



Synthesis and Crystal Structure of 4,4',6,6'-Tetrachloro-2,2'-[(1,4-butylene)dioxybis(nitrilmethylidene)]diphenol

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(Received: 17 April 2012;

Accepted: 24 December 2012)

AJC-12609

The compound 4,4',6,6'-tetrachloro-2,2'-[(1,4-butylene)dioxybis(nitrilmethylidene)]diphenol with the molecular formula $C_{18}H_{16}Cl_4N_2O_4$, was synthesized by the reaction of 3,5-dichloro-2-hydroxybenzaldehyde with 1,4-bis(aminoxy)butane in ethanol. The molecule is disposed about a crystallographic centre of symmetry and the molecule adopts an extended conformation where the two phenoldoxime moieties are apart from each other. There are strong O-H...N intramolecular hydrogen bond and weak C-H...O and C-H...Cl intermolecular hydrogen bonds and strong intermolecular π - π stacking interactions (the nearest benzene rings are parallel to that of another molecule, are *ca.* 2.512(2) Å), stabilizing the two dimensional structures of the present compound.

Key Words: Salen-type bisoxime compound, Synthesis, Crystal structure, Supramolecular interaction.

INTRODUCTION

Salen-type compounds have been intensively used as versatile chelating ligands in the formation of transition metal complexes¹. Some of them or their metal complexes are used in various organic reaction processes as catalysts², models of reaction centers of metalloenzymes³, have fascinating magnetic properties⁴ and are non-linear optical materials⁵. They can also be used as biological models in understanding the structure of biomolecules and biological processes^{6,7}. Most of their important features of these compounds are their preparative accessibility, diversity and structural variability, which make them more attractive. Some salen-type bisoxime derivatives, such as 2,2'-[(1,4-butylene)dioxybis(nitrilomethylidene)]dinaphthol⁸, 4,4'-dibromo-2,2'-[ethylenedioxybis(nitrilomethylidene)]diphenol⁹, 4,4'-dibromo-2,2'-[(1,3-propylene)dioxybis(nitrilomethylidene)]diphenol¹⁰, 2,2'-[(1,4-butylene)dioxybis(nitrilomethylidene)]diphenol¹¹, 4,4'-dichloro-2,2'-[(1,4-butylene)dioxybis(nitrilomethylidene)]diphenol¹², 4,4',6,6'-tetra(*tert*-butyl)-2,2'-[(1,4-butylene)dioxybis(nitrilomethylidene)]diphenol¹³, 2,2'-[(1,4-butylene)dioxybis(nitriloethylidene)]diphenol¹⁴, 2,2'-[(propane-1,3-diyl)bis(nitrilomethylidene)]diphenol¹⁵ and 5,5'-bis(diethylamino)-2,2'-[ethylenedioxybis(nitrilomethylidene)]diphenol¹⁶ have been studied in recent years. In this paper, a new bisoxime compound, 4,4',6,6'-tetrachloro-2,2'-[(1,4-butylene)dioxybis(nitrilmethylidene)]diphenol was synthesized and characterized structurally.

EXPERIMENTAL

2-Hydroxybenzaldehyde was purchased from Alfa Aesar and used without further purification. 1,4-bis(Aminoxy)butane was synthesized according to an analogous method reported earlier³. The other reagents and solvents were analytical grade reagents from Tianjin Chemical Reagent Factory. C, H and N analyses were carried out with a GmbH VariuoEL V3.00 automatic elemental analyzer. IR spectra in the range 4000-400 cm^{-1} were recorded on a VERTEX70 FT-IR spectrophotometer using KBr pellets. The ¹H NMR spectra were recorded on a Mercury-400BB spectrometer at room temperature using $CDCl_3$ as solvent. X-Ray single crystal structure was determined on a Bruker Smart 1000 CCD area detector. Melting points were measured by the use of a microscopic melting point apparatus made in Beijing Taike Instrument Limited Company and the thermometer was uncorrected.

General procedure

4,4',6,6'-Tetrachloro-2,2'-[(1,4-butylene)dioxybis(nitrilmethylidene)]diphenol: To an ethanol solution (6 mL) of 3,5-dichloro-2-hydroxybenzaldehyde (573.0 mg, 3.00 mmol) was added an ethanol (2 mL) solution of 1,4-bis(aminoxy)butane (180.1 mg, 1.50 mmol). The mixture solution was stirred at 328 K for 4 h. The precipitate was filtered and washed successively with ethanol and ethanol/hexane (1:4), respectively. The product was dried under vacuum and to yield 480.30 mg of the title compound. Yield,

68.7 %. m.p. 475-476 K. Colourless needle-shaped single crystals suitable for X-ray diffraction studies were obtained after four weeks by slow evaporation from a tetrahydrofuran/ethyl acetate/ethanol (2:1:2) solution of 4,4',6,6'-tetrachloro-2,2'-[(1,4-butylene)dioxybis(nitrilmethylidyne)]diphenol. Anal. calcd. (%) for: $C_{18}H_{16}Cl_4N_2O_4$: C, 46.45; H, 3.40; N, 5.87. Found (%): C, 46.38; H, 3.46; N, 6.01.

X-Ray structure determination: The single crystal of the title compound, with approximate dimensions of 0.45 mm \times 0.40 mm \times 0.17 mm was placed on a Bruker Smart 1000 diffractometer equipped with Apex CCD area detector. The diffraction data were collected using a graphite monochromated MoK_{α} radiation ($\lambda = 0.71073 \text{ \AA}$) at 293(2) K. The structure was solved by using the program SHELXS-97 and Fourier difference techniques and refined by full-matrix least-squares method on F^2 using SHELXL-97. Details of the data collection and refinements of title compound are given in Table-1. The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were added theoretically. CCDC: 684099.

TABLE-1
CRYSTAL DATA AND REFINEMENT
PARAMETERS FOR THE TITLE COMPOUND

Empirical formula	$C_{18}H_{16}N_2O_4Cl_4$
Formula weight	466.13
Temperature	293(2) K
Wavelength	0.71073 \AA
Crystal system	Triclinic
Space group	P-1
Cell dimensions	$a = 4.2525(5) \text{ \AA}$, $b = 7.9844(10) \text{ \AA}$, $c = 14.3333(19) \text{ \AA}$, $\alpha = 74.5330(10)^\circ$, $\beta = 84.330(6)^\circ$, $\gamma = 80.533(2)^\circ$
Volume	$461.88(10) \text{ \AA}^3$
Z	1
Density (calculated)	1.676 mg/m^3
Absorption coefficient	0.671 mm^{-1}
$F(000)$	238
Index ranges	$-5 \leq h \leq 4$, $-9 \leq k \leq 4$, $-16 \leq l \leq 16$
Reflections collected/unique	1819/1555 [$R_{\text{int}} = 0.0387$]
Independent reflections	1395
Data/restraints/parameters	1555/0/128
Goodness of fit indicator	1.046
$R [I > 2\sigma(I)]$	$R_1 = 0.0664$, $wR_2 = 0.1252$
Largest diff. peak and hole	0.427 and $-0.419 \text{ e. \AA}^{-3}$

RESULTS AND DISCUSSION

X-Ray crystallographic analysis revealed the crystal structure of the title compound. And the structure is shown in Fig. 1. Selected bond distances and angles are listed in Table-2. The single crystal structure of the title compound is built up by only the $C_{18}H_{16}Cl_4N_2O_4$ molecules (Fig. 1), in which all bond lengths are in normal ranges. The X-ray crystallography reveals the title compound crystallizes in the triclinic system, space group P-1 with $a = 4.2525(5) \text{ \AA}$, $b = 7.9844(10) \text{ \AA}$, $c = 14.3333(19) \text{ \AA}$, $\alpha = 74.5330(10)^\circ$, $\beta = 84.330(2)^\circ$, $\gamma = 80.533(2)^\circ$ and $Z = 1$. The molecule is disposed about a crystallographic centre of symmetry and the molecule adopts an extended conformation where the two phenoldoxime moieties are apart from each other. The distance ($1.365(2) \text{ \AA}$) of two parallel benzene rings in the title compound is shorter than

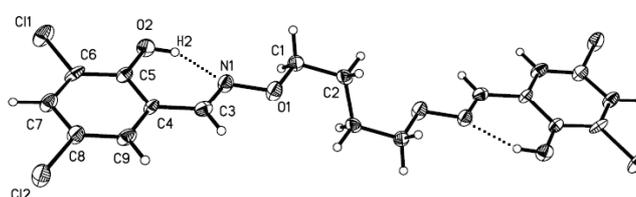


Fig. 1. Molecular structure of the title compound with atom numbering scheme. Displacement ellipsoids for non-hydrogen atoms are drawn at the 30 % probability level

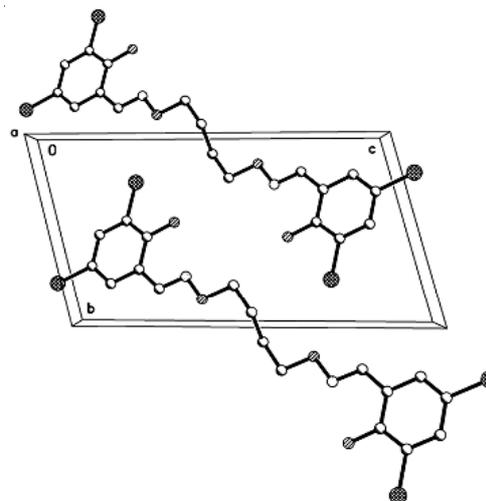


Fig. 2. Molecular packing arrangement in the unit cell

TABLE-2
SELECTED BOND LENGTHS (\AA) AND ANGLES ($^\circ$) FOR THE TITLE COMPOUND

Bond	Lengths	Bond	Lengths	Bond	Lengths
Cl(1)-C(6)	1.702(8)	O(2)-C(5)	1.315(9)	C(4)-C(5)	1.397(10)
Cl(2)-C(8)	1.714(9)	C(1)-C(2)	1.443(11)	C(5)-C(6)	1.349(12)
N(1)-C(3)	1.251(10)	C(2)-C(2)	1.532(16)	C(6)-C(7)	1.340(12)
N(1)-O(1)	1.365(9)	C(3)-C(4)	1.407(11)	C(7)-C(8)	1.378(12)
O(1)-C(1)	1.402(9)	C(4)-C(9)	1.367(11)	C(8)-C(9)	1.326(12)
Bond	Angles	Bond	Angles	Bond	Angles
C(3)-N(1)-O(1)	110.8(7)	C(5)-C(4)-C(3)	122.1(7)	C(5)-C(6)-Cl(1)	118.8(6)
N(1)-O(1)-C(1)	109.6(6)	O(2)-C(5)-C(6)	119.6(7)	C(6)-C(7)-C(8)	118.2(8)
O(1)-C(1)-C(2)	105.9(6)	O(2)-C(5)-C(4)	121.7(7)	C(9)-C(8)-C(7)	122.4(8)
C(1)-C(2)-C(2)	115.6(10)	C(6)-C(5)-C(4)	118.6(7)	C(9)-C(8)-Cl(2)	119.6(7)
N(1)-C(3)-C(4)	120.5(7)	C(7)-C(6)-C(5)	121.9(8)	C(7)-C(8)-Cl(2)	118.0(7)
C(9)-C(4)-C(5)	119.8(7)	C(7)-C(6)-Cl(1)	119.3(7)	C(8)-C(9)-C(4)	119.1(8)
C(9)-C(4)-C(3)	118.1(7)	-	-	-	-

the distance (2.028(3) Å) of parallel benzene ring planes in 4,4'-dichloro-2,2'-[(1,4-butylene)dioxybis(nitrilomethylidyne)]-diphenol¹². The oxime, chloro groups and phenolic alcohols have anticongformation, which is similar to what is observed in our previously reported salen-type bisoxime of 2,2'-[(1,4-butylene)dioxybis(nitriloethylidyne)] diphenol¹⁴. There are strong O-H...N intramolecular hydrogen bond between N1 atom and O2 atom of hydroxyl. Three C atoms, one N atom, one O atom and one H atom generate a six-membered ring through hydrogen bonds ($d(\text{O2-H2}) = 0.82 \text{ \AA}$, $d(\text{H2}\cdots\text{N1}) = 1.83 \text{ \AA}$, $d(\text{O2}\cdots\text{N1}) = 2.555(3) \text{ \AA}$, $\angle\text{O2-H2}\cdots\text{N1} = 146^\circ$), which with weak C-H...O and C-H...Cl intermolecular hydrogen bonds and strong intermolecular pi-pi stacking interactions (the nearest benzene rings are parallel to that of another molecule, are *ca.* 2.512(2) Å), stabilizing the two dimensional structures of the title compound¹⁷.

ACKNOWLEDGEMENTS

This work was supported by the Foundation of Preparative Research of Jin-Chuan Corporation (Grant No. 420032), which is gratefully acknowledged.

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