

NOTE

Crystal Structure of (NH₄)₉(P₂VW₂₂O₇₈)(H₂O)₂

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The title compound, $(NH_4)_9(P_2VW_{22}O_{78})(H_2O)_2$, is prepared by hydrothermal method. It contains PVW heteropolyacid anion, ammonium cations and water molecular. The crystal packing is stabilized by intermolecular N-H…O and N-H…N hydrogen bonds interaction.

Key Words: Hydrothermal, (NH₄)₉(P₂VW₂₂O₇₈)(H₂O)₂.

Keggin-type polyoxometalates $[XM_{12}O_{40}]^{n-}$ (X = B, P, Si, etc.; M = Mo, W) and their derivatives have been investigated for over a century because of rich structural chemistry and diverse physicochemical properties¹. Potential activities of Keggin anions as catalysts is shown by previous workers²⁻⁵. The polyoxometalates may provide structurally well-characterized surfaces formed by approximately coplanar. we sythesize the (NH₄)₉(P₂VW₂₂O₇₈)(H₂O)₂ and report its crystal structure here.



Fig. 1. Molecular structure of the title compound with atom-labling scheme

All commercially obtained reagent-grade chemicals were used without further purication. A mixture of CdCO₃ (0.1 mmol, 0.018 g), $(NH_4)_2WO_4$ (0.1 mmol, 0.029 g), NH_4VO_3 (0.1 mmol, 0.012 g), H_3PO_4 (0.2 mL) and triethylamine (1 mL) were added into 20 mL water with 20 % (v/v) ethanol and heated for 12 h at 140 °C. The solution was obtained by filtration after cooling to room temperature. Colourless block



Fig. 2. Packing diagram of three-dimensional of the title complex

single crystals suitable for X-ray measurements were obtained after a few weeks. Triethylamine and Cd^{2+} do not take part in the reaction.

The crystal structure of $(NH_4)_9(P_2VW_{22}O_{78})(H_2O)_2$ (Fig. 1) is built up of heteropolyacid cluster, ammonium cation and water molecules. The packing diagram of three-dimensional structure of the title complex is shown in Fig. 2. The crystal

data and structure refinement is shown in Table-1. In the heteropolyacid cluster, the P atom is coordinated by four O atoms(O1, O2, O3, O4) to form a tetrahedron. There are eleven terminal O atoms, twenty-four μ_2 -O atoms, two μ_3 -O atoms and one μ_4 -O atom. The μ_2 -O3 bridges P1 and W5, while μ_2 -O5 bridges W1 and V1. The μ_2 -O10 bridges W5 and V1, while μ_2 -O17 bridges W9 and V1. The μ_3 -O1 bridges P1, W1 and W2, while μ_3 -O2 bridges P1, W3 and W4. The μ_4 -O4 bridges P1, W9, W10 and W11. The five atoms (W1, W2, W4, W5, V1) are on the equatorial plane. The distance from O1 to the plane is 0.332 Å, while the distance from O2 (or O3) to the plane is 0.371 Å. The distance from O4 to the plane is 1.684 Å. The d (W-O) are in the range of 1.58-2.45 Å. The d (V-O) are in the range of 2.33-2.48 Å. Some bond lengths are shown in Table-2.

TABLE-1		
REFINEMENT FOR 100113 g		
Identification code	100113g	
Empirical formula	$H_{40}N_9O_{80}P_2VW_{22}$	
Formula weight	5603.99	
Temperature	298(2) K	
Wavelength	0.71073 Å	
Crystal system, space group	Orthorhombic, Pbcm	
Unit cell dimensions	$a = 12.9700(10) \text{ Å}, \alpha = 90^{\circ}$	
	$b = 23.120(2) \text{ Å}, \beta = 90^{\circ}$	
	$c = 38.810(3) \text{ Å}, \gamma = 90^{\circ}$	
Volume	11637.8 (16) Å ³	
Z, Calculated density	4, 3.198 Mg/m ³	
Absorption coefficient	21.826 mm ⁻¹	
F ₍₀₀₀₎	9696	
Crystal size	0.31 mm × 0.18 mm x 0.16 mm	
Theta range for data collection	1.57-25.02°	
Limiting indices	$-15 \Leftarrow h \Leftarrow 13, -27 \Leftarrow k \Leftarrow 21, -46$	
	$\Leftarrow 1 \Leftarrow 42$	
Reflections collected/unique	56934/10396 [R(int) = 0.2447]	
Completeness to theta = 25.02	99.5 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.1279 and 0.0564	
Refinement method	Full-matrix least-squares on F ²	
Data/restraints/parameters	10396/0/684	
Goodness-of-fit on F ²	1.018	
Final R indices $[I > 2\sigma(I)]$	$R1 = 0.1076$, $wR_2 = 0.2705$	
R indices (all data)	$R1 = 0.2568$, $wR_2 = 0.3720$	
Largest diff neak and hole	3 252 and -4 696 eÅ-3	

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TABLE-2	
BOND LENGTHS [Å] FOR 100113 g	
V(1)-O(10)	2.33(3)
V(1)-O(5)	2.37(3)
V(1)-O(17)	2.46(3)
V(1)-O(11)	2.48(3)
W(1)-O(23)	1.70(3)
W(1)-O(1)	2.37(3)
W(2)-O(24)	1.67(3)
W(2)-O(1)	2.34(3)
W(3)-O(25)	1.58(4)
W(3)-O(2)	2.39(3)
W(4)-O(26)	1.61(3)
W(4)-O(2)	2.39(3)
W(5)-O(27)	1.71(3)
W(5)-O(3)	2.36(3)
W(6)-O(11)	1.82(3)
W(6)-O(12)	1.98(3)
P(1)-O(1)	1.54(3)
P(1)-O(4)	1.55(3)
N(1)-H(1C)	0.9002
N(2)-H(2C)	0.9002

The crystal packing is stabilized by intermolecular N-H…O and N-H…N hydrogen bonds interaction.

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