

## Synthesis of Some Dihydropyridino Azitidinones and Their Antimicrobial Activity

S.N. THORE, D.B. SHINDE†, D.V. MANE, S.B. BHAWSAR and M.S. SHINGARE\*

Department of Chemistry

Dr. B.A. Marathwada University, Aurangabad-431 004, India

Hydrazides of substituted 1,4-dihydropyridines have been synthesised by condensing substituted dihydropyridine dicarboxylic acid ester with hydrazine hydrate. Hydrazides were condensed with different aromatic aldehydes to give different Schiff bases. These schiff bases on treatment with chloroacetyl chloride gave azitidinones. The synthesised compounds were screened for antimicrobial activity.

### INTRODUCTION

2-Azitidinones and 4-thiazolidinone derivative are associated with diverse pharmacological and biological activities<sup>1-3</sup>. These observations prompted us to synthesise some azitidinones incorporating substituted dihydro pyridino<sup>4</sup> moiety and to study their biological activity. The 1,4-dicarboxylic acid hydrazide (I) on condensing with aldehydes gives schiff bases. The Schiff bases on treatment with chloroacetyl chloride gives 2-azitidinones 1–20. (Table-I). The steps involved in the synthesis are shown in Scheme-I.

All the compounds were screened for their antimicrobial activity against *Alternaria brassicicola*, *Fusarium udam*, *Staphylococcus* gram(+ve) and *Lactobacillus* gram(-ve). The activities are discussed in experimental part.

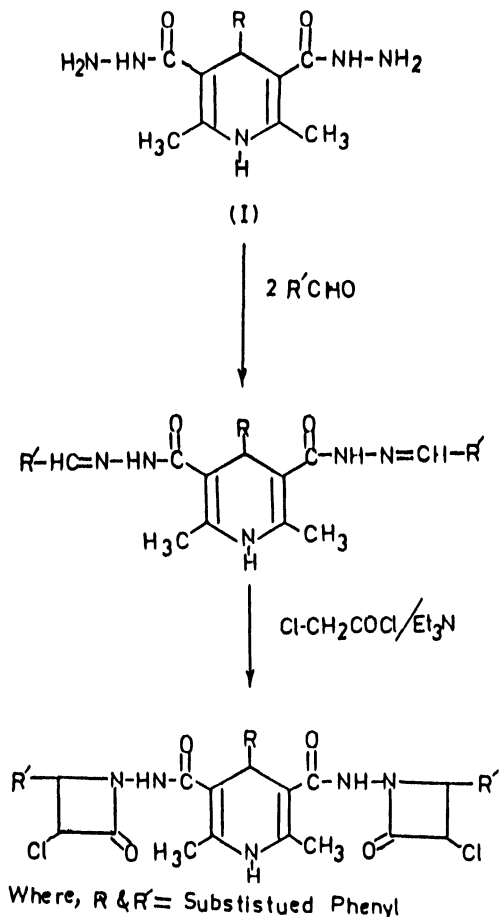
### EXPERIMENTAL

All melting points were taken in open capillary in a liquid paraffin bath and the uncorrected. The purity of all compounds was checked by TLC. IR spectra were recorded in nujol on Perkin-Elmer 1420 spectrophotometer while PMR spectra in CDCl<sub>3</sub> using TMS as an internal standard.

3,5-Bis(3'-chloro-4'-aryl-2'-azitidino-1'-yl carbonyl)-2,6-dimethyl-4-phenyl-1,4,-dihydro pyridine (1)

Schiff base of the hydrazide (I) was prepared by known procedure. The prepared Schiff base (0.01 M), chloroacetyl chloride (0.2 M) and triethanolamine (0.03 M) were refluxed in dioxane for 3 h. The reaction mixture was cooled and poured over crushed ice. The solid separated was filtered and crystallised from methanol. Yield: 75%; m.p.: 178°C (Compd No. 1.) (Table-I); IR:  $\nu_{\max}$  3320 cm<sup>-1</sup> (—NH), 1670 cm<sup>-1</sup> (C=O), 1620 cm<sup>-1</sup> (C=N), 1605 cm<sup>-1</sup> (C=C).

†Deogiri College, Aurangabad, India.



#### SCHEME-I

PMR  $\delta$  2.23 (s, 6H, —CH<sub>3</sub>), 3.59 (s, 2H, Ar—CH), 4.74 (s, 2H, 2  $\times$  CH—Cl), 5.75 (s, CH pyridyl); 7.1–7.5 (m, 15 aromatic proton), 8.2 (s, 2H, 2  $\times$  CO—NH) and 8.5–9.5 (s, 1H, —NH pyridyl).

#### Antimicrobial Activity

Compounds 1, 8, 10, 11 and 17 from Table-I were screened for their antifungal activity against *Alternaria brassicicola* and *Fusarium udum*, while for antibacterial activity against *Lactobacillus* and *E. coli* by paper disc method<sup>5</sup> at 250 ppm and 500 ppm concentration in dimethyl sulfoxide. Standard Zapkes medium was used. Filter paper discs of 5 mm size were used and the diameters of zones of inhibition formed around each disc after incubating for a period of 48 h at 30°C were recorded. Results were compared with reference to fungicide (carben-diazium) and bactericide (streptomycine). The compounds 10 and 17 were showing good activity when compared with standards because these two nuclei were incorporated with more percentage of chlorine atom which is known for its toxicity.

TABLE-I  
CHARACTERISATION DATA OF AZITIDINONES (1-20)

Comp No.	R	R'	Yield (%)	m.p.* (°C)	Mol. Formula	% N	
						Found	(Calcd)
1	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	75	178	C <sub>31</sub> H <sub>29</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>2</sub>	11.25	(11.55)
2	C <sub>6</sub> H <sub>5</sub>	4-Cl C <sub>6</sub> H <sub>4</sub>	73	169	C <sub>31</sub> H <sub>27</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>4</sub>	20.60	(20.88)
3	C <sub>6</sub> H <sub>5</sub>	4-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	74	163	C <sub>33</sub> H <sub>33</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>2</sub>	11.00	(11.19)
4	C <sub>6</sub> H <sub>5</sub>	4-OH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	76	171	C <sub>33</sub> H <sub>33</sub> N <sub>5</sub> O <sub>6</sub> Cl <sub>2</sub>	10.40	(10.66)
5	2-Cl C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	80	178	C <sub>31</sub> H <sub>28</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>3</sub>	16.50	(16.62)
6	2-Cl C <sub>6</sub> H <sub>4</sub>	4-Cl C <sub>6</sub> H <sub>4</sub>	68	168	C <sub>31</sub> H <sub>26</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>5</sub>	24.90	(25.01)
7	2-Cl C <sub>6</sub> H <sub>4</sub>	4-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	71	157	C <sub>33</sub> H <sub>32</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>3</sub>	15.70	(15.93)
8	2-Cl C <sub>6</sub> H <sub>4</sub>	4-OCH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	78	148	C <sub>33</sub> H <sub>32</sub> N <sub>5</sub> O <sub>6</sub> Cl <sub>3</sub>	15.00	(15.20)
9	4-Cl C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	73	139	C <sub>31</sub> H <sub>28</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>3</sub>	16.02	(16.22)
10	4-Cl C <sub>6</sub> H <sub>4</sub>	4-Cl C <sub>6</sub> H <sub>4</sub>	78	175	C <sub>31</sub> H <sub>26</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>5</sub>	24.70	(25.01)
11	4-Cl C <sub>6</sub> H <sub>4</sub>	4-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	70	153	C <sub>33</sub> H <sub>32</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>3</sub>	15.70	(15.93)
12	4-Cl C <sub>6</sub> H <sub>4</sub>	4-OCH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	76	172	C <sub>33</sub> H <sub>32</sub> N <sub>5</sub> O <sub>6</sub> Cl <sub>3</sub>	14.90	(15.20)
13	4-OCH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	75	158	C <sub>32</sub> H <sub>31</sub> N <sub>5</sub> O <sub>5</sub> Cl <sub>2</sub>	19.82	(20.14)
14	4-OCH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	4-Cl C <sub>6</sub> H <sub>4</sub>	78	152	C <sub>32</sub> H <sub>29</sub> N <sub>5</sub> O <sub>5</sub> Cl <sub>4</sub>	18.10	(18.34)
15	4-OCH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	4-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	71	162	C <sub>34</sub> H <sub>35</sub> N <sub>5</sub> O <sub>7</sub> Cl <sub>2</sub>	11.70	(11.91)
16	4-OCH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	4-OCH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	73	165	C <sub>34</sub> H <sub>35</sub> N <sub>5</sub> O <sub>7</sub> Cl <sub>2</sub>	11.10	(11.30)
17	3-Cl C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	76	162	C <sub>31</sub> H <sub>28</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>3</sub>	16.25	(16.62)
18	3-Cl C <sub>6</sub> H <sub>4</sub>	4-Cl C <sub>6</sub> H <sub>4</sub>	78	170	C <sub>31</sub> H <sub>26</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>5</sub>	24.55	(25.01)
19	3-Cl C <sub>6</sub> H <sub>4</sub>	4-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	72	179	C <sub>33</sub> H <sub>32</sub> N <sub>5</sub> O <sub>4</sub> Cl <sub>3</sub>	15.72	(15.93)
20	3-Cl C <sub>6</sub> H <sub>4</sub>	4-OCH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	73	168	C <sub>33</sub> H <sub>32</sub> N <sub>5</sub> O <sub>6</sub> Cl <sub>3</sub>	15.00	(15.20)

\*All compounds were crystallised from methanol.

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### REFERENCES

1. F.A. Bouffar and B.G. Christenses, *J. Am. Chem. Soc.*, **100**, 313 (1978).
2. J.C. Sheehan and E.J. Corey, *Org. React.*, **9**, 393 (1957).
3. N.S. Isaac, *Chem. Soc. Rev.*, **5**, 181 (1976).
4. B.M. Khadiolkar, V.R. Mestha and S.R. Bhayade, *Indian J. Chem.*, **33B**, 451 (1974).
5. H.W. Seely and P.J. Van Denmark, *Microbes in Action : A Laboratory Manual of Microbiology*, DBT Taraporevala Sons and Co. Pvt. Ltd., Bombay, pp. 55, 80 (1975).