

**NOTE****Synthesis and Crystal Structure of  
4,4'-Di-2-pyridyl-2,2'-dithiodipyrimidine (C<sub>18</sub>H<sub>12</sub>N<sub>6</sub>S<sub>2</sub>)**

HUA-ZE DONG\* and HAI-BIN ZHU†

*Department of Chemistry and Chemical Engineering,  
Hefei Normal University, Hefei-230061, P.R. China  
E-mail: dapdong@163.com*

A new heterocyclic sulfide ligand with molecular formula C<sub>18</sub>H<sub>12</sub>N<sub>6</sub>S<sub>2</sub>, named as 4,4'-di-2-pyridyl-2,2'-dithiodipyrimidine, was formed by the reaction of 4-pyridin-2-yl-pyrimidine-2-thiol with ferric nitrate. The crystal is monoclinic, space group P2<sub>1</sub>/c with unit cell parameters: a = 11.777 (2) Å, b = 7.2830 (15) Å, c = 20.182 (4) Å, α = 90°, β = 98.832 (4)°, γ = 90°, V = 1710.5 (6) Å<sup>3</sup>, Z = 4, Mr = 376.47, Dc = 1.470 g/cm<sup>3</sup>, μ = 0.327 mm<sup>-1</sup>, F(000) = 776, R = 0.0378, wR = 0.0749 for 1820 reflections with I > 2σ(I).

**Key Words:** 4-Pyridin-2-yl-pyrimidine-2-thiol, Hydrogen bonds, Heterocyclic sulfide.

There has been increasing interest of heterocyclic sulfide ligands in the field of coordination chemistry<sup>1-6</sup>. In our previous paper, we reported two disulfide ligands derived from 4-pyridin-pyrimidine-2-thiol<sup>5,6</sup>. Herein reported is a new disulfide complex C<sub>18</sub>H<sub>12</sub>N<sub>6</sub>S<sub>2</sub>.

All reagents for synthesis were commercially available and employed as received or purified by standard methods prior to use. 4-Pyridin-2-yl-pyrimidine-2-thiol was prepared by general procedure reported in the literature<sup>7</sup>. Analyses for carbon, hydrogen and nitrogen were performed on a Perkin-Elmer 1400C analyzer.

**Synthesis:** Fe(NO<sub>3</sub>)<sub>3</sub>·9H<sub>2</sub>O metal salt (0.1 mmol) was mixed with 4-pyridin-pyrimidine-2-thiol (0.1 mmol) and NaOH (0.1 mmol) in 20 mL of CH<sub>3</sub>CN. The reaction mixture was stirred for 1 h and then filtered. The purple mother solution was allowed for one week to evaporate slowly and obtained colorless crystals block-shaped single crystals 1 suitable for X-ray analysis. Yield 57.1 %. Anal. calcd. for C<sub>18</sub>H<sub>12</sub>N<sub>6</sub>S<sub>2</sub>: C, 57.43; H, 3.21; N, 22.32 %. Found: C, 57.49; H, 3.30; N, 22.17.

**Crystal structure determination:** A single crystal of compound with dimensions of 0.2 mm × 0.2 mm × 0.2 mm was selected for crystallographic data collection at 291(2)K and structure determination on a Bruker SMART CCD-4K diffractometer employing graphite-monochromated MoKα radiation (λ = 0.71073 Å). A total of 8261 reflections were collected in the range of 2° ≤ q ≤ 25°, of which 3009 reflections

†School of Chemistry and Chemical Engineering, Southeast University, Najing-211189, P.R. China, E-mail: zhuhaibin@seu.edu.cn.

were unique with  $R_{\text{int}} = 0.055$ . The data were collected using SMART and reduced by the program SAINT. All the structures were solved by direct methods and refined by full-matrix least squares method on  $F_o^2$  by using SHELXTL-PC software package. Non-hydrogen atoms were placed in geometrically calculated positions. Hydrogen atoms were added according to theoretical model. The final full-matrix least-squares refinement including 235 variable parameters for 3009 reflections with  $I > 2\sigma(I)$  and converged with unweighted and weighted agreement factors of

$$R_1 = \Sigma(|F_o| - |F_c|) / \Sigma|F_o| = 0.0378 \quad (1)$$

$$\text{and} \quad wR_2 = \{ \Sigma[w(F_o^2 - F_c^2)^2] / \Sigma w(F_o^2)^2 \}^{1/2} = 0.0749 \quad (2)$$

where  $w = 1/[\sigma^2(F_o^2) + (0.0280P)^2]$  and  $P = (F_o^2 + 2F_c^2)/3$ . The maximum and minimum peaks on the final difference Fourier map are corresponding to 0.15 and  $-0.19 \text{ e}/\text{\AA}^3$ , respectively.

The selected bond lengths and bond angles in Table-1. Respecting, Fig. 1 shows the molecular structure of the title compound. Fig. 2 shows the packing diagram of the title compound. In Fig. 1, the C-S-S-C torsion angle of  $82.72 (9)^\circ$  is similar with that in its analogue, namely 2,2'-dithiobis(4-pyridin-3-yl-pyrimidine)<sup>5</sup>. The S-S bond length of  $2.0113 (9) \text{ \AA}$  in Fig. 1 is within the normal range<sup>1,2,5,6</sup>. It is noted that the pyrimidinyl ring and the pyridyl ring of title complex are obviously not coplanar, the dihedral angles between the two heterocyclic rings is *ca.*  $4.0^\circ$  or  $11.1^\circ$ . In crystal packing, it is interesting to observe that the C-H...N intermolecular hydrogen bonds are formed between adjacent molecules resulting in a 3D supramolecular framework.

TABLE-1  
SELECTED BOND DISTANCES (Å) AND ANGLES (°)

S1-S2	2.0113 (9)	N3-C5	1.343 (3)	N4-C13-N5	128.6 (2)
S1-C13	1.7690 (2)	N3-C9	1.329 (3)	N6-C14-C10	115.7 (2)
S2-C4	1.7830 (2)	S2-S1-C13	104.63 (8)	N6-C14-C19	122.9 (2)
N1-C1	1.3430 (3)	S1-S2-C4	105.66 (8)	S1-C13-N5	120.23 (17)
N1-C4	1.3280 (3)	C1-N1-C4	115.37 (19)	S1-C13-N4	111.17 (15)
N2-C3	1.3380 (3)	C3-N2-C4	113.6 (2)	C5-N3-C9	116.5 (2)
N2-C4	1.3140 (3)	—	—	—	—

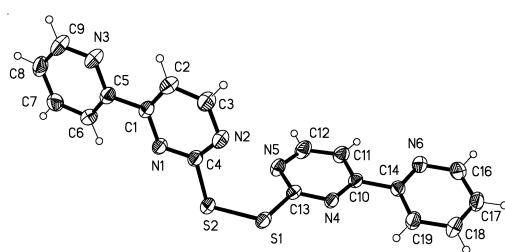


Fig. 1. Molecular structure of the present compound

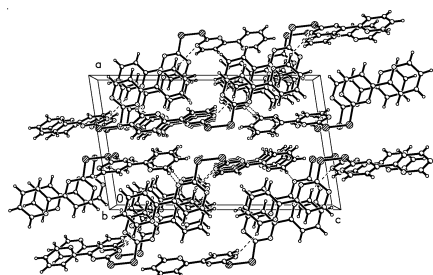


Fig. 2. View of a 3D supramolecular framework of present compound showing the intermolecular hydrogen bonding

TABLE-2  
HYDROGEN BOND DISTANCES (Å) AND ANGLES (°)

Type (D-H...A)	d(D-H)	d(H...A)	∠(DHA)	d(D...A)	A
C9-H9...N5	0.9300	2.6100	149.00	3.437(3)	-x, -1/2 + y, 3/2-z
C16-H16...N1	0.9300	2.5900	174.00	3.521(3)	x, 5/2-y, 1/2 + z

### Conclusion

Crystal structure of a new heterocyclic disulfide complex has been synthesized and characterized by elemental analysis and X-ray diffraction analysis.

### Supplementary material

Crystallographic data for the structure reported in this communication have been deposited with the Cambridge Crystallographic Data Center as supplementary publication No. CCDC 705137.

### ACKNOWLEDGEMENT

This work is financially supported by National Natural Science of Foundation of China (project No. 20801011).

### REFERENCES

1. L.S. Higashi, M. Lundeen and J. Seff, *J. Am. Chem. Soc.*, **100**, 8101 (1978).
2. R. Horikoshi and T. Mochida, *Coord. Chem. Rev.*, **250**, 2595 (2006).
3. F.M. Tabellion, S.R. Seidel, A.M. Arif and P.J. Stang, *J. Am. Chem. Soc.*, **123**, 7740 (2001).
4. G.M. Sheldrick, *Acta Cryst.*, **A64**, 112 (2008).
5. J.-F. Ji, L. Li and H.-B. Zhu, *Acta Cryst.*, **E65**, o1253 (2009).
6. H.B. Zhu, H. Wang and L. Li, *Acta Cryst.*, **E65**, o1588 (2009).
7. H.Z. Dong, X. Liu and S.H. Gou, *Transition Met. Chem.*, **32**, 518 (2007).

(Received: 6 October 2009;

Accepted: 2 March 2010)

AJC-8500