

Studies on Aromatic Amide of Root of *Carissa carandas* Linn

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The objective of the work is to study the chemical constituents of benzene extract of root of *Carissa carandas* Linn. Phytochemical investigation conducted on root revealed the presence of carbohydrates, alkaloids, flavanoids, saponins, steroids and fatty acids. The aromatic amide was detected in benzene extract and methanol extract of the root. The extracts were carried out by Soxhlet extractor and the yield of benzene extract was 2.5 %. TLC study of benzene extract showed the presence of 5 spots, out of which one spot was identified as aromatic amide. The present study deals with chemical investigation of benzene extract of root of *Carissa carandas* Linn. The five compounds were separated by column chromatography using benzene:ethyl acetate:chloroform (4:2:1). TLC study of separated compounds with R_f value 0.45 (B_1), 0.58 (B_2), 0.93 (B_3), 0.62 (B_4) and 0.59 (B_5). From UV, IR, 1H NMR and LC-MS analysis of the compound B_5 , it could be characterized as aromatic amide.

Key Words: *Carissa carandas* Linn., Benzene extract, Aromatic amide.

INTRODUCTION

Carissa carandas Linn (Family: Apocynaceae) commonly known as Karaunda, is a large evergreen shrub with a short stem, glabrous shrub found almost through out India¹. *C. carandas* Linn is a large evergreen shrub with short stem and strong thorns in pairs, bark light grey, scaly leaves simple, opposite, elliptic or obovate, shortly mucronate, glabrous shining and coriaceous; flowers white, in pubescent terminal corymbose cymes; fruit ellipsoid or globose berry, purplish black when ripe enclosing two or more seeds². The root has reputation of bring better stomachic. Used in the konkan, pounded with horse urine, lime juice and camphor as remedy for itching³⁻⁵. The presence of cardio tonic activity of water soluble fraction has been attributed to the presence of glycosides of odorside H. Presence of alkaloids is also reported in root and stem bark⁴.

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EXPERIMENTAL

The fresh root samples of *Carissa carandas* Linn was collected, dried and cut into pieces, crushed into powder then passed through sieve no 40 to obtain uniform particles and extracted with the solvents of increasing polarity⁶, petroleum ether, benzene, chloroform, ethyl acetate and methanol for 48 h in 50 batches of 100 g for each batch by successive solvent extraction method. The extracts were collected to dryness in a rotary evaporator under reduced pressure and controlled temperature (50-60°). After drying the extracts were weighed.

Thin layer chromatographic (TLC) studies of the fatty acids of benzene extract: The crude benzene extract was subjected to TLC studies using precoated silica gel GF₂₅₄ plates of 0.2 thickness⁷. A suitable mobile phase was developed consisting of benzene:ethyl acetate:chloroform (4:2:1). Which gave maximum resolution of the spots. The spots were viewed under UV light and by spraying with dilute sulphuric acid. TLC study of benzene extract showed the presence of 5 spots.

Column chromatography of benzene extract: The benzene extract was chromatographed in a column with a aluminium oxide neutral built in petroleum ether and eluted with benzene:ethyl acetate:chloroform (4:2:1) five fractions were collected and concentrated. Fraction-1 upon concentration yielded reddish semisolid mass (0.3 g) and named them as B₁. Similarly fractions 2, 3, 4 and 5 are concentrated and yielded (0.2, 0.25, 0.3 and 0, 2 g, respectively). They were labeled as B₂, B₃, B₄ and B₅, respectively. The collected fractions are subjected to thin layer chromatography using the solvent system benzene:ethyl acetate:chloroform (3:2:1), spraying the developed plates with dilute sulphuric acid yellow colour spots were observed. The R_f value of five compounds were calculated as B₁ R_f value = 0.45, B₂ R_f value = 0.58, B₃ R_f value = 0.93, B₄ R_f value = 0.62 and B₅ R_f value = 0.55. All the five compounds were subjected for antifungal activity the compound B₅ was showed the moderate anti fungal activity hence the compound B₅ was subjected for spectral analysis for further studies to know the chemical constituent responsible for the activity.

Chemical and spectroscopic analysis of compound B₅: The compound B₅ was examined for its colour, odour, nature, melting point and solubility. The melting point was determined by open capillaries of glass and they are uncorrected. The UV spectrum of the compound was recorded by using UV spectrophotometer. The IR spectrum was recorded in KBr pellet using FTIR spectrophotometer. ¹H NMR of the compound was recorded in CDCl₃ using Bruker NMR spectrophotometer.

Antifungal activity of aromatic amide of root of *Carissa carandas*: The crude aromatic amide isolated from benzene extract were tested for antifungal activity by cup plate method⁸ at 1 and 2 % concentration in dimethyl sulphoxide and fluconazole was used as standard drug. The potato agar media was used for this work⁸. The test organisms used were (*A. niger* and *A. fumigatus*).

RESULTS AND DISCUSSION

From benzene extract of root of *Carissa carandas* Linn aromatic amide was separated by column chromatography. Tables 1 and 2 shows the results of spectral analysis and antifungal activity of B₅ compound.

TABLE-I
SPECTROSCOPIC ANALYSIS OF COMPOUND B₅

Compound	Spectrum	Characteristics peaks
B ₅	UV spectrum	$\lambda_{\max} = 320$ nm
	IR Spectrum (KBr) cm ⁻¹	3442 (NH ₂ group), 3000-3100 (aromatic stretching), 2400-2923 (C-H stretching of the CH ₂ and CH ₃ groups), 1732 (C=O group), 1652 (N-H bending and C=N), 1600 (C=C).
	¹ HNMR (CDCl ₃)	0.6-3.0 δ ppm (presence of CH ₂ group), 3.5-4.5 δ ppm (presence of CH ₃ group), 5.1 C δ ppm (NH ₂ group), 6.7-7.7 δ ppm (presence of aromatic ring)
	LC-mass	m/z 803.6 (the basic peak)

TABLE-2
ANTIFUNGAL ACTIVITY OF COMPOUND B₅

Microorganism	Zone of inhibition in mm.			
	B ₅ compound		Fluconazole	
	1 % w/w	2 % w/w	1 % w/w	2 % w/w
<i>A. niger</i>	14	16	16	18
<i>A. fumigatus</i>	15	17	17	18

TLC studies of benzene extract showed five spots and developed yellow colour with dilute sulphuric acid. The aromatic amide with R_f value of B₅ (0.59) was isolated by preparative TLC.

In the spectroscopic analysis of B₅ compound⁹ showed an absorption bands. In IR spectrum KBr, cm⁻¹. Table-1 suggested presence of 3442 cm⁻¹ (hydrogen bonded NH₂ group), the weak band around 3100-3000 cm⁻¹ (aromatic stretching), 2923-2400 cm⁻¹ (C-H stretching of the CH₂ and CH₃ groups), 1732 cm⁻¹ (C=O group), 1652 cm⁻¹ (N-H bending and C=N) and 1600 cm⁻¹ (due to C=C).

The major peaks observed in the ¹H NMR spectrum at δ 0.6-3.0 and at δ 3.5-4.5 indicates the presence of CH₂ and CH₃ groups. The peak at δ 5.1 may be due to NH₂ group and the peak at δ 6.7-7.7 may be due to aromatic proton indicating the presence of aromatic ring. The IR and ¹H NMR spectral studies of component B₅ suggests the presence of aromatic ring, amino group, carbonyl group and methylene and methyl group. Hence, the compound B₅ may contain an aromatic with amide chain having CH₂ and CH₃ groups and containing amide group. Hence, the compound B₅ is an aromatic amide¹⁰. In contrary, the mechanism of amides will not depend on chain length and number of double or triple bonds. It has been hypothesized

to produce its effects through conjugation with peptides and proteins of surface membranes thereby increase the permeability of the membrane and effective against all types of microbes. This could be the reason behind the antifungal activity of B₅ compound. The phytochemical investigation and spectral analysis of compound B₅ has shown the presence of aromatic amide in benzene extract. The aromatic amide could be the reason for its effectiveness as antifungal activity.

The LC-mass spectrum of compound B₅ showed the base peak at m/z 803.6. The molecular peak not recorded. The antifungal activity of the compound B₅ because of aromatic amide.

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