segraves, T.A. Johnson, S. Whitman, J.K. Sui, V.A. Kenyon, R.H. Cichewicz, P. Crews and T.R. Holman, J. Med. Chem., 47, 4060 (2004).
- 6. J. Salvá and D.J. Faulkner, J. Nat. Prod., 53, 757 (1990).
- D. Handayani, R.A. Edrada, P. Proksch, V. Wray, L. Witte, R.W.M. Van Soest and A. Kunzmann, *J. Nat. Prod.*, **60**, 1313 (1997).
- 8. Y.L. Wang, M.L. Ma and S.H. Chen, Youji Huaxue, 25, 734 (2005).
- 9. K.Q. Wu, Y.L. Wang and H. Zheng, CN 1597657 (2005).
- 10. H. Jin, S. Chen and R.T. Hou, Youji Huaxue, 26, 1424 (2006).
- 11. S. Chen, Y. Xin and Y. Zhan, Youji Huaxue, 3, 498 (2008).
- N.K. Utkina, V,A. Denisenko, O.V. Scholokova, M.V. Virovaya, A.V. Gerasimenko, D.Y. Popov, V.B. Krasokhin and A.M. Popov, *J. Nat. Prod.*, 64, 151 (2001).
- N.K. Utkina, V.A. Denisenko, M,V. Virovaya, O,V. Scholokova and N.G. Prokof'eva, J. Nat. Prod., 65, 1213 (2002).
- X. Tang, M. Xie, Y.X. Sun, J.H. Liu, Z.C. Zhong and Y.L. Wang, *Chin. Chem. Lett.*, **20**, 435 (2009).
- 15. Y. Zou, J. Wu and R.T. Hou, *Youji Huaxue*, 1, 111(2008).
- 16. Z.X. Yang, H. Chen and Y.L. Wang, Youji Huaxue, 3, 432 (2008).
- 17. T. Saitoh and J. Ichikawa, J. Am. Chem. Soc., 127, 9696 (2005).