NOTE

# Synthesis and Antimicrobial Activity of New 1H-3-(2-Acetoxy-4-Methoxy) Phenyl-5-Substituted-Phenyl-Pyrazolines and Their Derivatives

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Condensation of 2-acetoxy-4,4'-dimethoxy chalcone with hydrazine hydrate either in ethanol or ethylene glycol medium gave the title compounds (3). Their N-substituted derivatives 4 and 5 have been obtained by acetylation and benzoylation. The compounds 3, 4 and 5 have been screened for antimicrobial activity.

Pyrazolines are important nitrogen containing heterocycles possesing diverse biological activity<sup>1, 2</sup>. In view of potential pharmacological values, the new pyrazoline derivatives are designed to contain —COCH<sub>3</sub>/COC<sub>6</sub>H<sub>5</sub> groups at position 1 and [2-acetoxy-4-methoxy] phenyl group at position 3. 2-Hydroxy-4-methoxy-acetophenone (paenol) was condensed with various aromatic aldehydes to obtain chalcones (1), which on acetylation give 2'-acetoxychalcones (2). The 2'-acetoxy-chalcones (2) on treatment with hydrazine hydrate in either ethanol or ethylene glycol medium give the pyrazolines (3). The pyrazolines (3) were N-acetylated by their reaction with acetic acid to obtain N-acetyl derivatives (4) and N-benzoylated with benzoyl chloride to obtain N-benzoyl derivatives (5).

The compounds 3, 4 and 5 have been screened for their antimicrobial activity against Escherichia coli, Staphylococcus aureus, Pseudomonas aeruginosa, Salmonella typhi, Aerobacter aerogenes, Vibrio cholerae, Klebsiella pneumoniae, Proteus vulgris and Shigella dysentery at a concentration of 50 µg/mL by agar cup method<sup>3</sup>.

Compounds 5a-c showed higher activity against all the bacteria studied except V. cholerae. Compound 4a was effective against S. aureus, P. aeruginosa and S. typhi. Compounds 3a and 3c were effective only against P. aeruginosa and S. typhi.

All the melting points have been determined in open capillaries and are uncorrected. IR spectra were taken on a Perkin-Elmer spectrophotometer and NMR on Jeol-90 spectrometer. Compounds were routinely checked for their purity on silica gel-G TLC plates.

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CH<sub>3</sub>O OCOCH<sub>3</sub>
C-CH=CH

R<sub>1</sub>

$$R_1$$
 $R_2$ 
 $R_1$ 
 $R_2$ 
 $R_1$ 
 $R_2$ 
 $R_1$ 
 $R_2$ 
 $R_3$ 
 $R_4$ 
 $R_5$ 
 $R_6$ 
 $R_7$ 
 $R_7$ 

### Formation of 1H-3-(2-Acetoxy-4-Methoxy) Phenyl-5-Anisyl-Pyrazoline (3a)

2-Acetoxy-4,4'-dimethoxy-chalcone (2a), (0.01 mol, 3.2 g) and 99% hydrazine hydrate (0.015 mol, 0.6 mL) in ethanol (50 mL) was refluxed for 2 h. The reaction mixture was then concentrated and allowed to cocl. The resulting solid was filtered, washed with ethanol and crystallised from ethanol to obtain yellow solid (3a). m.p. 142°C, yield 80%. Found: C 70%, H 5%, N 9.03%. Calculated for  $C_{18}H_{17}O_3N_2$ : C 69.9%, H 5.5%, N 9.06%. IR (nujol) (cm<sup>-1</sup>): 3380 v(—NH); 1540–1550 v(C=N); 1620 v(O—COCH<sub>3</sub>); 1050 (O—CH<sub>3</sub> str.); 1310–1320 (O—CH<sub>3</sub> deformation). PMR (CDCl<sub>3</sub>):  $\delta$  2.35 (s, 3H, —COCH<sub>3</sub>); 3.25 (dd, 1H, >CHH<sub>A</sub>-, J<sub>AB</sub> = 18 Hz, J<sub>AX</sub> = 4 Hz); 3.52 (dd, 1H, >CHH<sub>B</sub>-, J<sub>AB</sub> = 18 Hz, J<sub>BX</sub> = 11 Hz); 3.8 (s, 6H, —OCH<sub>3</sub>); 4.8 (dd, 1H, >CH<sub>X</sub>-, J<sub>AX</sub> = 4 Hz, J<sub>BX</sub> = 11 Hz); 6.5 to 7.4 (m, 7H, Ar–H).

On the basis of properties, chemical reactions and spectral data, 3a was considered to be 1H-3 (2-acetoxy-4-methoxy) phenyl-5-anisyl pyrazoline (3a). Similarly 3b and 3c were prepared.

### Formation of 1-Acetyl-3-(2-Acetoxy-4-Methoxy) Phenyl-5-Anisyl-Pyrazoline (4a)

A mixture of 3a (0.001 mol, 0.34 g) and acetic acid (20 mL) was refluxed for 2 h. The solution was then concentrated. On cooling the solid separated was filtered, washed with water and recrystallised from ethanol to obtain white crystalline compound (4a). Yield 80%, m.p. 158°C. Found: C 68, H 5.5, N 7%. Calculated for  $C_{20}H_{20}O_4N_2$ : C 68.18, H 5.68, N 7.95%), IR (nujol) cm<sup>-1</sup>: 1665 v(N—COCH<sub>3</sub>); 1615 v(C=O); 1515 v(C=N); 1215 v(C—N). PMR (CDCl<sub>3</sub>):  $\delta$  2.35 (s, 6H, —COCH<sub>3</sub>); 3.25 (dd, 1H, >CHH<sub>A</sub>-J<sub>AB</sub> = 18 Hz, J<sub>AX</sub> = 4 Hz); 3.52 (dd, 1H, >CHH<sub>B</sub>-, J<sub>AB</sub> = 18 Hz, J<sub>BX</sub> = 11 Hz); 5.5 (dd, 1H, CH<sub>x</sub>-, J<sub>AB</sub> = 4 Hz, J<sub>BX</sub> = 11 Hz); 3.8 (s, 6H, —OCH<sub>3</sub>) 6.3 to 7.4 (m, 7H, Ar—H).

On the basis of properties, chemical reactions and spectral data, 4a was considered to be 1-acetyl-3-(2-acetoxy-4-methoxy) phenyl-5-anisyl-pyrazoline (4a). Similarly 4b and 4c were prepared.

## Formation of 1-Benzoyl-3-(2-Acetoxy-4-Methoxy) Phenyl-5-Anisyl-Pyrazoline (5a)

A mixture of 3a (0.001 mol, 0.3 g) and benzoyl chloride (0.002 mol) was dissolved in dry pyridine (10 mL) and stirred at room temperature for 1 h, after which the reaction mixture was treated with cold dil. HCl. The resulting solid was filtered, washed successively with water, cold NaOH (2%) and water and recrystallised from glacial acetic acid to get compound (5a); yield 85%, m.p. 120°C. (Found: C 72, H 5.6, N 6.00%; Calculated for  $C_{25}H_{21}O_4N_2$ : C 72, H 5.09, N 6.79%). IR (nujol) cm<sup>-1</sup>: 1660 v(C=O, N=C=O); 1610 v(C=N); 1600 (benzene ring); 1210 v(C=N); 1750 (-OCOCH<sub>3</sub>); 1350 (-OCH<sub>3</sub> str.); 1040 (-OCH<sub>3</sub> deformation). PMR (CDCl<sub>3</sub>):  $\delta$  3.35 (dd, 1H, >CHH<sub>A</sub>-, J<sub>AB</sub> = 16 Hz, J<sub>AX</sub> = 4 Hz) 3.55 (dd, 1H, >CH<sub>B</sub>A, J<sub>AB</sub> = 16 Hz, J<sub>BX</sub> = 11 Hz); 3.75 (s, 3H, -COCH<sub>3</sub>); 4.1 (s, 6H, -OCH<sub>3</sub>); 5.85 (dd, 1H, >CH<sub>x</sub>-, J<sub>AX</sub> = 4 Hz, J<sub>BX</sub> = 11 Hz); 6.5 to 8.3 (m, 12H, Ar=H).

On the basis of properties, chemical reactions and spectral data, 5a, was considered to be 1-benzoyl-3(2-acetoxy-4-methoxy) phenyl-5-anisyl-pyrazoline. Similarly 5b and 5c were prepared.

PHI SICAL DATA OF THE COMPOUNDS						
S.No.	R <sub>1</sub>	R <sub>2</sub>	m.p. (°C)	Yield in alcohol (%)	Yield in ethylene glycol (%)	Molecular formula
3a	—ОСН3	Н	142	80	85	C <sub>18</sub> H <sub>17</sub> O <sub>3</sub> N <sub>2</sub>
3b	Н	Н	105	85	88	$C_{19}H_{20}O_4N_2$
3c	—O—СH <sub>2</sub> —О—		130	78	80	$C_{19}H_{19}O_5N_2$
4a	—OCH <sub>3</sub>	Н	130	75	76	$C_{25}H_{21}O_4N_2$
4b	Н	Н	135	82	84	$C_{26}H_{24}O_5N_2$
4c	—O—CH <sub>2</sub> —О—		165	80	85	$C_{26}H_{23}O_6N_2$
5a	OCH <sub>3</sub>	Н	158	76	80	$C_{20}H_{20}O_4N_2$
5b	Н	Н	178	74	78	$C_{20}H_{20}O_4N_2$
5c	_O_CH <sub>2</sub> _O_		182	78	85	C <sub>21</sub> H <sub>21</sub> O <sub>6</sub> N <sub>2</sub>

TABLE-1
PHYSICAL DATA OF THE COMPOUNDS

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