

## NOTE

## Lignans from the Leaf of Smallanthus sonchifolius

JING-JIAO XUE<sup>1,2</sup>, LAN WANG<sup>2</sup>, YONG-KUANG CHZN<sup>2</sup>, ZHEN LIAO<sup>2</sup>, WEI LIU<sup>2</sup>, GUANG-YU YANG<sup>1,2</sup> and QIU-FEN HU<sup>1,2,\*</sup>

<sup>1</sup>School of Chemistry and Biotechnology, Yunnan Nationalities University, Kunming 650031, P.R. China <sup>2</sup>Key Laboratory of Tobacco Chemistry of Yunnan Province, Yunnan Acedemy of Tobacco Science, Kunming 650106, P.R. China

\*Corresponding author: E-mail: huqiufena@yahoo.com.cn

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Phytochemical investigation of the leaf of *Smallanthus sonchifolius* led to the isolation and identification of a new lignan (smallanlignan A) and five known lignans. The structures were elucidated by the analysis of spectroscopic data. The antioxidant activity of smallanlignan A was evaluated and it showed antioxidant activity with an IC<sub>50</sub> value of 1.65  $\mu$ g/mL.

Key Words: Smallanthus sonchifolius, Lignan, Antioxidant activity.

*Smallanthus sonchifolius* is a member of asteraceae family. It is a perennial plant grown in the Andes of Perú for its crisp, sweet-tasting tubers. The texture and flavour have been described as a cross between a fresh apple and watermelon. This is the reason, it is sometimes referred to as the apple of the earth. The tuber is composed mostly of water and fructooligo-saccharides<sup>1-3</sup>. It has recently been introduced into farmer's markets and natural food stores in Yunnan Province, P.R. China. The leaf and root of *Smallanthus sonchifolius* has also received more and more attentions because of their biological activities<sup>3-7</sup>.

In order to investigates the components of the aerial parts and search for potential leads for drug development, phytochemical investigation on the leaf of *Smallanthus sonchifolius* was carried out. This led to the isolation of one new lignans (smallanlignan A), together with five known lignans *i.e.*, epiwulignan A (**2**)<sup>8</sup>, epischisandrone (**3**)<sup>8</sup>, 8'- $\beta$ -hydroxyhinokinin (**4**)<sup>9</sup>, (-)-hinokinin (**5**)<sup>10</sup> and (-)-zuonin A (**6**)<sup>11</sup>. The structure was established by means of extensive NMR spectra and the antioxidant activity of mallanlignan A was evaluated (Fig. 1).

Optical rotation was measured in Horiba SEPA-300 high sensitive polarimeter. IR spectra were obtained in KBr disc on a Bio-Rad Wininfmred spectrophotometer. ESI-MS were measured on a VG Auto Spec-3000 MS spectrometer. <sup>1</sup>H, <sup>13</sup>C NMR and 2D NMR spectra were recorded on Bruker DRX-500 instruments with TMS as internal standard. On second separate used Agilent 1100 HPLC high performance liquid chromatography equipped with Zorbax-C<sub>18</sub> (9.4 nm × 250 nm, 5.0 µm) column and DAD detector. Column chromatography was performed on silica gel (200-300 mesh) or on silica gel H (10-40 µm, Qingdao Marine Chemical Inc., P.R. China).

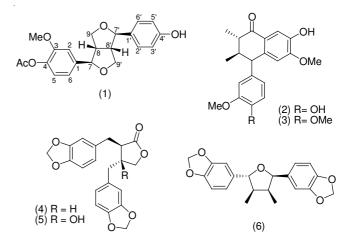


Fig. 1. Structure of lignans in Smallanthus sonchifolius

The leaf of *Smallanthus sonchifolius* were collected in Kunming of Yunnan Province, P.R. China, in September 2009 and was identified by Prof. S.F. WU. A voucher specimen (No. YNNI 09-18-09) was deposited in our laboratory.

**Extraction and isolation:** The dried sample (5.6 kg) was extracted with MeOH at room temperature  $(4 \times 5 \text{ L})$  to obtain 0.92 kg of the solid extract. The MeOH extract was suspended in H<sub>2</sub>O and extracted successively with hexane (3 × 3 L) and EtOAc (3 × 3 L) to give the hexane- (63.5 g) and EtOAc-soluble fractions (218 g), respectively. The EtOAc-soluble fraction (218 g) was chromatographed on a silica gel column eluted with a stepwise gradient of CHCl<sub>3</sub> and MeOH to yield five fractions (A-E: 35.4, 18.5, 23.6, 11.2 and 8.86 g, respec-

tively). Fraction B was further purified by HPLC with mobile phase (MeOH-H<sub>2</sub>O 65:35) to yield **3** (36.2 mg), **4** (14.6 mg), **6** (15.4 mg). Fraction C was further purified by HPLC with mobile phase (MeOH-H<sub>2</sub>O 60:40) to yield **1** (15.4 mg), **2** (64.5 mg), **5** (128.6 mg).

**Antioxidant activity assay:** Antioxidant activity was determined by the detection of the oxidative products with the 2',7'-dichlorofluorescin diacetate (DCFH) method reported previously<sup>12</sup>.

Smallanlignan A (1) was obtained as a white amorphous solid, showing  $[\alpha]_{2D}^{23.2}$  -55.02 (c 0.28, CHCl<sub>3</sub>). The molecular formula of **1** was determined to be  $C_{21}H_{22}O_6$  from high resolution ESIMS (positive model). IR spectrum of 1 showed an absorption at 3326 cm<sup>-1</sup> for hydroxyl groups. The <sup>1</sup>H NMR spectrum (Table-1) of **1** showed signals of 1,3,4-trisubstituted phenyl protons at  $\delta$  7.02 (d, 1.5), 6.99 (d, 8.2) and 6.95 (dd, 8.2, 1.5) and 1,4-substituted phenyl protons at  $\delta$  7.19 (2H, d, 8.4) and 6.79 (2H, d, 8.4), a signal due to a methoxyl protons at  $\delta$  3.89 (3H, s) and an acetyl methoxyl protons at  $\delta$  1.95 (3H, s). The <sup>13</sup>C NMR spectrum revealed 21 carbon signals typical to the lignan skeleton besides a methoxyl carbon signal at  $\delta$  55.6 and acetyl carbon signal at  $\delta$  21.4 q, 169.5 s. The methoxyl group was thought to be attached at C-3 position by consideration of heteronuclear multiple bond connectivity (HMBC) correlations between the protons at  $\delta$  3.89 (3H, s) and the carbon at  $\delta$  148.2 (C-3). In addition, the planar structure of 1 was also supported by other HMBC correlations, *i.e.*, between the hydroxyl proton at  $\delta$  5.48 (1H, brs) and  $\delta$  154.6 (C-4'), 116.1 (C-3', 5') (Figs. 2 and 3).

TABLE-1 <sup>1</sup> H AND <sup>13</sup> C NMR DATA OF SMALLANLIGNAN-A C IN CDCl <sub>3</sub>					
No.	$\delta_{\rm C}$ (mult.)	$\frac{\delta_{\rm H} \text{ (mult,}}{J, \text{Hz})}$	No.	$\frac{\delta_{\rm C}}{({\rm mult.})}$	$\frac{\delta_{\rm H} ({\rm mult},}{J,{\rm Hz})}$
1	135.6 s		3-OMe	55.6 q	3.89 s
2	110.5 d	7.02, d, 1.5	1'	133.5 s	
3	148.2 s		2'	126.8 d	7.19, d, 8.4
4	140.4 s		3'	116.1 d	6.79, d, 8.4
5	121.8 d	6.99, d, 8.2	4'	154.6 s	
6	119.8 d	6.95, dd, 8.2, 1.5	5'	116.1 d	6.79, d, 8.4
7	86.1 d	4.72 m	6'	126.8 d	7.19, d, 8.4
8	54.6 d	3.13, m	7'	85.9 d	4.70 m
9α	71.8 t	3.85 m	8'	54.2 d	3.11, m
9β		4.26 m	9'α	71.3 t	3.83 m
4-OAc	169.5 s		9'β		4.25 m
	21.4 q	1.95 s	4'-OH		5.48 brs

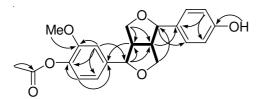


Fig. 2. Selected HMBC (®) and <sup>1</sup>H-<sup>1</sup>H COSY (-) of compound 1

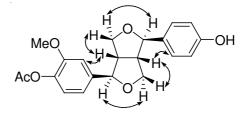


Fig. 3. Selected NOESY correlations of compound 1

By the comparison of circular dichroism (CD) spectrum of  $\mathbf{1}$  ( $[\theta]_{242} + 268.6$ ,  $[\theta]_{280} - 1016$ ) with that of known compound ((-)-pinoresinol), ( $[\theta]_{242} + 228$ ,  $[\theta]_{282} - 1038$ ), it is indicated that the absolute stereochemistry of  $\mathbf{1}$  was as same as that of(-)-pinoresinol<sup>13</sup>. Thus, the structure of  $\mathbf{1}$  was finally concluded to be the acetyl derivative of (-)-pinoresinol: (8S,2R, 5S,6R)-2-(4-hydroxyphenyl)-6-(3-methoxy-4-acetylphenyl)-3,7-dioxabicyclo[3.3.O]octane.

The antioxidant activity of smallanlignan A was determined by the detection of the oxidative products with the 2',7'dichlorofluorescin diacetate (DCFH) method reported previously<sup>12</sup>. It shows antioxidant activity with an IC<sub>50</sub> value of 1.65  $\mu$ g/mL. Smallanlignan A shows high antioxidant activity.

**Smallanactone A:** White amorphous solid;  $[α]_D^{223.2}$ -55.02 (c 0.28, CHCl<sub>3</sub>); UV  $λ_{max}$  (CDCl<sub>3</sub>) nm (log ε): 282 (1.61), 226 (5.18); CD (EtOH, c = 0.006) [θ]<sub>242</sub> +268.6, [θ]<sub>280</sub> -1016; IR (KBr,  $v_{max}$ , cm<sup>-1</sup>): 3418, 3326, 2874, 1757, 1602, 1587, 1526, 1476, 1342, 1268, 1118, 1027, 934, 855; <sup>1</sup>H and <sup>13</sup>C NMR data are shown in Table-1. HRESIMS (positive ion mode) m/z 393.1311 [M + Na]<sup>+</sup> (calcd. (%) 393.1314 for C<sub>21</sub>H<sub>22</sub>O<sub>6</sub>).

## REFERENCES

- 1. A. Moure, J.M. Cruz, D. Franco and J.M. Domnguez, *Food. Chem.*, **72**, 145 (2001).
- S. Graefe, M. Hermann, I. Manrique, S. Golombek and A. Buerkert, *Field. Crop. Res.*, 86, 157 (2004).
- 3. V.A. Neves and M.A. da Silva, J. Agric. Food. Chem., 55, 2424 (2007).
- 4. A. Narai-Kanayama, N. Tokita and K. Aso, J. Food. Sci., 72, 381 (2007).
- X. Zheng, G. Kuo, D. De-Qiang, K. Ting-Guo, S. Yu-Yuan and D. Feng, *Nat. Prod. Commun.*, 4, 1201 (2009).
- M. Adam, M. Juklova, T. Bajer, A. Eisner and K. Ventura, J. Chromatogr. A, 1084, 2 (2005).
- 7. K. Schorr and F.B. Costa, Phytochem. Anal., 16, 161 (2005).
- 8. J.S. Liu, Y. Tao and M.F. Huang, Acta. Chim. Sin., 46, 483 (1988).
- 9. Y.H. Kuo, C.H. Chen and Y.L. Lin, Chem. Pharm. Bull., 50, 978 (2002).
- L.M.X. Lopes, M. Yoshida and O.R. Gottlieb, *Phytochemistry*, 22, 1516 (1983).
- 11. A. Urzua, A.J. Freyer and M. Shamma, *Phytochemistry*, **27**, 1509 (1987).
- S. Takamatsu, A.M. Galal, S.A. Ross, D. Ferreira, M.A. Elsohly, A.R. Ibrahim and F.S. El-Feraly, *Phytother. Res.*, **17**, 963 (2003).
- A. Hosokawa, M. Sumino, T. Nakamura, S. Yano, T. Sekine, N. Ruangrungsi, K. Watanabe and F. Ikegami, *Chem. Pharm. Bull.*, 52, 1265 (2004).