NOTE

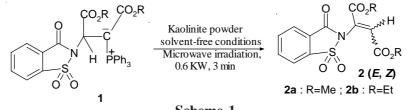
Microwave Induced Stereoselective Conversion of Dialkyl-2-(1,1,3-trioxo-1,3-dihydro-2*H*-1,2-benzisothiazol-2-yl)-3-(triphenylphosphoranylidene)succinates to Dialkyl-2-(1,1,3trioxo-1,3-dihydro-2*H*-1,2-benzisothiazol-2-yl)-2-butendioates

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Kaolinite powder was found to catalyze conversion of stabilized phosphorus ylides to corresponding electron-poor *N*-vinylated isothiazoles in solvent-free conditions under microwave irradiation at microwave power 0.6 KW in 3 min in fairly good yields.

Key Words: Microwave irradiation, Catalyst, Kaolinite, Saccharin, Phosphorus ylide, Solvent-free conditions.

Silica gel as an additive promotes the Wittig reactions^{1,2} of phosphorus ylides with aldehydes, including sterically hindered aldehydes to increase the rate and yields of alkenes^{3,4}. Earlier, a one-pot method for preparing stabilized phosphorus ylides utilizing *in situ* generation of the phosphonium salts is reported⁵. In this paper, the catalytic role of kaolinite powder in the stereoselective conversion of dialkyl-2-(1,1,3-trioxo-1,3-dihydro-2*H*-1,2-benzisothiazol-2-yl)-3-(triphenylphosphoranylidene)succinates (1) to dialkyl-2-(1,1,3-trioxo-1,3-dihydro-2*H*-1,2-benzisothiazol-2-yl)-2-butendioates (2) in solvent-free conditions under microwave irradiation at microwave power 0.6 KW in 3 min in fairly good yields is reported (Scheme-1).



Scheme-1

Commerical oven Butane M245 was used for microwave irradiation. Melting points were measured on an Electrothermal 9100 apparatus and are uncorrected. IR spectra were recorded on a Shimadzu IR-460

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spectrometer. ¹H and ¹³C NMR spectra were measured with a Bruker DRX-500 Avance spectrometer at 500 and 125 MHz, respectively.

General procedure for the preparation of compounds 6a-b: Mixture of Kaolinite powder (2 g) and the ylide 1 (1 mmol) were heated under microwave irradiation at microwave power 0.6 KW in 3 min and then placed over a column of silica gel (10 g). The column chromatography was washed using ethyl acetate-light petroleum ether (1:9) as eluent. The solvent was removed under reduced pressure and products (2a-b) were obtained (2a: White crystals, m.p. 130-137°C; Yield: 46%. Z-isomer: 83%, *E*-isomer: 17 %; **2b**: Viscous colourless oil; Yield: 42%. *Z*-isomer: 66%, E-isomer: 34%). The characterization data of the compounds (6a-b) are given in our previous reports⁷.

Kaolinite powder was found to catalyze stereoselective conversion of dialkyl-2-(1,1,3-trioxo-1,3-dihydro-2H-1,2-benzisothiazol-2-yl)-3-(triphenylphosphoranylidene)succinates (1) to dialkyl-2-(1,1,3-trioxo-1,3-dihydro-2H-1,2-benzisothiazol-2-yl)-2-butendioates (2) in solvent-free conditions under microwave irradiation at microwave power 0.6 KW in 3 min in fairly good yields (Scheme-1). Several catalysts have also been used e.g., K₂HPO₄, KH₂PO₄, MgO, MnSO₄, KHSO₄, ZnO, Al₂O₃, NaHCO₃ and MgSO₄, but vields of corresponding products 2 in cases MgO, K₂HPO₄ and KH₂PO₄ were low and in the others cases no product were observed. The structures **2a-b** were elucidated from their IR, ¹H and ¹³C NMR spectra.

In short, the kaolinite powder is found to catalyze stereoselective conversion of ylides 1 to compounds 2 in solvent-free conditions under microwave irradiation at microwave power 0.6 KW in 3 min in fairly good yields. Other aspects of this process are under investigation.

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