Synthesis of New 5-Arylazo-3-Benzyl-2-Arylimino-4-Thiazolidinones

B.N. SINGH

Chemistry Department, Kisan Post-Graduate College Bahraich, India-271 801

A number of 5-arylazo-3-benzyl-2-arylimino-4-thiazolidinones have been synthesised by coupling 3-benzyl-2-arylimino-4-thiazolidinones with aryldiazonium chlorides. Structure of these compounds was supported by spectral studies.

INTRODUCTION

The 4-thiazolidinone nucleus has drawn the attention of various workers due to significant biological properties such as antibacterial^{1,2}, anticonvulsant³ and antifungal^{4,5}. It has been reported that introduction of azo group into 4-thiazolidinones considerably enhances their antibacterial and fungicidal activities⁶. Further sulphonamides and their derivatives possess antibacterial activity and introduction of this group into 4-thiazolidinones also augments their antibacterial activity^{7,8}.

In view of these findings, the author synthesised two series of 5-arylazo-3-benzyl-2-arylimino-4-thiazolidinones(IIa and IIb) possessing the structural requirements essential for the significant activity from 3-benzyl-2-arylimino-4-thiazolidinones(I)⁹. The compounds of the type IIa were synthesised by coupling I with phenyl diazonium chloride and that of type of IIb by coulping I with p-sulphonamidophenyl diazonium chloride.

The structure of these compounds was supported with help of I.R. seectral data. The compounds are described in Table 1.

OC — N—CH₂Ph OC — N—CH₂Ph

$$H_2C$$
 C— N—Ar Ar' —N—N—HC C— N—Ar

S
(I)

(II)

$$(II_a)$$
, $(Ar' = C_6H_5-)$ (II_b) $(Ar' = H_2NO_2S-pC_6H_4-)$

Preparation of 5-Phenylazo-3-Benzyl-2-Benzylimino-4-Thiazolidinone

To an ice-cold solution of freshly distilled aniline (2 g; 0.22 mole) in hydrochloric acid (5 ml) was added a pre-cooled solution of sodium nitrite (1.2 g; 0.017 mole). The mixture was stirred well and a few crystals of sodium acetate were added to it. The resulting diazo solution was poured slowly with stirring

TABLE 1 5-ARYLAZO-3-BENZYL-2-ARYLIMINO-4-THIAZOLIDINONES (II)

		TANTE LALVE DE LA LE PARTE LIMINA "4" ILINALO LA	-ONITATION	TIMESOF	TOTAL (II)		
Par mo) ** Y	(50) # #	Viold @	Motomior Formula	,)N	N(%)*
	₹	₹	III. pt. (C)	i iciu %	Molecular Formitula	Found	Required
1-	Phenyl	Phenyl	53	88	C22H18N4OS	14.37	14.51
2.	o-Tolyl	Phenyl	78	59	C23H20N4OS	14.09	14.00
ж	p-Tolyl	Phenyl	92	62	C23H20N4OS	14.12	14.00
4	p-Anisyl	Phenyl	92	99	C23H20N4O2S	13.54	13.46
5.	p-Phenetyl	Phenyl	82	9/	C24H22N4O2S	13.09	13.02
9	p-Fluorophenyl	Phenyl	8	25	C22H17FN4OS	14.17	13.86
7.	o-Chlorophenyl	Phenyl	8	88	C22H17CIN4OS	13.02	13.32
∞	p-Chlorophenyl	Phenyl	73	8	C22H17CIN4OS	13.28	13.32
6	p-Iodophenyl	Phenyl	117	22	C22H17IN4OS	11.16	11.14
10.	o-Hydroxyphenyl	Phenyl	105	59	C22H18N4O2S	13.86	13.93
11.	m-Hydroxyphenyl	Phenyl	145	26	C22H18N4O2S	14.15	13.93
12.	1-Naphthyl	Phenyl	110	61	C26H20N4OS	13.06	12.84
13.	2-Pyridyl	Phenyl	115	99	C21H17N5OS	17.81	18.09
14.	Cyclohexyl	Phenyl	71	73	C22H24N4OS	14.17	14.28
15.	p-Bromophenyl	p-Sulphonamidophenyl	125	73	C22H18BrN5O3S2	12.98	12.87
16.	p-lodophenyl	p-Sulphonamidophenyl	165^d	69	C22H19IN5O3S2	11.78	11.84
17.	o-Hydroxyphenyl	p-Sulphonamidophenyl	103	62	C22H19N5O4S2	14.64	14.55
18.	m-Hydroxyphenyl	p-Sulphonamidophenyl	143	26	C22H19N5O4S2	14.46	14.55
19.	p-Hydroxyphenyl	p-Sulphonamidophenyl	140^d	62	C22H19N5O4S2	14.24	14.55
20.	4-Hydroxy-2-methylphe	4-Hydroxy-2-methylphenyl p-Sulphonamidophenyl	147	73	C23H21N5O4S2	14.08	14.14
21.	5-Chloro-2-hydroxypher	-hydroxyphenyl p-Sulphonamidophenyl	145	2.2	C22H18CIN5O4S2	13.75	13.58
*All th	ne compounds were also a	*All the compounds were also analysed for C, H and satisfactory results were obtained	ory results were c	obtained.			

to an ice-cold solution of 3-benzyl-2-benzylimino-4-thiazolidinones (2g; 0.007 mole) in acetone (25 ml) and glacial acetic acid (25 ml) and the stirring was continued for half an hour. The reaction mixture was diluted with water when a coloured-precipitate was obtained. The precipitate was washed with water, dried and recrystallised from ethanol, m.pt. 78°C, yield 67% (found: C 69.27, H 5.13 and N 13.73%; calcd. for C₂₃H₂₀N₄OS: C 69.00, H 5.00 and N 14.00%).

I.R. spectrum of the compound showed characteristic absorption peaks at (cm^{-1}) : 2960, 2885 (CH₂); 1725 (C—O); 1645 (C—N); 1610, 1460 (—N—N—) and 745 (C—S).

Other 5-phenylazo derivatives synthesised by this method are listed in Table 1 along with their relevant data.

Preparation of 5-p-Sulphonamidophenylazo-3-Benzyl-2-Arylimino-4-Thiazolidinone

A solution of sulphanilamide (1.2 g, 0.007 mole) dissolved in concentrated hydrochloric acid (5 ml) and water (10 ml) was cooled in ice-bath and diazotised with sodium nitrite (1.2 g, 0.017 mole) in water (5 ml), the p-sulphonamidophenyl diazonium chloride solution thus obtained was added slowly with stirring to a previously cooled solution of 3-benzyl-2-p-tolylimino-4-thiazolidinone (2 g, 0.007 mole) in ethanol (30 ml) and acetone (15 ml) containing crystals of sodium acetate (3 g, 0.022 mole). The stirring was continued for about 1 hr. Excess of water was added to the reaction mixture. After several hrs, a coloured precipitate was obtained which was collected, washed with water, dried and recrystallised from ethanol, m.pt. 86°C, yield 62% (found: C 57.87, H 4.31 and N 14.46%; calcd. for $C_{23}H_{21}N_5O_3S_2$: C 57.62, H 4.38 and N 14.61%).

I.R. spectrum of the compound showed the characteristic absorption peaks at (cm⁻¹: 3450, 3375 (NH); 2960, 2870 (CH₂); 1740 (C—O), 1650 (C—N); 1610, 1470 (—N—N—), 1350, 1150 (SO₂) and 745 (C—S).

Similarly other 5-p-sulphonamidophenylazo derivatives were prepared and are listed in Table 1 along with their relevant data.

Biological screenings for different activities of the compounds are in progress. The results will be communicated in another paper.

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REFERENCES

1. G. Fenech and P. Monforte, Am. Chim (Nome), 48, 975 (1958).

- J. Kunugava and E. Magase, Japanese Patent, 16, 285 (1966); Chem Abstr., 66, 10918c (1967)
- 3. H.D. Troutman and L.M. Long, J. Am. Chem. Soc., 70, 3436 (1948).
- 4. R.P. Rao, Curr. Sci. (India), 35, 641 (1966).
- N.K. Sundholma and J.B. Skaptason, U.S. Patent, 1950, 2, 510, 725 Chem. Abstr., 44, 8045h (1950).
- 6. R.P. Rao, J. Indian Chem Soc., 38, 784 (1961).
- 7. B.K. Pataniak and M.K. Rout, J. Indian Chem. Soc., 32, 563 (1955).
- 8. G.N. Mahapatra and M.K. Rout, J. Indian Chem. Soc., 32, 715 (1955).
- 9. B.N. Singh and R.P. Rao, Agri. Biol. Chem. (In press)

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Contact to:

Professor (Dr.) M. M. Khater Chemistry Department Faculty of Science Cairo University Giza, Cairo, Egypt.

Tel: 727213, 727435; Fax: 727288