NOTE

Stereoselective Synthesis of Dialkyl 1,1-diacetyl-8a-hydroxy-8-oxo-1,2,8,8a-tetrahydrocyclopenta[a]indene-2,3-dicarboxylates from the Reaction of Dialkyl 2-(1-acetyl-2-oxopropyl)-3-(tributylphosphoranylidene) Succinates with Indene-1,2,3-trione

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Protonation of the highly reactive 1:1 intermediates produced in the reaction between tributylphosphine and dialkyl acetylene-dicarboxylates by acetylacetone leads to sterically congested phosphorus ylides. Reaction between the sterically congested phosphorus ylides with ninhydrin leads to stereoselective synthesis of dialkyl 1,1-diacetyl-8a-hydroxy-8-oxo-1,2,8,8a-tetrahydrocyclopenta[a]indene-2,3-carboxylates in the presence of magnesium sulfate powder in solvent-free conditions.

Key Words: Magnesium sulfate, Wittig reaction, Solvent-free conditions, Ninhydrin, Tributylphosphine.

Organophosphorus compounds have been extensively used in organic synthesis^{1, 2}. A well known method for achieving alkenylation is the Wittig reaction¹. β-Additions of nucleophiles to the vinyl group of vinylic phosphonium salts leading to the formation of new alkylidenephosphoranes has attracted much attention as a very convenient and synthetically useful method in organic synthesis^{1, 2}. In the past, a convenient one-pot method for preparing stabilized phosphorus ylides utilizing *in situ* generation of the phosphonium salts¹ has been established. In this article, a facial one-pot stereoselective synthesis of dialkyl 1,1-diacetyl-8a-hydroxy-8-oxo-1,2,8,8a-tetrahydrocyclopenta[a]indene-2,3-dica rboxylates (10) by the reaction of dialkyl 2-(1-acetyl-2-oxopropyl)-3-(tributyl-phosphoranylidene) succinates (4) and indene-1,2,3-trione (6) (Scheme-1) in the presence of magnesium sulfate powder in solvent-free conditions i reported.

Melting points were measured on an Electrothermal 9100 apparatus and are uncorrected. IR spectra were recorded on a Shimadzu IR-460 spectrometer. ¹H, ¹³C NMR spectra were measured with a Bruker DRX-500 Avance spectrometer at 500, 125 and 470.6 MHz, respectively.

General procedure for the preparation of dialkyl 1,1-diacetyl-8a-hydroxy- $8-0 \times 0-1, 2, 8, 8a-tetra hydrocyclopenta [a] indene-2, 3-dicarboxylates$ To a magnetically stirred solution of tributylphosphine (1) (1 mmol) and acetylacetone (3) (1 mmol) in CH₂Cl₂ (4 mL) was added dropwise a mixture of 2 (1 mmol) in CH₂Cl₂ (3 mL) at -10°C over 15 min. The mixture was allowed to warm up to room temperature and the solvent was removed under reduced pressure. The homogeneous mixture of the dried residue (ylide 4), magnesium sulfate powder (2 g) and powdered indane-1,2,3-trione (1 mmol, 6, prepared from powdered ninhydrin (8) under thermal conditions (4 h at 120°C) in oven) were heated for 1 h at 90°C and then the mixture placed over a column of silica gel (12 g). The column chromatography was washed using ethyl acetate-light petroleum ether (1:9) as eluent. The solvent was removed under reduced pressure and products were obtained as white crystals (6a-b). The characterization data of dialkyl 1,1-diacetyl-8a-hydroxy-8-oxo-1,2,8,8a-tetrahydrocyclopenta[a]indene-2,3-dicarboxylates (10a-b) are given in a previous report⁵.

Reactions are known in which an α,β-unsaturated carbonyl compound is produced from a phosphorane and a carbonyl compound such as an aldehyde or ketone³. Thus, compounds 10 may be regarded as the product of an intramolecular Wittig reaction. Such addition-olefination products may result from an initial addition of tributylphosphine (1) to the acetylenic ester (2) and concomitant protonation of the 1:1 adduct, followed by attack of the acetylacetone anion on the vinylphosphonium cation to form phosphorane⁴ (4). Attack of the stabilized ylide (5) on the highly electron deficient carbonyl group of indane-1,2,3-trion (6), would lead to the stabilized ylide (9). Intramolecular Wittig reaction of stabilized ylide (9) leads to dialkyl 1,1-diacetyl-8a-hydroxy-8-oxo-1,2,8,8a-tetrahydrocyclopenta[a]indene-2,3-dicarboxylates (10) and tributylphosphine oxide (11) in the presence of magnesium sulfate powder in solvent-free conditions (Scheme-1). TLC indicated that the reactions were completed in the presence of magnesium sulfate powder in solvent-free conditions at 90°C after 24 h. The less reactive aldehydes (p-nitrobezaldehyde, etc.) and ketones (acetophenone, etc.) instead of indane-1,2,3-trion in this reaction have also been used, but no products were observed. TLC indicated that the mixture contained ylide (4) and the starting aldehyde or ketone. The structures 10a-c were deduced from their elemental analyses and their UV, IR, ¹H and ¹³C NMR spectra. All of these data are the same as the previously reported data⁵ for the compounds 10a-b. Since compound 10 possesses two stereogenic centers, four stereoisomers (2R, 8aR; 2S, 8aS; 2R, 8aS and 2S, 8aR) are possible⁵. The stereochemistry of compound 10a (2R, 8aR and its mirror image 2S, 8aS) via single crystal X-ray diffraction method has been proved⁶.

In summary, a convenient one-pot stereoselective method for preparing dialkyl 1,1-diacetyl-8a-hydroxy-8-oxo-1,2,8,8a-tetrahydrocyclopenta[a]indene-2,3-dica rboxylates (10) utilizing in situ generation of the phosphorane 4 is developed. Other aspects of this process are under investigation.

$$Bu_{3}P + RO_{2}CC \equiv CCO_{2}R + OHOO CH_{2}CI_{2}$$

$$1 \qquad 2 \qquad 3 \qquad CH_{2}CI_{2}$$

$$CH_{3} \qquad Bu_{3}P = C - CHCO_{2}R$$

$$CH_{3} \qquad Bu_{3}P = C - CHCO_{2}R$$

$$CO_{2}R \qquad A$$

$$CH_{3} \qquad Bu_{3}P = C - CHCO_{2}R$$

$$CO_{2}R \qquad A$$

$$CH_{3} \qquad Bu_{3}P = C - CHCO_{2}R$$

$$CO_{2}R \qquad A$$

$$CH_{3} \qquad Bu_{3}P = C - CHCO_{2}R$$

$$CO_{2}R \qquad Bu_{3}P = C - CHCO_{2}R$$

$$CH_{3} \qquad Bu_{3}P = C - CHCO_{2}R$$

$$CH_{3} \qquad CH_{3} \qquad CO_{2}R \qquad Bu_{3}P = C - CHCO_{2}R$$

$$CH_{3} \qquad CH_{3} \qquad CO_{2}R \qquad Bu_{3}P = C - CHCO_{2}R$$

$$CH_{3} \qquad CH_{3} \qquad CO_{2}R \qquad Bu_{3}P = C - CHCO_{2}R$$

$$CH_{3} \qquad CH_{3} \qquad CO_{2}R \qquad Bu_{3}P = C - CHCO_{2}R$$

$$CH_{3} \qquad CH_{3} \qquad CO_{2}R \qquad DH_{3} \qquad CO_{2}R \qquad DH_{3} \qquad D$$

ACKNOWLEDGEMENT

This work was supported by the Zanjan University.

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