

## NOTE

### Extraction and Characterization of Naringenin from *Pseudohandellia umbelliferae*

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*Pseudohandellia umbelliferae* collected from Kalat Naderi area, Khorasan Razavi province, Iran was investigated for the presence of naringenin, 5,7-dihydroxy 2-(4-hydroxyphenyl)chroman-4-one by spectroscopic methods.

**Key Words:** *Pseudohandellia umbelliferae*, Chromatography, Soxhlet apparatus, Naringenin.

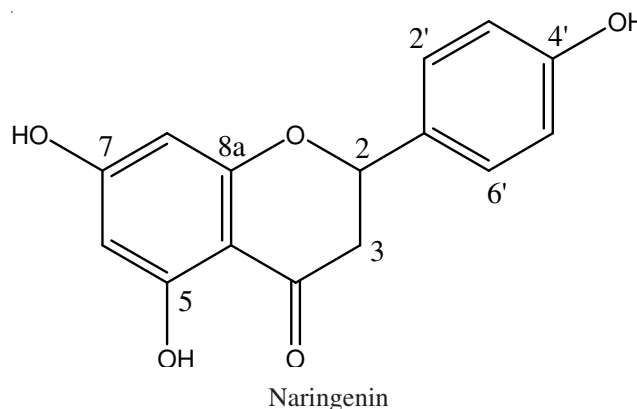
*Pseudohandellia umbelliferae* is a plant with a 20-100 cm high, leaves greyish with loose cobwebby tomentum, main rootstock thickened and heads numerous in dense umbel-shaped compound corymbs, flowering between May to July<sup>1</sup>. Following our research on the chemical constituents of this plant<sup>2</sup>, we describe in this paper the isolation and characterization of a flavonoid, naringenin, from methanol extract of the aerial parts of *Pseudohandellia umbelliferae*.

Melting point was measured on a Bamstead/electro-thermal 9200 melting point apparatus. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded in DMSO, on a Bruker AC 100 MHz NMR spectrometer. Chemical shift is expressed in  $\delta$  units relative to TMS ( $\delta = 0$ ) as internal standard. Mass spectra was recorded on a Shimadzu GC-17A spectrometer operating at 70 eV in electron impact mode. FT-IR was performed on a Shimadzu 8400.

The aerial parts of *Pseudohandellia umbelliferae* were collected from Kalat Naderi area, Khorasan Razavi province, Iran, in April 2006 by Mr M.R. Joharchi. A voucher specimen is deposited at the Herbarium of the Research Institute of Forests and Rangelands, in the University of Ferdowsi, Mashhad, Iran.

**Extraction and isolation:** The aerial parts of *Pseudohandellia umbelliferae* are dried at ambient temperature before being powdered. The powder was successively extracted with methanol in a Soxhlet apparatus over 6 h. The organic solvent was evaporated to dryness in vacuum to yield the corresponding extracts (11 %, w/w).

Phytochemical tests were carried out on the methanol extract and on the powdered specimens to identify the constituents using standard procedures<sup>3-5</sup>. The tests revealed the presence of a flavonoid.



Thin layer chromatographic (TLC) analyses were made on 0.25 mm thick silica gel 60G (Merck, 7731), prepared on glass plates. A mixture of chloroform/methanol (2:3) proved to be the best eluent for the TLC analysis. The plate was examined under UV light at 254 nm. Only one of the spot ( $R_f = 0.8$ ) out of four was observed as a dark blue colour. The methanol extract was also chromatographed on a silica gel column, using the same gradient elution system as mentioned above, to yield a pale yellow crystals (200 mg), m.p. 239-241 °C, IR (film,  $\nu_{\max}$ ,  $\text{cm}^{-1}$ ): 3600 (OH), 3400 (OH), 3050, 1650 (C=O) 1600 (C=C), 1250, 1300, 840, 730 and 690; <sup>1</sup>H NMR (100 MHz, DMSO)  $\delta$  2.6 (d, 1H), 3.2 (1H), 5.5 (dd, 1H, Ar), 6 (s, 2H, Ar), 6.7 (d, 2H,  $J = 8.5$ , Ar), 7.4 (d, 2H,  $J = 8.5$ , Ar), 9.7 (s, 1H, OH), 12.1 (s, 2H, OH); <sup>13</sup>C NMR (100 MHz, DMSO)  $\delta$ , 196 (C=O), 166, 163, 162, 157, 129, 128, 115, 101, 95, 94, 78, 42; MS (EI) ( $m/z$ , %) 272 [ $M^+$ ] (100), 179 (23), 166 (23), 153 (81), 124 (31), 120 (53), 107 (30), 91 (31), 69 (29), 50 (5) and 39 (26); UV ( $\lambda_{\max}$ , MeOH, nm) 219, 289, 330; which is a typical

absorptions of flavanones and carbonyl group ( $1650\text{ cm}^{-1}$ ). The  $^{13}\text{C}$  NMR spectrum showed the resonance of six aromatic methine carbons, C-3', C-5' ( $\delta$  115), C-2', C-6' ( $\delta$  128), C-8 ( $\delta$  94) and C-6 ( $\delta$  95), six quaternary aromatic carbon at C-7 ( $\delta$  166), C-5 ( $\delta$  163), C-8a ( $\delta$  162), C-4a ( $\delta$  101), C-1' ( $\delta$  128) and C-4' ( $\delta$  157), an oxymethine carbon signal C-2 ( $\delta$  78), an methylene carbon C-3 ( $\delta$  42) and finally a carbonyl carbon C-4 ( $\delta$  196).

### Conclusion

Naringenin, was extracted and isolated from the aerial parts of *Pseudohandellia umbelliferae* using methanol. The chemical structure of naringenin was characterised by spectroscopic methods.

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