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# Antidiabetic Activity of *Coccinia indica* in Streptozotocin Induced Diabetic Rats

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The purpose of this study was to investigate the antidiabetic potential of the *n*-hexane extracts of *Coccinia indica* for the treatment of diabetic mellitus in the streptozotocin included diabetic rats at 200 mg/kg body weight for 30 d period. The effect of the compound was found to be 206  $\pm$ 23.54 mg/dl of bood, after the extract was given to the diabetic rats.

Key Words: Coccinia indica, Antidiabetic activity.

#### **INTRODUCTION**

Diabetes mellitus is now recognized as serious global health problem<sup>1</sup>. *Coccinia indica* is grown abundantly in India and have been widely used in the traditional treatment of diabetes mellitus<sup>2</sup>. Contradictory findings on the antidiabetic activity of the different parts of *Coccinia indica* have been reported<sup>3,4</sup>.

Nikkhila and Kekki<sup>5</sup> have reported an increase in the cholesterol level in diabetic rats, which was control by fraction treated plant extract. Galletto *et al.*<sup>6</sup> have shown the antidiabetic effect of *Gymnema sylvestre*, similarly Shrivastava *et al.*<sup>7</sup> have reported that 70 % ethanolic extract of *Butea monosperma* where the extract showed increasing insulin in the  $\beta$ -cells of pancreas. Latha and Pari<sup>8</sup> have reported be the effect of aqueous extract of *Scoparia dulcis* on blood glucose level in the rats. Hussain<sup>9</sup> reported the reversal of diabetic retinopathy in STZ induced diabetic rats using traditional Indian antidiabetic plant *Azadirachta indica*. Venkateswaran *et al.*<sup>10</sup> reported the effects of *Coccinia indica* on blood glucose, insulin and hepatic key enzymes in experimental diabetes.

In the present study, it was planned to test a few fractions and delineate the most active fraction of *Coccinia indica*. The *n*-hexane extract of *Coccinia indica* was found to be more active in reducing blood glucose level: this extract was subjected to further fraction and evaluation on antidiabetic activity. 6480 Shakya

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## **EXPERIMENTAL**

*Coccinia indica* (Wight & Arn.) commonly known as Kanduri belonging to the family Cucurbitaceae. Aerial parts of *Coccinia indica* were collected, dried under shade, powdered and the alcoholic extract was prepared. The dried alcoholic extract was a semi-solid mass and was successively extracted with *n*-hexance, chloroform, ethyl acetate and *n*-butanol, concentrated and dried in a dessicator. The extracts were used for the present study as shown in Table-1.

TABLE-1 PERCENTAGE YIELD OF Coccinia indica BY SOXHLET EXTRACTION AT 40-50 ℃

Solvent used	Wt. of powder (g)	Wt. of extract (g)	Yield of crude drug (%)
<i>n</i> -Hexane	150	2.32	1.546
Chloroform	300	2.88	0.960
Ethyl acetate*	200	3.56	1.780
<i>n</i> -Butanol*	300	4.25	1.416

\*Both the solutions gave polar compounds.

**Chemical analysis of crude drug:** The biologically active compound was separated from the crude extract by column chromatography. It was determined by TLC using CHCl<sub>3</sub>:MeOH:H<sub>2</sub>O (2:2:1), CHCl<sub>3</sub>:CCl<sub>4</sub>: acetone (6:6:3) and CHCl<sub>3</sub>:MeOH:H<sub>2</sub>O (2:4:2). This gave three fractions<sup>11,12</sup>. Each fraction was further chromatographed till a single spot was obtained (Table-2).

TABLE-2

TLC OF <i>n</i> -HEXANE EXTRACT OF Coccinia indica							
Solvent system	Obtained	Spot	Behaviour		R <sub>f</sub> value of		
	fraction	No (s)	Visible	UV light	each spot		
CHCl <sub>3</sub> :MeOH:H <sub>2</sub> O (2:2:1)	$C_1$	Spot 1	Dark green	Dark green yellow	0.31		
CHCl <sub>3</sub> :CCl <sub>4</sub> :H <sub>2</sub> O (6:6:3)	C <sub>2</sub>	Spot 2	Light green Light yellow Dark green	Brown Green	0.42		
CHCl <sub>3</sub> :MeOH:H <sub>2</sub> O (2:4:2)	C <sub>3</sub>		Light brown Yellow brown	Dark brown Yellow	0.56		

**Acid hydrolysis of purified fraction:** The compound (fraction  $C_1$ ) was hydrolyzed with 10 %  $H_2SO_4$  in MeOH: $H_2O$  (1:1) at 80 °C for 4 h. The reaction mixture was neutralized with BaCO<sub>3</sub> and filtered. The filtrate was evaporated to dryness in vacuum to give a residue in which glucose was

**Methylation:** The acid hydrolyzed fractions were methylated after the removal of methanol, the solution was extracted with ethyl acetate thrice.

**Glycosidal analysis:** The four fractions were tested chemically for glycosides which show the presence of mono and disaccharides in the compound. Iodine and Benedict's tests confirmed the presence of glycosides.

**Spectral analysis:** The spectral analysis of the compound was carried out for NMR which showed no signals for an aldehylde group at C-23. Instead, signals for two carbonyl carbon ( $\delta$  177.6 and 187.9) are present in the <sup>13</sup>C NMR spectrum. The <sup>1</sup>H NMR spectrum showed resonance for only five methyl groups on tertiary carbon at  $\delta$  0.73, 0.81, 0.98, 1.14 and 1.19 (s, 3H each). A three proton singlets were shown at  $\delta$  3.73 with a one bond heteronuclear correlation to the carbon resonance at  $\delta$  52.3 and a correlation to the carbonyl signals at  $\delta$  178.9, additionally an oxymethylenic group ( $\delta$  3.75) was also shown. The compound has the highest molecular mass of the group ( $\delta$  3.75) was also shown. The compound has the highest molecular mass of the saponin isolated from 70 % ethanol<sup>13</sup>. The R<sub>f</sub> value of the compound was found 0.3-0.5 which indicates the presence of saponin. On the basis of this, a triterpenoid saponin was isolated from *Coccinia indica*.

**Antidiabetic activities:** The fresh fruits of *Coccinia indica* were washed with water and air-dried at room temperature. The air-dried fruits were grinded to powder about 40-60 mesh size. About 150 g known amount of powdered substance was then extracted successively with solvent ethanol to increase the polarity. The crude extract obtained was filtered using Whatmann filter paper No. 1 and the solvent was evaporated to dryness under reduced pressure in vacuum evaporator at 40-50 °C. The freshly prepared extract was chemically tested for the presence of different constituents using standard method<sup>14</sup>.

**Induction of diabetes:** The animals were starved overnight then diabetes was induced by a single interaperitoneal injection of a freshly prepared streptozotocin (STZ) solution (50 mg/kg body weight). Streptozotocin was dissolved in 0.1 M freshly prepared citrate buffer solution (pH 4.5). The animals were allowed to drink 5 % glucose solution overnight to overcome. After 5 d streptozotocin administration, rats showing diabetes. Control rats were injected with citrate buffer alone. The extract in aqueous solution was administered orally through a garage at a concentration of 200 mg/kg body weight/rat/day for 30 d.

**Experimental design:** The animals were divided into four groups for the analysis of biochemical parameters. Each group has six animals. Group I: Normal control rats; Group II: Diabetic control rats; Group III: Diabetic rats treated with *Coccinia indica* extract 200 mg/kg body weight orally and Group IV: Diabetic treated with standard drug.

## **RESULTS AND DISCUSSION**

Data are expressed as mean  $\pm$  SEM. The biochemical parameters were analyzed statistically using one-way Anova followed by Dunnett's multiple comparison test. The glucose levels before and after the administration of

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different fractions were compared using paired students t-test. The minimum level of significance was fixed at p < 0.05. The result of present study indicates that the *n*-hexane fraction was found to be quite active, decreasing the blood glucose level -27.2 %. This fraction was found to possess triterpenoid saponin (Table-3).

## TABLE-3 EFFECT OF VARIOUS EXTRACTS OF Coccinia indica ON THE BLOOD GLUCOSE LEVEL IN STREPTOZOTOCIN INDUCED DIABETIC RATS

Group	Doses (mg/kg)	Glucose level (mg/dl) a	% After drug	
		Before drug treatment	After drug treatment	% After drug treatment
Solvent control	-	$269.15 \pm 17.57$	$305.66 \pm 31.32$	+13.56
Positive control (Phenformin)	30	$284.67 \pm 30.16$	$140.60 \pm 33.14$	-50.60
<i>n</i> -Hexane	200	$283.00 \pm 21.28$	$206.00 \pm 23.54*$	-27.2
Chloroform	200	$257.00 \pm 15.31$	$223.12 \pm 12.36$	-13.18
Ethyl acetate	200	$288.12 \pm 17.05$	$238.13 \pm 12.34$	-17.35
<i>n</i> -Butanol	200	$261.67 \pm 26.00$	$281.67 \pm 21.15$	+07.64

The values are mean  $\pm$  SEM (n = 6 in each group. "+" Denotes increase in hyperglycemic activity. \*p < 0.05 in comparison to corresponding value before treatment.

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