

NOTE

Microwave Induced Stereoselective Conversion of Dialkyl 2-(Imido-*N*-yl)-3-(Triphenylphosphoranylidene)butanedioates to Electron-poor (Z)-*N*-Vinylimides in the Presence of Silica Gel Powder in Solvent-free Conditions

ALI RAMAZANI*, AKRAM ABBASI MOTEJADDED and EBRAHIM AHMADI

Chemistry Department, Zanjan University, P.O. Box 45195-313, Zanjan, Iran

E-mail: aliramazani@yahoo.com

Silica gel was found to catalyze stereoselective conversion of dialkyl-2-(imido-*N*-yl)-3-(triphenylphosphoranylidene)butanedioates to electron-poor (Z)-*N*-vinylimides in solvent-free conditions under microwave irradiation.

Key Words: Microwave, Silica gel, (Z)-*N*-Vinylimide, Phosphorus ylide, Solvent-free conditions.

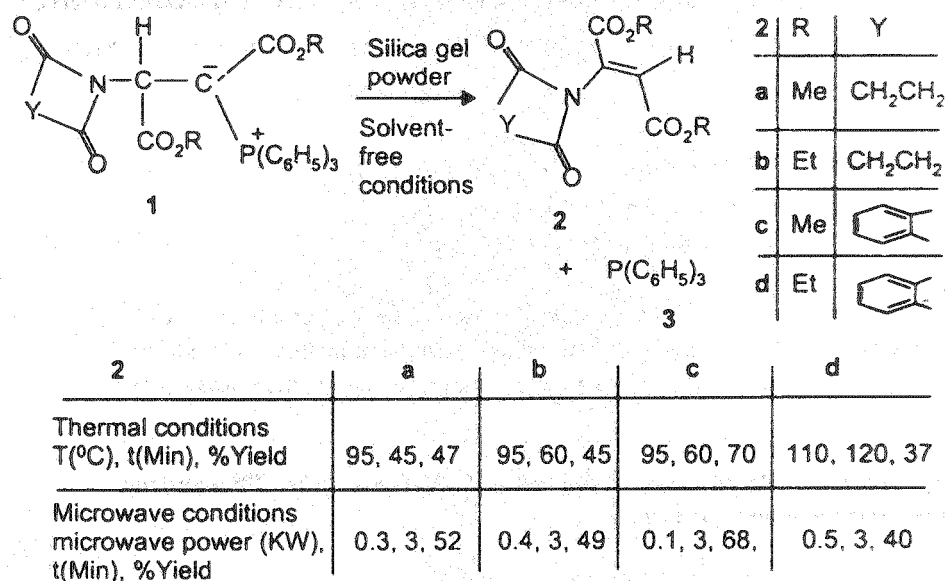
β -Additions of nucleophiles to the vinyl group of vinylic phosphonium salts leading to the formation of new alkylidenephosphoranes has attracted much attention as a very convenient and synthetically useful method in organic synthesis¹⁻⁸. Organophosphorus compounds have been extensively used in organic synthesis as useful reagents as well as ligands of a number of transition metal catalysts². Silica gel as an additive promotes the Wittig reactions of phosphorus ylides with aldehydes, including sterically hindered aldehydes to increase the rate and yields of alkenes^{9, 10}. In this article, the catalytic rule of silica gel powder in the stereoselective conversion of dialkyl 2-(imido-*N*-yl)-3-(triphenylphosphoranylidene)-butanedioates (**1**)¹⁰ to electron-poor (Z)-*N*-vinylimides (**2**)⁸ in solvent-free conditions¹⁴ under microwave irradiation with high conversions (**Scheme-1**) has been reported.

Commercial oven Butane M245 was used for microwave irradiation. Melting points were measured on an Electrothermal 9100 apparatus and are uncorrected. ¹H and ¹³C NMR spectra were measured with a Bruker DRX-500 AVANCE spectrometer at 500 and 125 MHz respectively.

Preparation of Compounds 2a–d: The homogeneous mixture of powdered dry silica gel (2 g) and ylide **1**¹⁰ (1 mmol) were irradiated in a microwave oven at microwave power 0.1–0.5 kW for 3 min (**Scheme-1**) and then placed over a column of silica gel (12 g). The column chromatography was washed using ethyl acetate-light petroleum ether (1 : 9) as eluent. The solvent was removed under reduced pressure and products (**2a–d**) were obtained. The characterization data of the compounds (**2a–d**) are already reported earlier^{8, 9}.

Silica gel powder was found to catalyze stereoselective conversion of ylides (**1**)¹⁰ to electron-poor (Z)-*N*-vinylimides (**2**)⁸ in solvent-free conditions¹⁴ under microwave irradiation at microwave power 0.1–0.5 kW with high conversions (**Scheme-1**)⁸⁻¹². TLC indicated that the reaction was completed after 3 min. The

reaction proceeds smoothly and cleanly under solvent-free conditions¹⁴ at microwave power 0.1–0.5 kW (in all cases the reaction works efficiently with high conversions) and no side reactions were observed.



Scheme-1

The structures **2a–d** were deduced from their ¹H NMR and ¹³C NMR spectra and also via X-ray single crystal (for **2c**) structure determination¹³.

ACKNOWLEDGEMENT

This work was supported by the Zanzan University.

REFERENCES

1. A. Ramazani and A. Bodaghi, *Tetrahedron Lett.*, **41**, 567 (2000).
2. I. Yavari and A. Ramazani, *Synth. Commun.*, **27**, 1449 (1997).
3. I. Yavari, A. Ramazani and A. Yahya-Zadeh, *Synth. Commun.*, **26**, 4495 (1996).
4. I. Yavari and A. Ramazani, *Phosphorus, Sulphur and Silicon*, **130**, 73 (1997).
5. A. Ramazani, N. Shajari and F. Gouranlou, *Phosphorus, Sulphur and Silicon*, **174**, 223 (2001).
6. A. Ramazani, A. Momeni-Movahhed and F. Gouranlou, *Phosphorus, Sulphur and Silicon*, **177**, 903 (2002).
7. A. Ramazani, B. Mohammadi and N. Noshiranzadeh, *Phosphorus, Sulphur and Silicon*, **178**, 545 (2003).
8. ———, *Phosphorus, Sulphur and Silicon*, **178**, 767 (2003); **178**, 761 (2003); **178**, 539 (2003).
9. A. Ramazani and E. Ahmadi, *Phosphorus, Sulphur and Silicon*, **178**, 2293 (2003).
10. A. Ramazani and H. Ahani, *Phosphorus, Sulphur, and Silicon*, **170**, 181 (2001).
11. V.J. Patil and U. Mavers, *Tetrahedron Lett.*, **37**, 1281 (1996).
12. C. Xu, G. Chen, C. Fu and X. Huang, *Synth. Commun.*, **25**, 2229 (1995).
13. A. Ramazani, F. Marandi, E. Ahmadi and A. Morsali, *Z. Kristallogr., NCS*, **219**, 179 (2004).
14. K. Tanaka and F. Toda, *Chem. Rev.*, **100**, 1025 (2000).