Synthesis and Characterization of Some Biologically Significant Thiazolidinones

NADENDLA RAMA RAO*, B. RAMU and C. GOPINATH

KVSR Siddhartha College of Pharmaceutical Sciences Polyclinic Road, Vijayawada520 010, India E-mail: nrrchandra@hotmail.com, nadendla2000@yahoo.com.in Website: www.geocities.com/sriramachandra2001

Some new 2-(aryl or substituted aryl)-3-(3'-fluoro-4'-chloro phenyl) thiazolidinones were synthesized by cyclocondensation of anils with thioglycollic acid. The synthesized compounds were established by spectral analysis and evaluated for antimicrobial properties.

Key Words: Fluorochloroaniline, Thioglycollic acid, Schiff bases, Antimicrobial activity.

Thiazolidinones have been found to be biologically versatile compounds which possess a broad spectrum of biological activities ¹⁻³, viz., antiinflammatory, antiparkinsonian, anthelmintic, antibacterial, anticonvulsant and hypotensive in animals as well as human beings. Thiazolidinones were synthesized by the cyclocondensation of anils of aromatic or heterocyclic systems by the action of thioglycollic acid⁴. In addition fluorine-containing compounds have widely spread into our modern life from the kitchen utensils to steroids, with quite broad applications in electronic, agricultural and medicinal industries. Keeping in view these valid observations and in continuation of our research for biologically active fluorine derivatives⁵⁻⁷, it was planned to synthesise some new fluorinated chalcones.

Preparation of Schiff bases (I)

3-Fluoro-4-chloroaniline (0.01 mol) was refluxed with benzaldehyde (0.01 mole) in ethanol (30 mL) at 80°C for 2 h. The liquid obtained was poured over crushed ice; the solid obtained was filtered off, dried and recrystallised from DMF. Similarly, other substituted Schiff bases were prepared with different aromatic aldehydes (salicylaldehyde, nitrobenzaldehyde and p-dimethylaminobenzaldehyde).

Syntheis of 2-(aryl or substituted aryl)-3-(-3'-fluoro-4'-chloro phenyl) thiazolidinones

The mixture of Schiff base (I) (0.01 mole) and thioglycollic acid (0.02 mole) in dry DMF with a pinch of zinc dust was refluxed on oil bath for 8 h. The excess solvent was distilled off and the reaction mixture was poured on to ice. The separated solid was filtered off, washed with distilled water, dried and recrystallized from DMF. Similarly, other thiazolidinones were synthesized and melting points were determined in open capillaries in a liquid paraffin bath and are uncorrected. IR spectra (KBr) were recorded on a Perkin-Elmer 783 spectrophotometer. The structures of the compounds were established on the basis of their elemental analysis and IR spectra. The IR spectra of the compound 3a show characteristic bands (cm⁻¹) at: 3040 v(aromatic C—H), 2250 v(CH₂ of thiazolidinone ring), 1140 v(C—S) and 1760 v(C=O of thiazolidinone ring). The yields, melting points and elemental analysis data are shown in Table-1.

TABLE-1						
ANALYTICAL AND PHYSICAL DATA OF THE SYNTHESISED C	OMPOUNDS					

Compd. No.	R	m.f.	m.w.	Yield (%)	Elemental analysis (%) Found (Calcd.)		
					С	Н	N
IIa	—C ₆ H ₅	C ₁₅ H ₁₁ CIFNOS	307.77	82	58.50 (58.54)	3.60 (3.58)	4.55 (4.50)
IIb	-2-OH—C ₆ H ₄	C ₁₅ H ₁₁ ClFNO ₂ S	323.77	70	55.64 (55.50)	3.42 (3.30)	4.33 (4.30)
IIc	-4-NO ₂ —C ₆ H ₄	C ₁₅ H ₁₀ CIFNO ₃ S	352.77	65	57.07 (57.00)	2.86 (2.75)	7.94 (7.80)
IId	-4-N[(CH ₃) ₂]—C ₆ H ₄	C ₁₇ H ₁₆ CIFN ₂ OS	350.84	68	58.19 (56.78)	4.59 (4.60)	7.98 (7.20)

Antibacterial activity

The final compounds were evaluated for antibacterial activity by cup-plate method at a concentration of 100 µg/mL against the test organism, *Staphylococcus aureus*. The zone of inhibition was compared with standard ampicillin (100 µg/mL). The results obtained are recorded in Table-2.

TABLE-2 ANTBACTERAL ACTIVITIES OF FLUORINATED THIAZOLIDINONES

Compound (100 µg/mL)	Zone of Inhibition (mm) S. aurėus			
Ampicillin (Standard)	25.00			
IIIa	10.00			
Шь	14.00			
IIIc	10.00			
IIId	16.00			

Some new fluorinated thiazolidinones have been prepared and IR studies have supported the constitution. The products have been screened for their antimicrobial activity. Compound IId showed significant activity against Staphylococcus aureus. Compound IIb showed moderate activity whereas compounds IIa and IIc were less active against the same bacteria.

ACKNOWLEDGEMENT

The authors are thankful to Darwin Research Laboratories (Vijayawada) for providing necessary facilities.

REFERENCES

- 1. V.K. Jain and J.T. Rao, Asian J. Chem., 15, 483 (2003)
- 2. D.P. Gupta, S. Ahmad, A. Kumar and K. Shankar, Indian J. Chem., 27B, 1060 (1998).
- 3. J. Korenet, T. Varia and W. Beaven, J. Heterocyclic Chem., 20, 1553 (1984).
- 4. M. Tyagi, V.K. Srivastava, R. Chandra and A. Kumar, Indian J. Chem., 41B, 2367 (2002).
- 5. N. Ramarao and G.S. Rao, J. Institution of Chemists (India), 36, (2002) (in press).
- -, Acta Ciencia Indica, 8624/c (2002) (in press).
- 7. N. Ramarao, G.S. Rao, E. Jayachandran, Presented in 54th Indian Pharmaceutical Congress at Pune (2002).

(Received: 7 April 2003; Accepted: 7 November 2003) AJC-3224