



Antibacterial and Antifungal Activities of *Bisbenzoylthiourea* Compounds from Benzoyl Isothiocyanate and Diamines

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Five *bisbenzoylthiourea* compounds have been synthesized from benzoyl isothiocyanate and different diamines in CH_2Cl_2 medium under solid-liquid phase transfer catalysis conditions. Structures of these compounds have been characterized by elemental analyses as well as IR and ^1H NMR spectroscopy. These compounds were tested for their antibacterial and antifungal activities. The results indicated most of the compounds have good antimicrobial activities, especially in against agriculture disease fungal.

Key Words: *Bisbenzoylthiourea*, Synthesis, Antibacterial and antifungal activity.

INTRODUCTION

As we know, thiourea compounds, especially acylthiourea compounds, are of great interest because of their varied biological activities such as herbicides, insecticides, plant-growth regulators, antifungal and pharmacodynamics¹⁻⁴. Some thiourea compounds are prominent organic reaction catalyst in the metal-catalyzed asymmetric reduction of carbonyl compounds and carbonylative cyclization of *o*-hydroxyaryl-acetylenes⁵⁻⁷. In recent years, with the rapid development of coordination chemistry and supramolecular chemistry, thiourea compounds are one of many important neutral receptors and their tendency to coordinate with metal ions⁸⁻¹³. Herein, in a novel series of *bisbenzoylthioureas* (**3a-3e**) have been synthesized and characterized by elemental analyses, IR and ^1H NMR spectroscopy, their antibacterial activities have also been studied.

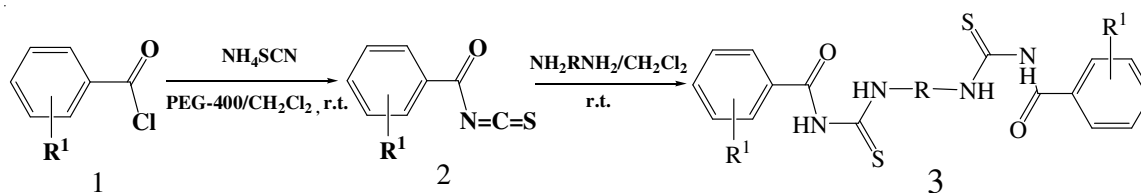
EXPERIMENTAL

Benzoyl chloride, *o*-chlorobenzoyl chloride, 1,2-dibromoethane, 1,3-dibromopropane and polyethylene glycol-400 were purchased from Alfa Aesar and used without further purification. The other reagents and solvents were analytical grade reagents from Tianjin Chemical Reagent Factory. 1,2-*Bis*(phthalimidooxy)ethane and 1,3-*bis*(phthalimidooxy)propane were synthesized according to a literature method¹⁴⁻¹⁶. C, H and N analyses were carried out with a GmbH VariuoEL V3.00 automatic elemental analyzer. IR spectra in the range 4000-400 cm^{-1} were recorded on a VERTEX70 FT-

IR spectrophotometer using KBr pellets. The ^1H NMR spectra were recorded on a Mercury-400BB spectrometer at room temperature using CDCl_3 as solvent. Melting points was measured by the use of a microscopic melting point apparatus made in Beijing Taike Instrument Limited Company and the thermometer was uncorrected.

The antimicrobial activity was determined using agar disc diffusion method by measuring the inhibition zone in mm. The microbiology stains were obtained from China Center of Industrial Culture Collection, including 4 species of bacteria: *Pseudomonas aeruginosa* NKCCMRNK10.PAO1 Δ rhII and *Escherichia coli* ACCC11864 as gram-negative Obligate aerobes bacteria, *Corynebacterium diphtheriae* CCTCCAB2010172 and *Staphylococcus aureus* ACCC01331 as gram-positive facultative aerobic bacteria. 3 species of fungi: *Gibberella* sp. CICIMF0137, *Pythium aphanidermatum* ACCC36125 and *Cochliobolus sativus* ACCC30139 as against agricultural disease fungal. Ampicillin and nystatin were served as standard antibacterial and antifungal agents, respectively.

General procedure: Benzoyl chloride and its derivatives was treated with ammonium thiocyanate under the condition of solid-liquid phase transfer catalysis using PEG-400 as the catalyst to give the corresponding benzoyl isothiocyanates, without isolation, the obtained benzoyl isothiocyanates were treated with differet diamines to afford the title compounds **3a-3e** in good-to-excellent yield^{17,18}. Synthetic route to benzoylthiourea derivatives **3a-3d** is shown in Fig. 1. All the synthesized compounds have been characterized by elemental analyses as well as IR and ^1H NMR spectroscopy (Tables 1 and 2).



3a: R¹=H, R=-CH₂-CH₂-;

3b: R¹=H, R=-CH₂-CH₂-CH₂-CH₂-;

3c: R¹=2-Cl, R=-CH₂-CH₂-CH₂-CH₂-CH₂-CH₂-;

3d: R¹=H, R=-O-CH₂-CH₂-O-;

3e: R¹=H, R=-O-CH₂-CH₂-CH₂-O-;

Fig. 1. Synthetic route to benzoylthiourea derivatives **3a-3e**

TABLE-1
COLOUR, YIELDS, MELTING POINTS AND ANALYTICAL DATA OF SYNTHESIZED COMPOUNDS **3a-3e**

Compound	Colour	m.p. (°C)	Yield (%)	m.f. (m.w.)	Elemental analysis (%): Found (calcd.)		
					C	H	N
3a	Colourless	214-215	82.3	C ₁₈ H ₁₈ N ₄ O ₂ S ₂ (386.49)	55.90 (55.94)	4.59 (4.69)	14.28 (14.50)
3b	Colourless	196-198	81.8	C ₂₀ H ₂₂ N ₄ O ₂ S ₂ (414.54)	57.91 (57.95)	5.47 (5.35)	13.36 (13.52)
3c	Colourless	176-178	90.6	C ₂₂ H ₂₆ N ₄ O ₂ S ₂ (442.6)	59.86 (59.70)	5.90 (5.92)	12.39 (12.66)
3d	Colourless	122-123	65.6	C ₁₈ H ₁₈ N ₄ O ₄ S ₂ (418.49)	51.43 (51.66)	4.60 (4.34)	13.71 (13.39)
3e	Colourless	117-118	51.8	C ₁₉ H ₂₀ N ₄ O ₄ S ₂ (432.52)	52.62 (52.76)	4.95 (4.66)	13.23 (12.95)

TABLE-2
IR AND ¹H-NMR SPECTRAL DATA OF SYNTHESIZED COMPOUNDS **3a-3e**

Compound	IR (ν, cm ⁻¹)	¹ HNMR (DMSO-d ₆ , δ/ppm)
3a	3422,3234(NH); 1668(C=O); 1158(C=S)	3.98 (t, 4H, CH ₂); 7.36-7.98 (m, 10H, C ₆ H ₅); 10.95 (s, 2H, NH); 11.31 (s, 2H, NH).
3b	3416,3222(NH); 1672(C=O); 1146(C=S)	1.70 (t, 4H, CH ₂); 3.68 (t, 4H, CH ₂); 7.45-7.92 (m, 10H, C ₆ H ₅); 10.91 (s, 1H, NH); 11.02 (s, 1H, NH).
3c	3243(NH); 1668(C=O); 1153(C=S)	1.45 (t, 4H, CH ₂); 1.65 (t, 4H, CH ₂); 3.60 (t, 4H, CH ₂); 7.45-7.93 (m, 10H, C ₆ H ₅); 10.89 (s, 2H, NH); 11.01 (s, 2H, NH).
3d	3329,3175(NH); 1680(C=O); 1178(C=S)	4.50 (s, 4H, CH ₂); 7.51 (t, J = 10.2 Hz, 4H, ArH); 7.61 (d, J = 10.2 Hz, 2H, ArH); 7.83 (d, J = 10.8 Hz, 4H, ArH); 8.93 (s, 2H, HN); 12.90 (s, 2H, HN).
3e	3203(NH); 1677(C=O); 1177(C=S)	2.18 (d, J = 8.8 Hz, 2H, CH ₂); 4.40 (t, J = 8.2 Hz, 4H, CH ₂); 7.52 (t, J = 10.2 Hz, 4H, ArH); 7.62 (d, J = 9.2 Hz, 2H, ArH); 7.85 (d, J = 9.2 Hz, 4H, ArH); 8.90 (s, 2H, HN); 12.78 (s, 2H, HN).

Synthesis of N,N'-(1,4-tetramethylene)bisbenzoylthiourea (3a), N,N'-(1,4-tetramethylene)bisbenzoylthiourea (3b) and N,N'-(1,6-hexamethylene)bisbenzoylthiourea (3c): 1.41 g (0.01 mol) of benzoyl chloride was reacted with 1.15 g (0.015 mol) of ammonium thiocyanate in 25 mL of CH₂Cl₂ under solid-liquid phase transfer catalysis conditions, using 0.18 g of 3% polyethylene glycol-400 as the catalyst, to give the corresponding benzoyl isothiocyanate after stirring for 1 h at the room temperature, a white precipitate was formed and the white solution turned to yellow, filtered and washed with CH₂Cl₂, which was reacted with a CH₂Cl₂ (15 mL) solution of 0.0045 mol of diamines (ethanediamine/1,4-butanediamine/1,6-hexamethylenediamine) at the room temperature, after stirring for 2.5 h. The solid isolated was separated from the liquid phase by filtration, washed successively with CH₂Cl₂ and H₂O, respectively, the product was dried under reduced pressure and purified with recrystallization from DMF to obtain the title compound **3a-3c**.

Synthesis of N,N'-(ethylenedioxy)bisbenzoylthiourea (3d) and N,N'-(1,3-propylenedioxy)bisbenzoylthiourea (3e): Benzoyl chloride (0.72 g, 0.005 mol) was reacted with ammonium thiocyanate (0.57 g, 0.0075 mol) in CH₂Cl₂ (15 mL) solution under solid-liquid phase transfer catalysis, using polyethylene glycol-400 (0.09 g) as the catalyst, to give a CH₂Cl₂ solution of benzoyl isothiocyanate, which was reacted with a CH₂Cl₂ (15 mL) solution of 0.0025 mol of diamines (1,2-bis(aminoxy)ethane¹⁹⁻²²/1,3-bis(aminoxy)propane²²⁻²⁴) at the room temperature, after stirring for 4 h. The solid isolated was separated from the liquid phase by filtration, washed successively with H₂O and CH₂Cl₂, respectively, the product was dried under reduced pressure to obtain the title compounds (**3d** and **3e**).

Antimicrobial activity: All the synthesized compounds were evaluated *in vitro* for their antibacterial and antifungal activities. nutrient agar and sabour's medium were used to culture bacteria and fungal, respectively. The hot nutrient agar

TABLE-3
RESULTS OF ANTIMICROBIAL ACTIVITY OF SYNTHESIZED COMPOUNDS **3a-3e**

Compound	Diameter of zone of inhibition (mm)						
	<i>Pseudomonas aeruginosa</i>	<i>Escherichia coli</i>	<i>Corynebacterium diphtheriae</i>	<i>Staphylococcus aureus</i>	<i>Gibberella sp.</i>	<i>Pythium aphanidermatum</i>	<i>Cochliobolus sativus</i>
3a	3.1 ± 0.1	5.4 ± 0.15	6.8 ± 0.25	5.9 ± 0.2	8.1 ± 0.3	6.3 ± 0.2	8.2 ± 0.1
3b	–	–	4.3 ± 0.2	5.6 ± 0.15	7.9 ± 0.3	–	8.4 ± 0.35
3c	5.4 ± 0.15	6.9 ± 0.25	7.7 ± 0.15	6.2 ± 0.2	8.8 ± 0.2	9.3 ± 0.4	8.9 ± 0.1
3d	14.8 ± 0.6	11.9 ± 0.45	15.1 ± 0.5	12.1 ± 0.3	17.6 ± 0.7	16.8 ± 0.5	17.5 ± 0.6
3e	10.4 ± 0.35	8.9 ± 0.3	11.2 ± 0.5	8.6 ± 0.4	13.1 ± 0.4	15.6 ± 0.3	14.3 ± 0.6
Ampicillin	16 ± 0.25	15.9 ± 0.7	13.7 ± 0.4	13.6 ± 0.5	–	–	–
Nystatin	–	–	–	–	15.4 ± 0.3	18.1 ± 0.6	17.3 ± 0.5

–: Indicates no activity.

and sabour's medium solution was poured into sterilized petridishes and allowed to attain room temperature. Seed layer medium which contains the previously grown subculture was lawned into the Petri dishes. Cups were made using sterile borer of 5 mm diameter. To these cups 0.5 mL of the drug solution (50 µg/mL), standard solution were added and allowed to cool for 1 h to facilitate diffusion. The plate was incubated at 37 °C for 48 h. Zone of inhibition around wells were measured. All tests were repeated three times to confirm the results. The results are showed in Table-3.

RESULTS AND DISCUSSION

A series of bisbenzoylthiourea compounds **3a-3e** have been synthesized with good yields and the structures are confirmed by elemental analyses, IR spectra and ¹H NMR data.

All the synthesized compounds were tested for their antimicrobial activities and 7 species of microbial have been choosen. Because two classes of the compounds synthesized above (**3d** and **3e**) were oxime compounds and they were reported as oxygen eliminating agent, so they were tested by two obligate aerobes bacteria and two facultative aerobic bacteria to prove their oxygen eliminating activities. Acylthiourea compounds also reported varied biological activities especially in the aspect of agricultural disease prevention and treatment, so they were tested by 3 species of commom agricultural disease fungal to prove their antifungal activities. The results presented: compound **3d** showed the highest antibacterial and antifungal activities and close to the standard drug. For some bacteria and fungal, compound **3d** even have higher antimicrobial activities than the standard drugs like the experiment of *Corynebacterium diphtheria*, *Gibberella* sp and *Cochliobolus sativus*. Compounds **3e** and **3c** also possess higher antibacterial and antifungal activities. Compounds **3d** and **3e** were shown higher inhibition on the two Obligate aerobes bacteria and lower inhibition on the two facultative aerobic bacteria because of their oxygen eliminating activities. At the concentration of 50 µg/mL, compound **3b** have no inhibition on gram-negative bacteria like the experiment of *Pseudomonas aeruginosa* and *Escherichia coli*, it possess the lowest antibacterial and antifungal activities as compared to the other compounds. In a word, the antimicrobial experiments had indicated most of the synthesized compounds have shown good antibacterial and antifungal activities as compared to their

standard drugs, but the antifungal activity is more prominent. So the bright prospects of the compounds synthesized above in against agricultural disease fungal and the potential of antimicrobial have been brought out.

REFERENCES

1. A. Jurasek, P. Safer and V. Zvak, *Chem. Pap.*, **41**, 693 (1987).
2. S.G. Teoh, S.H. Ang, H.K. Fun and C.-W. Ong, *J. Organomet. Chem.*, **580**, 17 (1999).
3. H. Arslan, N. Duran, N.O. Sahin and N. Kulcu, *Asian J. Chem.*, **18**, 1710 (2006).
4. R. Campo, J.J. Criado, E. Garcia, M.R. Hermosa, A. Jiménez-Sánchez, J.L. Manzano, E. Monte, E. Rodríguez-Fernández and F. Sanz, *J. Inorg. Biochem.*, **89**, 74 (2002).
5. Y. Nan, H. Miao and Z. Yang, *Org. Lett.*, **2**, 297 (2000).
6. J.A.J. Breuzard, M.L. Tommasino, F. Toucard, M. Lemaire and M.C. Bonnet, *J. Mol. Catal. A: Chem.*, **156**, 223 (2000).
7. F. Touchard, F. Fache and M. Lemaire, *Tetrahedron: Asymm.*, **8**, 3319 (1997).
8. P. Buhlmann, S. Nishizawa, K.P. Xiao and Y. Umezawa, *Tetrahedron*, **53**, 1647 (1997).
9. Y.M. Zhang, W.X. Xu, Y.Q. Zhou, H. Yao and T.-B. Wei, *Acta Chim. Sin.*, **64**, 79 (2006).
10. J. Valdés-Martínez, S. Hernández-Ortega, M. Rubio, D.T. Li, J.K. Swearingen, W. Kaminsky, D.R. Kelman and D.X. West, *J. Chem. Crystallogr.*, **34**, 533 (2004).
11. Y.M. Zhang, T.B. Wei, L. Xian and L.M. Gao, *Phosphorus, Sulfur Silicon Rel. Elem.*, **179**, 2007 (2004).
12. Y.M. Zhang, L. Xian, T.B. Wei and K.B. Yu, *J. Chem. Res.*, 798 (2003).
13. E. Guillon, I. DeÂchamps-Olivier and J.P. Barbier, *Polyhedron*, **17**, 3255 (1998).
14. W.K. Dong, Y.X. Sun, S.J. Xing, Y. Wang and X.H. Gao, *Z. Naturforsch.*, **67b**, 197 (2012).
15. W.K. Dong, J.F. Tong, Y.X. Sun, S.S. Gong and L. Li, *Synth. React. Inorg. Met.-Org. Nano-Met. Chem.*, **41**, 155 (2011).
16. W.K. Dong and J.G. Duan, *J. Coord. Chem.*, **61**, 781 (2007).
17. Y.M. Zhang, L. Xian and T.B. Wei, *Acta Cryst.*, **C59**, m473 (2003).
18. W.K. Dong, X.Q. Yang, L.Q. Chai, Y.Q. Tian and J.H. Feng, *Phosphorus, Sulfur Silicon Rel. Elem.*, **183**, 1181 (2008).
19. S. Akine, T. Taniguchi and T. Nabeshima, *Angew. Chem., Int. Ed.*, **41**, 4670 (2002).
20. W.K. Dong, G. Wang, S.S. Gong, J.F. Tong, Y.X. Sun and X.H. Gao, *Transition Met. Chem.*, **37**, 271 (2012).
21. W.K. Dong, Y.X. Sun, X.N. He, J.F. Tong and J.C. Wu, *Spectrochim. Acta A*, **76**, 476 (2010).
22. W.K. Dong, Y.X. Sun, Y.P. Zhang, L. Li, X.N. He and X.L. Tang, *Inorg. Chim. Acta*, **362**, 117 (2009).
23. W.K. Dong, X.N. He, H.B. Yan, Z.W. Lv, X. Chen, C.Y. Zhao and X.L. Tang, *Polyhedron*, **28**, 1419 (2009).
24. W.K. Dong, J.G. Duan, Y.H. Guan, J.Y. Shi and C.Y. Zhao, *Inorg. Chim. Acta*, **362**, 1129 (2009).