

Synthesis and Spectral Studies of Substituted Phenyl and Naphthyl imino-1,2,3,4-tetrahydrocarbazoles

R. VELMURUGAN and M. SEKAR*

Department of Chemistry, Sri Ramakrishna Mission Vidyalaya College of Arts and Science,
Coimbatore- 641 020, India
E-mail: mmsekar7@yahoo.co.in

The reaction of 1-oxo-1,2,3,4-tetrahydrocarbazole (**1**) and aniline (**2a-c**) in ethanol and refluxed for 1 h at 120 °C afforded phenyl imino-1,2,3,4-tetrahydrocarbazole derivatives (**3a-c**). The same reaction of 1-oxo-1,2,3,4-tetrahydrocarbazole and naphthylamine (**4**) in ethanol and refluxed for 1 h at 120 °C yielded naphthyl imino-1,2,3,4-tetrahydrocarbazole (**5**).

Key Words: 1-Oxo-1,2,3,4-tetrahydrocarbazole, Phenyl imino carbazole, Naphthyl imino carbazole.

INTRODUCTION

Carbazole alkaloids represent a new and interesting variant in the large number of existing indole alkaloids which yielded several important drugs¹⁻⁴. A large number of carbazole alkaloids have been isolated from plants⁵. The Indian Curry leaf plant *Murraya Koemigii Spreng (Rutaceae)* has been found to be the rich and rewarding source of many carbazole alkaloids⁶. The alkaloids ellipticine and olivacine have showed marked anticancer activity⁷, Olivacine is also known for its antiulcer, antiheumatic properties⁸. Carbamycin A and carbamycin B have been found to be useful antibacterial and antifungal agents. It has been reported that pyridocarbazoles, oxotetrahydrocarbazoles⁹, Mukonine isomers¹⁰, Girinimbine isomers¹¹, pyrazino(3,2,1-J,K)carbazoles, acetyl amino carbazoles¹² show marked anticancer and anti-HIV activities¹³.

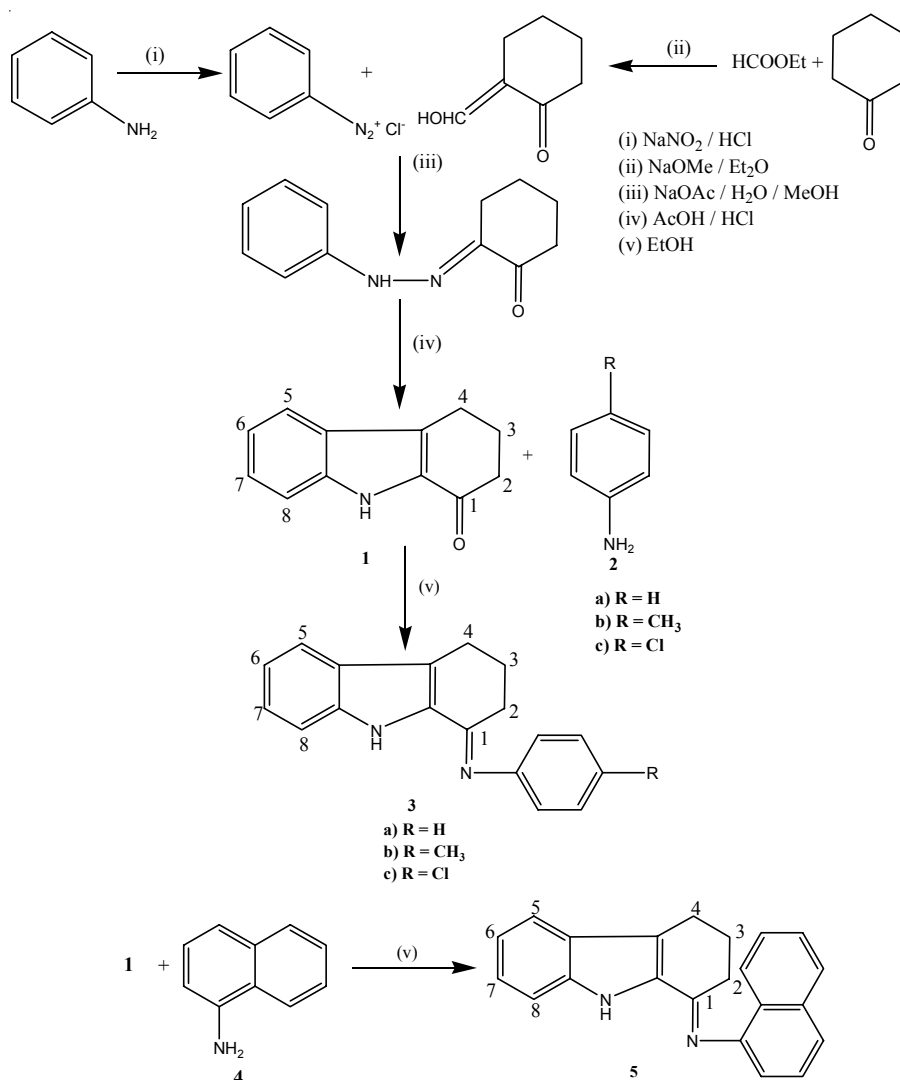
EXPERIMENTAL

The products were purified using column chromatographic method, packed with silica gel and petroleum ether, ethyl acetate mixture of solvents used for elution.

The melting points were determined using a metler FP-5 apparatus and are uncorrected. FT-IR spectra were recorded on a Perkin-Elmer-597, FT IR-8201 PC spectro-meter using potassium bromide. ¹H NMR and ¹³C NMR spectra were recorded in CDCl₃ on a Varian FX-90 FT-NMR and WH-270 NMR spectrometer using tetramethyl silane as an internal standard.

Preparation of phenyl imino-1,2,3,4-tetrahydrocarbazoles (3a-c): A mixture of the appropriate 1-oxo-1,2,3,4-tetrahydrocarbazole (**1**, 0.1 mol) and aniline (**2a-c**, 0.1 mol) both dissolved in ethanol and refluxed for 1 h. Then the contents were cooled and poured into crushed ice, the solid obtained was filtered off and recrystallized from petroleum ether and ethyl acetate (2:1) mixture (**Scheme-I**).

Preparation of naphthyl imino-1,2,3,4-tetrahydrocarbazoles (5): A mixture of the appropriate 1-oxo-1,2,3,4-tetrahydrocarbazole (**1**, 0.1 mol) and naphthylamine (**4**, 0.1 mol) both dissolved in ethanol and refluxed for 1 h. Then the contents were cooled and poured into crushed ice, the solid obtained was filtered off and recrystallized from petroleum ether and ethyl acetate (2:1) mixture (**Scheme-I**).



RESULTS AND DISCUSSION

1-Oxo-1,2,3,4-tetrahydrocarbazole and aniline (**2a-c**) both dissolved in ethanol and refluxed for 1 h to afford products (**3a-c**). Table-1 contains of the analytical data for compounds (**1-5**). FT-IR and ^1H NMR spectral values are given in Table-2.

TABLE-1

Compd.	Yield (%)	m.p. ($^{\circ}\text{C}$)	m.f. (m.w.)
1	85.00	158	$\text{C}_{12}\text{H}_{11}\text{NO}$ (185)
3a	76.92	148	$\text{C}_{18}\text{H}_{16}\text{N}_2$ (260)
3b	98.50	165	$\text{C}_{19}\text{H}_{18}\text{N}_2$ (274)
3c	84.80	Low melting point	$\text{C}_{18}\text{H}_{15}\text{N}_2\text{Cl}$ (294)
5	87.10	Low melting point	$\text{C}_{22}\text{H}_{18}\text{N}_2$ (310)

TABLE-2
IR, ^1H NMR AND ^{13}C NMR SPECTRAL DATA OF **1**, **3a-c**, **5**

Compd.	IR (ν_{max} , cm^{-1})	^1H NMR signals (δ ppm)
1	3300, 1640, 1540, 1480, 1340, 1260, 1200, 1140, 820 and 740	2.28 (p, 2H, C3-H, $J = 6.25$ Hz) 2.67 (t, 2H, C4-H, $J = 6.45$ Hz) 3.02 (t, 2H, C2-H, $J = 6.58$ Hz) 7.11-7.67 (m, 4H, C5, C6, C7, C8-H) 9.0 (singlet, N-H)
3a	3265, 2937, 1647, 1571, 1542, 1475, 1257, 1171, 746 and 731	3.018 (t, 2H, C2-H), 2.274 (p, 2H, C3-H) 2.676 (t, 2H, C4-H), 7.312-7.668 (m, 9H, C5, C6, C7, C8 and 2',3',4',5',6'-H).
3b	3285, 2928, 1642, 1571, 1541, 1616, 1474, 1329, 1188, 1169 and 748	3.15 (t, 2H, C2-H), 2.60 (s, 3H, CH_3), 2.30 (p, 2H, C3-H), 2.70 (t, 2H, C4-H), 7.37-7.78 (m, 8H, C5, C7, C8 and 2',3',5',6'-H)
3c	3380, 3289, 2924, 2854, 1288, 1616, 1494, 1181, 1088, 820, 748, 731 and 697	3.20 (t, 2H, C2-H), 2.35 (p, 2H, C3-H), 2.75 (t, 2H, C4-H), 7.40-7.85 (m, 8H, C5, C7, C8 and 2',3',5',6'-H)
5	3341, 3230, 3024, 2937, 1643, 1573, 1457, 1643, 1573, 1457, 1288, 1171, 1087, 790 and 770	2.235 (p, 2H, C3-H), 2.959 (t, 2H, C2-H) 2.623 (t, 2H, C4-H) 6.732-7.775 (m, 11H, C5-H, C6-H, C7-H, C8-H and C2'-H, C3'-H, C4'-H, C5'-H, C6'-H, C7'-H, C8'-H).

The FT-IR spectrum of the compound (**3a**) showed two strong absorption bands at 3265 and 2937 cm^{-1} for NH and aromatic C-H stretching's respectively and C=N stretching appeared at 1647 cm^{-1} . Its ^1H NMR spectrum showed a multiplet at δ 7.312 to 7.668 ppm for 9 aromatic protons of C₅, C₆, C₇ and C₈ of carbazole phenyl moiety, C_{2'}, C_{3'}, C_{4'}, C_{5'} and C_{6'} protons of imino phenyl moiety. Two protons of C₂-carbon atom appeared as triplet at δ 3.018 ppm. Another two protons at C₃ carbon atom appears as pentet centered at 2.274 ppm the splitting of the signal (pentet) is due to two adjacent C₂ and C₄ methylene protons. Triplet appearing at δ 2.676 ppm has been assigned to the C₄ methylene protons.

The presence of thirteen distinct peaks in the ^{13}C NMR spectrum of the synthesized compound **3a** confirms the molecular structure. The ipso carbon atoms present in the product appear as weak signals at δ 191.450 and 137.913 ppm. These data were in support of the structure of phenyl imino 1,2,3,4-tetrahydrocarbazole (**3a**).

The FT-IR spectrum of the compound (**3b**) showed a strong absorption band at 1642 cm⁻¹ corresponding to C=N stretching and the two strong absorption bands at 2928 and 3285 cm⁻¹ corresponds to C-H and N-H stretching frequencies, respectively. On the basis of spectral evidence (Table-2) the compound named as 1-(4'-methyl phenyl)imino-1,2,3,4-tetrahydrocarbazole (**3b**).

The FT-IR spectrum of the compound (**3c**) showed an absorption band at 3380 cm⁻¹ for N-H stretching and the absorption of aromatic C-H stretching bands appears at 3289 cm⁻¹. The two strong absorption bands at 2924 and 1616 cm⁻¹ are due to asymmetric C-H stretching of CH₂ group and C=N stretching. The C-Cl stretching is confirmed by the band at 697 cm⁻¹. On the basis of spectral evidence (Table-2) the compound was named as 1-(4'-chloro phenyl)imino-1,2,3,4-tetrahydrocarbazole (**3c**).

The FT-IR spectrum of the compound (**5**) showed absorption band at 3341 cm⁻¹ for NH stretching and two strong absorption bands at 3230 and 1643 cm⁻¹ for aromatic C-H stretching and C=N stretching. Its ¹H NMR spectrum showed a pentet at δ 2.25 ppm has been assigned to methylene protons at C₃ carbon atom of carbazole moiety. The triplet at δ 2.959 ppm assigned to two protons at C₂ carbon atom of the same moiety. Another triplet has its center at δ 2.623 ppm is due to two methylene protons of C₄ carbon and has been spilt in to triplet by two adjacent protons at C₃ carbon atom. The multiplet peaks appeared in the aromatic region from δ 6.732 to 7.775 ppm corresponds to eleven protons of the phenyl moiety in the carbazole and 1-naphthyl moiety in the compound. On the basis of spectral evidence of the compound was named as 1-(1'-naphthyl)imino-1,2,3,4-tetrahydrocarbazole (**5**).

ACKNOWLEDGEMENTS

One of the authors (MS) thanks to UGC, New Delhi for the award of Major Research Project F. No: 32-281/2006 (SR). The authors also express their thanks to The Principal, Secretary and the Management of SRMV College of Arts and Science, for providing laboratory facilities to carry out this research work.

REFERENCES

1. W.L. Alberecht, R.W. Fleming, W.S. Horgan and G.D. Mayer, *J. Med. Chem.*, **20**, 364 (1974).
2. (a) D. Sowmithran and K.J.R. Prasad, *Heterocycles*, **24**, 2195 (1986); (b) P. Bhattacharyya, D.P. Chakraborty and B.K. Chowdhury, *Indian J. Chem.*, **23B**, 849 (1984).
3. D. Sowmithran and K.J.R. Prasad, *Indian J. Chem.*, **25B**, 1179 (1986).
4. D. Sowmithran and K.J.R. Prasad, *Indian J. Chem.*, **26B**, 277 (1987).
5. D.P. Charkarabarty, *Fortschr. Chem. Org. Naturst.*, **34**, 299 (1977).
6. R.S. Kapil, in ed.: R.H.F. Manske, *The Alkaloids*, New York, London, Vol. 13, pp. 273-302 (1971).
7. M. Sainbury, *Synthesis*, 438 (1977).
8. T. Kamatani, Japan Patent, 7,712,196 (1975); *Chem. Abstr.*, **87**, 68591c (1977).
9. N.L. Andreeva, *Farmak Toks.*, **36**, 713 (1967); *Chem. Abstr.*, **68**, 57793r (1968).
10. D. Sowmithran and K.J.R. Prasad, *Heterocycles*, **24**, 711 (1986).
11. M. Sekar and K.J.R. Prasad, *Indian J. Chem.*, **33B**, 479 (1994).
12. M. Sekar, S. Vanitha and K.J.R. Prasad *Z. Naturforsch.*, **49**, 687 (1994).
13. A.E. Martin and K.J.R. Prasad, *Acta Pharm.*, **56**, 79 (2006).