Asian Journal of Chemistry

Reaction of 4-(4-Methylbenzoyl)-5-(4-methylphenyl)furan-2,3-dione with Various Amides

ILHAN OZER ILHAN* and FAHRETTIN ERSAN Department of Chemistry, University of Erciyes, Kayseri-38039, Turkey Fax: (90)(352)4378802; E-mail: ilhano@erciyes.edu.tr

The thermal decomposition of 4-acyl substituted furan-2,3-dione (1) leads to the α -oxo ketene (2) as an intermediate which undergoes nucleophilic addition reactions with amides. The open-chain α -*N*-acyl-oxo-amide derivatives (**4a-f**) which were obtained by these reactions show ketoenol tautomerism. The structures of all the products were confirmed by elemental analysis, IR, ¹H and ¹³C NMR.

Key Words: Furan-2,3-dione, α -Oxoketenes, Amide, Ketoenol, Oxo-amide.

INTRODUCTION

The furan ring systems, though are not found in animal metabolism, occur widely in secondary plant metabolites, especially in terpenoids. Some furandione derivatives are important in medicine¹. 2,3-Dihydro-2,3furandione chemistry attracted a wide interest due to versatile biological activities of 2,3-furandione derivatives, which belong to an essential group of heterocycles². Due to their high reactivity, numerous reports have highlighted their chemistry and use³. In furan-2,3-diones carbon atoms C-2, C-3, C-5 and C-6 represent electrophilic sites of different reactivity and could be used for the construction of condensed heterocyclic systems upon reaction with various nucleophiles and binucleophiles⁴. The reaction of substituted 2,3-furandiones with some dienophiles in different solvents at various temperatures have also been studied⁵. The derivatives of monocyclic heterocyclic 2.3-diones have been found to serve as versatile synthetic equivalents in thermolysis reaction⁶, cycloaddition reactions⁷ and nucleophilic addition reactions⁸. Recently several attempts to change functional groups in furan- or pyrrol-2,3-diones and related systems have been reported9. A convenient method for the synthesis and the mechanism of reactions and calculations on the interaction of 2,3-dihydro-2,3-furandiones with several semicarbazones, ureas and their thio-analogues have been reported recently¹⁰. The reactions are generally initiated by the nucleophilic attack of the nitrogen atom of semicarbazone, urea or hydrazines, directed on the oxygen atom of the furan ring¹¹. The reaction of furan-2,3-dione (1),

1380 Ilhan et al.

Asian J. Chem.

obtained easily from p, p'-dimethyldibenzoylmethane and oxalyl chloride¹², with α -naphthylamine, aniline, p-toluidine and o-toluidine derivatives in benzene at the room temperature led to the formation of the 2-oxo-3-butenoic acid derivative, the pyrole-2,3-dione, the pyrole-2-ones and the quinoxaline-2-ones derivatives, respectively¹³. Thermal decomposition of 2,3-furandiones lead to the formation of highly reactive α -oxoketene intermediates¹⁴. The general reactivity of ketenes and the mechanism of the reactions with nucleophiles have been reviewed recently and a few *ab initio* and semi-empirical (AM1 and PM3) calculations on a series of differently substituted α -oxoketenes, especially substituent effects on their stability as well as reactivity, have been published¹⁵. Therefore, these *vicinal* dione compounds (**1**) have been used as initial materials in the synthesis of the target heterocycles.

EXPERIMENTAL

Solvents were dried by refluxing over the appropriate drying agent and distilled before use. Melting points were determined on an Electrothermal 9200 apparatus and are uncorrected. Microanalyses were performed on a Carlo Erba elemental analyser, model 1108. The infrared spectra were recorded on a Jasco FT-IR spectrometer model 460, using potassium bromide pellets. The ¹H and ¹³C NMR spectra were recorded on Varian Gemini 200 instrument. The chemical shifts are reported in ppm from tetramethylsilane as an internal standard and are given in δ (ppm). All experiments were follwed by TLC using DC Alufolien Kieselgel 60 F₂₅₄ Merck and Camag TLC lamp (254/366 nm).

N-[(2Z)-3-Hydroxy-2-(4-methylbenzoyl)-3-(4-methylphenyl)prop-2-enoyl]benzamide (4a): 4-(4-Methylbenzoyl)-5-(4-methylphenyl)furan-2,3-dione (1) (0.2 g, 0.65 mmol) and benzamide (**3a**) (0.08 g, 0.65 mmol) were refluxed in toluene (50 mL) for 6 h. The solvent was evaporated and the remaining oily residue was treated with dry diethyl ether to give a yellow crude product which was recrystallized from methanol and allowed to dry on P₂O₅; yield 46% (0.12 g); m.p. 190°C; IR (KBr, cm⁻¹) 3400-3150 v(NH, OH), 1720, 1705, 1690, 1680 v(C=O); ¹H NMR (DMSO, δ): 11.00 (br, -NH), 9.47 (s, enol-OH), 8.10-7.39 (m, 13H, Ar-H), 7.28 (s, keto-CH), 2.52-2.28 ppm (s, 6H, CH₃); ¹³C NMR (d₆-DMSO, δ): 192.43 (t, ArCO), 172.68 (N-C=O), 167.81 (Cl=O), 137.54 (Ar-C3-OH), in enol form 134.22-126.91 (m, aromatic C), 106.40 (s, C2), 67.18 (diketo, CH) and 26.92 and 25.82 ppm (s, CH₃). Elemental analysis: found (calcd.): [C = 75.18 (74.93), H = 5.26 (5.55), N = 3.51 (3.31)].

N-[(2Z)-3-Hydroxy-2-(4-methylbenzoyl)-3-(4-methylphenyl)prop-2-enoyl]nicotinamide (4b): 4-(4-Methylbenzoyl)-5-(4-methylphenyl)furan-2,3-dione (1) (0.3 g, 0.98 mmol) and nicotinamide (3b) (0.12 g, 0.98 Vol. 19, No. 2 (2007) Reaction of 5-(4-methylphenyl)furan-2,3-dione with Various Amides 1381

mmol) were refluxed in benzene (50 mL) for 6 h. The solvent was evaporated and the remaining oily residue was treated with dry diethyl ether. The crude product was then filtered and recrystallized from methanol and allowed to dry on P₂O₅; yield 0.16 g (41%); m.p. 195°C; IR (KBr, cm⁻¹) 3255-3155 v(NH, OH), 1725, 1715, 1685, 1680 v(C=O); ¹H NMR (DMSO, δ): 12.45 (br, -NH), 9.28 (s, enol-OH), 8.86-7.36 (m, 12H, Ar-H), 7.25 (s, keto-CH), 2.53-2.39 ppm (s, 6H, CH₃); ¹³C NMR (d₆-DMSO, δ): 191.96 (t, ArCO), 170.52 (N-C=O), 165.46, 155.11 and 151.16 (C-N=C), (Cl=O), 138.04 (Ar-C3-OH, in enol form) 138.99-116.06 (m, aromatic C), 106.44 (s, C2), 68.33 (diketo, CH) and 26.82 and 25.82 ppm (s, CH₃). Elemental analysis: found (calcd.): [C = 72.01 (71.72), H = 5.04 (5.08), N = 7.09 (7.25)].

(2Z)-N-Acetyl-3-hydroxy-2-(4-methylbenzoyl)-3-(4-methylphenyl)acrylamide (4c): 4-(4-Methylbenzoyl)-5-(4-methylphenyl)furan-2,3-dione (1) (0.2 g, 0.65 mmol) and acetamide (3c) (0.04 g, 0.65 mmol) were refluxed in toluene (50 mL) for 6 h. The solvent was evaporated and the remaining oily residue was treated with dry diethyl ether. The crude product was then filtered and recrystallized from methanol and allowed to dry on P₂O₅; yield 0.095 (45%); m.p. 180°C; IR (KBr, cm⁻¹) 3270-3170 v(NH, OH), 1735, 1715, 1690, 1680 v(C=O); ¹H NMR (d₆-DMSO, δ): 11.27 (br, -NH), 9.23 (s, enol-OH), 7.84-7.33 (m, 8H, Ar-H), 7.05 (s, keto-CH), 2.52-2.28 ppm (s, 6H, arom, CH₃) 2.10 ppm (s, CH₃); ¹³C NMR (DMSO, δ): 190.37 (t, ArCO), 172.71 (N-C=O), 166.59 (Cl=O), 133.83 (enol-OH), 133.64-115.32 (m, aromatic C), 106.71 (s, C2), 66.96 (diketo, CH), 27.18 (methyl-C=O) and 26.92 and 25.82 ppm (s, CH₃). Elemental analysis: found (calcd.): [C = 71.21 (71.43), H = 5.63 (5.35), N = 4.15 (4.12)].

N-[(2Z)-3-Hydroxy-2-(4-methylbenzoyl)-3-(4-methylphenyl)prop-2-enoyl]buthanamide (4d): 4-(4-Methylbenzoyl)-5-(4-methylphenyl)furan-2,3-dione (1) (0.3 g, 0.98 mmol) and *n*-butyramide (3d) (0.085 g, 0.98 mmol) were refluxed in benzene (50 mL) for 6 h. The solvent was evaporated. The remaining oily residue was dissolved in dry diethyl ether. Petroleum ether was then added. The white product was then filtered and recrystallized from methanol and allowed to dry on P₂O₅; yield 0.16 g (45%); m.p. 205°C; IR (KBr, cm⁻¹) 3280-3150 v(NH, OH), 1735, 1705, 1695, 1675 v(C=O); ¹H NMR (d₆-DMSO, δ): 11.21 (br, -NH), 8.92 (s, enol-OH), 8.11-7.30 (m, 8H, Ar-H), 7.09 (s, keto-CH), 2.53-2.50 (s, 6H, arom, CH₃), 2.49-2.34 (-CO-CH₂-), 1.56-1.44 (-CH₂-), 0.89-0.82 ppm (-CH₃); ¹³C NMR (DMSO, δ): 191.55 (t, ArCO), 172.50 (N-C=O), 167.01 (Cl=O), 136.54 (enol-OH), 133.72-123.91 (m, aromatic C), 106.50 (s, C2), 67.18 (diketo, CH) and 26.91 and 25.80 ppm (s, CH₃). Elemental analysis: found (calcd.): [C = 72.32 (71.97), H = 6.30 (6.35), N = 3.83 (3.55)]. 1382 Ilhan et al.

Asian J. Chem.

N-[(2Z)-3-Hydroxy-2-(4-methylbenzoyl)-3-(4-methylphenyl)prop-2-enoyl]hexanamide (4e): 4-(4-Methylbenzoyl)-5-(4-methylphenyl)furan-2,3-dione (**1**) (0.2 g, 0.65 mmol) and *n*-hexanamide (**3e**) (0.075 g, 0.65 mmol) were refluxed in benzene (50 mL) for 6 h. The solvent was evaporated. The remaining oily residue was dissolved in dry diethyl ether. Petroleum ether was then added. The white product was filtered and recrystallized from methanol and allowed to dry on P₂O₅; yield 0.1 g (40%); m.p. 224°C; IR (KBr, cm⁻¹) 3425-3170 v(NH, OH), 1705, 1695, 1680, 1655 v(C=O); ¹H NMR (d₆-DMSO, δ): 11.28 (br, -NH), 9.42 (s, enol-OH), 8.11-7.24 (m, 8H, Ar-H), 6.69 (s, keto-CH), 2.49-2.40(s, 6H, arom, CH₃), 0.89-0.81 ppm (s, CH₃); ¹³C NMR (DMSO, δ): 192.37 (t, ArCO), 171.68 (N-C=O), 165.81 (Cl=O), 137.54 (enol-OH) 130.22-126.91 (m, aromatic C), 67.22 (diketo, CH) and 26.92 and 25.82 ppm (s, CH₃). Elemental analysis: found (calcd.): [C = 73.28 (73.55), H = 6.87 (6.55), N = 3.56 (3.33)].

(2Z)-N-Ethanethioryl-3-hydroxy-2-(4-methylbenzoyl)-3-(4methylphenyl)acrylamide (4f): 4-(4-Methylbenzoyl)-5-(4-methylphenyl)furan-2,3-dione (1) (0.2 g, 0.65 mmol) and thioacetamide (3f) (0.05 g, 0.65 mmol) were refluxed in benzene (50 mL) for 6 h. The solvent was evaporated and the remaining oily residue was treated with dry diethyl ether. The crude product was then filtered and recrystallized from methanol and allowed to dry on P₂O₅; yield 0.11 g (48%); m.p. 175°C; IR (KBr, cm⁻¹) 3300-3150 v(NH, OH), 1705, 1690, 1680 v(C=O), 1300 v(C=S); ¹H NMR (DMSO, δ): 11.11 (br, -NH), 9.30 (s, enol-OH), 8.11-7.36 (m, 8H, Ar-H), 7.29 (s, keto-CH), 2.53-2.41 ppm (s, 6H, CH₃). Elemental analysis: found (calcd.): [C = 67.98 (67.82), H = 5.38 (5.57), N = 3.97 (3.65) S = 9.07 (9.38)].

RESULTS AND DISCUSSION

In the present study, we carried out the reaction of the furan-2,3-dione (1) with amide derivatives (**3a-f**) yields α -*N*-acyl-oxo-amide derivatives (**4a-f**) in boiling benzene and toluene (Experimental). The reaction equations are shown in **Scheme**. Compounds **4a-f** were obtained by the addition of amides to α -oxoketene (2) which is formed by heating of 1. The strucutres of **4a-f** were confirmed by elemental analysis, IR, ¹H NMR and ¹³C NMR spectral data. The IR and NMR spectral data of compunds **4a-f** indicate the existence of keto-enol tautomerism in these compounds. Product **4a** was obtained by treating 1 with benzamide (**3a**) in 46% yield and refluxing the mixture in toluene for 6 h. In the IR spectra of compound **4a** the C=O absorption bands are found at about 1720, 1705, 1690, 1680 cm⁻¹. The -NH and -OH absorption bands are found at about 3400-3150 cm⁻¹. From the ¹H NMR spectrum of **4a**, important structural information can be obtained. The ¹H NMR peak of **4a** observed at 7.28 ppm belongs to the

-CH in the keto form for which the chemical shift values of very close analogs of α -oxo-carboxylic and derivatives are found between 6.82 and 6.47 ppm^{6a,d,f}. The peaks at 9.47 ppm is thought to represent the -OH in the enol form. The broad peak at 11.00 represents the -NH. This indicates the existence of a keto-enol tautomerism. In the ¹³C NMR spectrum of **4a** the peak at 26.92 and 25.82 ppm belongs to the methyl groups the other peak at 67.18 ppm represents the -CH in the diketo form. The peaks corresponding to 172.68 and 192.43 ppm indicate imide-C=O and benzoyl-C=O, respectively. More information about the **4a-f** spectrums is given in the experimental section.



ACKNOWLEDGEMENT

The authors wish to express their gratitude to Prof. Dr. I. Yildirim for helpful and encouraging discussions. This study was financially supported by the Research Center of Erciyes University.

REFERENCES

- (a) H. Zimmer, R. Palmer-S. and D. Ho, *J. Heterocycl. Chem.*, **30**, 161 (1993); (b) J.A. Joule, K. Mills and G.F. Smith, Heterocyclic Chemistry, Chapman & Hall, London, U.K., edn. 3, p. 278 (1995).
- 2. G. Kollenz and W. Heilmayer, Trends in Heterocycl. Chem., 3, 379 (1993).
- (a) E. Ziegler, M. Eder, C. Belegratis and E. Prewedourakis, *Monatsh Chem.*, 98, 2249 (1967) (b) G. Kollenz, C.O. Kappe and H.A.A. El-Nabi, *Heterocycles*, 32, 669 (1991);
 (c) R.W. Saalfrank, T. Lutz, B. Horner, J. Gundel, K. Peters and H.G. Von Schnering, *Chem. Ber.*, 124, 2289 (1991); (d) E. Saripinar, Y. Guzel, Z. Onal, I.O. Ilhan and Y. Akcamur, *J. Chem. Soc. Pak.*, 22, 308 (2000); (e) T. Hokelek, E. Saripinar, I. Yildirim,

1384 Ilhan et al.

Asian J. Chem.

M. Akkurt and Y. Akcamur, Acta Cryst., 58E, 30 (2002).

- (a) G. Kollenz, E. Ziegler, W. Ott and H. Igel, Z. Naturforschg, **31b**, 1511 (1976);
 (b) W. Ott, E. Ziegler, G. Kollenz, Synthesis, **7**, 477 (1976);
 (c) Y. Akcamur, G. Penn, E. Ziegler, H. Sterk, G. Kollenz, K. Peters, E.-M. Peters, H.G. Von Schnering, Monatsh Chem., **117**, 231 (1986);
 (d) Y. Akcamur and G. Kollenz, Org. Prep. Proced. Int., **19**, 52 (1987);
 (e) Y. Akcamur, A. Sener, A.M. Ipekoglu and G. Kollenz, J. Heterocycl. Chem., **34**, 221 (1997).
- (a) G. Kollenz, G. Penn, G. Dolenz, Y. Akcamur, K. Peters, E.-M. Peters and H.G. Von Schnering, *Chem. Ber.*, **117**, 1299 (1984); (b) G. Kollenz, W. Ott, E. Ziegler, E.-M. Peters, K. Peters, H.G. Von Schnering, V. Formacek and H. Quast, *Liebigs Ann. Chem.*, 1137 (1984); (c) E. Terpetschnig, W. Ott, G. Kollenz, K. Peters, E.-M. Peters, H.G. Von Schnering, *Monatsh Chem.*, **119**, 367 (1988); (d) G. Kollenz, H. Sterk and G. Hutter, *J. Org. Chem.*, **56**, 235 (1991).
- (a) E. Ziegler, G. Kollenz and W. Ott, *Synthesis*, 679 (1973); (b) E. Ziegler, G. Kollenz and W. Ott, *Justus Annalen Der Chemie*, **11**, 2071 (1976); (c) E. Ziegler, G. Kollenz, G. Kriwetz and W. Ott, *Liebigs Ann. Chem.*, **10**, 1751 (1977); (d) C.O. Kappe, G. Farber, C. Wentrup and G. Kollenz, *J. Org. Chem.*, **57**, 7078 (1992) (e) B. Fulloon, H.A.A. El-Nabi, G. Kollenz and C. Wentrup, *Tetrahedron Lett.*, **36**, 6547 (1995); (f) H.A.A. El-Nabi and G. Kollenz, *Monatsh Chem.*, **128**, 381 (1997); (g) G. Kollenz, W. Heilmayer, C.O. Kappe, B. Wallfisch and C. Wentrup, *Croat. Chim. Acta*, **74**, 815 (2001).
- (a) G. Kollenz, G. Penn, W. Ott, K. Peters, E.-M. Peters and H.G. Von Schnering, *Heterocycles*, 26, 625 (1987); (b) E. Terpetschnig, G. Penn, G. Kollenz, K. Peters, E.-M. Peters and H.G. Von Schnering, *Tetrahedron*, 47, 3045 (1991).
- (a) G. Kollenz, H. Igel and E. Ziegler, *Monatsh Chem.*, **103**, 450 (1972); (b) G. Kollenz, E. Ziegler and W. Ott, *Org. Prep. Proced. Int.*, **5**, 261 (1973); (c) G. Kollenz, E. Ziegler, W. Ott and G. Kriwetz, *Naturforschg*, **32B**, 701 (1977); (d) B. Altural, Y. Akcamur, E. Saripinar, I. Yildirin, G. Kollenz, *Monatsh Chem.*, **120**, 1015 (1989); (e) I. Yildirim, Y. Akcamur, E. Saripinar and G. Kollenz, *Kuwait J. Sci. Eng.*, **29**, 57 (2002).
- 9. (a) W.M.F. Fabian, G. Kollenz and C.O. Kappe, J. Mol. Struct., 313, 219 (1994);
 (b) M.W. Wong and C. Wentrup, J. Org. Chem., 59, 5279 (1994).
- (a) Y. Akcamur, B. Altural, E. Saripinar, G. Kollenz, C.O. Kappe, E.-M. Peters and H.G. Von Schnering, *J. Heterocycl. Chem.*, 25, 1419 (1988); (b) I. Yildirim, E. Saripinar, Y. Guzel, S. Patat and Y. Akcamur, *J. Mol. Struct.*, 334, 165 (1995); (c) I. Yildirim, M. Tezcan, Y. Guzel, E. Saripinar and Y. Akcamur, *Turk. J. Chem.*, 20, 27 (1996); (d) E. Saripinar, I. Yildirim, Y. Guzel and Y. Akcamur, *Monatsh Chem.*, 127, 505 (1996); (e) I. Yildirim, Y. Akcamur, E. Saripinar and G. Kollenz, *Kuwait J. Sci. Eng.*, 29, 57 (2002).
- (a) A. Sener, R. Kasimogullari, M.K. Sener, I. Bildirici and Y. Akcamur, *J. Heterocycl. Chem.*, **39**, 869 (2002); (b) I.O. Ilhan, Y. Akcamur, E. Saripinar and E. Aslan, *Asian J. Chem.*, **15**, 1373 (2003).
- 12. I. Yildirim and I. Koca, Kuwait J. Sci. Eng., 32, 49 (2005).
- 13. I. Yildirim and I. Koca, Asian J. Chem., 16, 899 (2004).
- (a) W.M.F. FAbian, G. Kollenz, Y. Akcamur, T.R. Kok, M. Tezcan, M. Akkurt and W. Hiller, *Monatsh Chem.*, **123**, 265 (1992); (b) R.L. Toung and C. Wentrup, *Tetrahedron Lett.*, **48**, 761 (1992); (c) I. Yildirim and I.O. Ilhan, *J. Heterocycl. Chem.*, **34**, 1047 (1997); (d) A. Stadler, K. Zangger, F. Belaj and G. Kollenz, *Tetrahedron*, **57**, 6757 (2001).
- (a) R. Janoschek, W.M.F. Fabian and G. Kollenz, *J. Comput. Chem.*, **15**, 132 (1994);
 (b) D.M. Birney, *J. Org. Chem.*, **59**, 2557 (1994);
 (c) F.P. Cassio, J.M. Ugalde, X. Lopez, B. Lecea and C. Palomo, *J. Am. Chem. Soc.*, **115**, 995 (1993);
 (d) L. Gong, M.A. Allister and T.T. Tidwell, *J. Am. Chem. Soc.*, **113**, 6021 (1991).

(Received: 2 January 2006; Accepted: 18 September 2006) AJC-5103