NOTE

Synthesis and Crystal Structure of a Tetaraazamacrocyclic Compound

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The tetaraazamacrocyclic compound of iron, {L·[Fe(CN)₅NO]} (L = 5,5,7,12,12,14-hexamethyl-1,4,8,11-tetraazacyclotetradeca-4,11-diene) has been synthesized and determined by X-ray diffraction. The crystal is triclinic, space group P-1 with unit cell parameters: a = 8.167(3) Å, b = 9.345(2) Å, c = 9.702(3)A, α = 63.823(12)°, β = 73.863(14)°, γ = 73.877 (14)°, V = 627.9(3)ų, Z = 1, Mr = 496.42, Dc = 1.313 Mg/m³, μ = 0.634 mm¹, F(000) = 262, T = 293(2) K, R = 0.0790, wR = 0.1513 for 4885 reflections with I > 2σ(I).

Key Words: Tetaraazamacrocyclic compound, Crystal structure.

In recent years, the metal-ion coordination compounds formed by polyaza macrocycles has been extensively investigated due to the ability of these systems to interact with different substrates such as metal ions or anionic species¹ to form complexes with applications in catalysis, models for metalloenzymes, mechanism and molecular recognition research²⁻³. Recently, in our laboratory, a series of transition metal compounds have been synthesized and studied⁴⁻⁷.

In this communication, the synthesis and the structure of the tetaraazamacrocyclic compound $\{L\cdot[Fe(CN)_5NO]\}\ (L=5,5,7,12,12,14-hexa methyl-1,4,8,11-tetraazacyclotetradeca-4,11-diene)$ are reported.

All the reagents were of AR grade and used without further purification. IR spectra was record on a Nexus-870 spectrophotometer. Elemental analyses for C, H and N were performed on a Elementar Vario EL-III analyzer. The crystal structure was determined by Siemens SMART CCD area-detector diffractometer.

Synthesis: The 5,5,7,12,12,14-hexamethyl-1,4,8,11-tetraazacyclotetradeca-4,11-diene [L·(ClO₄)₂] was synthesized according to the literature^{8,9}. A mixture of 25 mL methanol solution of L·(ClO₄)₂ (10 mmol) were respectively added 25 mL water solution of K₃Fe(CN)₅NO (10 mmol), then refluxed for 1 h. After filtered, the solution was kept at room condition for 6 d and then the brown colour rhombus

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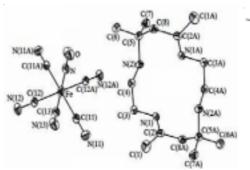
crystals resulted from it. IR (KBr, cm⁻¹): 3150(s), 2130, 1890(s), 1660, 1540(s). Anal. Calcd. for $C_{21}H_{32}N_{10}OFe$: C, 50.76; H, 6.45; N, 28.20 %. Found: C, 50.81; H, 6.44; N, 28.16 %.

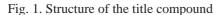
Structure determination: A single crystal (0.35 mm × 0.30 mm × 0.10 mm) was selected for crystallographic data collection at 293(2) K and structure determinated with graphite-monochromatic MoK_{α} radiation (λ = 0.71073 Å). A total of 4885 reflections were collected in the range of 2.67° ≤ θ ≤ 27.48°, of which 2833 reflections were unique with R_{int} = 0.0124. The structure was solved by direct methods and expanded using Fourier techniques and SHELXS-97 program system was used in the solution and refinements of the structure. The final full-matrix least-squares refinement including 151 variable parameters for 2833 reflections with I > 2 σ (I) and converged with unweighted and weighted agreement factors of R = 0.0790 and wR = 0.1513, where w = 1/[σ ²(F₀²) + (0.0000P)² + 0.6705P], and P = (F₀² + 2F_c²)/3. The maximum and minimum peaks on the final difference Fourier map are corresponding to 0.539 and -1.079e/ų (CCDC No. 646129), respectively.

The atomic coordinates and thermal parameters are given in Table-1 and the selected bond lengths and bond angels are in Table-2. The molecular structure of compound is shown in Fig. 1. Fig. 2 shows the packing diagram in the unit cell, in which there are positive negative charge interactions between L^{2+} cations and $[Fe(CN)_5NO]^{3-}$ anions.

TABLE-1 ATOMIC COORDINATES (× 10^4) AND THERMAL PARAMETERS (× 10^3 Å²)

Atom	X	Y	Z	U (eq)
Fe	0	0	0	54(1)
N(1)	4886(4)	5004(4)	-1813(3)	33(1)
C(1)	3458(7)	6508(7)	-4116(6)	68(2)
C(2)	4655(5)	6239(5)	-3073(4)	36(1)
C(11)	2088(5)	578(4)	-1507(4)	39(1)
N(11)	3319(5)	918(5)	-2390(4)	58(1)
<u> </u>	1992(6)	-1776(5)	2551(5)	70(1)





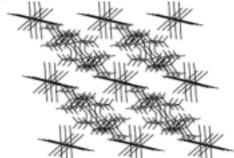


Fig. 2. Molecular packing arrangement

TABLE-2
SELECTED BOND DISTANCES (Å) AND ANGLES (°) OF {L[Fe(CN)₅NO]}

Bond	Dist.	Bond	Dist.	Angles	(°)	Angles	(°)
Fe-N	1.806(3)	C(1)-C(2)	1.493(5)	C(13)#1-Fe-N	180.0(4)	N(1)-C(2)-C(1)	126.1(4)
Fe-C(11)	1.941(4)	N(1)-C(2)	1.274(4)	N#1-Fe-C(11)#1	90.66(17)	C(2)-N(1)-C(3)	119.2(3)
O-N	1.135(4)	N(1)-C(3)	1.473(5)	N-Fe-C(11)#1	89.34(17)	N(1)-C(3)-C(4)	109.9(3)
N(11)-C(11)	1.142(5)	C(3)-C(4)	1.513(5)	O-N-Fe	178.6(4)	C(1)-C(2)-C(8)#2	115.1(3)

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REFERENCES

- R.M. Izatt, J.S. Bradshaw, S.A. Nielsen, J.D. Lamb, J.J. Christensen and D. Sen, *Chem. Rev.*, 85, 271 (1985).
- 2. T. Koike, M. Inoue and E. Kimura, J. Am. Chem. Soc., 118, 3091 (1996).
- 3. C. Bazzicaluppi, A. Bencini, A. Bianchi, M. Cecchi, B. Escuder, V. Fusi, E. Garcia-Espana, C. Giorgi, S.V. Luis, G. Maccagni, V. Marcelino, P. Paoletti and B. Valtancoli, *J. Am. Chem. Soc.*, **121**, 6807 (1999).
- 4. J.H. Bi, L.T. Kong, Z.X. Huang and J.H. Liu, *Inorg. Chem.*, 47, 4564 (2008).
- 5. J.H. Bi, Y.F. Sun, Z.X. Huang and H.B. Wang, Asian J. Chem., 20, 5788 (2008).
- 6. J.H. Bi, R.J. Ding, Z.X. Huang, Y. Chen and N.L. Hu, Asian J. Chem., 20, 4963 (2008).
- 7. J.H. Bi, F.H. Yao, Z.X. Huang, H.L. Wang and N.L. Hu, Asian J. Chem., 19, 5360 (2007).
- 8. J.H. Bi, L.Q. Chen, Z.Q. F.X. Xie, X.D. Zhao, S.S. Ni and J. D. Xu, *Asian J. Chem.*, **14**, 1621 (2002).
- 9. J.H. Bi, F.X. Xie, X.D. Zhao, Q. Chen, J. D. Xu and S.S. Ni, Asian J. Chem., 16, 137 (2004).
- 10. O.H. Hankovszky, K. Hideg, D. Lloyd and H. McNab, J. Chem. Soc. Chem. Commun., 378 (1974).

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