Asian Journal of Chemistry

Vol. 11, No. 3 (1999), 1048-1050

NOTE

Methylation of Acacia auriculiformis gum polysaccharide

R.B. SINGH

Department of Zoology, Social Forestry and Environment Institute of Basic Science Dr. B.R. Ambedkar University, Agra-282 002, India

Polysaccharide has been extracted from *Acacia auriculiformis* gum with water as L-arabinose and D-galactose in 1:4 molar ratio. Methylation of gum afforded methyl sugars as 2,3,4,6-tetra-O-methyl-D-galactose; 2,3-di-O-methyl-L-arabinose; 2,3,4-tri-O-methyl-D-galactose; 2,4-di-O-methyl-D-galactose and 2,3,4-tri-O-methyl-D-glucuronic acid in 1:1:1:2:1 molar ratio.

Acacia auriculiformis plant¹ (Mimoseae) occurs in Northern India, and the gum contain a water soluble polysaccharide. Present manuscript deals with methylation studies alongwith a proposed gum polysaccharide structure.

Paper chromatography was carried out by descending technique² on Whatman No. 1 and 3 mm paper with upper phase of the following solvent system (v/v): (S_1) *n*-butanol, ethanol, water $(4:1:5)^3$, (S_2) *n*-butanol, acetic acid, water $(4:1:5)^3$ and (R_1) *p*-anisidine phosphate⁴ used as spray reagent for the detection of methyl sugars.

Methylation of gum polyaccharide: Polysaccharide (15 g) was methylated by Hakomari's method⁵ 3 times with water (40 mL), dimethyl sulphate (175 mL) and sodium hydroxide (30%, 150 mL) in atmospheric nitrogen at 5–10°C giving partly methylated product (6.50 g, —OCH₃, 38%). It was remethylated 5 times by Purdie's reagent⁶ with methanol (30 mL), methyl iodide (30 mL) and silver oxide (12 g), which gave fully methylated product (5.024 g, —OCH₃, 42.58%), $[\alpha]_D^{27}$ –39.8° (H₂O).

Hydrolysis of methylated gum polysaccharide: Methyl sugar (5 g) was refluxed with methanolic hydrogen chloride (4%, 15 mL), then evaporated off to dryness and saponified with barium hydroxide (0.03 N, 10 mL) at 60°C for 3 h. Excess Ba(OH)₂ was removed by CO₂ in the solution. Saponified product after

being worked up give an ether soluble portion (A) (yield 2.002 g and an ether insoluble portion (B) (yield 0.3756 g).

Identification of ether soluble portion (A): It (2.002 g) was hydrolysed with HCl (1 N, 40 mL) on water-bath (10 h). Hydrolysate was cooled, neutralised (BaCO₃) and evaporated to syrup. It was resolved into 4 components on Whatman No. 3 mm paper (S₁) and paper strips corresponding to individual sugars were eluted with water⁷.

I: 2,4-di-O-methyl-D-galactorse: Syrup (0.5992 g) gave single spot on paper chromatography (S₁), R_f (0.40), $[\alpha]_D^{27}$ +81.2°C (H₂O). Found: —OCH₃, 29.8%. Derivative was prepared by refluxing syrup (0.3 g) with aniline (0.162 g) and ethanol (5 mL) giving crystals of 2,4-di-O-methyl-N-phenyl-D-galactosyl amine, m.p. 208°C.⁸

II: 2,3,4-tri-O-methyl-D-galactose: Syrup (0.351 g) having $[\alpha]_D^{27}$ +114° (H₂O) showed single spot on paper chromatography (R_f 0.62 in S₁), Found: —OCH₃, 39.7%, calcd. for C₉H₁₈O₆, —OCH₃, 41.9%. Derivative was prepared as 2,3,4-tri-O-methyl-N-phenyl-D-galactosyl amine, m.p. 163°C.⁸

III: 2,3-di-o-methyl-L-arabinose: Syrup (0.2814 g) gave single spot (R_f 0.63 in S_1) on paper chromatography, [α] $_D^{27}$ +98° (H_2O). Found: —OCH $_3$, 34.9%, calcd, for $C_7H_{14}O_5$, 34.8%. Sugar was converted into 2,3-di-O-methyl-N-phenyl-L-arabinose amine derivative, m.p. 138°C. 9 .

IV: 2,3,4,6-tetra-O-methyl-D-galactose: Syrup (0.310 g) had R_f , 0.86 (S_1), [α] $_D^{27}$ +112° (H_2O). Found: —OCH $_3$, 51.99%; calcd. for $C_{10}H_{20}O_6$, 52.5%. It was converted into 2,3,4,6-tetra-O-methyl-N-phenyl-D-galactosyl amine, m.p. 190–191°C (192°C).

Identification of ether insoluble portion (B): Syrup (0.3756 g) on paper chromatography in S_2 used (R_1) as spray reagent to reveal single spot (R_f , 0.82) Found: —OCH₃, 38.4%, calcd. for 2,3,4-tri-O-methyl-D-glucuronic acid, $C_9H_{16}O_6$, 39.4%. Derivative was prepared by usual manner as 2,3,4-tri-O-mel-thyl- γ -D-glucopyranoside uronomide, m.p. 183°C.⁸

Acacia auriculiformis gum polysaccharide was methylated by Hakomari and Purdie's method. Methyl sugars were identified as 2,3,4-tetra-O-methyl-D-galactose, 2,3-di-O-methyl-L-arabinose, 2,3,4-tri-O-methyl-D-galactose, 2,4-di-O-methyl-D-galactose and 2,3,4-tri-O-methyl-D-glucuronic acid in 1:1:1:2:1 molar ratio. 2,3,4,6-tetra-O-methyl-D-galactose suggest that side chain R is terminated by D-galactopyranose moiety. Side chain R may be regarded as composed of L-arabofuranose unit joined to a D-galactopyranose unit $(5\rightarrow 1)$ - α -type linkage, accordingly the R = L-Araf $(5\rightarrow 1)$ -D-Galp. The uronic acid moiety is justified an acidic sugar, 2,3,4-tri-O-methyl-D-glucuronic acid. It may be pointed out here that in gum structure, L-fucose is not included because it was obtained in traces. Hydrolysis of methylated gum did not furnish any methylated L-fucose unit.

The tentative structure (Fig. 1) suggested for A. auriculiformis gum polysaccharide clearly indicates that gum is highly branched in nature and contains $(1\rightarrow6)-\beta$, $(1\rightarrow3)-\beta$ and $(1\rightarrow5)-\alpha$ -type linkages. 1050 Singh Asian J. Chem.

D-Galp 1
$$\xrightarrow{\beta}$$
 6 D-Galp 1 $\xrightarrow{\beta}$ 6 D-Galp 1 $\xrightarrow{\beta}$ 6 D-Galp 1 $\xrightarrow{\beta}$ 6 D-Galp 1 $\xrightarrow{\beta}$ 7 D-Galp (3 $\xrightarrow{\alpha}$ 1)-R $\xrightarrow{\beta}$ 6 D-Galp (3 $\xrightarrow{\alpha}$ 1)-R $\xrightarrow{\beta}$ 6 D-GpA D-GpA

where: Galp = Galactopyranose; GpA = Glucopyranosyl uronic acid $R = L\text{-Araf} (5 \xrightarrow{\alpha} 1)\text{-D-Galp}$

Fig. 1 Tentative structure of Acacia auriculiformis gum polysaccharide

REFERENCES

- 1. Y.R. Chadha, The Wealth of India: Raw Materials, CSIR, New Delhi, vol. 1-A, p. 23 (1985).
- 2. S.M. Partridge, Nature (London), 158, 270 (1946).
- 3. S.M. Partridge and R.G. Westall, Biochem. J., 42, 238 (1948).
- 4. S. Mukherjee and H.C. Srivastava, Nature (London), 169, 330 (1952).
- 5. S. Hakomari, J. Biochem. (Tokyo), 55, 205 (1964).
- 6. T. Purdie and J.C. Irvine, J. Chem. Soc., 83, 1021 (1903).
- 7. C.E. Dent, Biochem. J., 41, 240 (1947).
- 8. S. Bose and K.C. Gupta, Indian J. Chem., 2, 156 (1964).
- 9. E.L. Hirst and J.K.N. Jones, J. Chem. Soc., 120 (1948).
- 10. F. Smith, J. Chem. Soc., 1035 (1940).

(Received: 2 December 1998; Accepted: 1 June 1999) AJC-1744