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# Synthesis and Pharmacological Activity of Some Phenoxy Chalcones

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Some new phenoxy chalcones were prepared by condensing different aromatic aldehydes with phenoxy acetophenones. All the synthesized compounds were confirmed by <sup>1</sup>H NMR and MASS spectral studies and were screened for their antiinflammatory activity. Some of the synthesized compounds show potent antiinflammatory activity.

Key Words: Phenoxy acetophenones, Aromatic aldehyde, Phenoxy acetophenone, Antiinflammatory activity.

## **INTRODUCTION**

Chalcones a class of naturally occurring pigments, have a broad spectrum of biological activities such as antiinflammatory<sup>1</sup>, antibacterial<sup>2,3</sup>, antifungal<sup>4</sup>, anthelmintic<sup>5</sup> and antitumor<sup>6</sup> activities. In view of the above mentioned facts, we intend to synthesize some substituted 4'-phenoxy chalocones and screen them for antiinflammatory activity.

## EXPERIMENTAL

The structures of all these compounds were established by <sup>1</sup>H NMR and mass spectral studies. Melting points were determined in capillaries in liquid paraffin and are uncorrected. <sup>1</sup>H NMR spectra were recorded on Bruker DRX (300 MHz, FT NMR) spectrometer using TMS as an internal standard ( $\delta$  ppm) and mass spectra on Jeol 5x102/DA-6000 mass spectrometer.

**Synthesis of phenoxy acetophenone (I):** To a cooled mixture of biphenyl ether (0.33 mol, 55 mL) and aluminium chloride (18.8 g) in carbon disulphide (87.5 mL), acetic anhydride (12 mL) was added and refluxed for 2 h. After 2 h, the reaction mixture was cooled to room temperature and poured into a mixture of crushed ice and conc. HCl. It was then extracted with diethyl ether. After removing the ether, the residue was recrystallized

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in ethanol. Yield: 78.72 %; m.p.: 48-50°C; IR (KBr, cm<sup>-1</sup>): 1680 v(C=O), 1230 v(C-O-C), 1580, 740, 680; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  2.56 (s, 3H, -COCH<sub>3</sub>), 6.98~7.39 (m, 9H, aromatic protons); MS: m/z 212 (M<sup>+</sup>), 197, 141, 115, 77.

**General synthesis of substituted 4'-phenoxy chalcones (Ia-d):** To a stirred solution of potassium hydroxide (3 g) in water (3 mL) and ethanol (50 mL) mixture, 4'-phenoxy acetophenone (0.03 mol, 6.37 g) and aromatic aldehyde (0.01 mol) were added. The reaction mixture was stirred vigorously for 3 h and then poured into crushed ice, a solid mass separates out. The product was filtered, washed with cold water (50 mL) and recrystallized from methanol (Table-1) (**Scheme-I**).

TABLE-1 PHYSICAL CHARACTERISTICS OF THE SYNTHESIZED COMPOUNDS

Compound	R	m.p. (°C)	Yield (%)
Ia	Н	80-82	77.46
Ib	4-Methoxy	60-64	74.36
Ic	3,4-Dimethoxy	106-108	76.47
Id	3,4-Methylene dioxy	126-128	74.06



Aromatic aldehyde Potassium hydroxide Ethanol



Scheme-I

## Spectral data

**4'-Phenoxy chalcone (1a):** <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  8.04 (d, 2H, phenoxy substituted phenyl protons), 7.82 (d, 1H, alkenyl β-proton), 7.52 (d, 1H, alkenyl α-proton), 7.2 ~ 7.04 (m, 12H, aromatic protons); MS: m/z 300 (M<sup>+</sup>), 272, 207, 197, 154, 103, 77.

**1b:** <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.77 (d, 1H, alkenyl β-proton), 7.55(d, 2H, phenoxy substituted phenyl protons), 7.39 ~ 7.02 (m, 10H, aromatic protons and α-proton), 6.88 (d, 2H, phenyl protons ortho to methoxy group), 3.78 (s, 3H, methoxy); MS : m/z 330 (M<sup>+</sup>), 237, 197, 170, 154, 77.

**1c:** <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  8.03~7.09 (m, 13H, aromatic protons and alkenyl protons), 6.88 (d, 1H, phenyl proton ortho to methoxy group), 3.78 (s, 3H, methoxy); MS: m/z 360 (M<sup>+</sup>), 346, 330, 267, 212, 197, 157, 77.

**1d:** <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  8.01~7.14 (m, 13H, aromatic protons and alkenyl protons), 6.85 (d, 1H, phenyl proton ortho to methylene dioxy group), 6.01 (s, 2H, methylene dioxy group); MS: m/z 344 (M<sup>+</sup>), 316, 251, 199, 77.

**Antiinflammatory activity:** All the synthesized compounds were screened for antiinflammatory activity by carrageenan induced rat paw edema method<sup>7</sup> in Albino rats at a dose of 20 mg/kg. The test compounds were made into homogeneous suspension with distilled water, 1 % CMC and were administered orally. The percentage inhibition was noted at the end of 3 h of administration of carrageenan. Carrageenan induced paw edema method of the test compounds was compared with known standard compound (Indomethacin). Among the synthesized compounds, **Ic** and **Id** showed significant antiinflammatory activity.

# **RESULTS AND DISCUSSION**

All the synthesized compounds were characterized by <sup>1</sup>H NMR and mass spectral data. In <sup>1</sup>H NMR of compounds the characteristic alkenyl protons merges with the aromatic region only in 4'-unsubstituted phenoxy chalcones, they show at  $\delta$  7.52 and  $\delta$  7.82. Further, the methoxy protons and methylene dioxy protons appear at  $\delta$  3.78 and  $\delta$  6.01, respectively. By analyzing the mass spectra it was found that the molecular ion peaks were strong probably due to the rigid ring structure. Among the synthesized compounds, **Ic** and **Id** were found to be potent than the other compounds. Compound **Ic** and **Id** show 37.11 and 50.60 % inhibition, respectively after 3 h when compared with standard drug indomethacin, which show 70.43 % inhibition (Table-2).

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# TABLE-2 ANTIINFLAMMATORY ACTIVITY OF THE SYNTHESIZED COMPOUNDS

Compound	Dose (mg/kg)	% Inhibition of edema at 3 h
Ι	20	15.90
Ia	20	21.10
Ib	20	31.40
Ic	20	37.11*
Id	20	50.60*
Std. (Indomethacin)	10	70.43*

\*p < 0.01 and n = 6, where n = number of rats used in each group.

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