# Gossypetin-7-0-β-D-Glucopyranoside, A Novel Flavonoid Glycoside from *Impatiens Scabrida* D.C.

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Phytochemical examination of *Impatiens scabrida* resulted in the isolation and identification of a novel flavonoid glycoside Gossypetin-7-0- $\beta$ -D-glucopyranoside.

Impatiens scabrida<sup>1</sup> D.C. (N.O. Balsaminaceae) is distributed throughout India. It is an annual herb with sessile and narrow leaves. The Ayurvedic system of medicines describes the oil of the plant to be used as a semidrying oil. The present paper deals the isolation and structural elucidation of a novel flavonoid glycoside.

The methanolic extract of the water soluble part of the 95% concentrated ethanolic extract of dried and powdered plant, when worked up by C. C. gave crude glycoside (0.48%). Methanolic extract on TLC examination showed two spots which was separated by column chromatography on Si-gel G. One of the compd. eluted with ethylacetate: methanol (1:3) gave a brown amorphous compound m.pt. 208°C, molecular formula  $C_{21}H_{20}O_{13}$ , M+480 ( $R_f = 0.64$ ). It gave positive Molisch test and responded to all the colour tests for flavonoidal glycoside<sup>2,3</sup> [1], the study of another compd is in progress.

[1] On acid hydrolysis with 7% ethanolic H<sub>2</sub>SO<sub>4</sub> gave an aglycone [2] and the sugar moiety, which was indentified as D-glucose (by Co-Pc and Co-TLC).

The aglycone  $C_{15}H_{10}O_8$ , m.pt. 308°, M+318, m/e = 290, 288, 169, 168, 140, 143, 132, crystallised in yellow needle, UV.  $\lambda_{max}^{MeOH}$  256, 270, 375 nm,  $\lambda_{max}^{AlCl_3}$  257, 272, 446 nm,  $\lambda_{max}^{NaoAC+H_3BO_3}$  266, 395 nm, IR  $\nu_{max}^{KBr}$  3388, 2916, 1680, 1600, 1280, 1114, 810 cm<sup>-1</sup>, <sup>1</sup>HNMR spectrum of hexa-acetyl derivative of the aglycone showed signals at  $\delta = 6.48$  (d, J = 2, C<sub>6</sub>-1H),

 $\delta = 7.53$  (d, J = 2.5, C<sub>2</sub>,  $-^{1}$ H),  $\delta = 6.88$  (d, J = 8.5, C<sub>5</sub>,  $-^{1}$ H),  $\delta = 7.63$  (d, d, J = 2.5, 9, C<sub>6</sub>,  $-^{1}$ H),  $\delta = 2.34$  (S, 3H, C<sub>3</sub>, -OAC),  $\delta = 2.36$  (S, 3H, C<sub>4</sub>, -OAC),  $\delta = 2.48$  (S, 3H, C<sub>3</sub>-OAC),  $\delta = 2.45$  (S, 3H, C<sub>5</sub>-OAC),  $\delta = 2.41$  (S, 3H, C<sub>7</sub>-OAC),  $\delta = 2.39$  (S, 3H, C<sub>8</sub>-OAC).

A peak at 3388 cm<sup>-1</sup> in the IR spectrum of the [2] showed the presence of OH groups in it. The number of OH groups were estimated by the acetylation with AC<sub>2</sub>O/pyridine yielded hexa acetyl derivative, thereby confirming the presence of six hydroxyl groups in it.

Formation of 4:5 dihydroxy-benzoic acid (protocate cheuic acid) on alkaline degradation of the [2] confirmed the presence of OH groups at C-3' and C-4' in the [2].

A bathochromic shift of 38 nm in the band I and hypsochromic shift of

34 nm upon addition of AlCl<sub>3</sub> and H<sub>3</sub>BO<sub>3</sub> in the UV Spectrum of [2] further confirmed the presence of OH groups at C<sub>3</sub>, and C-4' respectively<sup>4</sup>.

Formation of phloroglucinol on alkaline degradation of the [2] showed the presence of two -OH groups at C-5 and C-7 respectively.

Spectral shift<sup>5</sup> with AlCl<sub>3</sub> (bathochromic shift 24 nm) and CH<sub>3</sub>COOH (bathochromic shift 34 nm) indicated the presence of -OH groups at C-5 and C-7 respectively.

Yellow fluorescence of the [2] in UV light<sup>6</sup> and the spectral shift with AlCl<sub>3</sub> in presence of HCl relative to band (I) in MeOH indicating a free 3-OH group in the aglycone.

A characteristic colour reaction with Zn/Hg and zirconium oxychloride in citric acid<sup>7</sup> further suggested the presence of -OH group at C-3.

The [2] gave a red complex with zirconium nitrate solution soluble in HCl showing the presence of a hydroxyl group at position  $C_8^8$ , which was further confirmed by a signal at  $\delta = 2.39$  in the <sup>1</sup>H NMR spectrum of acetylated aglycone.

Formation of above degradation products established the identity of the [2] as 3:5:7:8:3':4'—hexahydroxy flavone (Gossypetin).

The aqueous hydrolysate on chromatographic study revealed the presence of only D-glucose (by CO-PC and CO-TLC). The sodium metaperiodate oxidation<sup>9</sup> of the [1] consumed 2.01 mole of periodate and liberated 1.07 mole of formic acid indicating the presence of a monosaccharide and also that it was in the pyranose form.

[1] on permethylation<sup>10</sup> followed by hydrolysis yielded 2:3:4:6-tetra-O-methyl-D-glucose (by CO-PC and CO-TLC) confirmed that the sugar was attached via its  $C_1$ -OH to the  $C_7$ -OH of the aglycone.

[1] on hydrolysis with emulsion<sup>11</sup> indicated  $\beta$ -linkage between the sugar and the aglycone thereby confirming that the glycoside was Gossypetin-7-O- $\beta$ -D-glucopyranoside which was further supported by its spectral analysis, UV  $\lambda_{\max}^{\text{MeOH}}$  254, 272, 363 nm,  $\lambda_{\max}^{\text{AlCl}_3}$  295, 364 nm,  $\lambda_{\max}^{\text{H}_3\text{BO}_3}$  224, 270, 362 nm, 1R  $\nu_{\max}^{\text{KBr}}$  3400, 2910, 1675, 1598, 1112, 1275, 815 cm<sup>-1</sup>, <sup>1</sup>H NMR  $\delta$  = 6.46 (d, J = 2, C<sub>6</sub>-1H).

 $\delta = 7.54$  (d, J = 2.5,  $C_2$ , -1H),  $\delta = 6.86$  (d, J = 8.5,  $C_5$ , -1H),  $\delta = 7.61$  (d, d, J = 2.5, 9,  $C_6$ , -1H),  $\delta = 2.31$  (s, 3H,  $C_3$ , -OAC),  $\delta = 2.35$  (S, 3H,  $C_4$ , -OAC),  $\delta = 2.47$  (s, 3H,  $C_3$ -OAC),  $\delta = 2.44$  (s, 3H,  $C_5$ -OAC),  $\delta = 2.40$  (S, 3H,  $C_8$ -OAC),  $\delta = 5.7$  (d, J = 7,  $C_{1'}$ -anomeric proton),  $\delta = 5.46$  (m, 6-protons of sugar residue),  $\delta = 2.06$  (S, 3H,  $C_{2'}$ -OAC),  $\delta = 2.10$  (S, 3H,  $C_{3'}$ -OAC),  $\delta = 2.02$  (S, 3H,  $C_{4'}$ -OAC),  $\delta = 3.92$  (S, 3H,  $C_{6'}$ -OAC), M+480, M/e = 318, 290, 288, 169, 168, 140, 134, 132.

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