

## Synthesis of 1,2,4-Trihalophenothiazin-3-ones and Their Conversion into (1,4)-Benzothiazino-(2,3-b)-Phenothiazines

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1,2,4-Trihalophenothiazin-3-ones (III) were prepared by condensing 2-amino benzenethiols (I) with chloranil/bromanil (II) in 1 : 1 molar ratio. Compounds (III) were further condensed with the same or different 2-aminobenzenethiol (IV) which afforded 1,4-benzothiazino-(2,3-b)-phenothiazines (V).

### INTRODUCTION

Phenothiazinone and phenoxazine derivatives containing amino quinone system have been studied for biological and pharmaceutical activities and to obtain useful pigments<sup>1-5</sup>. Triphenodithiazines form an important class of compounds among the dyes of thiazine series<sup>6</sup>. Dyes of this series are useful for dyeing cotton, cellulose materials, rubber, paper and plastics in various shades<sup>7-12</sup>. In continuation of our interest in the synthesis and usages of these compounds, we wish to report the synthesis of some new phenothiazinone (III) and (1,4)-benzothiazino-(2,3-b)-phenothiazine derivatives (triphenodithiazines) (V).

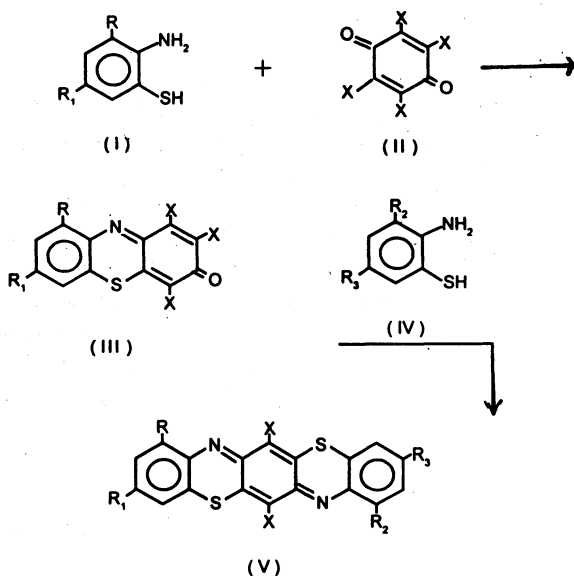
In the present work 1,2,4-trihalophenothiazin-3-ones (III) were prepared by the reaction of substituted 2-aminobenzenethiols (I) and chloranil/bromanil (II). The compound (III) was further condensed with same or different 2-aminobenzenethiol (IV) which afforded symmetrical substituted (1,4)-benzothiazino-(2,3-b) phenothiazines (Va,b) and unsymmetrical (1,4)-benzothiazino-(2,3-b)-phenothiazines (Vc-j Scheme-I). Symmetrical triphenodithiazines (Va,b) were also prepared by an alternative single step procedure. In this method chloranil/bromanil was condensed with substituted 2-amino benzenethiol in a molar ratio of 1 : 2 (Scheme-II). The products prepared by single step method were identical (TLC, IR and mixed melting point) with those obtained by two step reaction.

### RESULTS AND DISCUSSION

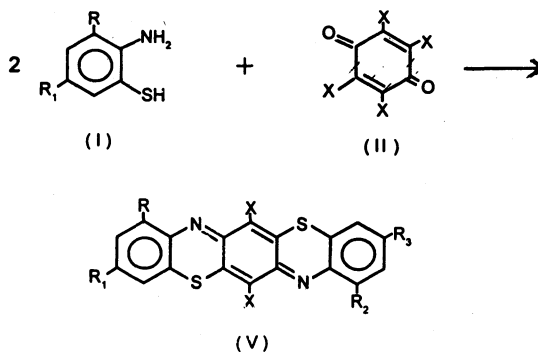
1,2,4-Trihalophenothiazin-3-ones exhibit strong band in the region 1675–1620  $\text{cm}^{-1}$  due to C=C stretching vibration. In chloro derivatives band in the region

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765–745  $\text{cm}^{-1}$  is due to C—Cl stretching vibrations and in bromoderivatives band in the region 580–560  $\text{cm}^{-1}$  is due to C—Br stretching vibrations. The bands in the region 1600–1200  $\text{cm}^{-1}$  can be assigned to C—C and C—N ring vibrations. Similarly 1,4-benzothiazeno-(2,3-b)-phenothiazines show sharp and medium bands in the region 1625–1600  $\text{cm}^{-1}$  which are due to C—C, C—N ring vibrations and 580–560  $\text{cm}^{-1}$  band is appeared due to C—Br stretching vibrations.



Scheme-I



Scheme-II

### EXPERIMENTAL

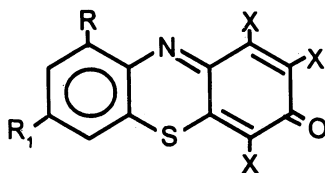
Melting points of all the compounds were determined in open capillaries and are uncorrected. IR spectra were recorded on Perkin-Elmer model 781 in

KBr/nujol. The purity of the compounds was checked on thin layers of silica-gel in various nonaqueous solvent systems. Bromanil was prepared by the method described by Torry and Hunter<sup>13</sup>. Substituted 2-aminobenzenthiois were prepared by the method reported in literature<sup>14-16</sup>.

(i) Preparation of substituted 1,2,4-trihalophenothiazin-3-ones (IIIa,b)

To a suspension of chloranil/bromanil (II, 0.01 mole) in ethanol (20 mL) was added to a solution of substituted 2-aminobenzenthiois (I, 0.01 mole) in ethanol (10 mL) and anhydrous sodium acetate (0.05 mole). The reaction mixture was refluxed for 6 h. After cooling to room temperature the solid separated was filtered, washed with water and finally with 30% ethanol. The products were recrystallised from benzene. Physical data are given in Table-1.

TABLE-1  
PHYSICAL DATA OF 1,2,4-TRIHALOPHENOTHIAZIN-3-ONES (III)



S.No.	Compound			m.p. (°C)	Yield (%)	m.f.	N % Found (Calcd.)
	R	R <sub>1</sub>	X				
(a)	Br	Br	Br	154	52	C <sub>12</sub> H <sub>2</sub> NOSBr <sub>5</sub>	2.27 (2.30)
(b)	Br	Br	Cl	128	54	C <sub>12</sub> H <sub>2</sub> NOSBr <sub>2</sub> Cl <sub>3</sub>	2.90 (2.95)

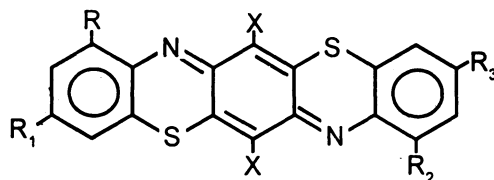
\*All the compounds gave the satisfactory C, H and S analysis.

(ii) Preparation of 1,4-benzothiazino-(2,3-b)-phenothiazines (Vc-j)

**Method A:** To a mixture of 1,2,4-trihalophenothiazin-3-ones (III, 0.01 mole) and substituted 2-aminobenzenthiois (I, 0.01 mole) in ethanol (20 mL) was added anhydrous sodium acetate (0.05 mole). The reaction mixture was refluxed for 6 h, cooled to room temperature and the solid separated out was filtered, washed with water followed by 30% ethanol and recrystallised from benzene. Physical data are given in Table-2.

**Method B:** To a stirred suspension of chloranil/bromanil (0.01 mole) and 2-aminobenzenthiois (0.02 mole) in ethanol (20 mL) was added anhydrous sodium acetate (0.01 mole). The mixture was refluxed for 6 h, cooled to room temperature and the solid separated out was filtered, washed with water followed by 30% ethanol and recrystallised from benzene. Physical data are given in Table-2.

TABLE-2  
 PHYSICAL DATA OF SOME SUBSTITUTED 1,4-BENZOTHAZINO-  
 (2,3-b)-PHENOTHIAZINES (V)



S.No.	Compound					m.p. (°C)	Yield (%)	m.f.	N % Found (Calcd.)
	R	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	X				
(a)	Br	Br	Br	Br	Br	247	59	C <sub>18</sub> H <sub>4</sub> N <sub>2</sub> S <sub>2</sub> Br <sub>6</sub>	3.50 (3.53)
(b)	Br	Br	Br	Br	Cl	198	59	C <sub>18</sub> H <sub>4</sub> N <sub>2</sub> S <sub>2</sub> Br <sub>4</sub> Cl <sub>2</sub>	3.85 (3.90)
(c)	H	H	Br	Br	Br	223	52	C <sub>18</sub> H <sub>6</sub> N <sub>2</sub> S <sub>2</sub> Br <sub>4</sub>	4.39 (4.41)
(d)	H	H	Br	Br	Cl	218	55	C <sub>18</sub> H <sub>6</sub> N <sub>2</sub> S <sub>2</sub> Br <sub>2</sub> Cl <sub>2</sub>	5.08 (5.13)
(e)	H	Cl	Br	Br	Br	261	60	C <sub>18</sub> H <sub>5</sub> N <sub>2</sub> S <sub>2</sub> Br <sub>4</sub> Cl	4.13 (4.18)
(f)	H	Cl	Br	Br	Cl	185	58	C <sub>18</sub> H <sub>5</sub> N <sub>2</sub> S <sub>2</sub> Br <sub>2</sub> Cl <sub>3</sub>	4.80 (4.83)
(g)	Br	Br	CH <sub>3</sub>	Cl	Cl	201	53	C <sub>19</sub> H <sub>7</sub> N <sub>2</sub> S <sub>2</sub> Br <sub>2</sub> Cl <sub>3</sub>	4.65 (4.71)
(h)	Br	Br	CH <sub>3</sub>	Cl	Br	214	48	C <sub>19</sub> H <sub>7</sub> N <sub>2</sub> S <sub>2</sub> Br <sub>4</sub> Cl	4.06 (4.10)
(i)	CH <sub>3</sub>	Cl	Br	Br	Br	240	49	C <sub>19</sub> H <sub>7</sub> N <sub>2</sub> S <sub>2</sub> Br <sub>4</sub> Cl	4.06 (4.10)
(j)	CH <sub>3</sub>	Cl	Br	Br	Cl	209	52	C <sub>19</sub> H <sub>7</sub> N <sub>2</sub> S <sub>2</sub> Br <sub>2</sub> Cl <sub>3</sub>	4.65 (4.71)

\*All the compounds gave the satisfactory C, H and S analysis.

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