

NOTE**One-Pot Stereoselective Synthesis of Dimethyl-(Z)-2-(ethanimidoysulfanyl)-2-butenedioate from Dimethyl Acetylenedicarboxylate and Thioacetamide in Presence of Zinc Oxide Powder**

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Thioacetamide reacts with dimethyl acetylenedicarboxylate in presence of acetone to form dimethyl (Z)-2-(ethanimidoysulfanyl)-2-butenedioate. Zinc oxide powder was found to catalyze the reaction in solvent-free conditions at 90°C in 1 h.

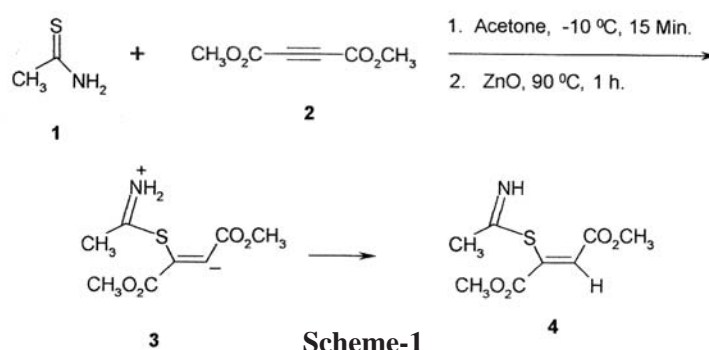
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Organosulfur chemistry has provided organic chemists with a wealth of reactions, many of which have found general application in organic synthesis¹⁻³. Organosulfur compounds are important heterocycles in bio-organic chemistry and are present in many pharmaceuticals⁴⁻⁶. Many studies on the synthesis of the organosulfur compound have been reported¹⁻¹⁰. Dimethyl acetylenedicarboxylate (**2**) is reactive system, which take part in many chemical reactions¹⁰⁻¹³. These results promoted us to examine the one-pot reaction of dimethyl acetylenedicarboxylate (**2**) with thioacetamide (**1**) (**Scheme-1**) in the presence of zinc oxide powder in solvent-free conditions.

Melting points were measured on an Electrothermal 9100 apparatus and are uncorrected. ¹H and ¹³C NMR spectra were measured with Bruker DRX-500 Avance spectrometer at 500 and 125 MHz, respectively.

Preparation of dimethyl-(Z)-2-(ethanimidoysulfanyl)-2-butenedioate (4): To a magnetically stirred solution of thioacetamide (**1**) (0.075 g, 1 mmol) in acetone (3 mL) was added dropwise a mixture of dimethyl acetylenedicarboxylate (**2**) (1 mmol) in acetone (2 mL) at 10°C over 15 min. The mixture was then stirred at 10°C for 15 min. The mixture was allowed to warm up to room temperature. Thermally activated dry zinc oxide powder (2 g) was added and the solvent was evaporated. Zinc oxide powder and the residue were heated at 90°C for 1 h. (**Scheme-1**) and then placed over a column of silica gel (10 g). The column chromatography was washed using ethyl acetate-light petroleum ether as eluent. The solvent was removed under reduced pressure and product **4** was obtained as light yellow crystals.

The compound **4** may result from initial addition of thioacetamide (**1**) to the acetylenic ester (**2**) and concomitant proton transfer of the 1:1 adduct **3**. Zinc oxide powder was found to catalyze the reaction in solvent-free conditions at 90°C in 1 h. (**Scheme-1**), in fairly high conversion. In the absence of zinc oxide powder, the reaction was completed in acetone in 12 h¹⁰. Zinc oxide powder may have efficient roll in the proton transfer step of the reaction (**Scheme-1**).



In summary, a new and high efficient, one-pot stereoselective method has been developed for preparing of compound **4** in the presence of zinc oxide powder in solvent-free conditions (**Scheme-1**). Other aspects of this process are under investigation.

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