## **NOTE**

## A Convenient Synthesis of Benzo[h] Cyclopenta[b][1,6]-Naphthyridin-6(5H)Ones

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Substituted benzo[h]cyclopenta[b][1,6]naphthyridin-6(5H)ones (2) have been synthesized by the condensation of 4-amino-3-formylquinoline-2(1H)ones (1) with cyclopentanone in presence of acetic acid and sulphuric acid.

Key Words: Synthesis, Benzo[h]cyclopenta[b][1,6]naphthyridin-6(5H)ones.

Many of the naphthyridines have shown bactericidal and fungicidal activities<sup>1</sup>. Very few reports have so far appeared in the literature on the synthesis of dibenzo[b, h][1,6]naphthyridines and their pharmacological activities<sup>1-11</sup>. We have already reported the synthesis of 8,9,10,11-tetrahydrodibenzo[b, h][1,6]naphthyridine-6(5H)ones<sup>12</sup>. Herein, we report a convenient method to synthesise benzo[h]cyclopenta[b][1,6]naphthyridine-6(5H)ones starting from 4-amino-3-formylquinoline-2(1H)ones. (Scheme-1).

Scheme-1

$$R_1$$
 $R_2$ 
 $R_3$ 
 $R_3$ 
 $R_4$ 
 $R_5$ 
 $R_6$ 

(i) Cyclopentanone, CH<sub>3</sub>COOH/H<sub>2</sub>SO<sub>4</sub>

(a)  $R_1 = R_2 = R_3 = H$  (b)  $R_1 = CH_3$ ;  $R_2 = R_3 = H$  (c)  $R_1 = R_3 = H$ ;  $R_2 = CH_3$  (d)  $R_1 = OCH_3$ ;  $R_2 = R_3 = H$  (e)  $R_1 = R_3 = H$ ;  $R_2 = OCH_3$ 

Melting points were determined on a Boetius microheating table and are uncorrected. IR spectra were recorded on a Perkin-Elmer-597 infrared spectro-

photometer as KBr pellets. <sup>1</sup>H NMR spectra were recorded on a Bruker WH-270 (270 MHz) NMR spectrometer or on an EM-390 (90 MHz) NMR spectrometer in CDCl<sub>2</sub> unless otherwise specified.

Typical procedure, Benzo[h]cyclopenta[b][1,6]naphthyridin-6(5H)ones (2a-e): 4-Amino-3-formylquinolinE-2(1H)one<sup>12</sup> (1) (0.01 mole) was dissolved in a mixture of cyclopentanone (0.02 mole) and acetic acid. Sulphuric acid (0.1 mole) was added and refluxed for 10 h. The cold solution was poured on to a mixture of conc. aqueous ammonia (40 mL) in (20 g) of ice, which gave a brown tarry product. After extraction with chloroform, drying, evaporation and addition of diethyl ether, the brown solid obtained was purified by chromatography over silica gel (60-120 mesh, 50 g) using pet. ether-ethyl acetate (95:5 v/v) as eluant. The product was recrystallized from ethyl acetate (Table-1).

TABLE-1 PHYSICAL AND SPECTROSCOPIC DATA OF COMPOUND 2(a-e)a

Compd.	m.p.°C (Yield %)	$IR \\ (v_{max}) cm^{-1}$	<sup>1</sup> H NMR (δ) ppm	MS m/z (M <sup>+</sup> )
2a	310 (74)	3100 v(NH) 1630 v(NHCO) 1445 v(CH)	3.07 (m, 4H, C <sub>10</sub> -2H and C <sub>8</sub> -2H); 2.13 (m, 2H, C <sub>9</sub> -2H) 7.6 (m, 3H, C <sub>2</sub> -H, C <sub>3</sub> -H and C <sub>4</sub> -H); 8.15 (s, 1H, C <sub>7</sub> -H); 8.82 (s, 1H, C <sub>1</sub> -H); 8.96 (s, 1H, NH)	
<b>2b</b>	300(d) (67)	3080 v(NH) 1640 v(NHCO) 1446 v(CH)	$\begin{array}{l} 2.85(s,3H,C_2\text{-}CH_3);3.05(m,4H,C_{10}\text{-}2H\text{ and}\\ C_8\text{-}2H)2.15(m,2H,C_9\text{-}2H);7.6\text{-}7.89(m,3H,C_1\text{-}H,C_3\text{-}H\text{ and }C_4\text{-}H);8.2(s,1H,C_7\text{-}H);\\ 8.91(s,1H,NH) \end{array}$	
<b>2</b> c	278–280 (68)	3080 v(NH) 1640 v(NHCO) 1446 v(CH)	$\begin{array}{l} 2.6 \ (s,\ 3H,\ C_3\text{-}CH_3);\ 3.0 \ (m,\ 4H,\ C_{10}\text{-}2H\ and \\ C_8\text{-}2H);\ 2.21 \ (m,\ 2H,\ C_9\text{-}2H);\ 7.50\text{-}7.69 \ (m,\ 2H,\ C_1\text{-}H\ and\ C_2\text{-}H);\ 8.5 \ (s,\ 1H,\ C_4\text{-}H);\ 8.71 \\ (s,\ 1H,\ C_7\text{-}H);\ 9.0 \ (s,\ 1H,\ NH) \end{array}$	
2d	281–283 (65)	3080 v(NH) 1640 v(NHCO) 1446 v(CH)	<del>-</del>	250
2e	294–295 (65)	3080 v(NH) 1640 v(NHCO) 1446 v(CH)	$\begin{array}{llllllllllllllllllllllllllllllllllll$	

<sup>(</sup>a) Recrystallised from ethyl acetate.

The compound 1a on condensation with cyclopentanone with acetic acid and sulphuric acid at 120°C for 10 h gave a product which on purification furnished a brown compound (m.p. 310°C) in 74% yield. Its IR spectrum showed disappearance of peak at 1680 cm<sup>-1</sup>. The compound showed negative tests for aldeyhde and amino group. The <sup>1</sup>H NMR spectrum of the compound showed signals at  $\delta$  3.07 (m, 4H, C<sub>10</sub>-2H & C<sub>8</sub>-H); 2.13 (m, 2H, C<sub>9</sub>-2H); 7.6 (m, 3H, C<sub>2</sub>-H,  $C_3$ -H,  $C_4$ -H); 8.82 (s, 1H,  $C_1$ -H); 8.15 (m, 1H,  $C_7$ -H); 8.96 (s, 1H, NH). The mass spectrum gave molecular ion peak at m/z 236. The compound was identified as benzo[h]cyclopenta[b][1,6]naphthyridin-6(5H)one (2a).

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The reactions sequence leading to 2a was then extended to synthesise hitherto unknown compounds 2b-2e.

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