

NOTE

Synthesis of Some Biologically Active Isoxazolines

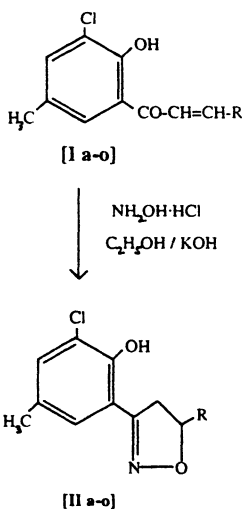
V.M. BAROT,* M.R. PATEL† and H.B. NAIK†

Department of Chemistry, Smt. S.M. P. Science College, Talod-383 215, India

Various 2-hydroxychalcones have been condensed with hydroxylamine hydrochloride to yield isoxazolines. The compounds have been characterised by elemental analysis, IR and NMR spectra.

The unsaturation present in the chalcones is responsible for many novel reactions¹. Among them the pyrazolines and isoxazolines are of prominence. Isoxazolines are reported to be active fungicides and insecticides. These compounds owe their activities to the heterocyclic ring present in the structure²⁻⁴. The reaction is not simple; besides the oxime and isoxazolines line, other products like hydroxylmine ketone, hydroxylamino oxime, disubstituted hydroxylamine, etc. may be formed depending upon the nature of substituents and the proportion of the reactants⁵⁻⁷.

Melting points were determined in open capillary and are uncorrected. IR spectra were recorded in KBr on Perkin-Elmer-377 spectrophotometer and NMR were recorded on Varian model EM-360L spectrophotometer. Satisfactory elemental analyses were obtained.



†South Gujarat University, Surat, India.

Preparation of 3{3'-chloro-2'-hydroxy-5'-methyl phen-1'-yl}-5-sub-2-isoxazoline

A mixture of 3'-chloro-2'-hydroxy-5'-methyl-chalcone² (I, 0.01 mol) and hydroxylamine hydrochloride (0.02 mol) in ethanol (25 mL) was refluxed on water bath at 60–70°C for 4 h. The reaction mixture was then cooled and acidified with glacial acetic acid. The solid separated was filtered, washed with water, dried and crystallized from ethanol (Table-1 a–o).

TABLE-1
PHYSICAL DATA OF COMPOUNDS 1 (a–o)

No.	R	m.f.	m.p. (°C) (Yield %)	Antibacterial activity (24 h)		Antifungicidal activity (48h)
				<i>S. aureus</i>	<i>E. coli</i>	
a	Phenyl	C ₁₆ H ₁₂ O ₂ NCl	150 (75)	17	NA	NA
b	2-furfuryl	C ₁₄ H ₁₀ O ₃ NCl	131 (70)	19	15.	12
c	2-chlorophenyl	C ₁₆ H ₁₁ O ₂ NCl ₂	114 (80)	NA	11	9
d	4-chlorophenyl	C ₁₆ H ₁₁ O ₂ NCl ₂	148 (82)	7	NA	8
e	4-N,N-dimethyl aminophenyl	C ₁₈ H ₁₇ O ₂ N ₂ Cl	166 (60)	NA	NA	NA
f	3,4-methylenedioxyphenyl	C ₁₇ H ₁₂ O ₂ NCl	108 (68)	NA	NA	NA
g	2-nitrophenyl	C ₁₆ H ₁₁ O ₄ N ₂ Cl	125 (72)	12	7	8
h	3-nitrophenyl	C ₁₆ H ₁₁ O ₄ N ₂ Cl	139 (72)	14	9	7
i	4-nitrophenyl	C ₁₆ H ₁₁ O ₄ N ₂ Cl	163 (78)	13	16	NA
j	4-hydroxyphenyl	C ₁₆ H ₁₂ O ₃ N ₂ Cl	121 (72)	11	13	12
k	3-aminophenyl	C ₁₆ H ₁₃ O ₂ N ₂ Cl	171 (67)	18	14	11
l	4-methoxyphenyl	C ₁₇ H ₁₄ O ₃ NCl	133-35 (72)	16	19.	10
m	4-methylphenyl	C ₁₇ H ₁₄ O ₃ NCl	159 (78)	12	17.	8
n	2-hydroxyphenyl	C ₁₆ H ₁₂ O ₃ NCl	172 (68)	11	13	7
o	4-hydroxy-3-methoxyphenyl	C ₁₇ H ₁₄ O ₄ NCl	181 (70)	11	15	9
<i>Standard Drugs</i>						
	Amoxicillin			25	–	–
	Cloxacillin			–	28	–
	Nystatin			–	–	27

IR (KBr): The isoxazolines show sharp band around 830 and 880 cm⁻¹ in addition to a medium intensity band around 1620 cm⁻¹. These bands are due to (N—O) and (C=N) stretching vibrations. Disappearance of (C=O) in the spectra of the compounds marks the condensation followed by ring closure.

NMR (CDCl₃): 2.25 (—CH₂ of methylenedioxy), 3.8 (—OCH₃), 6.99

(—OH), 7.17–8.43 (Ar—H). 2.47 (CH₂— of isoxazolines), 3.34 (CH isoxazolines) (δ_{PPM}).

Antibacterial activity: The antibacterial activity of title compounds is evaluated against *Staphylococcus aureus* and *Escherichia coli* by paper disc method and compared with standard drugs like amoxycillin and cloxacilin. Compound Nos. **b, k, l, m** have shown maximum activity against *S. aureus* and *E. coli*.

c and **a, d** are inactive against *S. aureus* and *E. coli* respectively while rest of the compounds have shown poor activity against both the bacteria.

Antifungal activity: The antifungal activity of the title compounds was evaluated against *C. albicans* by paper disc technique and compared with standard drug nystatin.

Compounds No. **a, e, i and f** are inactive while rest of the compounds have shown weak activity against fungi, *i.e.*, *C. albicans*.

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REFERENCES

1. G. Buethner, E. Klonke, E.P. Frohberger and I. Hamman, Ger. Pat. 2218329 (1973); *Chem. Abstr.*, **86**, 14912 (1974).
2. S.R. Mod and H.B. Naik, *Oriental J. Chem.*, **10**, 85 (1994).
3. M.D. Ankhiwala and H.B. Naik, *J. Indian Chem. Soc.*, **67**, 258 (1990).
4. P.F. Devitt, A. Timoney and M.A. Vickars, *J. Org. Chem.*, **26**, 1941 (1961).
5. V. Pietro and B. Aurom, *Chem. Abstr.*, **69**, 16007 (1968).
6. H. Richard, Valley (cct), *The Chemistry of Heterocyclic Compounds*, Interscience Publication, **7**, 95 (1962).
7. B. Unterhalf, *Chem. Abstr.*, **70**, 57351 (1969).
8. H. Singh and L.D.S. Yadav, *J. Indian Chem. Soc.*, **54**, 1143 (1977).

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