

## Synthesis and Crystal Structure of Cobalt(II) Complex of 4,4'-Diamino-1-[(cyclohexa-2,4-dienyl)methyl]benzene

ZHENHUA MEI<sup>†</sup>, XIANG NIU, JIE SUN and BIN QU\*

College of Chemistry and Molecular Engineering  
Qingdao University of Science and Technology, Qingdao-266042, P.R. China  
Tel/Fax: (86)(532)84022750; E-mail:qustchemistry@126.com

The crystal structure of cobalt(II) complex of 4,4'-diamino-1-[(cyclohexa-2,4-dienyl)methyl]benzene has been determined by single crystal X-ray diffraction method. The crystal belongs to orthorhombic system, space group  $P_1$  with unit cell constants  $a = 9.2500(8)$ ,  $b = 11.6619(10)$ ,  $c = 11.9660(11)$  Å,  $V = 1273.36(19)$  Å<sup>3</sup>,  $Z = 1$ ,  $D_c = 1.363$  g/cm<sup>3</sup>,  $\mu = 0.406$  mm<sup>-1</sup>,  $F(000) = 550$ ,  $R$  and  $wR$  are 0.0499 and 0.1498, respectively for 5981 unique reflections with 5095 observed reflections ( $I > 2\sigma(I)$ ). In the crystal structure, molecules are linked into chains along the  $a$  axis by N-H···O intermolecular hydrogen bonds along the  $a$  axis.

**Key Words:** Synthesis, Crystal structure, Cobalt(II) complex, 4,4'-Diamino-1-[(cyclohexa-2,4-dienyl)methyl]benzene.

### INTRODUCTION

The investigation of transition metal complexes become a very important field in the development of DNA molecule probes and chemotherapeutics<sup>1-6</sup>. Among these complexes, metals or ligands can be varied in an easily controlled way to facilitate the individual applications<sup>7-10</sup>. Trace element cobalt is an important element of life. The cobalt complexes as metal ions in biological/enzyme/substrate interaction simulation study has been very active. Cobalt complexes are good indicators for the hybridization detection of DNA in electrochemical biosensors<sup>11</sup>. In search for new indicators, a new complex of cobalt(II) with 4,4'-diamino-1-[(cyclohexa-2,4-dienyl)methyl]benzene was synthesized and its structure is elucidated.

### EXPERIMENTAL

All chemicals were of analytical reagent grade and used directly without further purification. 5 mL of aqueous solution of  $\text{Co}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$  (0.5 mmol, 0.1456 g) was added dropwise to 30 mL of ethanol solution of 4,4'-diaminobiphenyl methane (2 mmol, 0.3968 g) with constant stirring. The mixture was filtered after refluxed for 2 h. The filtrate was left to stand undisturbed and slow evaporation at room temperature for 2 weeks. Crystal suitable for X-ray diffraction analysis were obtained. Anal. data for the title compound: C, 59.76; H, 5.88; N, 13.40 ; Found: C, 59.11; H, 6.01; N, 13.76.

<sup>†</sup>College of Life Science and Pharmaceutical Engineering, Nanjing University of Technology, Nanjing 210009, P.R. China.

**Crystal data and structure determination:** A colourless single crystal with approximate dimension of  $0.21\text{ mm} \times 0.19\text{ mm} \times 0.16\text{ mm}$  was mounted on glass fibre in a random orientation. The data were collected by Bruker Smart 1000 CCD diffractometer with graphite minochromated MoK $\alpha$  radiation ( $\lambda = 0.71073\text{ \AA}$ ) using  $\omega$  scan mode in the range of  $2.27 \leq \theta \leq 25.56^\circ$  at temperature  $293(2)\text{ K}$ . A total of 7113 reflections were collected with 5981 unique ones ( $R_{\text{int}} = 0.012$ ), of which 5095 reflections with  $I > 2\sigma(I)$  were considered to be observed and used in the succeeding refinements. Intensity data were corrected for Lp factors and empirical absorption. The structure was solved by direct methods<sup>12</sup> and expanded by using Fourier differential techniques with SHELXL-9712. All non-hydrogen atoms were located with successive difference Fourier syntheses. The structure was refined by full-matrix least-squares method on  $F^2$  with anisotropic thermal parameters for all non-hydrogen atoms. Hydrogen atoms were added according to the theoretical models. Full matrix least-squares refinement gave the final  $R = 0.0449$  and  $wR = 0.1498$ ,  $w = 1/[\sigma^2(F_o)^2 + (0.1000P)^2]$  where  $P = (F_o^2 + 2F_c^2)/3$ .

## RESULTS AND DISCUSSION

The asymmetric unit of the title compound is composed of 3 parts, *viz.*, one crystallographically independent  $[\text{Co}(\text{H}_2\text{O})_2\text{L}_4]^{2+}$  cations ( $\text{L} = \text{C}_{13}\text{H}_{14}\text{N}_2$ ), two water molecules and two nitrate anions. Fig. 1 shows a perspective view of the monomeric unit with the atomic numbering scheme of the title compound. Fig. 2 shows a perspective view of the crystal packing in the unit cell for the complex. Atomic coordinates and equivalent isotropic displacement paraments are presented in Table-1. Selected bond lengths and angles are presented in Tables 2 and 3.

TABLE-1  
ATOMIC COORDINATES AND EQUIVALENT ISOTROPIC  
DISPLACEMENT PARAMETERS ( $\text{\AA}^2$ )

| Atom | x           | y           | z           | Ueq        |
|------|-------------|-------------|-------------|------------|
| Co1  | 0.68066(15) | 0.94367(11) | 0.83349(11) | 0.0359(2)  |
| O1W  | 0.6299(7)   | 0.7688(5)   | 0.8671(5)   | 0.0553(19) |
| O2W  | 0.7400(6)   | 1.1161(4)   | 0.8035(4)   | 0.0469(19) |
| N1   | 0.6627(11)  | 0.5022(6)   | 1.6504(7)   | 0.071(3)   |
| N2   | 0.4669(7)   | 0.9856(6)   | 0.9093(5)   | 0.043(2)   |
| N3   | 0.3181(11)  | 1.4269(8)   | -0.0377(7)  | 0.071(3)   |
| N4   | 0.5682(8)   | 0.9529(6)   | 0.6748(5)   | 0.046(2)   |
| N5   | 1.0427(10)  | 0.4658(6)   | 1.7057(6)   | 0.056(3)   |
| N6   | 0.7966(8)   | 0.9358(6)   | 0.9938(5)   | 0.041(2)   |
| N7   | 0.6998(11)  | 1.3889(8)   | 0.0203(7)   | 0.081(4)   |
| N8   | 0.8978(8)   | 0.8984(6)   | 0.7533(5)   | 0.045(2)   |
| C1   | 0.5748(8)   | 0.7358(7)   | 1.4191(6)   | 0.045(2)   |
| C2   | 0.6381(10)  | 0.6431(8)   | 1.4819(7)   | 0.050(3)   |
| C3   | 0.5848(12)  | 0.5902(8)   | 1.5840(7)   | 0.052(3)   |
| C4   | 0.4535(10)  | 0.6316(8)   | 1.6225(7)   | 0.054(3)   |

| Atom | x          | y          | z         | Ueq      |
|------|------------|------------|-----------|----------|
| C5   | 0.3839(11) | 0.7174(10) | 1.5612(7) | 0.060(3) |
| C6   | 0.4397(9)  | 0.7788(7)  | 1.4560(6) | 0.043(3) |
| C7   | 0.3559(10) | 0.8836(9)  | 1.3952(6) | 0.058(3) |
| C8   | 0.3904(10) | 0.9121(9)  | 1.2667(7) | 0.046(3) |
| C9   | 0.4582(11) | 1.0141(7)  | 1.2167(8) | 0.056(3) |
| C10  | 0.4814(10) | 1.0380(8)  | 1.1024(7) | 0.046(3) |
| C11  | 0.4396(8)  | 0.9629(7)  | 1.0303(6) | 0.037(2) |
| C12  | 0.3703(9)  | 0.8574(7)  | 1.0833(7) | 0.044(3) |
| C13  | 0.3470(10) | 0.8336(8)  | 1.1969(7) | 0.045(3) |
| C14  | 0.6163(10) | 1.1300(8)  | 0.5441(6) | 0.053(3) |
| C15  | 0.5190(9)  | 1.0611(7)  | 0.6151(6) | 0.037(2) |
| C16  | 0.3786(8)  | 1.1002(7)  | 0.6318(5) | 0.041(2) |
| C17  | 0.3339(10) | 1.2064(8)  | 0.5761(7) | 0.050(3) |
| C18  | 0.4272(9)  | 1.2799(8)  | 0.5020(7) | 0.042(3) |
| C19  | 0.5692(10) | 1.2384(9)  | 0.4910(7) | 0.052(3) |
| C20  | 0.3837(10) | 1.3933(7)  | 0.4418(6) | 0.048(3) |
| C21  | 0.3735(8)  | 1.4003(7)  | 0.3138(6) | 0.040(2) |
| C22  | 0.3271(11) | 1.3081(8)  | 0.2674(7) | 0.052(3) |
| C23  | 0.3049(10) | 1.3138(8)  | 0.1508(7) | 0.052(3) |
| C24  | 0.3340(9)  | 1.4128(8)  | 0.0785(7) | 0.048(3) |
| C25  | 0.3902(12) | 1.5067(8)  | 0.1232(7) | 0.055(3) |
| C26  | 0.4073(9)  | 1.4995(7)  | 0.2403(6) | 0.041(2) |
| C27  | 0.9541(11) | 0.3878(8)  | 1.4277(7) | 0.052(3) |
| C28  | 0.9706(10) | 0.3765(7)  | 1.5416(7) | 0.046(3) |
| C29  | 1.0214(9)  | 0.4670(7)  | 1.5875(6) | 0.047(3) |
| C30  | 1.0560(10) | 0.5728(7)  | 1.5119(7) | 0.046(3) |
| C31  | 1.0437(9)  | 0.5815(6)  | 1.4005(7) | 0.040(2) |
| C32  | 0.9915(9)  | 0.4885(7)  | 1.3518(7) | 0.044(3) |
| C33  | 0.9822(11) | 0.4898(7)  | 1.2273(7) | 0.056(3) |
| C34  | 0.9299(9)  | 0.6085(7)  | 1.1615(6) | 0.042(3) |
| C35  | 0.7915(10) | 0.6454(9)  | 1.1789(7) | 0.053(3) |
| C36  | 0.7485(10) | 0.7505(8)  | 1.1228(7) | 0.051(3) |
| C37  | 0.8431(9)  | 0.8238(7)  | 1.0521(6) | 0.038(3) |
| C38  | 0.9799(10) | 0.7882(8)  | 1.0381(7) | 0.053(3) |
| C39  | 1.0236(9)  | 0.6769(8)  | 1.0916(7) | 0.044(3) |
| C40  | 0.8793(11) | 0.8501(8)  | 0.5653(8) | 0.053(3) |
| C41  | 0.9277(9)  | 0.9269(7)  | 0.6325(6) | 0.040(2) |
| C42  | 0.9946(9)  | 1.0243(8)  | 0.5884(6) | 0.043(3) |
| C43  | 1.0183(10) | 1.0528(8)  | 0.4707(7) | 0.047(3) |
| C44  | 0.9715(10) | 0.9803(8)  | 0.3973(7) | 0.044(3) |
| C45  | 0.9038(9)  | 0.8810(8)  | 0.4466(6) | 0.048(3) |
| C46  | 0.9978(11) | 1.0071(9)  | 0.2703(6) | 0.065(4) |
| C47  | 0.9226(9)  | 1.1078(9)  | 0.2080(7) | 0.048(3) |
| C48  | 0.9787(10) | 1.1626(8)  | 0.1063(7) | 0.048(3) |
| C49  | 0.9034(12) | 1.2578(9)  | 0.0453(7) | 0.066(4) |
| C50  | 0.7750(11) | 1.2959(9)  | 0.0853(7) | 0.052(3) |
| C51  | 0.7147(10) | 1.2405(9)  | 0.1885(7) | 0.054(3) |
| C52  | 0.7985(11) | 1.1475(9)  | 0.2491(6) | 0.058(3) |

| Atom | x          | y         | z         | Ueq      |
|------|------------|-----------|-----------|----------|
| O1   | 1.2239(8)  | 1.2022(6) | 0.8701(6) | 0.067(3) |
| O2   | 1.0174(8)  | 1.2022(6) | 0.7898(5) | 0.060(3) |
| O3   | 1.1026(8)  | 1.0436(5) | 0.8815(5) | 0.064(3) |
| N9   | 1.1172(7)  | 1.1483(6) | 0.8483(5) | 0.042(2) |
| O4   | 0.2615(8)  | 0.8429(6) | 0.7861(6) | 0.066(3) |
| O5   | 0.3462(8)  | 0.6827(6) | 0.8744(5) | 0.058(3) |
| O6   | 0.1407(8)  | 0.6844(7) | 0.7950(6) | 0.065(3) |
| N10  | 0.2479(9)  | 0.7372(7) | 0.8185(6) | 0.054(3) |
| O3W  | 0.8150(10) | 0.5982(7) | 0.8291(8) | 0.100(4) |
| O4W  | 0.5542(10) | 0.2862(7) | 1.8446(8) | 0.092(3) |

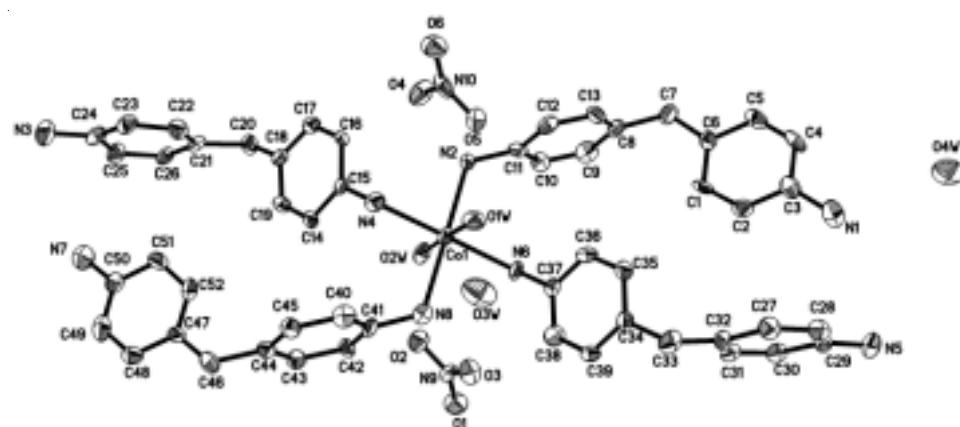


Fig. 1. The molecular structure of the title compound with the atomic numbering scheme

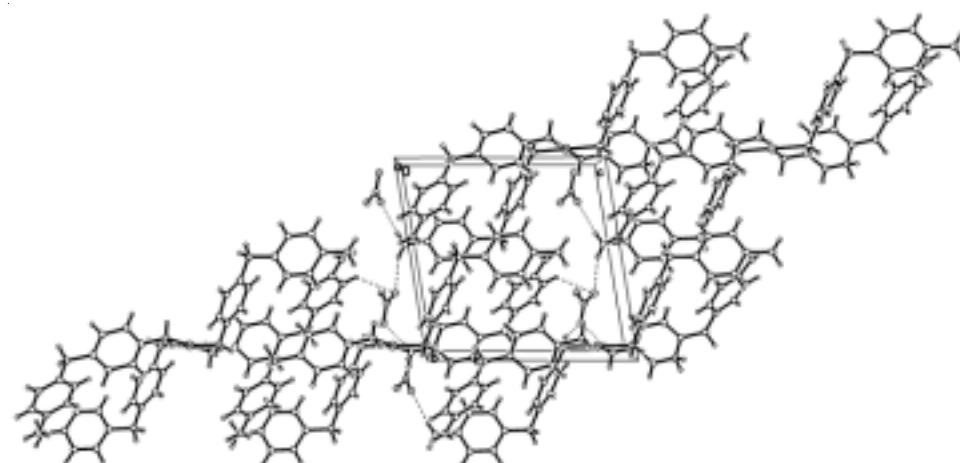


Fig. 2. A view of the crystal packing for the title compound

TABLE-2  
SELECTED BOND LENGTHS (Å)

| Bond    | Dist.     | Bond    | Dist.     |
|---------|-----------|---------|-----------|
| Co1-O1W | 2.069(6)  | C17-C18 | 1.405(12) |
| Co1-O2W | 2.061(5)  | C18-C19 | 1.405(13) |
| Co1-N2  | 2.226(7)  | C18-C20 | 1.463(12) |
| Co1-N4  | 2.189(6)  | C20-C21 | 1.527(10) |
| Co1-N6  | 2.224(6)  | C21-C22 | 1.366(12) |
| Co1-N8  | 2.285(7)  | C21-C26 | 1.367(11) |
| O1-N9   | 1.233(10) | C22-C23 | 1.409(12) |
| O2-N9   | 1.280(9)  | C23-C24 | 1.350(13) |
| O3-N9   | 1.229(9)  | C24-C25 | 1.405(13) |
| O4-N10  | 1.238(11) | C25-C26 | 1.406(11) |
| O5-N10  | 1.258(11) | C27-C28 | 1.363(12) |
| O6-N10  | 1.237(11) | C27-C32 | 1.402(12) |
| N1-C3   | 1.406(13) | C28-C29 | 1.360(12) |
| N2-C11  | 1.442(9)  | C29-C30 | 1.439(11) |
| N3-C24  | 1.388(12) | C30-C31 | 1.330(12) |
| N4-C15  | 1.428(11) | C31-C32 | 1.408(11) |
| N5-C29  | 1.434(10) | C32-C33 | 1.493(12) |
| N6-C37  | 1.450(11) | C33-C34 | 1.563(12) |
| N7-C50  | 1.424(14) | C34-C35 | 1.366(13) |
| N8-C41  | 1.447(9)  | C34-C39 | 1.352(12) |
| C1-C2   | 1.357(12) | C35-C36 | 1.365(14) |
| C1-C6   | 1.423(11) | C36-C37 | 1.391(12) |
| C2-C3   | 1.354(12) | C37-C38 | 1.341(12) |
| C3-C4   | 1.392(14) | C38-C39 | 1.416(13) |
| C4-C5   | 1.322(14) | C40-C45 | 1.418(12) |
| C5-C6   | 1.425(12) | C40-C41 | 1.383(12) |
| C6-C7   | 1.540(12) | C41-C42 | 1.322(12) |
| C7-C8   | 1.542(11) | C42-C43 | 1.403(11) |
| C8-C9   | 1.386(13) | C43-C44 | 1.395(13) |
| C8-C13  | 1.405(13) | C44-C46 | 1.512(11) |
| C9-C10  | 1.360(13) | C44-C45 | 1.359(13) |
| C10-C11 | 1.390(12) | C46-C47 | 1.473(14) |
| C11-C12 | 1.435(11) | C47-C48 | 1.366(12) |
| C12-C13 | 1.352(12) | C47-C52 | 1.339(13) |
| C14-C15 | 1.382(12) | C48-C49 | 1.422(14) |
| C14-C19 | 1.396(13) | C49-C50 | 1.360(15) |
| C14-C19 | 1.396(13) | C50-C51 | 1.397(12) |
| C15-C16 | 1.390(11) | C51-C52 | 1.444(14) |
| C16-C17 | 1.378(12) |         |           |

TABLE-3  
SELECTED BOND ANGLES (°)

| Angle       | (°)      | Angle       | (°)      |
|-------------|----------|-------------|----------|
| O1W-Co1-O2W | 177.4(2) | C14-C19-C18 | 123.9(8) |
| O1W-Co1-N2  | 89.8(3)  | C18-C20-C21 | 115.0(7) |
| O1W-Co1-N6  | 92.6(3)  | C20-C21-C22 | 121.4(7) |
| O1W-Co1-N8  | 89.5(3)  | C22-C21-C26 | 116.7(7) |
| O2W-Co1-N2  | 91.8(2)  | C20-C21-C26 | 121.9(7) |

| Angle       | (°)      | Angle       | (°)      |
|-------------|----------|-------------|----------|
| O2W-Co1-N4  | 94.2(2)  | C21-C22-C23 | 123.2(8) |
| O2W-Co1-N6  | 85.3(2)  | C22-C23-C24 | 120.1(9) |
| O2W-Co1-N8  | 88.9(2)  | N3-C24-C25  | 117.5(8) |
| N2-Co1-N4   | 85.7(2)  | C23-C24-C25 | 117.8(8) |
| N2-Co1-N6   | 94.6(2)  | N3-C24-C23  | 124.6(9) |
| N2-Co1-N8   | 178.8(2) | C24-C25-C26 | 120.8(8) |
| N4-Co1-N6   | 179.4(2) | C21-C26-C25 | 121.3(8) |
| N4-Co1-N8   | 93.3(2)  | C28-C27-C32 | 123.4(8) |
| N6-Co1-N8   | 86.4(2)  | C27-C28-C29 | 120.1(8) |
| Co1-N2-C11  | 120.7(5) | C28-C29-C30 | 117.7(7) |
| Co1-N4-C15  | 121.3(5) | N5-C29-C30  | 117.1(7) |
| Co1-N6-C37  | 119.0(5) | N5-C29-C28  | 125.2(7) |
| Co1-N8-C41  | 121.2(5) | C29-C30-C31 | 121.7(7) |
| Co1-N8-C41  | 121.2(5) | C30-C31-C32 | 121.3(7) |
| O1-N9-O2    | 119.6(7) | C31-C32-C33 | 124.2(7) |
| O1-N9-O3    | 122.1(7) | C27-C32-C33 | 119.9(7) |
| O5-N10-O6   | 119.7(8) | C27-C32-C31 | 115.9(8) |
| O4-N10-O5   | 119.2(8) | C32-C33-C34 | 113.6(7) |
| O4-N10-O6   | 121.0(8) | C33-C34-C35 | 119.9(8) |
| C2-C1-C6    | 120.2(7) | C35-C34-C39 | 120.1(8) |
| C1-C2-C3    | 124.0(9) | C33-C34-C39 | 119.9(8) |
| N1-C3-C4    | 121.8(8) | C34-C35-C36 | 119.2(8) |
| N1-C3-C2    | 121.4(9) | C35-C36-C37 | 122.0(8) |
| C2-C3-C4    | 116.7(9) | C36-C37-C38 | 118.3(8) |
| C3-C4-C5    | 121.3(8) | N6-C37-C36  | 121.6(8) |
| C4-C5-C6    | 123.6(9) | N6-C37-C38  | 120.1(7) |
| C1-C6-C5    | 114.0(8) | C37-C38-C39 | 120.0(8) |
| C1-C6-C7    | 126.2(7) | C34-C39-C38 | 120.2(8) |
| C5-C6-C7    | 119.7(8) | C41-C40-C45 | 117.0(8) |
| C6-C7-C8    | 114.4(7) | N8-C41-C40  | 117.3(7) |
| C7-C8-C13   | 118.8(8) | C40-C41-C42 | 121.7(7) |
| C7-C8-C9    | 122.6(9) | N8-C41-C42  | 121.0(7) |
| C9-C8-C13   | 118.6(8) | C41-C42-C43 | 120.4(8) |
| C8-C9-C10   | 120.7(8) | C42-C43-C44 | 121.3(8) |
| C9-C10-C11  | 122.5(8) | C43-C44-C46 | 122.7(8) |
| N2-C11-C10  | 123.1(7) | C45-C44-C46 | 121.1(8) |
| C10-C11-C12 | 116.2(7) | C43-C44-C45 | 116.1(8) |
| N2-C11-C12  | 120.7(7) | C40-C45-C44 | 123.5(8) |
| C11-C12-C13 | 121.4(8) | C44-C46-C47 | 118.0(8) |
| C8-C13-C12  | 120.6(8) | C46-C47-C52 | 121.1(8) |
| C15-C14-C19 | 118.1(8) | C46-C47-C48 | 120.3(8) |
| C14-C15-C16 | 120.0(8) | C48-C47-C52 | 118.6(9) |
| N4-C15-C16  | 121.3(7) | C47-C48-C49 | 119.9(9) |
| N4-C15-C14  | 118.6(8) | C48-C49-C50 | 121.6(8) |
| C15-C16-C17 | 120.6(7) | N7-C50-C49  | 120.2(8) |
| C16-C17-C18 | 122.0(8) | C49-C50-C51 | 119.6(9) |
| C19-C18-C20 | 121.2(8) | N7-C50-C51  | 120.1(9) |
| C17-C18-C20 | 123.5(8) | C50-C51-C52 | 116.5(8) |
| C17-C18-C19 | 115.2(8) | C47-C52-C51 | 123.7(8) |

In the molecule of the title compound (Fig. 1), the bond lengths and angles are within normal ranges<sup>13</sup>. The Co atom is hexa-coordinated by four N atoms from the 4,4'-diaminodiphenylmethane ligand and two O atoms from two water molecules. This  $\text{CoO}_2\text{N}_4$  coordination forms a distorted octahedral geometry (Fig. 1). The four N atoms and Co atom are coplanar, while the line through the two O atoms is almost perpendicular to the plane, with  $\text{O}1\text{W}-\text{Co}1-\text{O}2\text{W}$  bond angle of  $177.4(3)^\circ$ . Two nitrate anions is outside the coordination sphere, balancing the charge. In the crystal structure, molecules are linked into chains along the  $a$  axis by N–H…O intermolecular hydrogen bonds along the  $a$  axis (Table-4).

TABLE 4  
HYDROGEN BOND DISTANCES (nm) AND  
ANGLE ( $^\circ$ ) OF THE TITLE COMPOUND

| D  | H   | A   | Symm          | D-H  | H…A   | D…A       | D-H…A  |
|----|-----|-----|---------------|------|-------|-----------|--------|
| N2 | O4  | H2C | x, y, z       | 0.90 | 2.207 | 3.101(10) | 171.39 |
| N3 | H3A | O1  | -1+x, y, -1+z | 0.86 | 2.290 | 3.142(12) | 171.74 |
| N3 | H3B | O5  | x, 1+y, -1-z  | 0.86 | 2.201 | 3.026(12) | 156.81 |
| N4 | H4C | O4  | x, y, z       | 0.90 | 2.434 | 3.250(11) | 154.23 |
| C4 | H4B | O5  | x, y, 1+z     | 0.93 | 2.579 | 3.278(10) | 132.32 |

## ACKNOWLEDGEMENT

The authors thank the Natural Science Foundation of Shandong Province (grant Nos. Z2006B01 and Y2006B07).

## REFERENCES

1. P.J. Dardier, R.E. Holmlin and J.K. Barton, *Science*, **275**, 1465 (1997).
2. D.B. Hall, R.E. Holmlin and J.K. Barton, *Nature*, **382**, 731 (1996).
3. M.V. Keek and S. Lippard, *J. Am. Chem. Soc.*, **114**, 3386 (1992).
4. A.E. Friedman, J.C. Chamborn, J.P. Sauvage, N.J. Turro and J.K. Barton, *J. Am. Chem. Soc.*, **114**, 5919 (1992).
5. R.M. Hartshorn and J.K. Barton, *J. Am. Chem. Soc.*, **112**, 4960 (1990).
6. A.S. Sitlani, E.C. Long, A.M. Pyle and J.K. Barton, *J. Am. Chem. Soc.*, **114**, 2230 (1992).
7. D.S. Sigman, A. Mazumder and D.M. Perrin, *Chem. Rev.*, **93**, 2295 (1993).
8. G. Pratviel, J. Bernadou and B. Mcunicr, *Adv. Inorg. Chem.*, **45**, 251 (1998).
9. L.N. Ji, X.H. Zou and J.G. Liu, *Coord. Chem. Rev.*, **216**, 513 (2001).
10. J.A. Cowan, *Curr. Opin. Chem. Biol.*, **5**, 634 (2001).
11. X.M. Li, H.Q. Ju, L.P. Du and S.S. Zhang, *J. Inorg. Biochem.*, **101**, 1165 (2007).
12. G.M. Sheldrick, SHELXS-97, Program for X-ray Crystal Structure Solution, Göttingen University, Germany (1997).
13. F.H. Allen, O. Kennard, D.G. Watson, L. Brammer, A.G. Orpen and R. Taylor, *J. Chem. Soc. Perkin Trans. II*, S1 (1987).

(Received: 14 March 2008;

Accepted: 12 November 2008)

AJC-7022