

Synthesis and Biological Activities of Some New Formazans, Part-I

S.D. BHARADWAJ

Cancer Hospital & Research Institute, Gwalior-474 009, India

1-(Phenyl-3-(2-methoxy-4-N,N'-bis-2'-cyanoethylamino phenyl) *p*-tolyl formazans have been synthesised by first reacting 4-N,N'-bis-2-cyanoethylamino-2-methoxy benzaldehyde with *p*-tolyl hydrazide which gave the acid hydrazone. The acid hydrazone on treatment with diazotised aromatic amines in pyridine medium furnished fifteen formazans in 40 to 54% yield. The acid hydrazone and the formazans were found to be devoid of anticancer and anti-HIV activities.

Key words: Synthesis, Biological activities, Formazans.

INTRODUCTION

Formazans are used as dyes^{1,2} and belong to azo dye² family and as chelating agents³. Synthesis, antiviral⁴, antimicrobial^{5,6}, and anti-inflammatory⁷ activities of formazans have been given in literature. Some products from 2-methyl-4-N,N'-bis-2'-cyanoethylaminobenzaldehyde have shown anticancer activity which prompted the use of these aldehydes for synthesis of new formazans. Recently Jolly and others^{8,9} have synthesised new formazans for assessing their antiviral, anticancer activity^{10,11} and anti-HIV activities¹².

The present study records the reaction of 4-N,N'-bis-2-cyanoethylamino-2-methoxy benzaldehyde⁹ (I) with *p*-tolyl hydrazide¹³ (II) which gave the acid hydrazone (III). The acid hydrazone on treatment with diazotised aromatic amines in pyridine medium furnished formazans (IV) (Table-1). The reaction sequence has been outlined in Scheme-I.

EXPERIMENTAL

Preparation of *p*-methyl benzhydrazide (*p*-tolylhydrazide)

Methyl *p*-toluate: Methyl ester of *p*-toluic acid was readily prepared by refluxing a mixture of *p*-toluic acid (27.2 g, 0.2 mole), absolute methanol (80 mL) and concentrated sulphuric acid (4 mL) for 4 h heat on steam bath and worked up as for methyl *p*-chlorobenzoate. The ester was crystallised from petroleum ether (60°-80°C); m.p. 35°C; yield about 70%.

Hydazide: Methyl *p*-toluate and hydrazine hydrate dissolved in methanol (25 mL) was refluxed for 30 min on a hot water bath. On cooling, a white compound was obtained which was crystallised from aqueous ethanol as very light small leaflets; m.p. 116–117°C; yield about 70%.

Synthesis of *p*-tolyl hydrazone of 2-methoxy-4-*N,N'*-bis-2'-cyanoethylaminobenzaldehyde: Solutions of 2-methoxy-4-*N,N'*-bis-2'-cyanoethylaminobenzaldehyde (2.57 g, 0.01 mole) in ethanol (5 mL) and *p*-tolyl hydrazide (1.5 g, 0.01 mole) in ethanol (5 mL) were mixed and the mixture was refluxed for 15 min. On cooling, the liquid hydrazone separated as solid. It was filtered under suction and recrystallised from rectified spirit; m.p. 180°C; yield 98%.

Molecular formula $C_{22}H_{23}N_5O_2$, Analysis: found C 64.2, H 6.4, N 20.0%; required C 67.86, H 5.91, N 17.99%.

IR-spectra were recorded on Perkin-Elmer model 1720 and FTIR infracord spectrophotometer. The product shows IR bands, $\nu(N-H)$ 3325–3320, $\nu(C=O)$ 1680–1670, $\nu(C=N)$ 1610–1600, $\nu(N=N)$ 1570–1560, $\nu(C-N)$ 1350–1330, $\nu(C-OH)$ 3500 cm^{-1} .

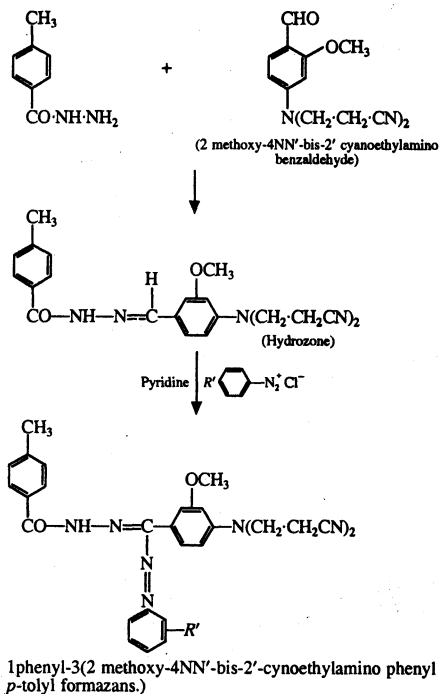
Synthesis of 1-(phenyl-3-(2-methoxy-4-*N,N'*-bis-2'-cyanoethylamino-phenyl)-*p*-tolyl formazans: Aniline (0.46 g, 0.01 mole) was dissolved in aqueous hydrochloric acid (4 mL 1 : 1). The contents were cooled and aqueous sodium nitrite (0.3 g in 2 mL water) was slowly added; 2-methoxy-4-*N,N'*-bis-2'-cyanoethylaminobenzalidine-*p*-tolyl hydrazone (0.49 g, 0.01 mole) was dissolved in dry pyridine (10 mL) and sodium acetate (0.3 g) was added. The contents were cooled in an ice bath and stirred. The solution of benzene diazonium chloride was added dropwise for 30 min, maintaining low temperature (0°C). The reaction mixture was kept in an ice bath for 4 h and then poured with stirring in ice water. The resulting dark brown coloured solid was washed with water till free from pyridine, filtered under suction and dried. The product was crystallised from ethanol; m.p. 160°C; yield 50%; colour brown.

Molecular formula $C_{28}H_{27}N_7O_2$, Analysis: found C 70.00, H 6.70, N 18.00%; required C 68.15, H 5.47, N 19.8%.

The product shows IR bands, $\nu(N-N)$ 3325–3320, $\nu(C-O)$ 1680–1670, $\nu(C-N)$ 1610–1600, $\nu(N=N)$ 1570–1560, $\nu(C-N)$ 1350–1330, $\nu(C-OH)$ 3500 cm^{-1} .

Anticancer activity: Six formazans (Nos. 3, 4, 9, 10, 11, 13) and one hydrazone (No. 1) were tested for their anticancer activity. None of the products was found active at the dose level tested.

Anti-HIV activity: Six formazans (Nos. 3, 4, 9, 10, 11, 13) and one hydrazone (No. 1) have been assessed under *in vitro* anti-HIV activity. All the compounds (seven) did not show significant activity. Tests were conducted at National Cancer Institute, Maryland, USA.



Scheme-1

TABLE-1
PHYSICAL DATA OF FORMAZANS

| S.No. | —R' | Yield (%) | m.p. (°C) | Colour |
|-------|-------------------------------|-----------|-----------|--------------|
| 1. | Hydrazone | 65 | 180 | light yellow |
| 2. | H | 50 | 160 | deep brown |
| 3. | CH ₃ (<i>o</i>) | 54 | 170 | deep brown |
| 4. | CH ₃ (<i>m</i>) | 50 | 170 | light yellow |
| 5. | CH ₃ (<i>p</i>) | 48 | 170 | deep brown |
| 6. | Cl (<i>o</i>) | 40 | 145 | brown |
| 7. | Cl (<i>m</i>) | 40 | 135 | deep brown |
| 8. | Cl (<i>p</i>) | 40 | 140 | deep brown |
| 9. | OCH ₃ (<i>o</i>) | 50 | 165 | deep brown |
| 10. | OCH ₃ (<i>m</i>) | 40 | 150 | deep brown |
| 11. | OCH ₃ (<i>p</i>) | 45 | 165 | light yellow |
| 12. | NO ₂ (<i>o</i>) | 45 | 122 | deep brown |
| 13. | NO ₂ (<i>m</i>) | 55 | 125 | deep brown |
| 14. | NO ₂ (<i>p</i>) | 40 | 120 | deep brown |
| 15. | COOH (<i>o</i>) | 55 | 175 | white |
| 16. | COOH (<i>p</i>) | 50 | 165 | yellow |

All compounds gave satisfactory elemental analysis; solvent for crystallisation: ethanol.

ACKNOWLEDGEMENTS

The authors are extremely grateful to Dr. B.R. Shrivastava, Director, Cancer Hospital and Research Institute, Gwalior for providing research facilities. We feel indebted to Dr. V.L. Narayanan, Chief, Drug Synthesis and Chemistry Branch, National Cancer Institute, Bethesda, Maryland, U.S.A. for anti-HIV and anti-cancer screening of the products. We are also thankful to the Director, Defence Research and Development Establishment, Gwalior (M.P.), India for spectral analysis. The financial assistance given by Madhya Pradesh Council of Science and Technology, Kisan Bhawan, Jail Road, Bhopal (M.P.), India for the research work is gratefully acknowledged.

REFERENCES

1. A. Uchumi and Terakiy, *Biomed. Res. Trac. Elem-2*, 141 (1991).
2. Claué Manchner and Patsch Manfred (BAST-A-9). Geroffen DE4, 230095 (1994).
3. H. Von Pechmann, *Ber.*, 25, 3175 (1982); Bemberger and Wheel Wright, *ibid.*, 25, 3201 (1982).
4. Gok, Yasar, Santurk and H. Bastri, *Dyes Pigm.*, 15, 279 (1991).
5. Archana Shrivastava, Sanjay Swaroop, V.K. Saxma and B.L. Chaudhary, *J. Indian Chem. Soc.*, 68, 658 (1991).
6. Nailesh Joshi, Atul Prakash, Baporda and Hames, *Inet. Med. Chem*, 338, 662 (Eng.) (1994).
7. B.N. Trivedi and V.M. Shah, *J. Indian Chem. Soc.*, 69, 765 (1992).
8. H.G. Garg and M. Kaur, *J. Med. Chem.*, 15, 554 (1972).
9. S.D. Bharadwaj, P. Pathak, and V.S. Jolly, *Oriental J. Chem.*, 11, 183 (1995).
10. V.S. Jolly, Ph.D. Thesis, Agra University, Agra (1966).
11. S.D. Bharadwaj and V.S. Jolly, *Asian J. Chem.*, 9, 48 (1997).
12. S.D. Bharadwaj, *Asian J. Chem.*, 10, 39 (1998).
13. A.I. Vogel, *Practical Organic Chemistry*, 3rd Edn., Longmans, Green & Co., London, pp. 787-88 (1956).

(Received: 15 November 2001; Accepted: 5 February 2002)

AJC-2584

6TH EUROPEAN LASER ABLATION ICP-MS WORKSHOP

UTRECHT, THE NETHERLANDS

JUNE 24-25, 2002

Visit the website:

<http://laicpms.geo.uu.nl>

or contact:

MARJOLEIN VAN WIJK

Faculty of Earth Sciences

Budapestlaan 4, De Uithof, Utrecht 3584 CD, The Netherlands

Tel.: +31 30 253 5122 Fax: +31 30 253 5030

E-mail: m.vanwijk@geo.uu.nl