Microwave Assisted Synthesis of Novel Azetidin-2-one and Thiazolidin-4-one Derivatives of Triazolyl Thiophene

MAHENDRA SHIRADKAR* and H.N. SHIVAPRASAD

Department of Pharmaceutical Chemistry, PES College of Pharmacy Hanumanthnagar, Bangalore-560 050, India Fax: (91)(080)26506928; Tel: (91)(080)26507428, 9845545353 E-mail: rrshiradkar@rediffmail.com

N-[3-(4-Amino-5-mercapto-4*H*-[1,2,4]triazol-3-yl)-4,5-dimethyl thiophen-2-yl]-acetamide (Ia) and N-[3-(4-amino-5-mercapto-4*H*-[1,2,4]triazol-3-yl)-4,5-dimethyl thiophen-2-yl]-benzamide (Ib) were condensed with different aromatic aldehydes to yield Schiff bases (IIa-j; IIIa-j). The Schiff bases on cyclization with chloroacetyl chloride in presence of triethyl amine furnished azetidin-2-ones (IVa-j; Va-j). The Schiff bases on cyclization with mercaptoacetic acid offered thiazolidin-4-ones (VIa-j; VIIa-j). The compounds thus synthesized in good yields by microwave irradiation method and were characterized by their TLC, chemical analysis and spectral data. The title compounds were subjected to antimicrobial and antitubercular activities.

Key Words: Synthesis, Microwave irradiation, Azetidin-2-one and Thiazolidin-4-one derivatives.

INTRODUCTION

In the past few years, many azetidin-2-one derivatives have been found to possess significant antibacterial¹ activity. Thiazolidin-4-ones are reported for antibacterial² and antitubercular^{3, 4} activites. Thiophenes exhibit a wide range of biological activities⁴ like antioxidants, antiinvasive, antiviral, etc. Schiff bases of triazoles have been reported to possess various pharmacological activities such as antibacterial⁵, antifungal⁵ activities. Microwave assisted reactions⁶ using dry media⁷ have attracted much interest because of the simplicity in operation, greater selectivity and rapid synthesis of a variety of heterocyclic compounds⁸. Keeping this in mind, it was worth while to develop rapid syntheses of title compounds under solvent-free conditions using microwave assisted organic reaction enhancement method.

EXPERIMENTAL

The melting points were recorded on electrothermal apparatus and are uncorrected. IR spectra were recorded in KBr on a Perkin-Elmer-983 and ¹H NMR spectra on Bruker-Avance 300 MHz instrument using CDCl₃ as solvent

(chemical shifts in δ ppm), using TMS as internal standard. Mass spectra were charted on Finning LCQ mass spectrometer. Microwave irradiations were carried out in Padmini Essentia oven, model Brownie, at 2450 MHz. Elemental analyses were performed on a Heracus CHN rapid analyser.

N-[3-(4-Amino-5-mercapto-4H-[1,2,4]triazol-3-yl)-4,5-dimethyl-thiophen-2-yl]-acetamide (Ia) and <math>N-[3-(4-amino-5-mercapto-4H-[1,2,4]triazol-3-yl)-4,5-dimethyl-thiophen-2-yl]-benzamide (Ib) were prepared as per literature method^{9, 10}

$N-{3-[4-Benzylideneamino)-5-mercapto-4}H[1,2,4]triazol-3-yl]-4,5-dimethyl-thiophen-2-yl}acetamide (IIa)$

Compound Ia (0.01 mol), a pinch of p-toluene sulphonic acid and appropriate aromatic aldehyde (0.01 mol) were mixed. Acidic alumina was added to the above mixture at room temperature. The reaction mixture was adsorbed, dried and kept inside the alumina bath and irradiated for 40–80 s. The mixtrue was cooled and the product was extracted with dry methanol and poured on to crushed ice. The solid thus separated was filtered, washed thoroughly with water and recrystallized from aqueous ethanol. Other Schiff bases for Ia and 1b series were obtained in a similar manner.

$N-{3-[4-(3-Chloro-2-oxo-4-phenyl-azetidin-1-yl)-5-mercapto-4H-[1,2,4] triazol-3-yl]-4,5-dimethyl-thiophen-2-yl}acetamide (IVa)}$

Schiff base (IIa) (0.01 mol) was mixed with triethylamine (2.80 mL, 0.02 mol) and chloroacetyl chloride (1.60 mL, 0.02 mol) was added dropwise over a period of 30 min. Acidic alumina was added to the above solution at room temperature. The reaction mixture was adsorbed, dried and kept inside the alumina bath and irradiated for 40–80 s. The mixture was cooled and the product was extracted with absolute ethanol and poured on to crushed ice. The solid thus separated was filtered, washed thoroughly with water and recrystallized from aqueous ethanol. The product (IVa) obtained was filtered, washed with water and crystallized. Other azetidine-2-ones for Ia and 1b series were obtained in a similar manner.

N- $\{3-[5-Mercapto-4-(4-oxo-2-phenyl-thiazolidin-3-yl)-4H-[1,2,4]$ triazol-3-yl]-4,5-dimethyl-thiophen-2-yl $\}$ acetamide (VIa)

Schiff base (IIa) (0.01 mol) was added to mercaptoacetic acid (1.40 mL, 0.02 mol) anhydrous aluminium chloride (0.05 g). Acidic alumina was added to the above solution at room temperature. The reaction mixture was mixed, adsorbed, dried and kept inside the alumina bath and irradiated for 40–80 s. The reaction mixture was then cooled and triturated with an excess of 10% sodium bicarbonate solution. The product obtained was filtered, washed several times with water and crystallized with isopropanol. Other thiazolidin-4-ones for Ia and 1b series were obtained in a similar manner.

Antimicrobial activity: All the compounds were screened for antibacterial activity against S. aureus and E. coli by paper disc technique ¹³. Gentamycin was used as standard. The antifungal activity of all the compounds was evaluated against C. albicans using the same technique. Nystatin was used as standard. The concentration of the test compound used was 100 μ g.

Antitubercular activity: The title compounds were tested in vitro for their antitubercular activity against M. tuberculosis $H_{37}Rv$. The antitubercular evaluation of compounds was carried out at "Tuberculosis Antimicrobial Acquisition and Coordinating Facility (TAACF)", USA. Primary screening of the compounds for antitubercular activity was conducted using the CABTEC 460 radiometric system. Compounds demonstrating at least > 90% inhibition in the primary screening were retested at lower concentration against M. tuberculosis $H_{37}Rv$ to determine the actual minimum inhibitory concentration (MIC) in CABTEC 460. The data were compared with the standard drug Rifampin at 0.03 $\mu g/mL$ concentration, which showed 97% inhibition.

RESULTS AND DISCUSSION

Looking at SAR, marked inhibition in bacteria was observed in the compounds bearing Ar = $2 \cdot C_6H_4Cl$, $2 \cdot C_6H_4OH$, $3 \cdot C_6H_4NO_2$ substituents whereas others showed moderate to least activity. Fungicidal screening data also revealed that compounds bearing Ar = $2 \cdot C_6H_4Cl$, $2 \cdot C_6H_4OH$, $3 \cdot C_6H_4NO_2$ substituent imparted maximum activity. Compounds Ar = $2 \cdot C_6H_4Cl$, $2 \cdot C_6H_4OH$, $3 \cdot C_6H_4NO_2$ were most active against M. tuberculosis $H_{37}Rv$ (> 90% inhibition) that will be retested at lower concentration to determine the actual minimum inhibitory concentration (MIC). Other compounds, viz, Ar = $4 \cdot N(CH_3)_2C_6H_4$, $2 \cdot BrC_6H_4$, $4 \cdot OHC_6H_4$ were moderately active against M. tuberculosis $H_{37}Rv$ strain (> 50% inhibition). The physical data of title compounds given in Tables 1 and 2 indicates different substitutents present.

The title compounds of the above scheme were confirmed by the following H¹ NMR spectra.

- (Ia) δ 2.0 (s, 2H, NH₂), δ 2.1 (s, 3H, CH₃), δ 2.2 (s, 3H, CH₃), δ 2.4 (s, 3H, CH₃), δ 3.0 (s, 1H, SH), δ 8.0 (s, 1H, NH).
- (Ib) δ 2.0 (s, 2H, NH₂), δ 2.2 (s, 3H, CH₃), δ 2.4 (s, 3H, CH₃), δ 3.0 (s, 1H, SH), δ 7.4–7.9 (M, 5H, ArH), δ 8.0 (s, 1H, NH).
- (IIa) δ 2.1 (s, 3H, CH₃), δ 2.2 (s, 3H, CH₃), δ 2.4 (s, 3H, CH₃), δ 3.0 (s, 1H, SH), δ 7.3–7.6 (M, 5H, ArH), δ 8.0 (s, 1H, NH), δ 8.1 (s, 1H, CH=N).
- (IIIa) δ 2.2 (s, 3H, CH₃), δ 2.4 (s, 3H, CH₃), δ 3.0 (s, 1H, SH), δ 7.3–7.9 (M, 10H, ArH), δ 8.0 (s, 1H, NH), δ 8.1 (s, 1H, CH=N).
- (IVa) δ 2.1 (s, 3H, CH₃), δ 2.2 (s, 3H, CH₃), δ 2.4 (s, 3H, CH₃), δ 3.0 (s, 1H, SH), δ 5.0 (s, 1H, CH-lactam ring), δ 5.4 (s, 1H, CH-lactam ring), δ 7.1–7.3 (m, 5H, ArH), δ 8.0 (s, 1H, NH).
- (Va) δ 2.2 (s, 3H, CH₃), δ 2.4 (s, 3H, CH₃), δ 3.0 (s, 1H, SH), δ 5.0 (s, 1H, CH-lactam ring), δ 5.4 (s, 1H, CH-lactam ring), δ 7.1–7.9 (m, 10H, ArH), δ 8.0 (s, 1H, NH).
- (VIa) δ 2.1 (s, 3H, CH₃), δ 2.2 (s, 3H, CH₃), δ 2.4 (s, 3H, CH₃), δ 3.0 (s, 1H, SH), δ 3.3 (s, 2H, CH₂-thiazolidine), δ 5.9 (s, 1H, CH-thiazolidine), δ 7.0–7.1 (m, 5H, ArH), δ 8.0 (s, 1H, NH).
- (VIIa) δ 2.2 (s, 3H, CH₃), δ 2.4 (s, 3H, CH₃), δ 3.0 (s, 1H, SH). δ 3.3 (s, 2H, CH₂-thiazolidine), δ 5.9 (s, 1H, CH-thiazolidine), δ 7.0–7.9 (m, 10H, ArH), δ 8.0 (s, 1H, NH).

TABLE-I Self CHARLES PHYSICAL DATA OF THE TITLE COMPOUNDS

S.No.	m.f.	m.p. (°C)	ing speci Silving	Yield (%)	% Carbon		% Nitrogen	
			m.w.		Calcd.	Found	Calcd.	Found
Ia	C ₁₀ H ₁₃ N ₅ OS ₂	187	283	77	42.40	42.42	24.73	24.67
Ib	C15H15N5OS2	192	345	84	52.17	52.02	20.28	20.45
IIa	C ₁₇ H ₁₇ N ₅ OS ₂	231	371	86	54.98	54.70	18.86	18.79
IIb	C17H16N5OS2CI	241	405	84	50.37	50.45	17.28	17.40
IIc	C ₁₇ H ₁₆ N ₅ OS ₂ Cl	251	405	85	50.37	50.25	17.28	17.16
IId	C ₁₇ H ₁₇ N ₅ O ₂ S ₂	231	387	82	52.71	52.53	18.08	18.23
He	$C_{19}H_{22}N_6OS_2$	262	414	89	55.07	55.17	20.28	20.42
IIC	C ₁₇ H ₁₆ N ₅ OS ₂ Br	247	449	87	45 43	45.36	15.59	15.48
IIIa	C22H19N5OS2	289	433	84	60.96	60.78	16.16	16.27
IIIP	C22H18N5OS2CI	268	467	89	56.53	56.47	14.98	14.85
IIIc	C22H18N5OS2CI	269	467	81	56.53	56.41	14.98	14.75
IIId	C22H19N5O2S2	274	449	85	58.79	58.62	15.59	15.82
IIIe	C24H24N6OS2	272	476	84.	60.50	60.43	17.64	17.74
IIIf	C ₂₂ H ₁₈ N ₅ OS ₂ Br	230	511	85	51.66	51.48	13.69	13.56
IVa	C19H18N5O2S2CI	285	447	86	51.00	51.16	15.65	15.43
IVb	C19H17N5O2S2Cl2	269	481	82	47.40	47.38	14.55	14.5
IVc	C19H17N5O2S2Cl2	247	481	87	47.40	47.65	14.55	14.74
IVd	C19H18N5O3S2CI	263	463	88	49.24	49.16	15.11	15.20
I Ve	C ₂₁ H ₂₃ N ₆ O ₂ S ₂ Cl	246	490	80	51.42	51.24	17.14	17.32
IVſ	C ₁₉ H ₁₇ N ₅ O ₂ S ₂ ClBr	276	525	79	43.42	43.21	13.33	13.4
Va	C24H20N5O2S2CI	241	509	78	56.58	56.31	13.75	13.58
Vb	C24H19N5O2S2Cl2	251	543	89	53.03	53.26	12.89	12.9
Vc	C24H19N5O2S2Cl2	287	543	85	53.03	53.14	12.89	12.69
Vd	C ₂₄ H ₂₀ N ₅ O ₃ S ₂ Cl	298	525	84	54.25	54.36	13.33	13.48
Ve	C ₂₆ H ₂₅ N ₆ O ₂ S ₂ Cl	283	552	86	56.52	56.42	15.21	15.47
VÍ	C ₂₄ H ₁₉ N ₅ O ₂ S ₂ ClBr	276	587	85	49.06	49.13	11.92	11.75
VIa	C19H19N5O2S3	243	445	82	51.23	51.37	15.73	15.56
VIb	C ₁₉ H ₁₈ N ₅ O ₂ S ₃ Cl	254	479	81	47.59	47.40	14.61	14.46
VIc	C19H18N5O2S3CI	268	479	83	47.59	47.62	14.61	14.49
VId	C19H19N5O3S3	281	461	84	49.45	49.21	15.18	15.23
VIe	C ₂₁ H ₂₄ N ₆ O ₃ S ₃	264	490	82	51.42	51.30	17.14	17.24
VIf	C ₁₉ H ₁₈ N ₅ O ₂ S ₃ Br	293	523	81	43.59	43.43	13.38	13.67
VIIa	C ₂₄ H ₂₁ N ₅ O ₂ S ₃	246	507	84	56.80	56.67	13.80	13.59
VIIb	C24H20N5O2S3CI	254	541	82	53.23	53.47	12.93	12.78
VIIc	C24H20N5O2S3CI	234	541	85	53.23	53.14	12.93	12.72
VIId	C24H21N5O3S3	265	523	84	55.06	55.21	13.38	13.65
VIIe	C ₂₆ H ₂₆ N ₆ O ₃ S ₃	249	566	87	55.12	55.02	14.84	14.67
VIII	C ₂₅ H ₂₀ N ₅ O ₂ S ₃ Br	284	597	87	50.25	50.46	11.72	11.54

TABLE-2

			B.B. D. Bus do		
Compd	. R	Ar	Compd	. R	Ar
Ia	NHCOCH ₃		IVj	NHCOCH ₃	2OCH ₃ C ₆ H ₄
Ib	NHCOC ₆ H ₅	5	Va	NHCOC ₆ H ₅	C ₆ H ₅
IIa	NHCOCH3	C_6H_5	Vb	NHCOC ₆ H ₅	4ClC ₆ H ₄
IIb	NHCOCH ₃	4CIC ₆ H ₄	Vc	NHCOC ₆ H ₅	2CIC ₆ H ₄
IIc	NHCOCH ₃	2CIC ₆ H ₄	Vd	NHCOC ₆ H ₅	20HC ₆ H ₄
IId	NHCOCH ₃	2OHC ₆ H ₄	Ve	NHCOC ₆ H ₅	4N(CH ₃) ₂ C ₆ H ₄
IIe	NHCOCH ₃	4N(CH ₃) ₂ C ₆ H ₄	Vſ	NHCOC ₆ H ₅	2BrC ₆ H ₄
IIC	NHCOCH ₃	2BrC ₆ H ₄	Vg	NHCOC ₆ H ₅	4BrC ₆ H ₄
IIg	NHCOCH ₃	4BrC ₆ H ₄	Vh	NHCOC ₆ H ₅	3NO₂C ₆ Ha
IIh	NHCOCH ₃	3NO ₂ C ₆ H ₄	Vi	NHCOC ₆ H ₅	2,4,6(OCH ₃) ₃ C ₆ H ₂
IIi	NHCOCH ₃	2,4,6(OCH ₃) ₃ C ₆ H ₂	Vj	NHCOC ₆ H ₅	20CH ₃ C ₆ H ₄
IIj	NHCOCH3	20CH ₃ C ₆ H ₂	VIa	NHCOCH ₃	C ₆ H ₅
IIIa	NHCOC ₆ H ₅	C ₆ H ₅	VIb	NHCOCH ₃	4ClC ₆ H ₄
IIIb	NHCOC ₆ H ₅	4CIC ₆ H ₄	VIc	NHCOCH ₃	2CIC₀H₄
IIIc	NHCOC ₆ H ₅	2ClC ₆ H₄	VId	NHCOCH ₃	20HC ₆ H ₄
IIId	NHCOC ₆ H ₅	20HC ₆ H ₄	VIe	NHCOCH ₃	4N(CH ₃) ₂ C ₆ H ₄
IIIe	NHCOC ₆ H ₅	4N(CH ₃) ₂ C ₆ H ₄	VII	NHCOCH ₃	2BrC ₆ H ₄
IIIf	NHCOC ₆ H ₅	2BrC ₆ H₄	VIg	NHCOCH ₃	4BrC ₆ H ₄
IIIg	NHCOC ₆ H ₅	4BrC ₆ H ₄	VIh	NHCOCH ₃	3NO ₂ C ₆ H ₄
IIIh	NHCOC ₆ H ₅	3NO ₂ C ₆ H ₄	VII	NHCOCH ₃	2,4,6(OCH ₃) ₃ C ₆ H ₂
IIIi	NHCOC ₆ H ₅	2,4,6(OCH ₃) ₃ C ₆ H ₂	V Ij	NHCOCH ₃	20CH ₃ C ₆ H ₄
IIIj	NHCOC ₆ H ₅	20CH ₃ C ₆ H ₄	VIIa	NHCOC ₆ H ₅	C ₆ H ₅
Va	NHCOCH ₃	C ₆ H ₅	VIIb	NHCOC ₆ H ₅	4ClC ₆ H ₄
Vb	NHCOCH ₃	4CIC ₆ H ₄	VIIc	NHCOC ₆ H ₅	2CIC ₆ H ₄
Vc	NHCOCH ₃	2CIC ₆ H ₄	VIId	NHCOC ₆ H ₅	20HC ₆ H ₄
Vd	NHCOCH ₃	20HC ₆ H ₄	VIIe	NHCOC ₆ H ₅	4N(CH ₃) ₂ C ₆ H ₄
Ve	NHCOCH ₃	4N(CH ₃) ₂ C ₆ H ₄	VIII	NHCOC ₆ H ₅	2BrC ₆ H ₄
Vſ	NHCOCH ₃	2BrC ₆ H ₄	VIIg	NHCOC ₆ H ₅	4BrC ₆ H ₄
Vg	NHCOCH ₃	4BrC ₆ H ₄	VIIh	NHCOC ₆ H ₅	3NO ₂ C ₆ H ₄
Vh	NHCOCH ₃	3NO ₂ C ₆ H ₄	VIIi	NHCOC ₆ H ₅	2,4,6(OCH ₃) ₃ C ₆ H ₂
Vi	NHCOCH ₃	2.4,6(OCH ₃) ₃ C ₆ H ₂	VIIj	-	20CH ₃ C ₆ H ₄

Scheme-1

ACKNOWLEDGEMENT

The authors are thankful to Dr. Cecil D. Kwong, Research Chemist, "Tuberculosis Antimicrobial Acquisition and Coordinating Facility (TAACF)" USA for antitubercular activity.

REFERENCES

- 1. R.F. Abdullah and H. Fuhr Knneth, J. Med. Chem., 18, 625 (1975).
- 2. M.B. Hogle, A.C. Uthale and B.P. Nikam, Indian J. Chem., 30B, 717 (1991).
- 3. N.C. Desai, R.R. Astik and K.A. Thaker, J. Indian Chem. Soc., 50, 771 (1982).
- 4. K.S. Malhotra, Indian J. Chem., 14B, 360 (2002).
- 5. S.N. Pandeya, D.S. Ram, G. Nath and E. De-Cleri, Arzneimittel Forschung, 50, 55 (2000).
- 6. D. Villemin, B. Martin and B. Garrigues, Synth. Commun., 23, 2251 (1993).
- 7. M. Kidwai, K.R. Bhushan and P. Kumar, Monatsh Chem., 130, 585 (1999).
- 8. M. Kidwai and N. Negi, Acta Pharm., 45, 511 (1995).
- 9. K. Gewald, E. Shinke and H. Bottcher, Chem. Ber., 99, 94 (1966).
- 10. R. Reid and N.D. Heidel, J. Heterocycl. Chem., 13, 925 (1976).
- 11. Aluminium oxide, acidic, Brockmann I, ca. 150 mesh, 58 Å CAMAG 506-C-1, Surface area 155 m²/g, pH = 6.0.
- 12. G. Bram, A. Loupy, M. Majdoub, E. Gutierrez and E. Ruiz-Hitzky, *Tetrahedron*, 45, 5167 (1990).
- 13. C. Jasper, J.C. Manizzella and P.A. Henry, J. Am. Pharm. Assoc., 471 (1958).

 (Received: 9 February 2005; Accepted: 26 September 2005) AJC-4405