

## Products from the Reactions of 4-(4-Methoxybenzoyl)-5-(4-Methoxyphenyl)-2,3-Furandione with Aryl Isocyanates

MUSTAFA SAÇMACI\* and YUNUS AKÇAMUR

*Department of Chemistry, Yozgat Faculty of Arts and Science*

*University of Erciyes, 66100-Yozgat, Turkey*

*E-mail: msacmaci@hotmail.com*

4-(4-Methoxybenzoyl)-5-(4-methoxyphenyl)-2,3-furandione (**1**) reacted with aryl isocyanates to give novel pyrrolo[2,3-d] pyrimidines (deazapurines) **2** and 1,3-oxazine-2,4-diones **3** under different conditions. The reaction pathways leading to **2** and **3** include [4 + 2] cycloaddition processes and decarboxylation across the (C=N) bond of the aryl isocyanate accompanied by rearrangements. N1-(1-naphthyl)-2-[2,4-di(4-methoxyphenyl)benzo[h]quinolin-3-yl]-2-oxoacetamide (**4**) was obtained from the thermolysis of **2d**. The formation of **4** proceeded *via* the iminobenzylfurandione as probable intermediate. The synthesis of new examples 1,3-oxazine-2,4-diones and pyrrolo[2,3d]pyrimidines following the unknown procedure and subsequent reactions using **1** and different aryl isocyanates has been reported.

**Key Words:** Pyrrolo[2,3-d]pyrimidines, 1,3-Oxazines, 2-Oxoacetamide, [4 + 2] Cycloaddition.

### INTRODUCTION

Pyrrolo[2,3-d]pyrimidines (7-deazapurines) are an important class of compounds, structurally and chemically related to nucleosides and some antibiotics<sup>1</sup>. The well known biological activity of these compounds has led to intensive investigation of their use as antitumor, anti-allergic, anti-viral and anti-inflammatory agents<sup>2</sup>. Fused 1,3-oxazine-2,4-diones have also been evaluated as cytotoxic or as potential anti-tumor agents<sup>3</sup>. Due to the interest in this class of compounds, we have previously reported<sup>7</sup> the reactivity and synthetic applications of these compounds.

4-Aryl substituted heterocyclic 2,3-diones, such as furandiones are versatile starting materials for a variety of reactions *via* generated cycloaddition of heterocumulenes across the oxa-1,3-diene subunit of  $\alpha$ -oxoketenes by thermolysis as well as addition of nucleophiles leading to a number of mono- and bicyclic heterocyclic systems<sup>4</sup>. The oxa-1,3-diene moiety in 4-benzoyl-5-phenylfuran-2,3-dione is well known to add isocyanides<sup>5, 6</sup> and several heterocumulenes *via* formal [4 + 1] or [4 + 2] cycloaddition processes affording various novel heterocyclic

systems. The oxa-1,3-diene system in 4-benzyl substituted five-membered heterocyclic 2,3-diones by the benzoyl group and the endocyclic C=C double bond is capable to add isocyanides, isocyanates, carbodiimides and ketenimines leading to various bicyclic heterocycles, and furan-2,3-diones are in general considered as convenient and versatile synthons in heterocyclic synthesis. They can serve as heterodienes in various cycloaddition processes, usually accompanied by surprising rearrangements<sup>7-9</sup>. Furthermore, these molecules are suitable precursors in generating highly reactive  $\alpha$ -oxoketenes during simple thermolysis in solution<sup>10, 11</sup>.

## EXPERIMENTAL

Melting points were obtained on an Electrothermal 9200 apparatus and not corrected. The FT-IR spectra were measured on a Jasco Plus Model 460 spectrometer, as potassium bromide pellets. The <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectra were recorded on Gemini-Varian 200 MHz spectrometer using tetramethylsilane as an internal standard. Elemental analyses were carried out by a Carlo-Erba 1108 Model 105 microanalyzer.

### General Procedure for the synthesis of 1*H*-pyrrolo[2,3*d*]pyrimidine-2,5,6-triones

1.0 g (2.96 mmol) of **1** were added to excess of (8.88 mmol) aryl isocyanates in a 25 mL round-bottomed flask equipped with a calcium chloride tube. The mixture was heated at 65°C for 24 h. After cooling to room temperature, the residue was triturated with anhydrous ether, and the crude product was recrystallized from corresponding solvents and dried on P<sub>2</sub>O<sub>5</sub>.

#### 4,7a-Di(4-methoxyphenyl)-1,3,7-triphenyl-2,3,5,6,7,7a-hexahydro-1*H*-pyrrolo[2,3*d*]pyrimidine-2,5,6-trione (**2a**)

Compound **2a** was prepared **1** and 1.050 g phenyl isocyanate. IR (cm<sup>-1</sup>) (KBr): 1730, 1686, 1674  $\nu$ (C=O), 1579  $\nu$ (C=C). <sup>1</sup>H-NMR (CDCl<sub>3</sub>):  $\delta$  7.67–6.20 (m, 23H, Ph), 3.89, 3.76 (s, 6H, CH<sub>3</sub>O); <sup>13</sup>C-NMR (CDCl<sub>3</sub>):  $\delta$  178.35, 165.37, 164.59 (C=O), 162.469–115.58 (C=C, arom. and aliph.), 81.36 (N—C—N), 57.53, 57.28 (CH<sub>3</sub>O).

#### 4,7a-Di(4-methoxyphenyl)-1,3,7-tri(4-methylphenyl)-2,3,5,6,7,7a-hexahydro-1*H*-pyrrolo[2,3-*d*]pyrimidine-2,5,6-trione (**2b**)

Compound **2b** was prepared **1** and 1.180 g *p*-tolyl isocyanate. IR (cm<sup>-1</sup>) (KBr): 1727, 1709, 1684  $\nu$ (C=O), 1586  $\nu$ (C=C). <sup>1</sup>H-NMR (CDCl<sub>3</sub>):  $\delta$  7.64–6.04 (m, 20H, Ph), 3.88, 3.77 (s, 6H, CH<sub>3</sub>O), 2.25, 2.18, 2.15 (s, 9H, CH<sub>3</sub>); <sup>13</sup>C-NMR (CDCl<sub>3</sub>):  $\delta$  178.51, 165.58, 164.47 (C=O), 162.35–115.52 (C=C, arom. and aliph.), 81.36 (N—C—N), 57.50, 57.28 (CH<sub>3</sub>O), 23.00, 22.94, 22.92 (CH<sub>3</sub>).

#### 1,3,7-Tri(4-chlorophenyl)-4,7a-di(4-methoxyphenyl)-2,3,5,6,7,7a-hexahydro-1*H*-pyrrolo[2,3-*d*]pyrimidine-2,5,6-trione (**2c**)

Compound **2c** was prepared **1** and 1.360 g 4-chlorophenyl isocyanate. IR

( $\text{cm}^{-1}$ ) (KBr): 1737, 1711, 1684  $\nu(\text{C}=\text{O})$ , 1584  $\nu(\text{C}=\text{C})$ .  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  7.58–6.23 (m, 20H, Ph), 3.89, 3.79 (s, 6H,  $\text{OCH}_3$ );  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  177.789, 164.919, 164.849 ( $\text{C}=\text{O}$ ), 162.705–115.82 ( $\text{C}=\text{C}$ , arom. and aliph.), 81.49 ( $\text{N}-\text{C}-\text{N}$ ), 57.56, 57.36 ( $\text{CH}_3\text{O}$ ).

**4,7a-Di(4-methoxyphenyl)-1,7-di(1-naphthyl)-3-(2-naphthyl)-2,3,5,6,7,7a-hexahydro-1H-pyrrolo[2,3-d]pyrimidine-2,5,6-trione (2d)**

Compound **2d** was prepared **1** and 1.5 g 1-naphthyl isocyanate. IR ( $\text{cm}^{-1}$ ) (KBr): 1750, 1711, 1689  $\nu(\text{C}=\text{O})$ , 1593  $\nu(\text{C}=\text{C})$ .  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  8.12–6.08 (m, 17H, Ph), 3.92, 3.67 (s, 6H,  $\text{CH}_3\text{O}$ );  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  178.757, 165.139, 164.050 ( $\text{C}=\text{O}$ ), 162.335–115.06 ( $\text{C}=\text{C}$ , arom. and aliph.), 82.87 ( $\text{N}-\text{C}-\text{N}$ ), 57.69, 57.14 ( $\text{CH}_3\text{O}$ ).

**1,3,7,7a-Penta(4-methoxyphenyl)-2,3,5,6,7,7a-hexahydro-1H-pyrrolo[2,3-d]pyrimidine-2,5,6-trione (2e)**

Compound **2e** was prepared **1** and 1.320 g 4-methoxyphenyl isocyanate. IR ( $\text{cm}^{-1}$ ) (KBr): 1729, 1712, 1682  $\nu(\text{C}=\text{O})$ , 1579  $\nu(\text{C}=\text{C})$ .  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  7.63–6.10 (m, 20H, Ph), 3.88, 3.65 (s, 15H,  $\text{CH}_3\text{O}$ );  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  178.519, 165.706, 164.449 ( $\text{C}=\text{O}$ ), 162.402–115.43 ( $\text{C}=\text{C}$ , arom. and aliph.), 81.27 ( $\text{N}-\text{C}-\text{N}$ ), 57.501, 57.453, 57.273 ( $\text{CH}_3\text{O}$ ).

**1,3,7-Tri(3,5-dimethylphenyl)-4,7a-di(4-methoxyphenyl)-2,3,5,6,7,7a-hexahydro-1H-pyrrolo[2,3-d]pyrimidine-2,5,6-trione (2f)**

Compound **2f** was prepared **1** and 1.300 g 3,5-dimethylphenyl isocyanate. IR ( $\text{cm}^{-1}$ ) (KBr): 1732, 1705, 1665  $\nu(\text{C}=\text{O})$ , 1587  $\nu(\text{C}=\text{C})$ .  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  7.64–5.81 (m, 17H, Ph), 3.88, 3.77 (s, 6H,  $\text{CH}_3\text{O}$ ), 2.06, 1.94 (s, 18H,  $\text{CH}_3$ );  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  178.577, 165.479, 164.549 ( $\text{C}=\text{O}$ ), 162.453–115.50 ( $\text{C}=\text{C}$ , arom. and aliph.), 81.36 ( $\text{N}-\text{C}-\text{N}$ ), 57.562, 57.253 ( $\text{CH}_3\text{O}$ ), 22.894 ( $\text{CH}_3$ ).

General Procedure for the synthesis of 2H-1,3-oxazine-2,4-diones

A mixture of 1.0 g (2.96 mmol) **1** and 2.96 mmol phenyl isocyanate was heated at 120°C until the evolution of CO has subsided (1 h). The cooled reaction mixture was triturated with dry ether and then recrystallized from corresponding solvents and dried on  $\text{P}_2\text{O}_5$ .

**5-(4-Methoxybenzoyl)-6-(4-methoxyphenyl)-3-phenyl-3,4-dihydro-2H-1,3-oxazine-2,4-dione (3a)**

Compound **3a** was prepared **1** and 0.350 g phenyl isocyanate. IR ( $\text{cm}^{-1}$ ) (KBr): 1774, 1690, 1646  $\nu(\text{C}=\text{O})$ .  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  7.95–6.81 (m, 13H, Ph), 3.84, 3.79 (s, 6H,  $\text{CH}_3\text{O}$ );  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  190.799, 166.598, 165.035 ( $\text{C}=\text{O}$ ), 162.364–113.47 (arom. and aliph.  $\text{C}=\text{C}$ ), 57.52, 57.44 ( $\text{CH}_3\text{O}$ ).

**5-(4-Methoxybenzoyl)-6-(4-methoxyphenyl)-3-(4-methylphenyl)-3,4-dihydro-2H-1,3-oxazine-2,4-dione (3b)**

Compound **3b** was prepared **1** and 0.39 g *p*-tolyl isocyanate. IR ( $\text{cm}^{-1}$ ) (KBr):

1773, 1688, 1660  $\nu(\text{C}=\text{O})$ ,  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  7.95–6.80 (m, 12H, Ph), 3.81, 3.76 (s, 6H,  $\text{CH}_3\text{O}$ ), 2.38 (s, 3H,  $\text{CH}_3$ );  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  190.719, 166.579, 165.014 ( $\text{C}=\text{O}$ ), 162.406–113.70 (arom and aliph.  $\text{C}=\text{C}$ ), 57.46, 57.38 ( $\text{CH}_3\text{O}$ ).

### 3-(4-Chlorophenyl)-5-(4-methoxybenzoyl)-6-(4-methoxyphenyl)-3,4-dihydro-2H-1,3-oxazine-2,4-dione (3c)

Compound **3c** was prepared **1** and 0.45 g 4-chlorophenyl isocyanate. IR ( $\text{cm}^{-1}$ ) (KBr): 1769, 1690, 1654  $\nu(\text{C}=\text{O})$ , 1598  $\nu(\text{C}=\text{C})$ .  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  7.95–6.78 (m, 12H, Ph), 3.82, 3.78 (s, 6H,  $\text{CH}_3\text{O}$ );  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  190.68, 166.65, 165.07, 162.469–113.33 (m, arom.  $\text{C}=\text{C}$ ), 57.55, 57.44 ( $\text{CH}_3\text{O}$ ).

### 5-(4-Methoxybenzoyl)-6-(4-methoxyphenyl)-3-phenyl-3,4-dihydro-2H-1,3-oxazine-2,4-dione (3d)

Compound **3d** was prepared **1** and 0.50 g 1-naphthyl isocyanate. IR ( $\text{cm}^{-1}$ ) (KBr): 1770, 1682, 1659  $\nu(\text{C}=\text{O})$ .  $^1\text{H-NMR}$  (DMSO):  $\delta$  8.20–7.02 (m, 11H, Ph), 3.86, 3.81 (s, 6H,  $\text{CH}_3\text{O}$ );  $^{13}\text{C-NMR}$  (DMSO):  $\delta$  190.72, 166.02, 164.03 ( $\text{C}=\text{O}$ ), 162.797–113.65 (m, arom.  $\text{C}=\text{C}$ ), 57.52, 57.44 ( $\text{CH}_3\text{O}$ ).

### N1-(1-Naphthyl)-2-[2,4-di(4-methoxyphenyl)benzo[h]quinolin-3y1]-2-oxoactamide (4)

4,7a-Di(4-methoxyphenyl)-1,7-di(1-naphthyl)-3-(2-naphthyl)-2,3,5,6,7,7a-h exahydro-1H-pyrrolo[2,3-d]pyrimidine-2,5,6-trione 1.0 g (1.32 mmol) was kept in an oil-bath at 220°C for 30 min, then the melted product was dissolved in ether/petrol ether (1 : 1), and stirred for a few days at room temperature. The red crude product was filtered and then dried on a  $\text{P}_2\text{O}_5$ . IR (KBr):  $\nu$  3233 (N—H), 1748, 1704  $\text{cm}^{-1}$  ( $\text{C}=\text{O}$ ).  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  8.22–6.04 (m, 11H, arom), 3.95, 3.48 (s, 6H,  $\text{CH}_3\text{O}$ ), NH (no detection);  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  182.702, 171.487 ( $\text{C}=\text{O}$ ), 164.608–115.00 ( $\text{C}=\text{C}$ ), 57.45, 56.93 ( $\text{CH}_3\text{O}$ ).

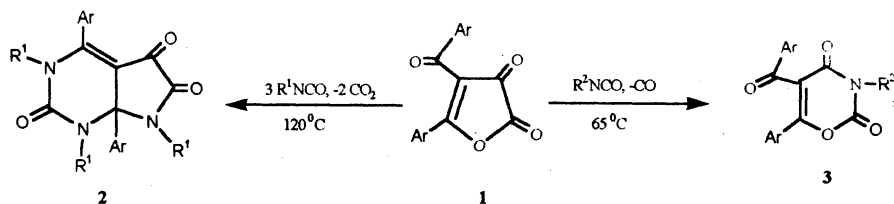
## RESULTS AND DISCUSSION

4-(4-Methoxybenzoyl)-5-(4-methoxyphenyl)-2,3-furandione (**1**) reacted with some aryl isocyanates in neat about 24 h at 65°C, releasing two moles of carbon dioxide, led to the formation of the compounds (**2a–f**) in 70–80% yield (**Scheme-1**). All the reaction steps obviously include formal [4 + 1] or [4 + 2] cyclo-addition processes accompanied by long-range Dimroth furandione rearrangements<sup>12</sup>. Previously, the mechanism of formation of pyrrolo-[2,3-d]pyrimidines from the furandione with aryl isocyanates was reported<sup>12</sup>. Compound **2a** shows characteristic IR absorption band at 1730, 1686, 1674  $\text{cm}^{-1}$  ( $\text{C}=\text{O}$ ). Its characteristic  $^{13}\text{C-NMR}$  signals at 178.35, 165.37, 164.59 ppm ( $\text{C}=\text{O}$ ), 81.36 ppm (N—C—N), are in full agreement with its proposed structures (Table-1).

TABLE-I  
PHYSICAL DATA AND ELEMENTAL ANALYSES OF COMPOUNDS 2 AND 3

Compds.	Yield (%)	m.p. (°C) (solvent)	m.f. (m.w.)	Elemental analysis (%), Calcd./ (Found)		
				C	H	N
<b>2a</b>	70	201 (acetic acid)	C <sub>38</sub> H <sub>29</sub> N <sub>3</sub> O <sub>5</sub> (607)	75.12 (74.85)	4.77 (4.83)	6.91 (7.01)
<b>2b</b>	65	225 (ethanol)	C <sub>40</sub> H <sub>35</sub> N <sub>3</sub> O <sub>5</sub> (637)	75.80 (76.02)	5.39 (5.69)	6.47 (6.22)
<b>2c</b>	75	235 (ethanol)	C <sub>38</sub> H <sub>26</sub> N <sub>3</sub> O <sub>5</sub> Cl <sub>3</sub> (710)	64.18 (63.98)	3.66 (3.40)	5.91 (5.60)
<b>2d</b>	80	217' (butanol)	C <sub>50</sub> H <sub>35</sub> N <sub>3</sub> O <sub>5</sub> (757)	79.20 (79.46)	4.62 (4.94)	5.54 (5.97)
<b>2e</b>	65	222 (butanol)	C <sub>41</sub> H <sub>35</sub> N <sub>3</sub> O <sub>8</sub> (697)	70.58 (70.86)	5.02 (4.94)	6.02 (5.97)
<b>2f</b>	60	219 (ethanol)	C <sub>44</sub> H <sub>41</sub> N <sub>3</sub> O <sub>5</sub> (691)	76.40 (76.15)	5.93 (5.90)	6.07 (6.01)
<b>3a</b>	75	232 (butanol)	C <sub>25</sub> H <sub>19</sub> NO <sub>6</sub> (429)	69.93 (69.80)	4.42 (4.51)	3.26 (3.14)
<b>3b</b>	65	195 (methanol)	C <sub>26</sub> H <sub>21</sub> NO <sub>6</sub> (443)	70.42 (70.18)	4.74 (4.51)	3.16 (2.93)
<b>3c</b>	55	220 (acetic acid)	C <sub>25</sub> H <sub>18</sub> NO <sub>6</sub> Cl (463)	64.72 (64.62)	3.88 (3.86)	3.02 (2.83)
<b>3d</b>	60	289 (acetic acid)	C <sub>29</sub> H <sub>21</sub> NO <sub>6</sub> (479)	72.65 (72.76)	4.38 (4.38)	2.92 (2.89)
<b>4</b>	80	225 (acetone)	C <sub>39</sub> H <sub>30</sub> N <sub>2</sub> O <sub>4</sub> (590)	79.32 (79.06)	5.08 (4.81)	4.74 (5.01)

The thermal decomposition of 4-benzoyl-5-phenylfuran-2,3-dione leads to the dibenzoylketene as an intermediate which undergoes cycloaddition reactions with hetero-cumulenes<sup>13</sup>. In a similar way, the thermal decomposition of novel 4-(4-methoxybenzoyl)-5-(4-methoxyphenyl)-2,3-furandione generates to the intermediate *p,p'*-dimethoxy dibenzoylketene<sup>14</sup>. The 1,3-oxazine-2,4-diones (**3a-d**) were obtained in 40–70% yields by the addition of aryl isocyanates to the  $\alpha$ -oxoketene generated by thermolysis of **1** (Scheme-1). The structures of compounds (**3a-d**) were confirmed by elemental analysis, IR, <sup>1</sup>H and <sup>13</sup>C-NMR spectral data, e.g., three carbonyl absorption bands 1774, 1690, 1646 cm<sup>-1</sup> (C=O) in the IR spectra as well as the corresponding signals in the <sup>13</sup>C-NMR spectra at  $\delta$  190.80 (anisoyl C=O), 165.03 (N—C=O), 166.58 (anisyl —C—O), 149.70 (N—CO—O), similar to very close analogues<sup>15</sup>.

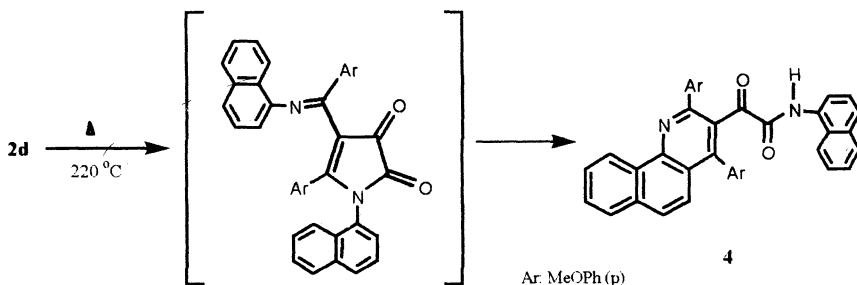


2,3	R <sup>1</sup>	R <sup>2</sup>
a	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>
b	C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub> -(4)	C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub> -(4)
c	C <sub>6</sub> H <sub>4</sub> Cl-(4)	C <sub>6</sub> H <sub>4</sub> Cl-(4)
d	C <sub>10</sub> H <sub>7</sub>	C <sub>10</sub> H <sub>7</sub>
e	C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub> -(4)	-
f	C <sub>6</sub> H <sub>3</sub> (CH <sub>3</sub> ) <sub>2</sub> -(3,5)	-

Ar: MeOPh (p)

Scheme-1

Finally, from thermolyze of **2d** at 220°C, compound **4** was furnished (Scheme 2). The product **4** was gained in 40% yield. It proceeded *via* iminobenzyl-furandiones as probable intermediate similar to the mechanism of thermolyze reaction of 7,7a-dihydro-1,3,4,7,7a-pentaphenyl-1H-pyrrolo[2,3-d]pyrimidine-2,5,6(3H)-trione to N,2,4-triphenyl-3-quinolin glyoxylamide<sup>7</sup>. In the IR spectrum of compound **4**, the (C=O) absorption bands were observed at 1748 and 1704 cm<sup>-1</sup>. The NH absorption band could not be identified in the <sup>1</sup>H-NMR spectrum of **4**, because of intramolecular hydrogen bonding between NH and oxygen at the 2-position<sup>16</sup>. That compound can be regarded as simple [4 + 2]-cycloreversion and this combined reaction sequence represents one example of the Ziegler-Hafner azulene synthesis<sup>17</sup>.



Scheme-2

## ACKNOWLEDGEMENTS

The authors thank the research fund (EUBAP: 01-66-1) of Erciyes University for financial support.

## REFERENCES

- (a) F. Seela, H.D. Winkeler, J. Ott, Q.H. Tran-thi, D. Hasselman, D. Franzen and W. Bubmann, in: *Nucleosides, Nucleotides and Their Biological Applications*, J.L. Rideout, D.W. Henry and L.M. Beachan III (Eds.), Academic Press Inc., New York, p. 181 (1983); (b) R.L. Tolmann, R.K. Robins and L.D. Townsend, *J. Am. Chem. Soc.*, **90**, 524 (1968); (c)

- H. Kassay, Z. Ohashi, F. Harada, S. Nishimura, N.J. Oppenheimer, P.F. Crain, J.G. Liehr, D.L. von Minden and J.A. McCloskey, *Biochemistry*, **14**, 4198 (1975); (d) T. Ongi, T. Kondo and G. Goto, *J. Am. Chem. Soc.*, **101**, 3629 (1979); (e) K. Anzai and S. Suzuki, *J. Antibiot. Ser. A*, **14**, 253 (1961); (f) K. Anzai, J. Nagatsu and S. Suzuki, *J. Antibiot. Ser. A*, **15**, 109 (1962); (g) R.J. Suhadolnic, *Nucleoside Antibiotics*, Wiley Interscience, New York (1970).
2. J.A. Montgomery and H.J. Thomas, in: E.D. Bergmann and B. Pullman (Eds.), *Proceedings of the 4th International Symposium on Quantum Chemistry and Biochemistry*, Academic Press, Jerusalem, p. 446 (1972); (b) L. B. Townsend and G.H. Milne, *Chemistry, Biology and Clinical Uses of Nucleoside Analogs*; A. Bloch, (Ed.), *Ann. N.Y. Acad. Sci.*, **91**, 225 (1975); (c) W. Hutzenlaub, R.L. Tolman and R. Krobins, *J. Med. Chem.*, **15**, 879 (1972); (d) C.W. Smith, R.W. Sidwell, R.K. Robins and R.L. Tolman, *J. Med. Chem.*, **15**, 833 (1972); (e) K. Eger, R. Fruchmann, H. Hurstmann, H. Jacabi and H.J. Roddath Roth, *German Offen. DE 3*, **287**, 3145 (1983); (f) *Chem. Abstr.*, **99**, 53780e (1983).
  3. M.R. Player and J.W. Sowell, *J. Heterocyclic Chem.*, **32**, 1537 (1995).
  4. G. Kollenz and W. Heilmayer, *Trends Heterocyclic Chem.*, **3**, 379 (1993); C. Wentrup, W. Heilmayer and G. Kollenz, *Synthesis* (Special Issue), 1219 (1994); Zs. Juhasz-Riedl, G. Hajos, G. Kollenz and A. Messmer, *Chem. Ber.*, **122**, 1935 (1989).
  5. G. Kollenz, W. Ott, E. Ziegler, K. Peters, H.G. von Schnering and H. Quast, *Liebigs Ann. Chem.*, 1801 (1980).
  6. G. Kollenz, W. Ott, E. Ziegler, E.M. Peters, K. Peters, H.G. von Schnering, V. Formacek and H. Quast, *Liebigs Ann. Chem.*, 1137 (1984).
  7. (a) G. Kollenz and K. Peters, *Chem. Ber.*, **114**, 1206 (1981); (b) G. Kollenz, G. Penn, G. Dolenz, Y. Akcamur, E.M. Peters, K. Peters and H.G. von Schnering, *Chem. Ber.*, **117**, 1299 (1984).
  8. G. Kollenz, G. Penn, W. Ott, K. Peters, E.M. Peters and H.G. von Schnering, *Chem. Ber.*, **117**, 1310 (1984).
  9. ———, *Heterocycles*, **26**, 624 (1987).
  10. G. Kollenz and W. Heilmayer, *Trends Heterocyclic Chem.*, **3**, 379 (1993).
  11. (a) Y.S. Andreichikov, Y.A. Nalimova and I.A. Rusakov, *Zh. Org. Khim.*, **14**, 2436 (1978); (b) O.N. Kolesnikova, L.I. Livantsova, Y.M. Shurov and A. Zaitseva, *ibid.*, **24**, 458, (1988); (c) T.N. Yanbarisov, S.N. Shurov, Y.S. Andreichikov, I.P. Rudokova, E.N. Semenova and G.N. Novoselova, *Khim. Farm. Zh.*, **23**, 1470 (1989); (d) E. Ziegler, G. Kollenz and H. Igel, *Monatsh. Chem.*, **102**, 1769 (1971); (e) E. Ziegler, G. Kollenz and G. Kriewetz, *Liebigs Ann. Chem.*, 1751 (1977); (f) G. Kollenz, E. Ziegler, W. Ott and G. Kriewetz, *Z. Naturforsch.*, **32b**, 701 (1977).
  12. G. Kollenz, H. Sterk and G. Hutter, *J. Org. Chem.*, **56**, 235 (1991).
  13. C. Wentrup, H.W. Winter, G. Gross, K.P. Netsch, G. Kollenz, W. Ott and A.G. Biedermann, *Angew. Chem. Int. Ed. Engl.*, **23**, 800 (1984).
  14. E. Saripinar, Y. Guzel, Z. Onal, I. Ozer Ilhan and Y. Akcamur, *J. Chem. Soc. Pak.*, **22**, 308 (2000).
  15. G. Kollenz, H. Igel and E. Ziegler, *Monatsh. Für Chemie*, **103**, 450 (1972).
  16. S.-D.C. Joo-Wha Chung, W.-Y. Choi, S.K. Kim and Y.-J. Yoon, *J. Heterocyclic Chem.*, **31**, 1199 (1994).
  17. K. Ziegler and K. Hafner, *Angew. Chem.*, **67**, 301 (1955).