Synthesis and Antibacterial Activity of 3-Cyano-2-(1H)-Pyridone Derivatives

M.D. DESAI and K.K. DESAI*

Department of Chemistry, South Gujarat University, Surat-395 007, India.

Several new 3-cyano-4-substituted aryl-6-[2'-hydroxy-4'-(p-trifluromethyl/nitro-phenoxy)-phen-1'-y1]-2-(1H)-pyridone (II) have been prepared by reaction of 1-[2'-hydroxy-4'-(p-trifluoromethyl/nitro phenoxy)-phen-1'-y]-3-substituted aryl-2-propene-1-ones (I) with ethyl cyanoacetate and ammonium acetate in ethanol. Few of these compounds are characterized by IR and NMR spectra. All these synthesized 3-cyano-2-pyridones have been screened against a few microorganisms for antibacterial activity.

Key Words: Synthesis, Antibacterial, 3-Cyano-2-(1H) Pyridone derivatives

INTRODUCTION

Pyridine and its homologues are commonly called pyridine bases. Various derivatives of pyridine are of pharmaceutical and agrochemical interest. Some workers synthesized 3-cyano-2-(1H)-pyridone derivatives from chalcones and cyano acetamide¹, while some synthesized it from chalcone and ethylcyanoacetate and ammonium acetate^{2,3}. Antibacterial⁴, antifungal and molluscicidal activity⁵ of 3-cyano-2-(1H)-pyridone derivatives were reported. Further, an introduction of fluorine atom or —CF₃ group into an organic molecule may alter the biological activities as reported⁶. Moreover 2-hydroxy-chalcones with —NO₂ group as an additional substituent are reported to exhibit very good antibacterial activity⁷. Therefore it was thought interesting to synthesize 3-cyano-2-(1H)-pyridones using ethyl cyanoacetate from chalcones containing —CF₃ and —NO₂ group in p-position of phenoxy substituent and to screen them as antibacterial agents.

EXPERIMENTAL

Preparation of 3-cyano-2-(1H)-pyridone derivatives

A mixture of chalcone (0.01 mol), ethyl cyanoacetate (0.01 mol) and ammonium acetate (0.08 mol) in ethanol (40 mL) was refluxed at 70–80°C on water bath for 6 h. It was then cooled and poured into ice-water. The product thus separated was filtered, washed with water, dried and crystallized from 1,4-dioxane.

Infrared spectra of the compounds were recorded in solid state using KBr pellet on Perkin-Elmer FT-IR spectrophotometer (Model-RX-1). The PMR spectra of

1384 Desai et al. Asian J. Chem.

compounds were recorded in DMSO-d₆ solvent at room temperature using TMS as reference compound. The spectra were recorded on Perkin-Elmer Model-32 NMR spectrometer, at 300 MHz at CDRI Lucknow.

The antibacterial activities of the synthesized compounds and standard drugs (ampicillin and tetracycline) were screened by agar cup method in DMF solvent. The activity was checked against gram positive bacteria B. subtilis and S. aureus and gram negative bacteria E. coli and S. typhi.

OH

C -CH = CH - Ar

Chalcone

$$C_2H_5OH$$

NCCH₂COOC₂H₅

CH₃COONH₄

NH

CN

(II)

R

(II)

RESULTS AND DISCUSSION

The main absorption bands observed in IR spectra are as under:

2230–2200 ν [C \equiv N]; 1710–1690 ν [C \equiv O]; 3235–3200 ν [N \rightarrow H]; 1575–1550 ν [N \rightarrow H]; 1225–1200 ν [C \rightarrow O \rightarrow C]; 3455–3420 ν [O \rightarrow H]; 1170–1150 ν [C \rightarrow F]; 1535–1525 and 1375–1365 cm $^{-1}\nu$ [N \equiv O].

The position of signals in PMR spectra of compd. IIr can be assigned to different types of protons as under: $\delta = 10.25$ (proton of —OH); $\delta = 3.80$ (three protons of —OCH₃ group); $\delta = 6.27$ to 8.00 (aromatic protons and proton of —NH).

Diameter of zone of inhibition (in mm) of standard drug ampicillin against B. subtilis, S. aureus (gram positive), E. coli and S. typhi (gram negative) were found to be 24, 22 17 and 16 respectively while tetracycline gave 18, 17, 21 and 22 respectively under identical conditions.

As compared to standards, among cyano-pyridones containing trifluoromethyl substitution compounds IIb and IIk containing 4-chloro-phenyl and 2-furyl substituent showed fairly good activity. Compounds IIc, IId and IIe containing 2-chloro-phenyl, 2,4-dichloro phenyl and 4-fluoro phenyl substituent respectively showed better activity than the previous two. Compound IIa containing phenyl substituent was found active against gram positive bacteria only. Compound IIh containing dimethoxy phenyl substituent was found inactive against B. subtilis and S. typhi. In case of nitro substituted cyano-pyridone compounds IIp and IIr containing 4-fluoro phenyl and 2-methoxy phenyl substituent showed maximum activity. Compounds IIn and IIu containing 2-chloro phenyl and 4-N,N-dimethylamino phenyl substituent showed moderate activity. Compound II

DATA SHOWING CHARACTERISTICS OF COMPOUNDS (II) AND RESULTS OF ANTIBACTERIAL ACTIVITY

		DAIA SHOWING CHARACHERS ITCS OF COMIT COINDS (II) AND RESOLES OF AN HEACH ENIAL ACTIVITY	כואים כאאר	100011	OINT COLVE	יייי (ווי) פי	KESOLLIS OF A	MIDACIEN	TALACITAL		
. (;		Elemental a	Elemental analysis % N		Antibacterial Activity	al Activity	
Compd	~ ~	Ar	m.w.	Yield	E é			Gram	Gram positive	Gram n	Gram negative
				(ox.)		Found	Calculated	B. subtilis	S. aureus	E. coli	S. typhi
IIa	CF ₃	-C ₆ H ₅	448	09	254	6.14	6.25	×	×	7	6
q ₁	CF3	4-CI-C,H4	482.5	89	265	5.78	5.80	17	18	91	15
IIc	CF3	-2-CI-C ₄ H ₄	482.5	99	260	5.74	5.80	19	20	18	21
PII	CF_3	-2,4-(CI) ₂ -C ₆ H ₃	517	<i>L</i> 9	172	3.39	5.41	21	17	19	50
IIe	CF3	4-F-C ₆ H ₄	466	89	268	5.90	90.9	20	18	21	19
Ш	CF_3	4-0CH ₃ -C ₆ H ₄	478	92	258	5.69	5.85	6	9	01	7
IIg	CF3	-2-0CH ₃ -C ₆ H ₄	478	19	255	5.81	5.85	5	∞	7	10
ΙΉ	CF3	-3,4-(OCH ₃) ₂ -C ₆ H ₃	208	92	279	5.40	5.51	×	9	6	×
Ħ	CF_3	-3,4,5-(0CH ₃) ₃ -C ₆ H ₂	425	65	281	5.51	5.20	∞	10	2	7
Ξ	CF3	4-{N(CH ₃) ₂ }-C ₆ H ₄	491	99	270	8.51	8.55	10	9	6	5
IIK	CF3	-C ₄ H ₃ O-(2-furyl)	438	2	261	6.29	6:39	16	17	16;	17
	NO ₂	-C ₆ H ₄	425	92	281	9.74	88.6	5	9	×	∞
IIm	NO ₂	4-CI-C ₃ H ₄	459.5	71	296	60.6	9.14	6	7	10	7
IIn	NO ₂	-2-CI-C ₆ H ₄	459.5	99	285	9.11	9.14	12	=	13	15
IIo	NO ₂	-2,4-(CI) ₂ -C ₆ H ₃	494	72	> 300	8.47	8.50	∞	5	6	9
IIp	NO ₂	-4-F-C ₆ H ₄	443	73	280	9.43	9.48	18	20	70	22
IIq	NO ₂	4-0CH ₃ -C ₆ H ₄	455	89	> 300	9.15	9.23	∞	×	7	5
H	NO_2	-2-0CH ₃ -C ₆ H ₄	455	<i>L</i> 9	281	9.10	9.23	17	61	18	21
IIs	NO ₂	-3,4-(OCH ₃) ₂ -C ₆ H ₃	485	72	274	8.59	8.65	6	01	7	5
Ħ	NO ₂	-3,4,5-(OCH ₃) ₃ -C ₆ H ₂	515	74	> 300	8.08	8.15	7	9	∞	6
IIu	NO ₂	4-{N(CH ₃) ₂ }-C ₆ H ₄	468	71	569	11.88	11.96	14	10	=	12
IIv	NO2	-C ₄ H ₃ O- (2-furyl)	415	70	700	10.08	10.12	∞	6	9	×

x = Found inactive

1386 Desai et al. Asian J. Chem.

containing phenyl substituent remained inactive against *E. coli*. Compound **IIq** containing 4-methoxy-phenyl substituent was found inactive against *S. aureus* while compound **IIv** containing 2-furyl substituent was found inactive against *S. typhi*.

ACKNOWLEDGEMENT

The authors are thankful to Head, Department of Chemistry, South Gujarat University, Surat, for providing necessary laboratory facilities. One of the authors (MDD) is also thankful to Government of Gujarat for award of research scholarship. Facilities provided by CDRI Lucknow for scanning PMR spectra is also acknowledged.

REFERENCES

- 1. L. Makhan, L.N. Bhatt and H. Junjappa, Synthesis, 6, 641 (1995).
- 2. H. Jahine, H.A. Zaher, A.A. Sayed and O. Sherif, Indian J. Chem., 11, 1122 (1973).
- 3. N. Latif, N. Mishriky, B. Haggag and W. Basyouni, Indian J. Chem., 24B, 1230 (1985).
- 4. N. Latif, N. Mishriky and N.S. Girgis, Indian J. Chem., 20B, 147 (1981).
- 5. N. Latif, M. Asaad and N.S. Girgis, Indian J. Chem., 20B, 463 (1981).
- 6. K.C. Joshi, A. Dandia and S. Khanna, Indian J. Chem., 29B, 1125 (1990).
- 7. M. Gabor, J. Sallai and T. Szell, Arch. Pharm. (Weinheim), 303, 593 (1970).

(Received: 27 February 2002; Accepted: 6 May 2002)

AJC-2697