Asian Journal of Chemistry

Vol. 21, No. 6 (2009), 4733-4736

# Synthesis and Biological Evaluation of 5-Pyridine-4-(arylidine amino)-3-mercapto-4(*H*)-1,2,4-triazoles

M.E. BHANOJIRAO\* and VIKAS G. RAJURKAR

Department of Pharmaceutical Chemistry, Roland Institute of Pharmaceutical Sciences Khodasingi, Berharmpur-760 010, Inda E-mail: drmebrao@yahoo.co.in

As triazoles have proven to be good antimicrobial agents, a new series of triazole derivatives were synthesized and characterized by <sup>1</sup>H NMR, IR, GCMS sophisticated analytical instruments and were evaluated for their antimicrobial activity. Out of several derivatives synthesized a few of 5-pyridine-4-(arylidine amino)-3-mercapto-4(*H*)-1,2,4-triazoles showed good antimicrobial activity by using cup-plate method.

# Key Words: Arylidine derivatives, 1,2,4-Triazole, Antimicrobial activity.

# **INTRODUCTION**

Microbial infections are most common infections and commonly used antibacterials, antifungals are antibiotics and recently used halogenated quinolones, azoles and other synthetic agents are being used to combat the newer and resistant microbial infections. Azoles in particular 1,2,4-triazoles have shown significant antimicrobial activity<sup>1-6</sup>. Compounds with the azomethine linkage possess biological activity such as antifungal<sup>7</sup>, antibacterial<sup>8</sup> and antiinflammatory<sup>9</sup> activity. Herein, an attempt is made to synthesize Schiff bases of 5-pyridine-4-(arylidine amino)-3-mercapto-4(H)-1,2,4-triazoles as potent antimicrobial agents.

# **EXPERIMENTAL**

All chemicals were used after purification and supplied by Loba Chemie, Qualigens and Research Labs. Melting points of newly synthesized compounds were determined in open capillary tube and were found uncorrected. The structure of the synthesized compounds was confirmed by spectral data. The IR spectra were recorded on FTIR 8400 F-Shimadzu spectrometer using KBr disc pellet method. <sup>1</sup>H NMR spectra were recorded on Verian mercury <sup>1</sup>H 300 using DMSO as solvent. GCMass spectra were recorded on GCMS-QP-5050 Schimadzu.

**Synthesis of isonicotinyl hydrazide (1):** In round bottom flask, 4-pyridine carboxylic acid (0.01 mol) and hydrazine hydrate 99 % (0.01 mol) was taken along with alcohol and the mixture was refluxed for 4 h. Then from the reaction mixture alcohol was removed under reduced pressure. Solid residue was obtained, recrystallized with ethanol.



Synthesis of potassium-pyridine-dithiocarbazate (2): In a 250 mL round bottom flask, aryl hydrazide 10.28 g (0.075 mol) was taken. To this a solution of potassium hydroxide 4.2 g (0.075 mol) in 100 mL of absolute alcohol and carbon disulphide were added agitated for overnight. The reaction mixture was diluted with 200 mL of dry ether. The solid obtained was 15.05 g (80 %). It was filtered and washed with dry ether.

**Synthesis of 5-pyridine-4-amino-3-mercapto-4**(*H*)**-1,2,4-triazole (3):** A mixture of potassium-pyridine-dithiocarbazate 25.1 g (0.1 mol) and hydrazine hydrate 5 mL (0.1 mol) was refluxed for 2 h with occasional shaking and the solution was poured into the cold water. The mixture was acidified with hydrochloric acid. The precipitate obtained was filtered, dried and recrystallized by using alcohol. IR (KBr,  $v_{max}$ , cm<sup>-1</sup>): 3309 (w, NH<sub>2</sub>, *str.*), 2677 (w, SH, *str.*), 2358 (s, CH<sub>2</sub>, *str.*). <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>): 1.5 (s, 1 H, SH), 5.38 (s, 2H, CH<sub>2</sub>).

Synthesis of 5-pyridine-4-(arylidine amino)-4-mercapto-4(*H*)-1,2,4-triazoles (4): A mixture of 5-pyridine-4-amino-3-mercapto-4(*H*)-1,2,4-triazole (0.01 mol) taken with various aromatic aldehydes (0.01 mol) and concentrated sulphuric acid (0.5 mL) in ethanol 100 mL. The mixture was refluxed on water bath for several hours with TLC monitoring. The solid was obtained on cooling the mixture and poured in cold water was afforded the Schiff's bases (**4a-4m**) (Table-1). IR (KBr,  $v_{max}$ , cm<sup>-1</sup>): 2677 (w, SH, *str.*), 2358 (s, CH<sub>2</sub>, *str.*), 1596 (w, C=N, *str.*). <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>): 14 (s, 1 H, SH), 5.38 (s, 2H, CH<sub>2</sub>), 9.43 (w, 1H, OH), 3.0 (s, 6H, 2CH<sub>3</sub>), 10.0 (w, 1H, N=CH). Mass: (i) 271 (M<sup>+</sup>-1), 44 (Base peak), 57, 71, 78, 91, 105, 119, 178.

Vol. 21, No. 6 (2009)

#### TABLE-1 PHYSICAL DATA OF SOME 5-PYRIDINE-4-(ARYLIDINE AMINO)-4-MERCAPTO-4(*H*)-1,2,4-TRIAZOLES

Recrystallization solvent: Ethanol

Compd.	Ar	m.f.	m.w.	m.p. (°C)	Yield (%)
4a	Furan	$C_{12}H_9N_5OS$	271	290	45
<b>4</b> b	o-Cl-Phenyl	$C_{14}H_{10}N_5SCl$	315	304	48
<b>4</b> c	<i>p</i> -OH-Phenyl	$C_{14}H_{11}N_5OS$	297	272	60
<b>4d</b>	<i>p</i> -OCH <sub>3</sub> -Phenyl	$C_{15}H_{13}N_5OS$	311	290	45
<b>4e</b>	<i>p</i> -Cl-Phenyl	$C_{14}H_{10}N_5SCl$	315	225	50
<b>4f</b>	<i>p</i> -N-Dimethyl-phenyl	$C_{16}H_{16}N_{6}S$	324	211	50
<b>4</b> g	Phenyl	$C_{14}H_{11}N_5S$	281	242	45
<b>4h</b>	p-OH-M-Methoxy-phenyl	$C_{15}H_{13}N_5O_2S$	327	270	50
<b>4</b> i	<i>p</i> -Nitro-phenyl	$C_{14}H_{10}N_6O_2S$	326	275	50
<b>4</b> j	o-Nitro-phenyl	$C_{14}H_{10}N_6O_2S$	326	230	50
<b>4</b> k	<i>m</i> -Hydroxy-phenyl	$C_{14}H_{11}N_5OS$	297	246	55
41	<i>m</i> -Nitro-phenyl	$C_{14}H_{10}N_6O_2S$	326	290	45
<b>4</b> m	o-OH-phenyl	$C_{14}H_{11}N_5OS$	297	208	60
3	Triazole	$C_8H_6N_5S$	204	260	60

#### TABLE-2

ANTIMICROBIAL ACTIVITY OF SOME 5-PYRIDINE-4-(ARYLIDINE AMINO)-4-MERCAPTO-4(*H*)-1,2,4-TRIAZOLES

Compd.	Zone of inhibition (mm)											
	B. su	btillis	S. aı	ireus	K. pneum	ioniae	P. aeri	ıginosa	C. al	bicans	<i>A. n</i>	iger
Conc. <sup>n</sup>	А	В	А	В	А	В	А	В	А	В	А	В
4a	2	5	3	6	3	5	3	5	4	8	5	9
<b>4</b> b	2	4	6	8	4	9	5	7	6	10	5	9
<b>4</b> c	4	9	5	7	6	8	5	8	4	6	3	7
<b>4d</b>	5	7	3	5	5	8	4	9	4	5	3	8
<b>4e</b>	2	5	4	8	6	9	4	6	5	10	4	9
<b>4f</b>	5	8	6	9	3	5	4	6	4	9	5	9
<b>4</b> g	3	7	3	5	5	9	6	7	8	11	10	12
<b>4h</b>	4	8	5	9	6	10	5	9	8	13	9	13
<b>4i</b>	5	8	7	9	6	10	7	10	3	7	5	8
<b>4</b> j	6	9	8	12	4	7	3	9	4	6	6	10
<b>4</b> k	3	8	4	6	7	13	4	9	5	9	6	8
41	3	5	2	7	5	7	5	9	3	8	4	10
<b>4</b> m	4	9	3	8	6	10	3	8	5	10	6	11
3	6	8	4	7	6	8	4	6	6	10	7	11
Norfloxacin 50 µg/0.1 mL			14 mm									
Griseofulvin 50 µg/0.1 m				13 mm								

A = 200  $\mu$ g/0.1 mL; B = 400  $\mu$ g/0.1 mL. All tests were performed in triplicate.

**Antimicrobial activity:** The antimicrobial activity was carried out by using cup-plate method<sup>10,11</sup> by using microbial strains as *Bacillus subtillis, Staphylococcus aureus, Pseudomonas aerugenosa* and *Kleibsella pneumonia* with incubation

4736 Bhanojirao et al.

Asian J. Chem.

period of 24 h at temperature 37 °C. The standard drug used was norfloxacin (50  $\mu$ g/0.1 mL) and the test compounds at concentrations of 200 and 400  $\mu$ g/0.1 mL.

Antifungal activity was carried out using cup-plate method<sup>10,11</sup>. By using fungal strains as *Aspergillus niger, Candida albicans*, for 48 h of incubation at 28 °C. The concentration of the standard drug griseofulvin was used 50  $\mu$ g/0.1 mL and the test compounds at concentrations of 200 and 400  $\mu$ g/0.1 mL (Table-2).

# **RESULTS AND DISCUSSION**

All the compounds tested showed some degree of antimicrobial activity **4h**, **4i**, **4j**, **4k** and **4m** showed good antibacterial activity and **4g**, **4h**, **4i**, **4j**, **4k**, **4l**, **4m** and **3** showed good antifungal activity due to presence of triazole nucleus in it. Due to presence of imino group and shifted SH group in compounds <sup>1</sup>H NMR spectra indicate completion of target reactions in this series.

### ACKNOWLEDGEMENTS

The authors dedicated this paper and thanks Prof. M.Y. Yedukondalu, Principal, College of Pharmacy, Mohuda, for his kind help, support and encouragement.

#### REFERENCES

- 1. A. Korolkovas, Essentials of Medicinal Chemistry, John Wiley and Son's, New York, edn. 2, p. 3 (1988).
- 2. D.A. Williams and T.L. Lekme, Foye's Principles of Medicinal Chemistry, Lippincott Williams and Wilkins, Philadelphia, edn. 5, p. 827 (2002).
- 3. G.K. Rao, S. Sen and B. Rajendra, *Indian Drugs*, **41**, 524 (2004).
- 4. B.S. Jayashree, A.R. Sahu, M. Srinivasa and K.N. Venugopala, Asian J. Chem., 19, 73 (2007).
- R.H. Udupi, V.M. Kulkarni, S.R. Shetty and P. Acharya, *Indian J. Hetrocycl. Chem.*, 11, 303 (2002).
- 6. G.K. Rao, S. Rajasekaran and M. Attimarad, Indian J. Pharm. Sci., 62, 475 (2000).
- R.M. Silverstein, G.C. Bassler and T.C. Morril, Spectrophotometric Identification of Some Compounds, Jhon Wiley and Son's, New York, edn. 5, p. 328 (1979).
- 8. A. Cremer, Microbiological Methods, Butter Worth and Co., London, edn. 6, p. 235 (1991).
- 9. L.M. Prescott, J.P. Harley and D.A. Klein, Microbiology, W.C. Brown Publishers, Oxford, England, edn. 2, p. 328 (1990).
- 10. K. Kavanagh, Analytical Microbiology, Academic Press, New York, p. 125 (1963).
- 11. C. Kokare, Pharmaceutical Microbiology, Experiments and Techniques, Career Publications, edn. 2, pp. 139-142.

(Received: 11 September 2008; Accepted: 31 March 2009) AJC-7387