

**NOTE****Intramolecular Cyclization of  
N'-Chloroacetylindole Hydrazide**

PRABHUODEYARA M. VEERESHA SHARMA

*Department of Chemistry, Gulbarga University, Gulbarga-585 106, India**E-mail: pmvsharma@gmail.com*

In this paper, some active class of compounds were synthesized, which are linked to indole nucleus. Various ethyl indole-2-carboxylates (**1a-c**) were prepared according to the Fischer method. These esters (**1a-c**) on reaction with hydrazine hydrate in ethanol yielded substituted indole-2-carboxyhydrazides (**2a-c**). Hydrazides (**2a-c**) on reaction with chloroacetyl chloride in dry dioxane at reflux temperature to get N'-chloroacetylindole hydrazide (**3a-c**). The compounds **3a-c** on reaction sodium hydroxide in dimethyl formamide at reflux temperature with constant stirring gave 5,6-dihydro-5-substituted-3-phenylindole-1,3,4-oxadiazin-5-one (**4a-c**).

**Key Words:** Synthesis, Cyclization, N'-Chloroacetylindole hydrazide.

Indole and its derivatives have attracted immense research interest due to their applications in pharmacology studies<sup>1,2</sup>. They can also be used in anti-HIV activity and varied biodynamic properties of indole analogues as reported in the literature earlier<sup>3</sup>. On continuation of interest in the synthesis of new medium-sized heterocycles<sup>4</sup> related hydrazine as potential psychopharmacological drug with monoaminoxidase inhibitory properties, the synthesis of 5,6-dihydro-5-substituted-3-phenylindole-1,3,4-oxadiazin-5-one (**4a-c**) is reported.

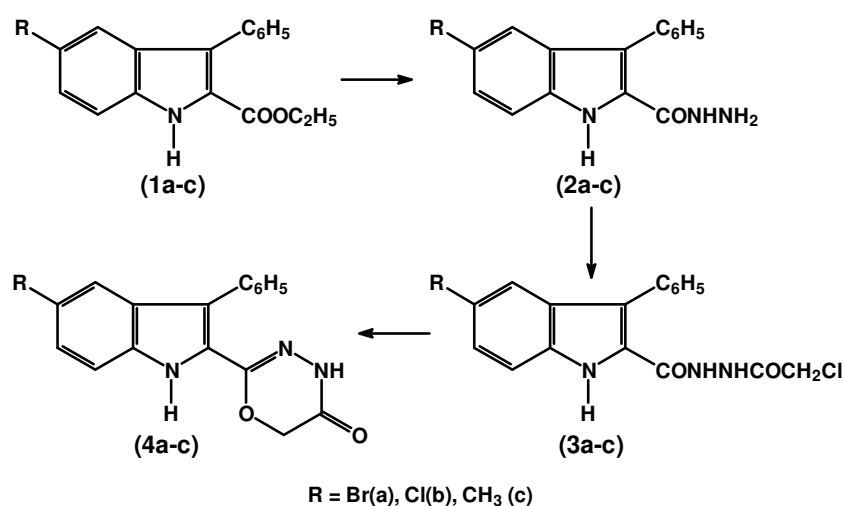
All melting points were recorded in open capillaries in liquid paraffin bath and are uncorrected. The reaction was monitored by TLC, IR spectra were recorded in KBr on Perkin-Elmer (one spectrum). <sup>1</sup>H NMR spectra were recorded on AMX-400 (in DMSO, using TMS as internal standard) at Sophisticated Instruments Facility, Indian Institute of Science, Bangalore, India. Resonances are assigned in ppm value. 5-substituted 3-phenylindole-2-carboxyhydrazide (**2a-c**) was prepared according to the literature procedure<sup>5</sup>.

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**Synthesis of N'-chloroacetylindole hydrazide (3a-c):** Mixture of 10 g of indole-2-carboxyaldehyde (2a-c), 8.5 g of chloroacetyl chloride in 40 mL of dry dioxane was refluxed for 2 h. After cooling, the crystals were separated and recrystallized from dioxane.

**Synthesis of 5,6-dihydro-5-substituted-3-phenylindole-1,3,4-oxadiazin-5-one (4a-c):** A mixture of N'-chloroacetylindole hydrazide or N'-substituted-N'-chloroacetylindole hydrazide (3a-c), 0.4 g of sodium hydroxide in 25 mL of dimethylformamide was heated at 130 °C in oil bath for 2 h with stirring and filtered. The solvent was evaporated, the solid obtained was crystallized from ethanol (Scheme-I).



**Scheme-I**

The structures of the synthesized compounds (3a-c) were confirmed by their spectral data and elemental analysis (Table-1). IR spectrum of 3a displayed a peak at 3231 cm<sup>-1</sup> for indole NH, peak at 3038 cm<sup>-1</sup> for NH of chloroacetylindole hydrazide and 1684 cm<sup>-1</sup> for carbonyl functionality. <sup>1</sup>H NMR of 3a exhibits singlet at 12.2 ppm which can be assigned to NH proton of indole moiety. Resonances at 10.6 ppm and 10.0 ppm can be ascribed to NH/NH of chloroacetyl hydrazide protons.

Compound (3a-c) on reaction with NaOH in DMF to yield 5,6-dihydro-5-substituted-3-phenylindole-1,3,4-oxadiazin-5-one (4a-c). The compound 4a exhibited 3312 cm<sup>-1</sup> for indole NH, 1682 cm<sup>-1</sup> for cyclic carbonyl group. The <sup>1</sup>H NMR spectrum of compound 4a exhibits indole NH peak at 12.5 ppm and 11.6 ppm for cyclic NH and 8.3 ppm for cyclic -CH<sub>2</sub>, the disappearance of NH group at 10.0 ppm confirms the cyclization of oxadiazine ring.

TABLE-1  
ANALYTICAL AND SPECTRAL DATA OF THE COMPOUNDS **3a-c** AND **4a-c**

Compd. (m.f.)	Yield (%) / (m.p. °C)	Elemental analysis (%):			IR (cm <sup>-1</sup> )	<sup>1</sup> H NMR (ppm)
		Calcd. (Found)				
		C	H	N		
<b>3a</b> (C <sub>17</sub> H <sub>13</sub> N <sub>3</sub> O <sub>2</sub> BrCl)	62 (240)	50.24 (50.20)	3.20 (3.18)	10.34 (10.32)	3231,NH; 3038,NH; 1684,C=O	12.2, indole NH; 10.6, NH; 10.0, NH; 4.3,CH <sub>2</sub>
<b>3b</b> (C <sub>17</sub> H <sub>13</sub> N <sub>3</sub> O <sub>2</sub> Cl <sub>2</sub> )	55 (260)	56.50 (56.20)	3.60 (3.58)	12.31 (12.30)	3200,NH; 3025,NH; 1690,C=O	12.1, indole NH; 10.4,NH;10.3,NH; 4.3,CH <sub>2</sub>
<b>3c</b> (C <sub>18</sub> H <sub>16</sub> N <sub>3</sub> O <sub>2</sub> Cl)	64 (232)	59.82 (59.80)	3.81 (3.80)	12.31 (12.30)	3219,NH; 3014,NH; 1684,C=O	12.2, indole NH; 10.5,NH;10.0,NH; 4.3,CH <sub>2</sub>
<b>4a</b> (C <sub>17</sub> H <sub>12</sub> N <sub>3</sub> O <sub>2</sub> Br)	70 (250)	54.94 (54.92)	3.23 (3.22)	11.31 (11.30)	3212,NH; 1682,C=O cyclic	12.5, indole NH; 11.6,cyclic NH; 8.3, cyclic CH <sub>2</sub>
<b>4b</b> (C <sub>17</sub> H <sub>13</sub> N <sub>3</sub> O <sub>2</sub> Cl)	54 (262)	62.57 (62.55)	3.68 (3.65)	12.88 (12.86)	3443,NH; 1692,C=O cyclic	12.4, indole NH; 11.6,NH; 8.3, cyclic CH <sub>2</sub>
<b>4c</b> (C <sub>18</sub> H <sub>16</sub> N <sub>3</sub> O <sub>2</sub> )	68 (258)	66.66 (66.64)	3.92 (3.91)	13.78 (13.76)	3323, NH; 1590, C=O cyclic	12.2, indole NH; 11.2,cyclic NH; 8.6, cyclic CH <sub>2</sub>

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